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**JM-20 COMO COMPOSTO MULTIALVO: AVALIAÇÃO DO
POTENCIAL ANTIOXIDANTE, INIBIDOR DE COLINESTERASES
E CITOTOXICIDADE**

Santa Maria, RS
2020

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Tese de doutorado apresentada ao Programa de Pós- Graduação em Ciências Biológicas: Bioquímica Toxicológica da Universidade Federal de Santa Maria como requisito parcial para a obtenção do título de **Doutora em Ciências Biológicas: Bioquímica Toxicológica**

Orientador: Prof Dr. João Batista Teixeira da Rocha

Santa Maria, RS
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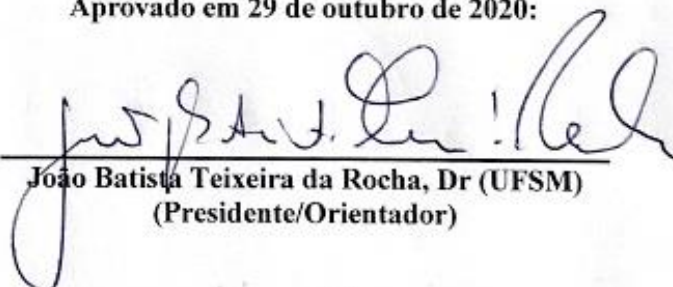
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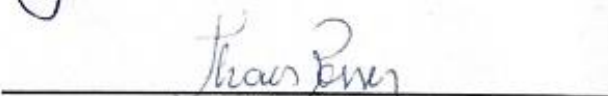
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Santa Maria, RS
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DEDICATÓRIA

A minha mãe e meus avós meus maiores incentivadores. Dedico também aos meus amigos e demais familiares que me apoiaram em todos os momentos.

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RESUMO

JM-20 COMO COMPOSTO MULTIALVO: AVALIAÇÃO DO POTENCIAL ANTIOXIDANTE, INIBIDOR DE COLINESTERASES E CITOTOXICIDADE

AUTOR: Fernanda D'Avila da Silva
ORIENTADOR: João Batista Teixeira da Rocha

Compostos multialvo tem despertado o interesse de muitos pesquisadores. Estas substâncias tem como principal característica a capacidade de agir através de diferentes mecanismos de ação, fato relevante em casos de patologias originadas por múltiplas causas. Neste contexto, em alguns casos de doenças neurodegenerativas pode ocorrer a perda neuronal em determinadas regiões cerebrais as quais contem neurônios colinérgicos, consequentemente causando comprometimento de funções cognitivas. O estresse oxidativo é caracterizado pelo desequilíbrio entre a produção das espécies reativas e as defesas antioxidantes e pode estar diretamente relacionado as causas ou consequências de patologias relacionadas ao SNC. Estudos anteriores mostraram que o JM-20, 1,5-benzodiazepina fundida a uma fração de di-hidropiridina, tem diferentes propriedades farmacológicas de interesse clínico. Neste sentido, o objetivo principal deste estudo foi avaliar o efeito do JM-20 na AChE e na BChE de diferentes fontes, identificar o tipo de inibição enzimática e compreender as interações entre o composto e as enzimas utilizando ferramentas *in silico de docking molecular*. Além disso, verificar o efeito protetor no estresse oxidativo induzido por Fe²⁺ em leucócitos e determinar a atividade *scavenger* de radicais livres. Do mesmo modo, averiguar a possível citotoxicidade utilizando células sanguíneas humanas e prever parâmetros ADMET utilizando ferramentas *in silico de screening* virtual. A BChE utilizada foi purificada de *Equus ferus* e presente no plasma humano, enquanto que a AChE foi de *Electrophorus electricus*, eritrócitos totais e presente nas membranas isoladas de eritrócitos humanos (ghost). As enzimas foram pré-incubadas durante 30 minutos na presença de diferentes concentrações de JM-20 (1 nM - 200 µM). Para avaliar o tipo de inibição enzimática, foi realizado um estudo cinético variando a concentração do substrato (0,05 - 1,6 mM). O sangue humano foi obtido de voluntários saudáveis. Imediatamente após a coleta do sangue, os leucócitos ou eritrócitos foram isolados, lavados e tratados com diferentes concentrações de JM-20 e avaliados de acordo com cada teste específico. Os resultados demonstraram o potencial efeito inibitório na atividade da AChE. Estes efeitos foram observados em todas as enzimas testadas (IC₅₀ = 123 nM ± 0,2 para *E. electricus*, 172 nM ± 0,2 para eritrócitos totais e 158 nM ± 0,1 para ghost). Além disso, sugerimos que o composto apresenta um tipo misto de inibição já que altera o K_m e V_{max} da AChE. O *docking molecular* demonstrou a existência de oito isômeros diferentes do JM-20 e os isômeros 4R interagem melhor com a HsAChE. O resultado do TBARS aponta para um potente efeito antioxidante e efeito *scavenger* de radicais livres observados na fase lenta da reação. Entretanto, a exposição a todas as concentrações de JM-20 testadas (10-50 µM) causou um aumento significativo nos níveis de espécies reativas intracelulares (ER). Embora tenha sido observada diminuição da viabilidade celular e aumento da produção de ER, a exposição ao JM-20 (10-50 µM) não alterou o ciclo celular, assim como não causou hemólise. Análises de *screening* virtual do JM-20 demonstraram similar perfil ADMET a nifedipina. Assim, nossas descobertas apoiam o potencial uso clínico do JM-20, para o tratamento de patologias associadas ao SNC.

Palavras-chave: JM-20, colinesterase, antioxidante, toxicidade, análise computacional

ABSTRACT

JM-20 AS MULTI-TARGET COMPOUND: EVALUATION OF ANTIOXIDANT POTENTIAL, CHOLINESTERASE INHIBITOR, AND CYTOTOXICITY

AUTHOR: Fernanda D'Avila da Silva
ADVISOR: João Batista Teixeira da Rocha

Multi-target compounds have aroused the interest of many researchers. The main characteristic of these substances is the ability to act through different mechanisms of action, a relevant fact in cases of pathologies caused by multiple causes. In this context, in some cases of neurodegenerative diseases, a neuronal loss may occur in certain brain regions that contain cholinergic neurons, consequently causing impairment of cognitive functions. Oxidative stress is characterized by the imbalance between the production of reactive species and antioxidant defenses and can be directly related to the causes or consequences of pathologies related to the CNS. Previous studies have shown that JM-20, 1,5-benzodiazepine fused to a dihydropyridine fraction, has different pharmacological properties of clinical interest. In this sense, the main objective of this study was to evaluate the effect of JM-20 on AChE and BChE from different sources, identify the type of enzyme inhibition and understand the interactions between the compound and the enzymes using silicon molecular docking tools. Besides, to verify the protective effect on oxidative stress induced by Fe^{2+} in leukocytes and to determine the scavenger activity of free radicals. Likewise, investigate possible cytotoxicity using human blood cells and predict ADMET parameters using in silico virtual screening tools. The BChE used was purified from *Equus ferus* and present in human plasma, while AChE was from *Electrophorus electricus*, total erythrocytes and present in membranes isolated from human erythrocytes (ghost). The enzymes were pre-incubated for 30 minutes in the presence of different concentrations of JM-20 (1 nM - 200 μM). To assess the type of enzyme inhibition, a kinetic study was performed varying the concentration of the substrate (0.05 - 1.6 mM). Human blood was obtained from healthy volunteers. Immediately after blood collection, the leukocytes or erythrocytes were isolated, washed, and treated with different concentrations of JM-20 and evaluated according to each specific test. The results demonstrated the potential inhibitory effect on AChE activity. These effects were observed in all enzymes tested ($\text{IC}_{50} = 123 \text{ nM} \pm 0.2$ for *E. electricus*, $172 \text{ nM} \pm 0.2$ for total erythrocytes and $158 \text{ nM} \pm 0.1$ for ghost). Besides, we suggest that the compound has a mixed type of inhibition as it changes the K_m and V_{max} of AChE. Molecular docking demonstrated the existence of eight different isomers of JM-20, and the 4R isomers interact better with *HsAChE*. The TBARS result points to a potent antioxidant and scavenger effect of free radicals seen in the slow phase of the reaction. However, exposure to all tested JM-20 concentrations (10-50 μM) caused a significant increase in the levels of reactive intracellular species (RS). Although decreased cell viability and increased RS production have been observed, exposure to JM-20 (10-50 μM) did not alter the cell cycle, nor did it cause hemolysis. Analyzes of virtual screening of the JM-20 demonstrated a similar ADMET profile to nifedipine. Thus, our findings support the potential clinical use of JM-20, for the treatment of pathologies associated with the CNS.

Keywords: JM-20, cholinesterase, antioxidant, toxicity, computational analysis

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APRESENTAÇÃO

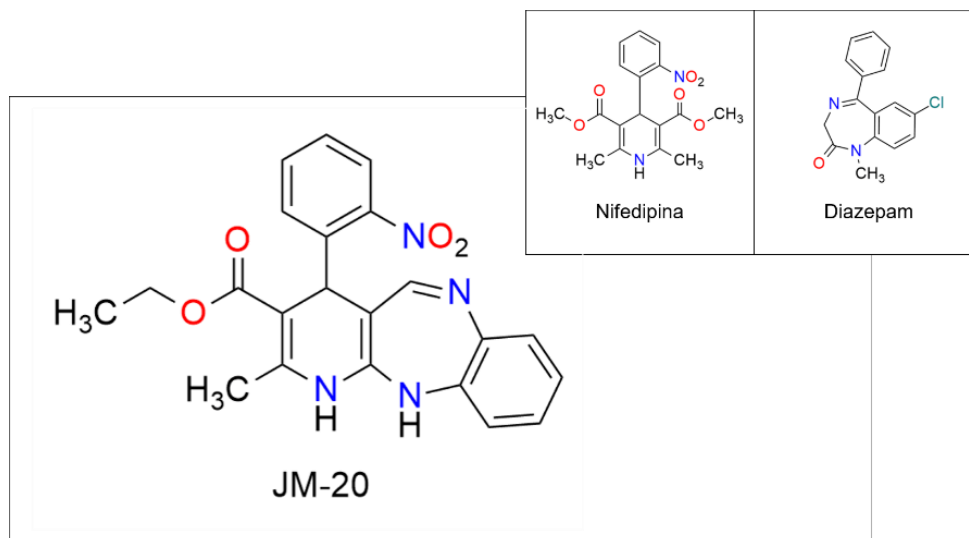
A tese apresentada está organizada de acordo com os seguintes tópicos: inicialmente é descrita uma introdução, com uma breve revisão sobre os temas abordados no trabalho, seguidos da justificativa e dos objetivos. Em seguida, o desenvolvimento na forma de um artigo publicado no periódico internacional “Biochimie” e um manuscrito que será submetido a um periódico igualmente internacional “Toxicology *in vitro*”. As seções Introdução, Materiais e métodos, Resultados, Discussão, Conclusão e Referências bibliográficas encontram-se no artigo e manuscrito e representam a íntegra deste estudo. Finalmente, no item discussão são apresentados interpretações e comentários gerais sobre os trabalhos. As referências bibliográficas apresentadas no final da tese referem-se às citações que aparecem nos itens introdução e discussão.

1 INTRODUÇÃO

O interesse por compostos multialvo para o tratamento de diferentes patologias, principalmente as associadas ao sistema nervoso central (SNC), tem despertado o interesse de muitos pesquisadores (DIAS et al., 2015; MONTANARI, 1995; SAMEEM et al., 2017). Estas moléculas têm como característica principal a capacidade de agir em diferentes mecanismos, fato relevante em casos de patologias multicausais (DIAS et al., 2015; LU et al., 2012; SPILOVSKA et al., 2017).

Dessa forma, o JM-20 (3-etoxicarbonil-2-metil-4-(2-nitrofenil) 4,11- di-hidro-1H-pirido [2,3-b] [1,5] benzodiazepina) é um composto derivado do 1,5-benzodiazepínico que apresenta uma porção 1,4- dihidropiridina (Figura 1) (NUÑEZ-FIGUEREDO et al., 2013, 2014a, 2014b). A partir de sua estrutura podemos observar a similaridade com moléculas atualmente utilizadas na prática clínica, como por exemplo, o Diazepam® e a Nifedipina®. O primeiro é um benzodiazepínico com propriedades anticonvulsivantes, ansiolítica, sedativa e relaxante muscular (MANDELLI; TOGNONI; GARATTINI, 1978). O Diazepam® age através do aumento da atividade do ácido gama-aminobutírico (GABA), um neurotransmissor presente no SNC (MANDELLI; TOGNONI; GARATTINI, 1978). Enquanto que a Nifedipina® (Figura 1), uma di-hidropiridina, foi descrita pela primeira vez na literatura em 1972 juntamente com outras di-hidropiridinas e atua como bloqueador dos canais de cálcio (LEE et al., 2014). A Nifedipina® é indicada para o tratamento da angina e hipertensão (LEE et al., 2014; OMS, 2019). Neste contexto, estudos prévios com o JM-20 demonstraram a manutenção dos efeitos como ansiolítico, sedativo, anticonvulsivante (semelhante ao Diazepam®) e bloqueador dos canais de cálcio (equivalente a Nifedipina®), além de apresentar outras propriedades farmacológicas de interesse clínico resumidas na Tabela 1.

Figura 1 – Estrutura do JM-20, nifedipina e diazepam.



Estrutura molecular do composto JM-20 (3 – etoxicarbonil – 2 – metil – 4 - (2 - nitrofenil) – 4 – 11 – dihidro – 1 – H – pirido [2, 3 - b] [1, 5] benzodiazepino), nifedipina e diazepam. Fonte: o autor

O JM-20 (2, 4, 8 e 10 mg/Kg) manteve o perfil ansiolítico, semelhante aos derivados dos benzodiazepínicos como o Diazepam (Figura 1), avaliados no teste do campo aberto e do labirinto em cruz elevada em camundongos. Nestes testes comportamentais, os animais apresentaram um aumento no número de entradas e tempo gasto na parte central do aparato (teste de campo aberto) e nos braços abertos (labirinto em cruz elevada) após a administração oral das drogas (NUÑEZ-FIGUEREDO et al., 2013). O JM-20 foi capaz de reduzir o comportamento agressivo de camundongos socialmente isolados, possivelmente, pela modulação positiva dos receptores GABAérgicos (NUÑEZ-FIGUEREDO et al., 2013). Semelhante aos sedativos, o JM-20 também aumentou significativamente o período de sono em ratos (NUÑEZ-FIGUEREDO et al., 2013). Além disso, este composto conferiu proteção em modelo de convulsões induzidas por pentilenotetrazol em camundongos, aumentou o período de latência para o início da crise convulsiva e a porcentagem de sobrevivência destes animais (NUÑEZ-FIGUEREDO et al., 2013).

Adicionalmente, o JM-20 pode atuar como um agente neuroprotetor de múltiplos alvos, reduzindo a lesão excitotóxica neuronal e protegendo as mitocôndrias da toxicidade induzida por Ca^{2+} , preservando, desse modo, o balanço energético celular (NUÑEZ-FIGUEREDO et al., 2014a, 2014c, 2014b). O JM-20 também foi capaz de atenuar os efeitos da lesão isquêmica, possivelmente, através da modulação da neuroinflamação e da via de sinalização celular antiapoptótica (NUÑEZ-FIGUEREDO et al., 2015, 2016, 2018; RAMÍREZ-SÁNCHEZ et al., 2015). Recentemente, foi avaliado o efeito do JM-20 em modelo de doença de Parkinson induzida por rotenona *in vitro* e *in vivo* (FONSECA-

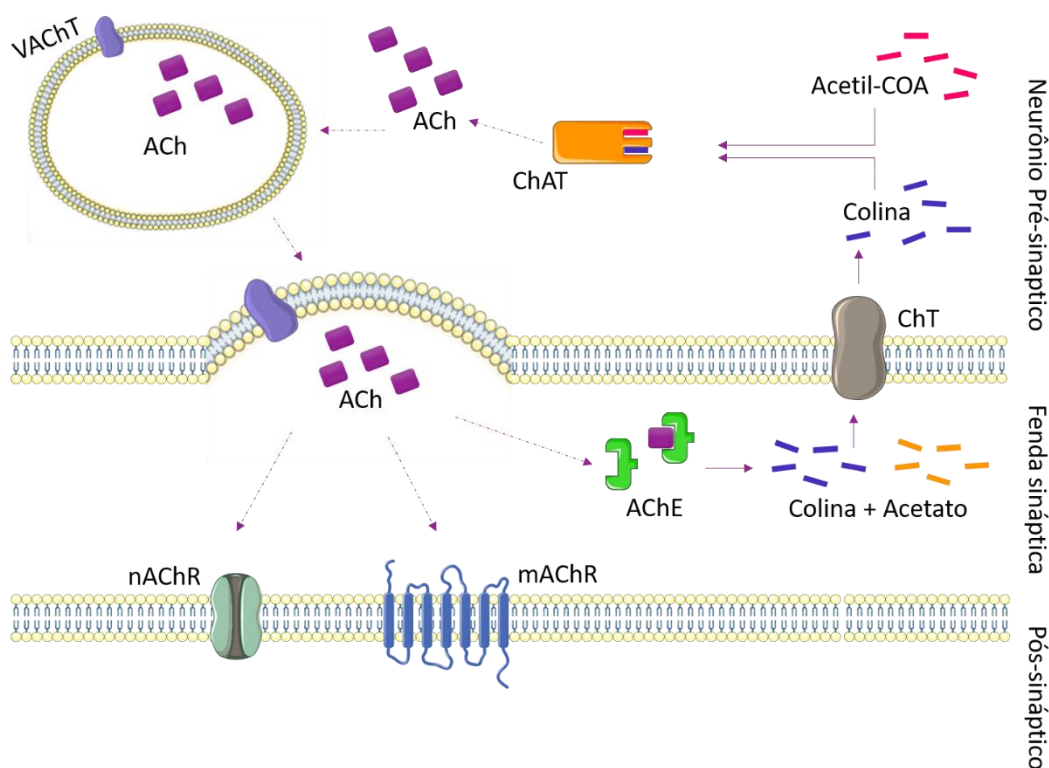
FONSECA et al., 2019a). Neste estudo foi possível observar, *in vitro*, o efeito protetor do composto ao diminuir a morte celular induzida pela rotenona. Nas análises *in vivo*, o JM-20 atenuou com sucesso as alterações comportamentais induzida pela rotenona em ratos (FONSECA-FONSECA et al., 2019a). O JM-20 aumentou a atividade de enzimas como a superóxido dismutase e a catalase, responsáveis pelo equilíbrio redox dos animais (FONSECA-FONSECA et al., 2019a). Além disso, o JM-20 teve efeito sobre a aquisição e consolidação de memória em ratos após o comprometimento cognitivo induzido por escopolamina (WONG-GUERRA et al., 2019). Estes fatos nos levam a considerar sua utilização como um composto promissor para uso terapêutico e incentiva o estudo dos efeitos do JM-20 em outras vias principalmente as associadas ao SNC.

Tabela 1. Efeitos do composto JM-20 descritos na literatura.

Efeito	Modelo	Referência
Ansiolítico	Camundongo	(NUÑEZ-FIGUEREDO et al., 2013b)
Redução do comportamento agressivo induzido por isolamento	Camundongo	(NUÑEZ-FIGUEREDO et al., 2013b)
Proteção em modelo de convulsões induzidas por PTZ	Camundongo	(NUÑEZ-FIGUEREDO et al., 2013b)
Sedativo	Camundongo	(NUÑEZ-FIGUEREDO et al., 2013b)
Neuroprotetor	Cultura celular de Feocromocitoma (PC-12) e mitocôndrias isoladas do fígado de ratos	(NUÑEZ-FIGUEREDO et al., 2014a)
	Mitocôndrias e sinaptossomas isolados do cérebro de ratos	(NUÑEZ-FIGUEREDO et al., 2014b)
Atenuou efeitos da lesão isquêmica	Cultura de fatias de hipocampo de ratos Ratos	(NUÑEZ-FIGUEREDO et al., 2014c)
	Vesículas sinápticas, sinaptossomas, partículas submitocondriais e culturas de células (astrócitos) de cérebro de ratos	(NUÑEZ-FIGUEREDO et al., 2015)
	Ratos	(NUÑEZ-FIGUEREDO et al., 2016)
	Cultura de fatias de hipocampo de ratos	(RAMÍREZ-SÁNCHEZ et al., 2015)
Proteção da memória após comprometimento induzido por escopolamina	Ratos	(WONG-GUERRA et al., 2019)
Proteção contra danos induzidos por rotenona	SHSY-5Y humano (ATCC® CRL-2266™) Ratos	(FONSECA-FONSECA et al., 2019b)

O SNC, sistema nervoso periférico (SNP) e a junção neuromuscular apresentam em comum o sistema colinérgico. (MCCORRY, 2007; PICCIOTTO; HIGLEY; MINEUR, 2013). Este sistema é constituído pelo neurotransmissor acetilcolina (ACh), seus receptores e as enzimas responsáveis pela sua síntese e degradação (MCCORRY, 2007). A ACh é sintetizada no neurônio pré-sináptico, a partir da colina (proveniente principalmente da recaptação de colina pelo transportador de colina (ChT) e da acetil-CoA (proveniente do piruvato), em uma reação catalisada pela enzima colina acetiltransferase (ChAT). Posteriormente, é transferida para vesículas e liberada na fenda sináptica onde irá atuar nos receptores muscarínicos ou nicotínicos iniciando a transmissão sináptica (Figura 2) (MCCORRY, 2007; PICCIOTTO; HIGLEY; MINEUR, 2013; POHANKA, 2014). A interrupção da neurotransmissão colinérgica se dá através da hidrólise da ACh em acetato e colina por enzimas colinesterases (ChE) (MCCORRY, 2007; PICCIOTTO; HIGLEY; MINEUR, 2013). Nos seres humanos, podem ser encontrados dois tipos de colinesterases: a butirilcolinesterase (BChE) e a acetilcolinesterase (AChE) (MCCORRY, 2007; PICCIOTTO; HIGLEY; MINEUR, 2013).

Figura 2 - Ação da acetilcolinesterase na fenda sináptica.



A ACh liberada na fenda sináptica pelos neurônios pré-sinápticos, através transportadores vesiculares de ACh (VACHT), pode interagir com os receptores nicotínicos (nAChR) ou muscarínicos (mAChR) presentes no neurônio pós-sinápticos. Ao término da neurotransmissão, a ACh é degradada em acetato e colina pela AChE ou BChE. A colina retorna ao neurônio pré-sináptico através do transportador de colina (ChT) e pode fazer parte da síntese de uma nova molécula de ACh pela ChAT. Fonte: o autor.

A AChE é uma hidrolase que apresenta alta eficiência catalítica (PETRONILHO; PINTO; VILLAR, 2011). Esta enzima possui um sítio ativo com diferentes subsítios (DA SILVA et al., 2018a; GRISARU et al., 1999). O subsítio esterático é onde encontra-se a tríade catalítica (serina, histidina e glutamato) diretamente responsável pela hidrólise da ACh (DA SILVA et al., 2018a; GRISARU et al., 1999; PETRONILHO; PINTO; VILLAR, 2011; SIEGEL et al., 2006). Também existe um subsítio aniônico (formado pelos resíduos de triptofano e glutamato), que interage com a parte catiônica da ACh (amina quaternária) e posiciona adequadamente o substrato para a hidrólise (DA SILVA et al., 2018a; GRISARU et al., 1999; PETRONILHO; PINTO; VILLAR, 2011; SIEGEL et al., 2006). Além disso, existe outra região denominada subsítio aniônico periférico, que influencia a conformação do subsítio catalítico, e pode estar envolvido na ação de alguns inibidores exógenos da enzima ou na inibição por excesso de substrato (DA SILVA et al., 2018a; GRISARU et al., 1999; PETRONILHO; PINTO; VILLAR, 2011; SIEGEL et al., 2006). Diferentes estudos tem demonstrado alterações na função da AChE relacionadas a sua inibição ou atividade acelerada (H. FERREIRA-VIEIRA et al., 2016; LÓPEZ; PASCUAL-VILLALOBOS, 2010; POHANKA, 2012; RICHENDRER; CRETON, 2015; WOREK et al., 2000). Um exemplo bastante comum de inibição da AChE é a intoxicação por pesticidas organofosforados, essas substâncias tem a capacidade de se ligar a enzima e impedir que a mesma realize sua função de hidrólise da ACh consequentemente impedindo a interrupção da neurotransmissão colinérgica, causando reações como perda do controle muscular, convulsões e até mesmo a morte (DA SILVA et al., 2018b; JOHN; SHAIKE, 2015; OCHOA; RODRIGUEZ; ZULUAGA, 2016). Por outro lado, quando ocorre a redução na liberação da ACh ou um aumento na produção ou atividade da AChE, a transmissão colinérgica pode ser prejudicada gerando sintomas como perda de memória. (LANCTÔT; RAJARAM; HERRMANN, 2009; MEHTA; ADEM; SABBAGH, 2012; NOGARA et al., 2015; RACCHI et al., 2004; YAN et al., 2008). Assim sendo, novas opções terapêuticas para patologias neurológicas, as quais apresentam uma disfunção no sistema colinérgico, tem sido amplamente estudadas (COLOVIC et al., 2013; FALSAFI et al., 2012).

A descoberta da síntese de ACh fora do sistema nervoso (não-neuronal) adicionou mais funções a esta molécula que estava sendo considerada apenas um neurotransmissor (GWILT; DONNELLY; ROGERS, 2007; WESSLER et al., 2003; WESSLER; KIRKPATRICK, 2001, 2016, 2008). Primeiramente, foi sugerido que poderia haver ACh fora no sistema nervoso devido a presença de seus precursores nas células. A acetil-CoA, é o principal produto do catabolismo de carboidratos, proteínas e lipídios (WESSLER; KIRKPATRICK; RACK, 1998)

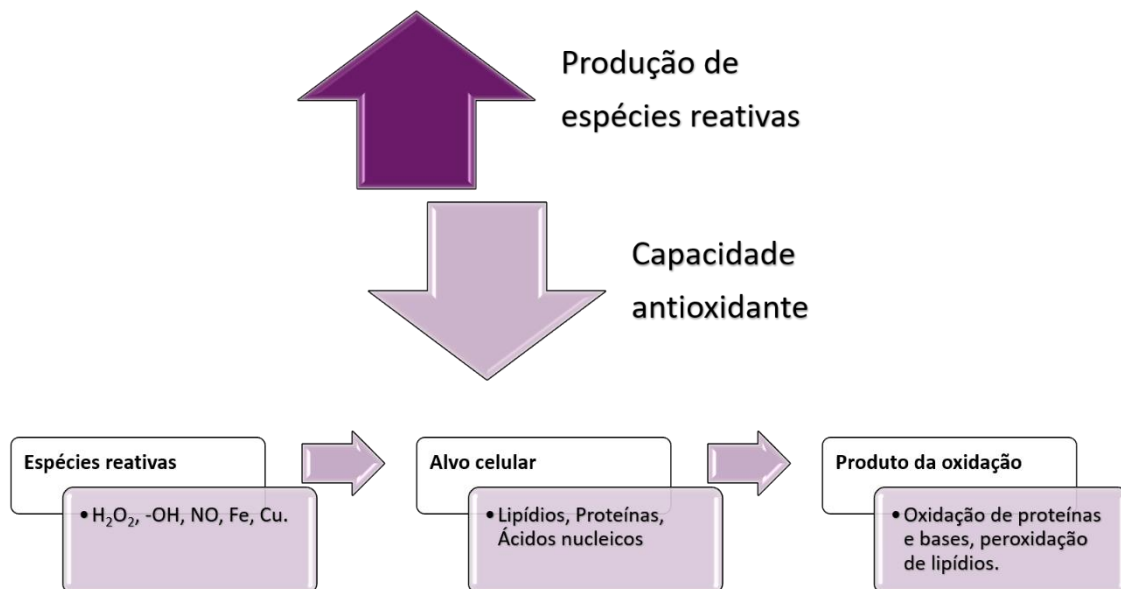
e a colina é proveniente da quebra intracelular de fosfolipídios ou da captação de colina extracelular por meio de transportador de colina (WESSLER et al., 2003; WESSLER; KIRKPATRICK, 2001, 2016, 2008; WESSLER; KIRKPATRICK; RACK, 1998). Uma vez que a acetil-CoA e a colina estão presentes na maioria das células, torna-se possível a síntese da ACh além do sistema nervoso (GWILT; DONNELLY; ROGERS, 2007; WESSLER et al., 2003; WESSLER; KIRKPATRICK, 2001, 2016, 2008). De fato, a expressão dos componentes do sistema colinérgico foi encontrada nas células epiteliais (pele, vias aéreas, intestino), nas células imunes (linfócitos) e nos eritrócitos (FUJII et al., 2017; SALDANHA, 2017; WESSLER; KIRKPATRICK, 2008). A ACh não-neuronal está envolvida na regulação de funções celulares básicas como proliferação, diferenciação, organização do citoesqueleto e liberação local de mediadores (óxido nítrico (ON)) e citocinas pró-inflamatórias, por exemplo) (WESSLER et al., 2003; WESSLER; KIRKPATRICK, 2016, 2008; WESSLER; KIRKPATRICK; RACK, 1998). Diferentemente da liberação de ACh através de vesículas pelos neurônios, o conhecimento sobre os mecanismos de liberação da ACh não-neuronal ainda é escasso (WESSLER; KIRKPATRICK, 2016). Entretanto, existem evidências que a ACh não-neuronal pode ser liberada imediatamente após sua síntese pela célula (WESSLER; KIRKPATRICK, 2001). Após a liberação a ACh pode interagir com receptores nicotínicos e muscarínicos da célula de origem ou de células vizinhas (WESSLER; KIRKPATRICK, 2001).

Outro mecanismo importante nos organismos aeróbios é o sistema antioxidante endógeno, o qual neutraliza as espécies reativas produzidas fisiologicamente (BIRBEN et al., 2012; RAHAL et al., 2014). Algumas dessas espécies reativas desempenham importantes papéis biológicos contra infecções, como mediadores da resposta inflamatória e, por este motivo, não são imediatamente neutralizadas (BIRBEN et al., 2012; RAHAL et al., 2014). Porém, quando há o desequilíbrio entre a produção de espécies reativas e a capacidade da célula de neutralizá-los, ocorre um o processo denominado estresse oxidativo (BIRBEN et al., 2012; RAHAL et al., 2014). O estresse oxidativo pode levar a dano aos lipídios, proteínas e ácidos nucleicos, devido à interação das espécies reativas com estas moléculas, induzindo necrose ou a apoptose (BIRBEN et al., 2012).

O alto consumo de oxigênio aliado ao baixo nível de antioxidantes e a alta concentração de ácidos graxos poli-insaturados resultam na grande susceptibilidade dos tecidos ao dano oxidativo (Figura 3) (ARUOMA, 1998; BIRBEN et al., 2012; FLOYD; HENSLEY, 2002). A interação entre as espécies reativas de oxigênio e de nitrogênio com os ácidos graxos insaturados das membranas celulares provocam um processo denominado peroxidação lipídica (BIRBEN et al., 2012; FLOYD; HENSLEY, 2002). Este processo está presente durante o

metabolismo celular normal e, em excesso nas doenças neurodegenerativas, é responsável pelo comprometimento celular, incluindo alteração na liberação de neurotransmissores. (ARUOMA, 1998; BIRBEN et al., 2012; FLOYD; HENSLEY, 2002; UTTARA et al., 2009). O estresse oxidativo em doenças neurodegenerativas se manifesta através da presença de proteínas oxidadas, de produtos de glicosilação avançada e da peroxidação lipídica (POHANKA, 2014). Conseqüentemente, há a formação de espécies tóxicas, tais como peróxidos, álcoois, aldeídos, carbonilas, cetonas e modificações no DNA nuclear e mitocondrial (BIRBEN et al., 2012; FLOYD; HENSLEY, 2002). Assim, a ingestão de antioxidantes exógenos pode colaborar no estado de equilíbrio (ARUOMA, 1998; BIRBEN et al., 2012).

Figura 3 - Esquema representativo do dano oxidativo.

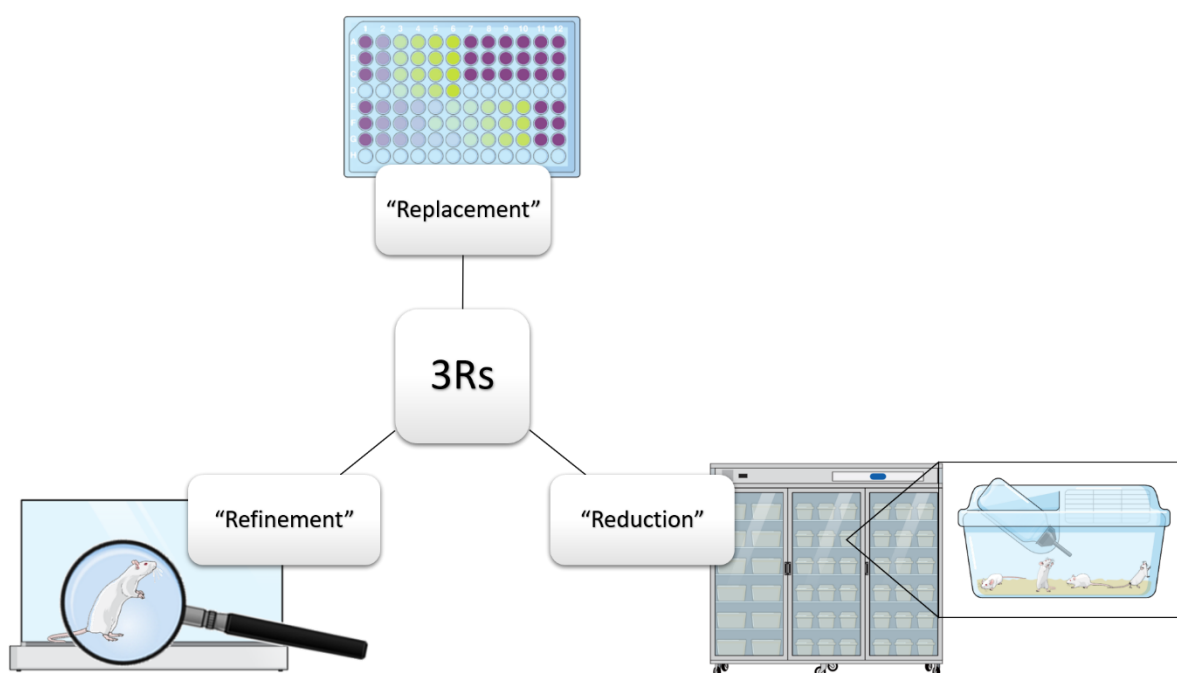


As espécies reativas descritas são produzidas e interagem com alvos celulares, formando produtos de oxidação responsáveis pelos danos no tecido. FONTE: o autor.

Para o estudo de doenças humanas, de moléculas terapêuticas potenciais e de toxicologia usa-se animais de experimentação, principalmente, modelos mamíferos de pequeno porte como camundongos e ratos. Entretanto, o uso de animais de experimentação para estudos toxicológicos de diferentes substâncias tem sido cada vez mais questionado (BHANUSHALI et al., 2015; JENNINGS, 2015; LIMONCIEL, 2014; ROBINSON, 2005). Têm-se preconizado a utilização de metodologias alternativas, as quais têm como principal objetivo a substituição, a redução no número de animais para este fim e quando necessário sua utilização o refinamento das técnicas empregadas (3Rs do inglês *replacement*, *reduction* e *refinement*) (BHANUSHALI et al., 2015; JENNINGS, 2015; LIMONCIEL, 2014;

ROBINSON, 2005) (Figura 4). Testes preliminares em células estão de acordo com a diretiva dos 3Rs, mas há a necessidade de que os testes *in vitro* sejam confiáveis e precisos para que as substâncias estudadas sejam adequadamente avaliadas antes de estudos clínicos e pré-clínicos (BRENDLER-SCHWAAB et al., 1994; LIMONCIEL, 2014). Ao mesmo tempo, é desejável que as avaliações de citotoxicidade *in vitro* sejam suficientemente rápidas e de baixo custo para permitir a triagem de um grande número de potenciais candidatos a medicamentos (BRENDLER-SCHWAAB et al., 1994; LIMONCIEL, 2014).

Figura 4 - Esquema representativo da diretiva dos 3Rs (“Replacement”, “Reduction” e “Refinement”).



“Replacement” refere-se à substituição de vertebrados, através da utilização de modelos alternativos, ensaios *in vitro* ou *in silico*. “Reduction” compreende a redução no número de animais utilizados quando necessário e “Refinement” é o refinamento das técnicas aplicadas a modelos de vertebrados evitando o estresse desnecessário e repetições experimentais. FONTE: o autor.

O *docking molecular*, é uma potente ferramenta utilizada mundialmente e tem a capacidade de prever a conformação (orientação) de um ligante de interesse quando este interage com um receptor ou uma enzima (CHEN, 2015; FERREIRA et al., 2015). Os programas usados na docagem molecular para descoberta de candidatos a fármacos em sua fase inicial elucidam as interações entre essas moléculas e o local alvo (DAINA; MICHIELIN; ZOETE, 2017; HU et al., 2020; PIRES; BLUNDELL; ASCHER, 2015). Da mesma forma, análises de triagem virtual de farmacocinética e a toxicidade também podem incluir abordagens computacionais utilizando plataformas online que possuem a capacidade

de prever se a substância avançará para o próximo estágio de pesquisa e será realmente usada como terapia no futuro. Estas plataformas detêm um banco de dados com embasamento em resultados prévios publicados utilizando modelos *in vivo* (DAINA; MICHIELIN; ZOETE, 2017; HU et al., 2020; PIRES; BLUNDELL; ASCHER, 2015; PIRES; KAMINSKAS; ASCHER, 2018).

1.1 JUSTIFICATIVA

O interesse por moléculas biologicamente ativas e com propriedades multialvo, capazes de desempenhar efeitos simultâneos frente a diferentes patologias é de grande relevância. Nesse contexto, o JM-20 é uma molécula nova que possui diferentes propriedades farmacológicas de interesse científico. Entretanto, ainda não foi totalmente elucidada quanto seu mecanismo de ação e a toxicidade. Assim, torna-se importante avaliar o efeito desse composto no sistema colinérgico neuronal e não-neuronal, via de diferentes patologias (principalmente neurodegenerativas), bem como, o potencial antioxidante do JM-20, pois o estresse oxidativo pode estar diretamente envolvido a causas ou consequências de diferentes doenças. Além disso, o estudo da citotoxicidade do JM-20 em células humanas saudáveis é de grande interesse. Neste sentido, o JM-20 também poderia inibir enzimas específicas, proteger do estresse oxidativo e quiçá apresentar efeitos adversos mínimos.

1.2 OBJETIVO GERAL

Avaliar o potencial multialvo do composto JM-20 como potencial antioxidante e seus efeitos na atividade de colinesterases e na toxicidade *in vitro* e *in silico*.

1.2.1 Objetivos específicos

→ Avaliar o efeito do composto JM-20 na atividade da enzima acetilcolinesterase de diferentes fontes.

→ Determinar o tipo de inibição enzimática causada pelo JM-20.

→ Avaliar o efeito do JM-20 na atividade da enzima butirilcolinesterase de diferentes fontes.

→ Realizar análises *in silico de docking molecular* afim de elucidar as interações entre o JM-20 e as enzimas.

→ Verificar o potencial antioxidante do composto, bem como efeito protetor em relação estresse oxidativo induzido por Fe^{2+} .

→ Avaliar a toxicidade do JM-20 em hemácias e leucócitos humanos saudáveis.

→ Prever a farmacocinética e toxicidade do JM-20 em através de um *screening virtual* utilizando ferramentas computacionais de simulação.

2.1 ARTIGO PUBLICADO – Molecular docking and *in vitro* evaluation of a new hybrid molecule (JM-20) on cholinesterase activity from different sources

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Research paper

Molecular docking and *in vitro* evaluation of a new hybrid molecule (JM-20) on cholinesterase activity from different sources



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ABSTRACT

The main function of AChE is the hydrolysis of the neurotransmitter acetylcholine (ACh) at the neuromuscular and in cholinergic brain synapses. In some pathologies, loss of cholinergic neurons may be associated with a deficiency of ACh in specific brain areas. Consequently, the study of new safe drugs that inhibit AChE is important, because they can increase ACh levels in the synaptic cleft without adverse effects. Here, we evaluated the effects of JM-20 (a benzodiazepine–dihydropyridine hybrid molecule) on cholinesterase (ChE) activities from distinct sources (AChE from *Electrophorus electricus* (EeAChE), human erythrocyte membranes (HsAChE (ghost)), total erythrocyte (HsAChE (erythrocyte)) and BChE from plasma (HsBChE) and purified enzyme from the horse (EcBChE)). Kinetic parameters were determined in the presence of 0.05–1.6 mM of substrate concentration. The interactions ChEs with JM-20 were performed using molecular docking simulations. JM-20 inhibited all tested AChE but not BChE. The IC₅₀ values were 123 nM ± 0.2 (EeAChE), 158 nM ± 0.1 (ghost HsAChE), and 172 nM ± 0.2 (erythrocytic HsAChE). JM-20 caused a mixed type of inhibition (it altered K_m and V_{max} of AChE). The molecular docking indicated the binding poses and the most plausible active isomer of JM-20. Besides giving important data for future drug design, our results help us understand the mode of action of JM-20 as a specific inhibitor of AChE enzymes.

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1. Introduction

Acetylcholine (ACh) is a neurotransmitter present in the central nervous system (CNS), peripheral nervous system (PNS) and at the neuromuscular junction [1–3]. ACh is synthesized by a specific enzyme (choline acetyltransferase) from choline and coenzyme A, transferred to vesicles and then released into the synaptic cleft, acting on muscarinic and/or nicotinic ACh receptors [1–4]. Physiologically, the interruption of cholinergic neurotransmission is performed by cholinesterases [1]. In humans, both butyrylcholinesterase (BChE) and acetylcholinesterase (AChE) can hydrolyze ACh [1–4].

AChE is the enzyme responsible for regulating ACh levels at synapse cleft by hydrolyzing ACh into choline and acetate [3–7]. However, in some pathologies (for example in Alzheimer's disease), dysfunctions in brain regions that contain cholinergic neurons can occur, resulting in the loss of cholinergic neurotransmission [2,8]. Thus, the discovery of new drugs that can inhibit AChE activity and increase ACh levels in the synaptic cleft (without causing adverse effects) is relevant. Multiple variants of AChE can be produced by alternative splicing, for instance, the synaptic, erythrocytic and readthrough AChE isoforms [3,6,9]. Thus, extensive research of the cholinergic system has been used to unravel novel therapeutic medicines for neurological and psychiatric pathologies, which present dysfunction in the cholinergic system [2,5,10,11].

JM-20 (3-ethoxycarbonyl-2-methyl-4-(2-nitrophenyl)-4,11-dihydro-1H-pyrido [2,3-b] [1,5] benzodiazepine) is a compound which derives from 1,5-benzodiazepines, by adding the 1,4-dihydropyridine moiety fused to the benzodiazepine ring (Fig. 1)

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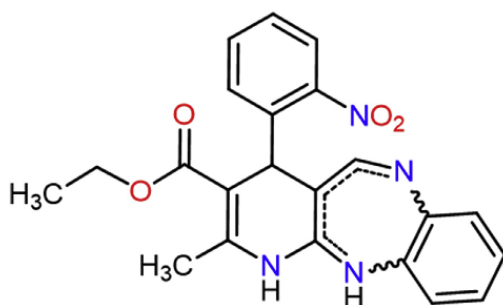


Fig. 1. Structural formula of the compound 3-ethoxycarbonyl-2-methyl-4-(2-nitrophenyl)-4,11-dihydro-1H-pyrido [2,3- b] [1,5] benzodiazepine (JM-20).

[12,13]. JM-20 reduces the aggressive behavior of socially isolated mice and increases the latency period for the beginning of the tonic crisis and the percentage of survival of animals with seizures caused by pentylenetetrazole [14]. JM-20 reduces anxiolytic-like responses in mice or rats treated orally with 2–10 mg/kg of the compound [15]. Furthermore, JM-20 may act as a multi-target neuroprotective agent by reducing the neuronal excitotoxic injury and protecting the mitochondria from Ca^{2+} -induced toxicity, thereby preserving the cellular energy balance [13,16–18]. Importantly, JM-20 does not induce mortality, suggesting the absence of toxic effects [15] which leads us to consider its use as a promising compound for therapeutic use. Here, we evaluated *in vitro*, the effect of JM-20 on AChE activity from different sources, e.g., the purified enzyme from *Electrophorus electricus* (EeAChE), human erythrocyte membranes (HsAChE (ghost)) and, human total erythrocyte (HsAChE (erythrocyte)). We also verified the effect of JM-20 in the BChE activity purified from *Equus caballus* (EcBChE) and in non-purified BChE from human plasma (HsBChE). Moreover, kinetic analyses and *in silico* molecular docking were performed to elucidate the interactions between the compound and the enzymes, as well as the type of inhibition.

2. Material and methods

2.1. Compounds and reagents

JM-20 was synthesized, purified and characterized as reported elsewhere [15]. The compound was dissolved in dimethyl sulfoxide (DMSO). All chemicals and purified enzymes used were purchased from Sigma-Aldrich (St. Louis, MO, USA).

2.2. Sample preparation

2.2.1. Isolation of human erythrocyte membrane (ghost)

Human erythrocyte were obtained from a healthy volunteer from the toxicological biochemistry laboratory at the Universidade Federal de Santa Maria. The Ethical Committee of the UFSM approved the protocol (n° 3.471.823) used in this study. Freshly drawn heparinized human blood was centrifuged at $200\times g$ for 10 min. Plasma was removed, isotonic KCl (150 mM) was added, and the sample was centrifuged at $200\times g$. This procedure was repeated thrice. Red cells were maintained at 0–4 °C for the erythrocyte ghosts preparation, according to Bjerrum, 1979 [19]. Cells (20 mL) were lysed by the addition of 20 mL hemolyzing buffer A (4 mM $MgSO_4$, 3.8 mM CH_3COOH , pH 3.6), resulting in 50% hematocrit. After 1 min incubation, the cellular suspension was centrifuged at $23,000\times g$ for 12 min, and posteriorly the

supernatant was removed. In this procedure, the erythrocyte ghosts content was reduced to ~3% (wt/wt). At this stage, samples can present a pink color. The tonicity of the 3% (wt/wt) hemoglobin containing packed ghost was increased by addition 100 μL of buffer C (KCl 2 M) per mL of erythrocyte ghosts suspension. The suspension was incubated for 2 min, and then washed through the addition of 220 mL of medium B (4 mM $MgSO_4$, 1.2 mM CH_3COONa , 2 mM NaH_2PO_4/Na_2HPO_4 , pH 7.0) and centrifuged at $23,000\times g$ for 12 min. This procedure was repeated four times until the erythrocyte ghosts had a greyish white color. The erythrocyte ghosts were resuspended in isotonic KCl (150 mM). HsAChE assays using these membranes were realized immediately after erythrocyte ghosts isolation. Protein was adjusted to approximately 165 $\mu g/mL$ and the final content of protein in the AChE reaction system was 3–4 μg per well (0.2 mL).

2.2.2. Plasma and erythrocyte preparation

Freshly drawn heparinized human blood was centrifuged at $200\times g$ for 10 min. Plasma was separated and diluted 40 times and the erythrocyte were washed using saline solution and centrifuged at $200\times g$ for 10 min, this procedure was repeated 3 times. After washing, red blood cells were diluted 600 times. HsBChE and HsAChE assays were performed immediately after dilution of the samples. The procedure was performed as described elsewhere [11,20].

2.2.3. ChE enzymatic assay

ChE activity was measured according to the method proposed by Ellman et al. [21]. The reaction medium consisted of the following reagents at final concentration: 10 mM of potassium phosphate buffer (pH 7.4), 1 mM DTNB [5,5'-dithio-bis(2-nitrobenzoic acid)], AChE enzyme (0.0125 U of EeAChE/0.2 mL, i.e., the final volume in the wells; 3–4 μg protein of HsAChE from human erythrocyte membranes/0.2 mL, 10–15 μg protein of HsAChE from human total erythrocyte/0.2 mL, 15–20 μg protein of HsBChE from human plasma/0.2 mL or 0.005 U of purified EcBChE/0.2 mL), and JM-20 (25 nM - 40 μM) or 0.05% DMSO (control group). This system was preincubated for 30 min at room temperature (26 ± 2). Then, we added acetylthiocholine iodide (0.4 mM, final concentration) to start the reaction. The increase in absorbance was monitored in a BioRad iMark microplate reader (Japan) at 415 nm using a plate of 96 wells. Some experiments were also performed in a cuvette (1 mL; in a Shimadzu UV-1800 spectrophotometer, Japan) for determining the effect of JM-20 in the EeAChE and EcBChE in the absence of pre-incubation. The final concentrations of protein used in the cuvette assays were the same used in 96 well plates.

2.2.4. Determination of IC_{50} values

The Half-maximal inhibitory concentration (IC_{50}) was determined by nonlinear regression log (inhibitor) vs. response – variable slope in accordance to equation (1), using GraphPad, Prism 6.01.

$$Y = Bottom$$

$$+ (Top - Bottom) / (1 + 10^{((LogIC50 - X) * HillSlope)}); \quad (1)$$

HillSlope describes the steepness of the family of curves (A Hill-Slope of -1.0 is standard).

Top and Bottom are plateaus in the units of the Y-axis.

2.2.5. Kinetic analysis

Kinetic parameters of the AChE activity were measured according to the method proposed by Ellman et al. [21] previously described by varying the concentration of acetylthiocholine as a

substrate (0.05, 0.1, 0.2, 0.4, and 1.6 mM) in the presence and absence of JM-20 (0, 75, 150, 250, and 400 nM) using a plate of 96 wells. Different plots were constructed for assessing the type of inhibition. The V_{max} constant was obtained through nonlinear regression enzyme kinetic – inhibition, substrate inhibition using GraphPad, Prism 6.01, and K_i were obtained through nonlinear regression enzyme kinetic – inhibition, mixed inhibition in accordance to equation (2), using GraphPad, Prism 6.01.

$$V_0 = V_{max} / (1 + I / (\text{Alpha} * K_i)) * [S] / K_m * (1 + I / K_i) / (1 + I / (\text{Alpha} * K_i)) + [S]; \quad (2)$$

Alpha determines the mechanism. Its value determines the degree to which the binding of inhibitor changes the affinity of the enzyme for the substrate.

K_i is the inhibition constant.

2.2.6. In silico analysis

To simulate the binding mode of interactions between the JM-20 molecules and the cholinesterase enzymes, Vina 1.1.1 program was used [22], with exhaustiveness of 200, using two docking protocols: a) one in the presence of water molecules [23–25] (the methodologies and results from the docking performed in the absence of water are presented as supplementary materials in supporting information) and b) other with the highly-conserved network of water molecules in the human acetyl- and butyryl-cholinesterases [26]. In the first moment, a blind docking was carried out with the enzymes, where were observed the binding of the ligands (JM-20 isomers) in the respective active sites. To improve the interactions between the enzyme and ligands, a local docking was further made. By the *HsAChE* structure from the Protein Data Bank – PDB – with the code 4EY6 [27], the gridbox was centered on the coordinates: $x = -13.25$, $y = -46.76$ and $z = 33.52$, with the dimensions: $30 \times 30 \times 30 \text{ \AA}$. For *HsBChE* docking, the structure PBD: 4BDS from the Protein Data Bank [28] was used. The gridbox was centered on the coordinates: $x = 138.8$, $y = 111.5$ and $z = 39.9$, with the dimensions: $30 \times 30 \times 30 \text{ \AA}$. For the protein structures, ions, ligands, and water were removed, while the hydrogens were added using the CHIMERA 1.8 program [29]. In the docking with water, and for *HsAChE* the HOH molecules 704, 705, 707, 773, 803, 805 and 815 were kept; while for *HsBChE*, the water molecules were 2045, 2064, 2089, 2090, 2093, 2131, and 2140 [26]. JM-20 molecules, galantamine, and tacrine were created using the Avogadro 1.1.1 software [30] followed by semi-empirical PM6 [31] geometry optimization using the MOPAC2016 program [32]. The ChE enzymes and the ligands were prepared for the docking using AutoDock Tools 4.2 [33], in which the compounds were considered flexible (with PM6 charges) and the enzymes rigid (with Gasteiger charges). The non-polar hydrogens were omitted. The structures with the most favorable binding free energy were selected and analyzed using Accelrys Discovery Studio 3.5 [34]. For more details, see the Supporting information.

2.2.7. Statistical analyses

Statistics have been performed using the GraphPad Prism 6 (version 6.01, GraphPad Software, Inc., USA). Results were expressed as the mean \pm standard error of the mean (SEM). ChE activity data were analyzed using nonlinear regression enzyme kinetic – inhibition, substrate inhibition (comparisons with parameters obtained with Michaelis-Menten see Supplementary Material Table S2) and tested when slopes and intercepts were significantly different. Values of $p < 0.05$ were considered significantly different.

3. Results and discussion

Here, we expand the knowledge regarding the JM-20 compound, a new alternative as AChE inhibitor, which belongs to a new family of 1,5-benzodiazepines, structurally different from currently available 1,5-benzodiazepines by the presence of a 1,4-dihydropyridine moiety fused to the benzodiazepine ring [12,14]. JM-20 has been evaluated as a possible inhibitor of *EeAChE*, *HsAChE* (ghost), *HsAChE* (erythrocyte) and of *EcBChE* and *HsBChE*. Regardless of the origin of AChE, compound JM-20 acts as an AChE inhibitor at nanomolar range and, surprisingly, IC_{50} values (Table 1) were lower than those previously reported for tacrine (reference drug in inhibition of AChE) that present IC_{50} around $180 \text{ nM} \pm 10$ for *EeAChE* and $150 \text{ nM} \pm 10$ for *HsAChE* [35–37]. The IC_{50} values give an indication about the potency of a substance in inhibiting AChE activity [38]. The IC_{50} and K_i values obtained here for JM-20 is within the range previously reported for tacrine using either the *HsAChE* or *EeAChE* [35,39–41].

Additionally, studies using compounds structurally similar to JM-20 (containing dihydropyridine or benzodiazepine group) demonstrated a potential inhibitory effect on *EeAChE* activity [35,37,42,43]. To understand the inhibitory mechanism of the JM-20 compound on AChE activity, reactions were measured in the presence of different concentrations of substrate and data were analyzed according to the nonlinear regression enzyme kinetic – inhibition (substrate inhibition; GraphPad, Prism), Lineweaver-Burke, Eadie Hofstee, and Dixon method. Three different concentrations of inhibitor were used. K_m and V_{max} values were calculated from the slope of regression (Table 2 and Table 3, respectively).

JM-20 inhibited the enzyme activity proportionally to the increase in the concentration of the compound. Besides, JM-20 increased the K_m values as a function of its concentration (Table 2); while V_{max} decreased in the presence of increasing JM-20 concentrations (Table 3). These data suggest a mixed type of enzymatic inhibition [39,44]. The same kinetic assay approach was used for other sources of AChE. We used DMSO (0.05%) as control or JM-20 at a final concentration of 75, 150 and 250 nM in *EeAChE* (Fig. 2). Nonlinear regression of the Lineweaver-Burke plot indicated significant differences between the slopes in the absence or presence of JM-20 ($F = 122.21$, $p < 0.0001$). Similar results were observed with Eadie Hofstee plot ($F = 27.94$, $p < 0.0001$).

HsAChE (ghost) (Fig. 3) was also inhibited by JM-20, but in this test, we used 100 nM, 250 nM and 400 nM of inhibitor. The statistical analysis also indicated a significant difference between slopes obtained in the absence and presence of JM-20 ($F = 30.18$, $p < 0.0001$ for the Lineweaver-Burk plot and $F = 3.58$, $p < 0.0001$ for the Eadie Hofstee plot). For *HsAChE* (erythrocyte), we tested 75, 150 and 250 nM of JM-20 (Fig. 4). The difference between the slopes in the absence or presence of JM-20 was significant ($F = 12.93$, $p < 0.0001$ for the Lineweaver-Burk plot and $F = 3.29$, $p = 0.0254$ for the Eadie Hofstee plot). Furthermore, JM-20 was specific for AChE because it did not alter *HsBChE* and *EcBChE* activities when

Table 1

JM-20 IC_{50} and K_i values for ChE activity. IC_{50} was determined by nonlinear regression (log inhibitor vs response) in accordance to equation $Y = \text{Bottom} + (\text{Top} - \text{Bottom}) / (1 + 10^{-(\text{Log}(IC_{50} - X) * \text{Hillslope})})$, using GraphPad, Prism 6.01, and K_i values were obtained through nonlinear regression enzyme kinetic – inhibition, mixed inhibition in accordance to equation $V_0 = V_{max} / (1 + I / (\text{Alpha} * K_i)) * [S] / K_m * (1 + I / K_i) / (1 + I / (\text{Alpha} * K_i)) + [S]$.

ChE	IC_{50} (nM)	K_i (nM)
<i>EeAChE</i>	123 ± 0.2	18 ± 2.5
<i>HsAChE</i> (ghost)	158 ± 0.1	165 ± 25
<i>HsAChE</i> (erythrocyte)	172 ± 0.2	99 ± 24

Table 2

K_m values for cholinesterase (ChE activity) from different sources in the presence or absence of JM-20. K_m expressed in mM (acetylthiocholine as substrate).

Concentration (nM)	<i>EeAChE</i>	<i>HsAChE</i> (ghost)	<i>HsAChE</i> (erythrocyte)	<i>HsBChE</i> (plasma)
Control	0.15 ± 0.03	0.28 ± 0.08	0.13 ± 0.02	0.27 ± 0.03
75	0.25 ± 0.09	—	0.21 ± 0.06	0.29 ± 0.06
100	—	0.32 ± 0.13	—	—
150	0.53 ± 0.13	—	0.17 ± 0.02	0.17 ± 0.05
250	0.38 ± 0.09	0.41 ± 0.24	0.43 ± 0.13	0.24 ± 0.03
400	0.36 ± 0.05	0.34 ± 0.05	—	0.25 ± 0.02

Table 3

V_{max} values for ChE activity from different sources in the presence or absence of JM-20. V_{max} activity expressed as Δ absorbance at 415 nm/mg protein (for *EeAChE*); Δ absorbance at 415 nm/min/mg protein (for *HsAChE*-Ghost); and, Δ absorbance at 415 nm/h/mg protein (for *HsAChE*-erythrocyte and *HsBChE*-plasma). Absorbance was determined in wells containing 0.2 ml of enzyme incubation media (for details see Material and Methods) in a BioRad iMark microplate reader (Japan).

Concentration (nM)	<i>EeAChE</i>	<i>HsAChE</i> (Ghost)	<i>HsAChE</i> (erythrocyte)	<i>HsBChE</i> (plasma)
Control	42 ± 4	78 ± 13	50 ± 2	156 ± 10
75	20 ± 4	—	50 ± 8	152 ± 20
100	—	60 ± 15	—	—
150	16 ± 2	—	32 ± 2	124 ± 11
250	9 ± 1	47 ± 18	31 ± 4	130 ± 5
400	5 ± 1	30 ± 2	—	139 ± 3

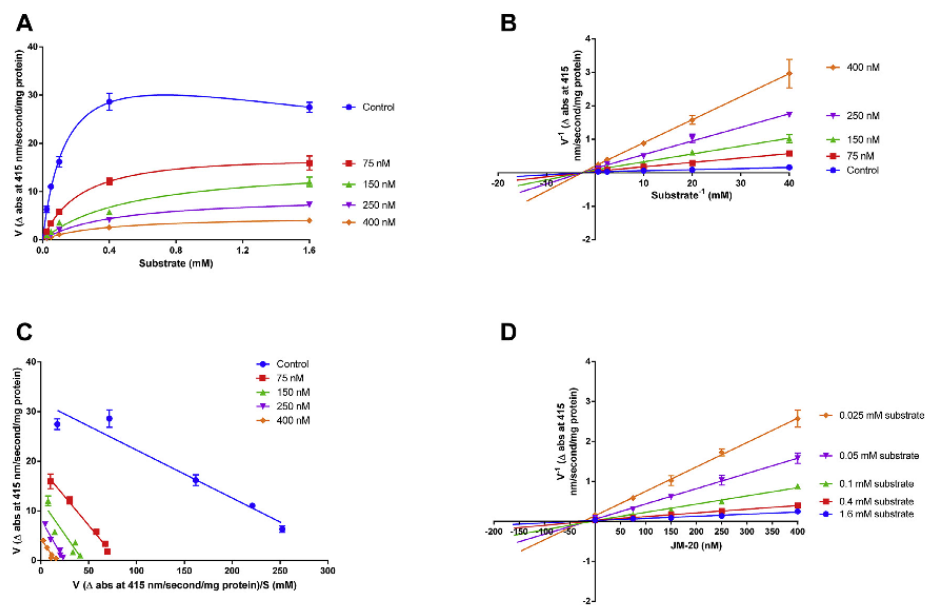


Fig. 2. Enzyme kinetic plots for acetylcholinesterase from the fish *Electrophorus electricus*: Nonlinear regression was calculated taken into consideration the substrate inhibition (acetylthiocholine) (A), Lineweaver-Burke (B), Eadie Hofstee (C) and Dixon plot (D) of *EeAChE*. The data were expressed as mean ± SEM. The differences between the slopes are statistically significant.

tested in the nM range (Fig. 5A and Fig. 5B, respectively). To our knowledge, we showed for the first time the inhibition profile of JM-20 on AChE activity using kinetic analyses. The value of K_m indicates a substrate concentration that is required to reach half the maximum reaction velocity [44,45]. Mixed-type inhibitors modify both the binding of substrate and the velocity of the catalyzed reactions. Consequently, in our case, the type of inhibition of AChE by JM-20 was of the mixed type, because it changed both kinetic parameters [44,45]. We also observed that the highest concentration

of the substrate used here (1.6 mM) reduced the velocity of the reaction catalyzed by *EeAChE* (Fig. 2A). The inhibition possibly resulted from the binding of more than one substrate molecule (acetylthiocholine) in the active site of *EeAChE* [46].

The effects of JM-20 (in the absence of pre-incubation) with *EeAChE* and *EcBChE* activities were also determined using a cuvette (Fig. 6). The results confirmed the specificity of JM-20 for AChE activity (Fig. 6A). Indeed, an inhibitory effect of *EcBChE* (Fig. 6B) enzyme was obtained only using μ M concentrations of JM-20 and

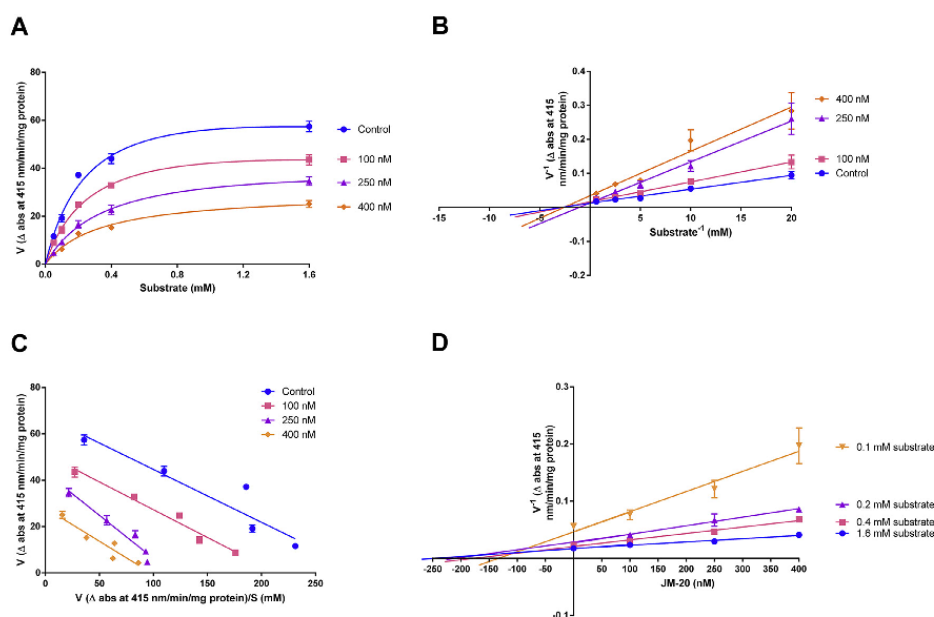


Fig. 3. Enzyme kinetic plots for acetylcholinesterase from the human erythrocyte ghosts: Nonlinear regression was calculated taken into consideration the substrate inhibition (acetylthiocholine) (A), Lineweaver-Burke (B), Eadie Hofstee (C) and Dixon plot (D) of *HsAChE* (ghost). The data were expressed as mean \pm SEM. The differences between the slopes are statistically significant.

the maximal inhibition was only about 50%, which was obtained with 10–20 μ M of JM-20 (Fig. 6B).

Analyzing the structure of JM-20, we observe a chiral carbon on the position 4 (position that binds the 2-nitrophenyl group), and also it could present resonance in this structure, with the hydrogen atom from the amine group of the diazepine ring interchanging between the positions 6 and 11, being thus a tautomer. Furthermore, the diazepine ring has a nitrogen atom in the sp^3 hybridization (in position 6 or 11), which makes this ring nonplanar. In this case, due to the non-planarity of the diazepine ring, the benzene moiety can present two different conformations in the JM-20 molecule, like to an atropisomer [47]. A similar observation was found to the diazepam analogs, where the R isomers present the highest recognition for the benzodiazepine receptor complex [48,49]. All possibilities of the isomers of JM-20 and the nomenclature used for its identification are showed in Fig. S1.

Because isomers may vary in binding specificity and activity, the determination of the active form is crucial to elucidate the mechanism of action and potential pharmacological strategies [50–55]. Considering the importance of the structure and conformation of the molecules that interact in a determined target, we perform the molecular docking with the isomers of JM-20 (a–h) and the cholinesterases enzymes (AChE from *Homo sapiens* (*HsAChE*) and *Electrophorus electricus* (*EeAChE*), and BChE from *Homo sapiens* (*HsBChE*) and *Equus caballus* (*EcBChE*)) to determine the binding pose of them and the probability of the most active isomer.

Here, we applied two docking protocols, one without water molecules [23–25] and other taking into account the highly-conserved network of water molecules [26] (Fig. S2). In the molecular docking in the absence of water molecules (using the *HsAChE*, *EeAChE*, *HsBChE*, and *EcBChE*), all JM-20 isomers (a–h) can access the active site of the enzymes, with a specific pattern of interactions, where the isomers that varied the amine group (NH)

position of 6–11 showed similar binding modes, generally (Fig. S3). In the absence of water, the JM-20 binds in the bottom of the gorge, near to the catalytic triad of both *HsAChE* and *HsBChE* enzymes.

Molecular docking simulations in the presence of the water molecules network (using the *HsAChE* and *HsBChE*), also demonstrated that the JM-20 interact in the active site of both cholinesterase, but with different binding pose when compared with the docking without water molecules (Figs. 7 and 8). These results are in accordance with the observations of Rosenberry et al. 2017 [26], where the exclusion of highly-conserved water molecules in the cholinesterase docking simulations resulted in unreliable binding poses. Furthermore, water molecules have potential importance in the catalysis of cholinesterases [26,56–58]. In this way, here we highlight the results from the docking in the presence of water. In the *HsAChE* the JM-20 isomers interact mainly in the Peripheral Anionic Subsite (PAS), interacting with the W286 by H-bonds and anion- π interaction, with Y341 and L289 by hydrophobic π - π and alkyl- π interactions, respectively, and with the S293 by H-bonds (Fig. 7). The analysis of the interactions and the ΔG_{bind} (Table 4) shown that the isomers a, b, e, and f present a very similar binding pose, and more favorable binding energy when compared with the isomers c, d, g, and h, indicating that the 4-R isomers interact better with the *HsAChE*. Besides, the benzodiazepine ring conformation in b and f allows a hydrophobic interaction with the L289 residue (Fig. 7B and F), which is correlated with its better binding energy (Table 4).

In the *HsBChE* docking, the JM-20 isomers interacted in the Peripheral Anionic Subsite (PAS), in Acyl Binding Pocket (ABP), and the Catalytic Triad (CT) subsites. Here, we observed that the water 2093 has an important role in the interactions with the inhibitor, by the H-bonds with the acetyl and nitro moieties from JM-20 (Fig. 8). The isomers a and e showed a very similar binding pose, while the other isomers demonstrated a more diverse binding mode (Fig. 8),

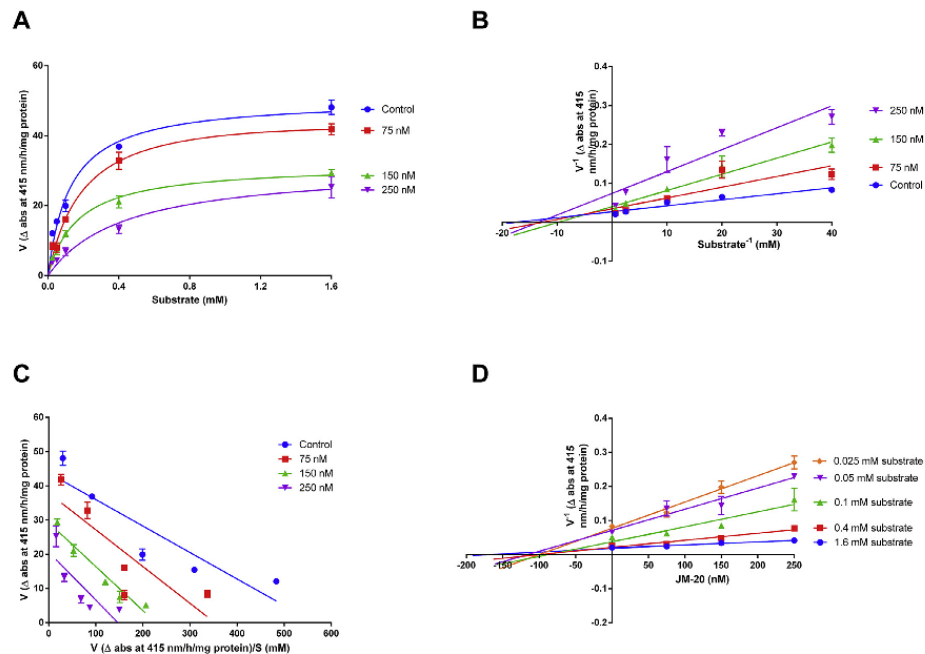


Fig. 4. Enzyme kinetic plots for acetylcholinesterase from the human erythrocyte: Nonlinear regression was calculated taken into consideration the enzyme inhibition by excess of substrate inhibition (A), Lineweaver-Burke (B), Eadie Hofstee (C) and Dixon plot (D) of *HsAChE* (erythrocyte). The data were expressed as mean \pm SEM. The differences between the slopes are statistically significant.

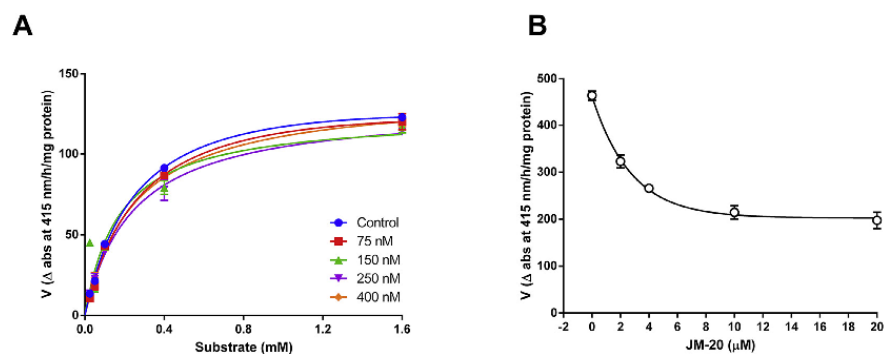


Fig. 5. *HsBChE* (A) and *EcBChE* (B) activities.

and as observed by the *HsAChE*, the isomers a and e presented a better ΔG_{bind} (Table 4). The isomers with the lowest ΔG indicate a more favorable interaction with the proteins and consequently could have a major contribution to the enzyme inactivation. However, the other isomers also could have an important role in the enzyme inactivation. In the case of a and e, they interact with the *HsBChE* mainly by hydrophobic interactions with the L286, F329, A328, and Y332 residues, and by H-bonds with S198, H438, and HOH2093 (Fig. 8A and E).

In general, position 4 (chiral center) of the JM-20 presented many binding modes in the *HsAChE* and *HsBChE*, while the

tautomerism of the amine group affects less the *HsAChE* than the *HsBChE*. Docking simulations also revealed that the planarity of the benzodiazepine group affects the binding pose of JM-20 in the enzymes (Figs. 7 and 8). JM-20 bind in the PAS region on *HsAChE*, while in the *HsBChE* it could interact in the bottom of the gorge, close to the CT. This distinct binding pose occurs probably due to the BChE active site are more voluminous than the AChE (a difference of about 200 \AA^3) [26,59,60]. The comparison between the residues of the AChEs and BChEs active sites shows specific amino acid replacements, such as the replacement of bigger and hydrophobic residues (Phe, Tyr, Trp) for smaller and hydrophobic

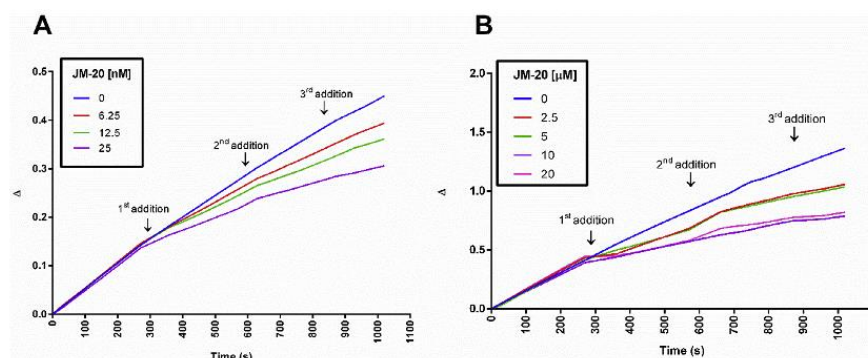


Fig. 6. *EeAChE* (A) and *EcBChE* (B). Each indicated point corresponds to the addition of JM-20 to the final concentration indicated for each addition. For instance, in the red line for AChE (A) the first addition corresponds to a final concentration of 6.25 nM of JM-20, the second, 12.5 nM and after the third 18.75 nM of JM-20. The *EeAChE* (A) activity was determined in a cuvette (volume of 1 ml) at 412 nm (Shimadzu UV-1800 spectrophotometer, Japan) in the presence of 12 μ g of purified enzyme. The *EcBChE* (B) was determined in a cuvette (volume of 1 ml) at 412 nm (Shimadzu UV-1800 spectrophotometer, Japan) in the presence of 5 μ g of purified enzyme.

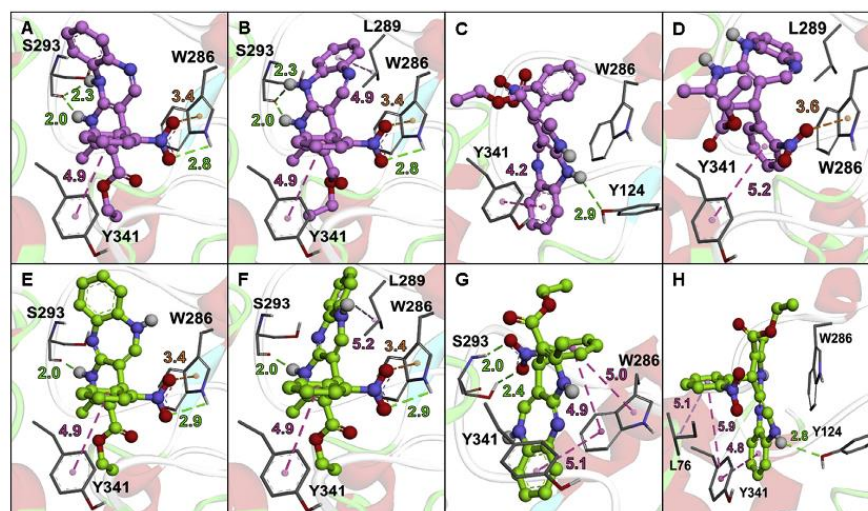


Fig. 7. Binding poses of JM-20 isomers a–h (A–H) in the *HsAChE* enzyme, from molecular docking in the presence of highly-conserved water molecules. The JM-20 isomers (a–h) are in a ball and stick model, with the carbon atoms from the isomers a–d and e–h in the pink and green color, respectively. Only the main residues involved in the interactions are shown in stick model. The respective interactions, H-bonds (in green), electrostatic π -anion (in orange), and hydrophobic π - π and alkyl- π (in pink) are represented by colored dot lines. The interaction distances are in Å.

residues (Leu, Ala, Val) (Fig. S4 and Table S1). The elucidation of these features is important to determine how these molecules interact with a specific target and help us in the development of new drugs. According to the docking simulations, in the case of the JM-20, the isomer b demonstrated to be the main structure involved in the AChE inhibition, but the coexisting isomers may allow a synergistic activity.

The inclusion of highly-conserved water molecules in the docking simulations lead to a different binding pose of ligands when compared with the docking without water, despite the ΔG (Table 4) and the redocking showed similar values (Fig. S5). However, the results obtained by the inclusion of water molecules method are more realistic [26]. JM-20 binding pose is similar to tacrine-dihydropyridine hybrids [35,36,61], where the R-isomers appear to be more active and shown π - π stacking with W286,

besides present IC_{50} for *HsAChE* in the range of nM. Coumarin hybrids also demonstrated important interactions with the PAS, which could help in the prevention of pro-aggregation of AChE toward beta-amyloid protein ($A\beta$) [62]. These data indicate that the interaction with the W286 in PAS is essential for these classes of molecules. As proposed by Leon et al. [35], the lack of the tryptophan residue in the *HsBChE* PAS (replaced by A277, Table S1) could explain the poor inhibition of this enzyme.

4. Conclusion

In conclusion, JM-20 can be an alternative therapy to treat neurological disorders related to cholinergic system disturbances because it has a potential inhibitory effect on AChE. Importantly, we reinforce that other studies are required to assess whether the

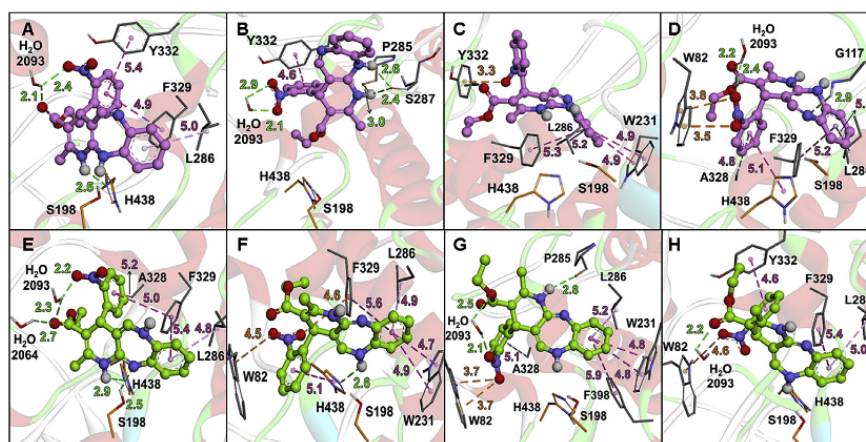


Fig. 8. Binding poses of JM-20 isomers a–h (A–H) in the HsBChE enzyme, from molecular docking in the presence of highly-conserved water molecules. The JM-20 isomers (a–h) are in a ball and stick model, with the carbon atoms from the isomers a–d and e–h in the pink and green color, respectively. Only the main residues involved in the interactions are shown in stick model. The respective interactions, H-bonds (in green), electrostatic π -anion (in orange), and hydrophobic π - π and alkyl- π (in pink) are represented by colored dot lines. The interaction distances are in Å.

Table 4

Predicted ΔG_{bind} (kcal/mol) data from molecular docking between ChE and the JM-20 isomers (a–h).

	a	b	c	d	e	f	g	h	Rd.
<i>With water</i>									
HsAChE	−9.2	−10.0	−8.1	−7.3	−9.0	−9.8	−8.3	−8.3	−10.5
HsBChE	−10.6	−9.1	−9.9	−10.6	−11.0	−9.7	−9.9	−9.8	−8.5
<i>Without water</i>									
HsAChE	−9.5	−11.1	−9.0	−7.3	−9.0	−10.5	−10.5	−10.5	−10.4
HsBChE	−10.3	−10.8	−10.2	−10.2	−10.2	−10.3	−10.3	−10.5	−8.1
EeAChE	−8.2	−10.6	−7.3	−7.7	−8.0	−9.2	−9.6	−10.0	−8.7
EcBChE	−9.8	−10.1	−9.6	−9.8	−10.4	−9.9	−10.1	−10.6	−8.1

Rd: redocking, for AChE the ligand is galantamine, and for BChE is tacrine.

AChE inhibitor will be maintained *in vivo*, as well as toxicity studies following different routes, doses and time of exposure. Besides, simulations by molecular docking indicate that the 4-R isomers (a, b, e, and f) interact better with the HsAChE, in the PAS region. These results help understand the mode of action of JM-20 on AChE, allowing more precise evidence for future pharmacological approaches.

Author contributions

Fernanda D'Avila da Silva: conceived and designed the analysis, collected the data, contributed data or analysis tools, performed the analysis, wrote the paper.

Pablo Andrei Nogara: collected the data, contributed data or analysis tools, performed the analysis, wrote the paper.

Estael Ochoa-Rodríguez: wrote the paper, synthesis and characterization of JM-20.

Yanier Nuñez-Figueroa: conceived and designed the analysis, wrote the paper, synthesis and characterization of JM-20.

Maylin Wong-Guerra: wrote the paper, synthesis and characterization of JM-20.

Denis Broock Rosemberg: contributed data or analysis tools, wrote the paper.

João Batista Teixeira Rocha: conceived and designed the analysis, contributed data or analysis tools, wrote the paper.

Declaration of competing interest

Material submitted entitled "Molecular docking and *in vitro* evaluation of a new hybrid molecule (JM-20) on cholinesterase activity from different sources" is original. This manuscript has not been published previously and it is not under consideration elsewhere. There is no conflict of interest. All authors agree to have the article published.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.biochi.2019.11.011>.

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2.1.1 Material de apoio do artigo - Molecular docking and *in vitro* evaluation of a new hybrid molecule (JM-20) on cholinesterase activity from different sources

Supporting Information

Molecular docking and *in vitro* evaluation of a new hybrid molecule (JM-20) on cholinesterase activity from different sources

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Molecular docking and protein homology modelling of AChE from *Electrophorus electricus* (*EeAChE*) and BChE from *Equus caballus* (*EcBChE*)

Three-dimensional structures of *EeAChE* and *EcBChE* were obtained by the protein homology modeling [1], and the Swiss-Model (<https://swissmodel.expasy.org>) was used to build them, using the amino acid sequence of the *EeAChE* and *EcBChE* taken from the National Center for Biotechnology Information – NCBI (<https://www.ncbi.nlm.nih.gov/pubmed/>) with the codes XP_026871422.1 and AAF61480.1, respectively. As the protein template, the *Tetronarce californica* AChE (PDB 3I6M) [2] and the *HsBChE* (PDB 4BDS) were used. The validation of the protein models by PROCHECK, Verify 3D and ProSA programs showed that the structures presented conformation similar to the native proteins (Table S3) [3–5]. For docking simulations, the *EeAChE* gridbox was centered on the coordinates: $x = 3.53$; $y = 65.57$; $z = 66.74$; with the dimensions 30^3 \AA , while the *EcBChE* gridbox was centered on the coordinates: $x = 3.76$; $y = 13.45$; $z = 13.57$; with the dimensions 30^3 \AA . Considering that the ChE models do not present water molecules, we used these structures in the docking simulations in the absence of water, only. For more details, see the section 2.3.6. *In silico analysis*.

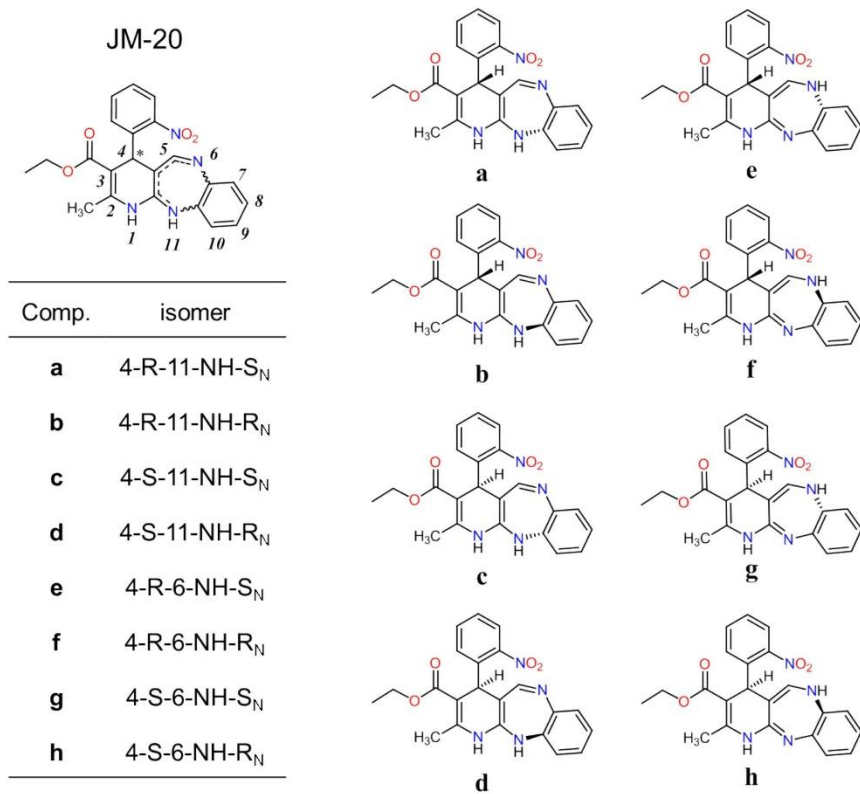


Figure S1. JM-20 isomers and the nomenclature used for their identification. To determine the configuration of the N sp³ of the diazepine ring, the N atom was considered a chiral center, with the dihydropyridine moiety and H atom in the plan, and the electron pair with low priority.

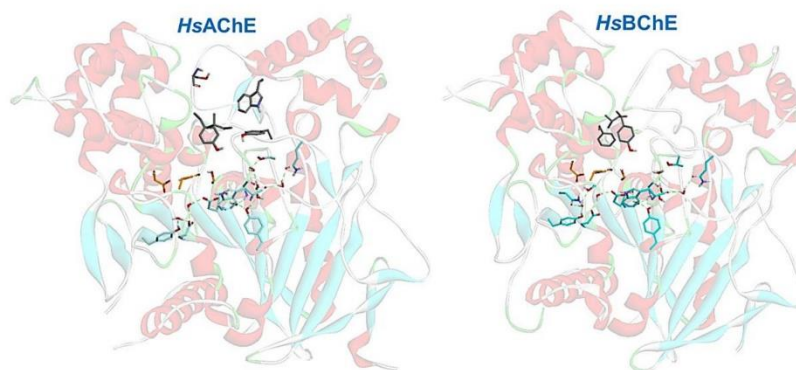


Figure S2. Structures of *HsAChE* (PDB 4EY6) and *HsBChE* (PDB 4BDS). The catalytic triad (Glu, His and Ser) are showed with carbon atoms in orange color, and the residues involved in the water molecules network with carbon atoms in blue color (*HsAChE*: Gln71, Trp86, Gly120, Ser125, Tyr133, Glu202, Tyr428, and Glu450; *HsBChE*: Gln67, Trp82, Gly115, Thr120, Tyr128, Glu197, Ser224, Asn322, Tyr419, and Glu441). For the *HsAChE* (4EY6) the water molecules are 704, 705, 707, 773, 803, 805 and 815, and for *HsBChE* (4BDS): 2045, 2064, 2089, 2090, 2093, 2131, and 2140. Some residues from Peripheral Anionic Subsite, PAS (*HsAChE*: Tyr124, Trp286, and Tyr341; *HsBChE*: Tyr332) and Acyl Binding Pocket, ABP (*HsBChE*: Leu286 and Phe329) are represented by grey carbon atoms.

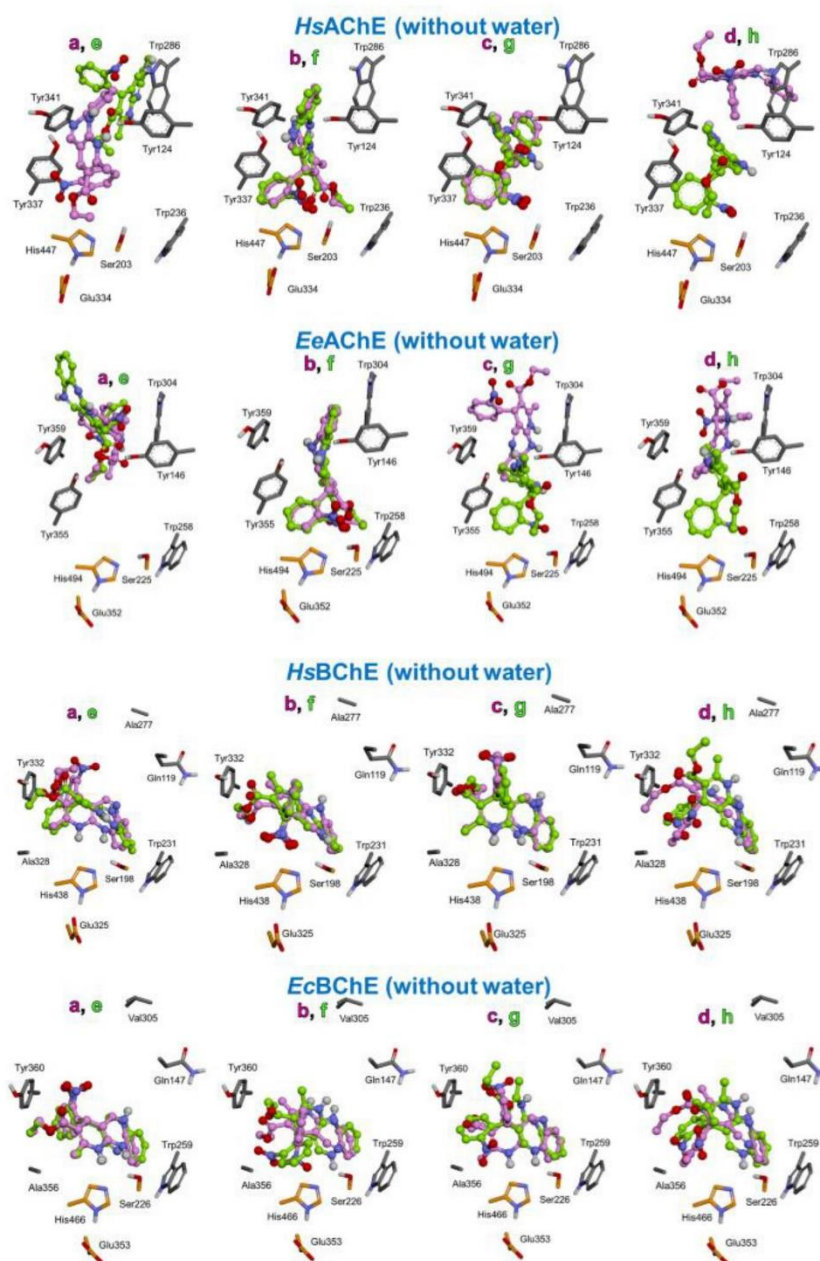


Figure S3. Comparison between the binding modes of each JM-20 isomer (a-h) in the cholinesterase enzymes (*HsAChE*, *EeAChE*, *HsBChE* and *EcBChE*), from molecular docking without water molecules. The JM-20 isomers (a-h) are in a ball and stick model, with the carbon atoms from the isomers a-d and e-h in the pink and green colors, respectively. The cholinesterase's catalytic triad is represented by Ser, His, and Glu residues (with the carbon atoms in the orange color).

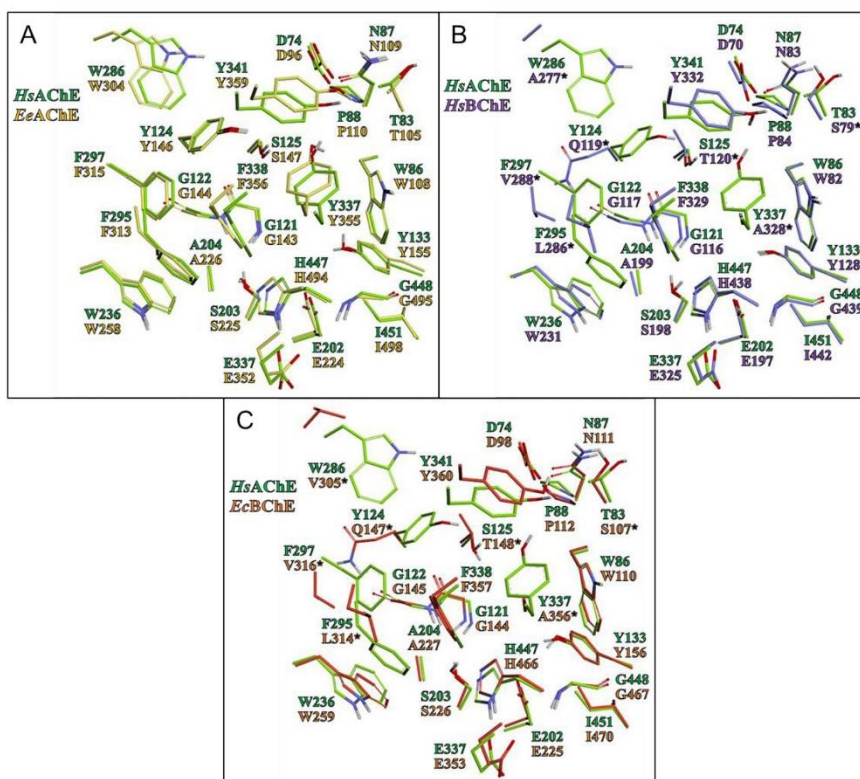


Figure S4. Comparison between the active site of *HsAChE* vs *EeAChE* (A), *HsAChE* vs *HsBChE* (B), and *HsAChE* vs *EcBChE* (C). The structures overlapping was made by the Discovery Studio Visualizer program (<https://www.3ds biovia.com>). * Indicates different amino acid residues.

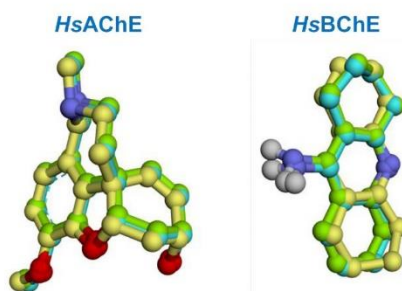


Figure S5. Root-mean-square deviation (RMSD) between X-ray structures (*HsAChE* (PDB 4ey6) and *HsBChE* (PDB 4bds)) and the molecular docking in the absence or presence of water molecules (with the carbons atoms shown in yellow, green and blue, respectively). For *HsAChE*, the galantamine presented the RMSD value of 0.36 Å (*HsAChE* x-ray structure vs docking without water) and 0.23 Å (*HsAChE* x-ray structure vs docking with water). For *HsBChE*, the tacrine presented the RMSD value of 0.48 Å (*HsBChE* x-ray structure vs docking without water) and 0.41 Å (*HsBChE* x-ray structure vs docking with water). A redocking with an RMSD value lower than 2.0 Å is considered a good docking protocol [6,7]. Only the heavy atoms are shown.

Table S1. Comparison between the amino acid residues from the active site of *HsAChE*, *EeAChE*, *HsBChE*, and *EcBChE*. For the visualization of each amino acid position, please see Figure S4.

Subsite	<i>HsAChE</i>	<i>EeAChE</i>	<i>DrAChE</i>	<i>HsBChE</i>	<i>EcBChE</i>
Catalytic triad (CT)	Ser203	Ser225	Ser225	Ser198	Ser226
	Glu334	Glu352	Glu352	Glu325	Glu353
	His447	His494	His495	His438	His466
Oxyanion Hole (OH)	Gly121	Gly143	Gly143	Gly116	Gly144
	Gly122	Gly144	Gly144	Gly117	Gly145
	Ala204	Ala226	Ala226	Ala199	Ala227
Anionic Subsite (AS)	Trp86	Trp108	Trp108	Trp82	Trp110
	Tyr133	Tyr155	Tyr155	Tyr128	Tyr156
	Glu202	Glu224	Glu224	Glu197	Glu225
	Gly448	Gly495	Gly496	Gly439	Gly467
	Ile451	Ile498	Ile499	Ile442	Ile470
Acyl Binding Pocket (ABP)	Trp236	Trp258	Trp258	Trp231	Trp259
	Phe295	Phe313	Phe313	Leu286	Leu314
	Phe297	Phe315	Phe315	Val288	Val316
	Phe338	Phe356	Phe356	Phe329	Phe357
Omega Loop (OL)	Thr83	Thr105	Ile105	Ser79	Ser107
	Asn87	Asn109	Asn109	Asn83	Asn111
	Pro88	Pro110	Pro110	Pro84	Pro112
Peripheral Anionic Subsite (PAS)	Asp74	Asp96	Asp96	Asp70	Asp98
	Tyr124	Tyr146	Tyr146	Gln119	Gln147
	Ser125	Ser147	Ser147	Thr120	Thr148
	Trp286	Trp304	Trp304	Ala277	Val305
	Tyr337	Tyr355	Tyr355	Ala328	Ala356
	Tyr341	Tyr359	Tyr359	Tyr332	Tyr360

Table S2. Comparison between enzyme kinetic parameters (Vmax and Km values) obtained by non-linear regression – inhibition (substrate inhibition) vs Michaelis-Menten, using the Graph Pad Prisma.

		Enzyme kinetic – inhibition (substrate inhibition) (nM)						Michaelis-Menten (nM)					
		Control	75	100	150	250	400	Control	75	100	150	250	400
		<i>EeAChE</i>	Vmax (Δ abs 415 nm/s/ mg protein)	42.12 ± 4.32	19.74 ± 4.13	-	15.64 ± 1.56	9.12 ± 0.71	4.93 ± 0.25	31.06 ± 1.34	18.19 ± 0.97	-	15.64 ± 1.52
	Km	0.15 ± 0.03	0.25 ± 0.09	-	0.53 ± 0.13	0.38 ± 0.09	0.36 ± 0.05	0.09 ± 0.01	0.21 ± 0.03	-	0.53 ± 0.13	0.43 ± 0.09	0.36 ± 0.05
<i>HsAChE</i> (Ghost)	Vmax (Δ abs 415 nm/min/ mg protein)	78.53 ± 13.38	-	60.17 ± 14.79	-	46.80 ± 18.07	30.18 ± 1.81	65.50 ± 2.44	-	50.02 ± 2.04	-	41.94 ± 2.52	29.92 ± 1.78
	Km	0.28 ± 0.08	-	0.32 ± 0.13	-	0.41 ± 0.24	0.34 ± 0.05	0.20 ± 0.02	-	0.22 ± 0.03	-	0.33 ± 0.05	0.33 ± 0.05
<i>HsAChE</i> (Erythrocytes)	Vmax (Δ abs 415 nm/h/ mg protein)	50.51 ± 1.94	50.03 ± 8.14	-	31.71 ± 1.48	31.33 ± 3.58	-	50.51 ± 1.88	46.91 ± 2.13	-	31.71 ± 1.44	31.33 ± 3.48	-
	Km	0.13 ± 0.02	0.21 ± 0.06	-	0.17 ± 0.02	0.43 ± 0.13	-	0.13 ± 0.02	0.18 ± 0.03	-	0.17 ± 0.02	0.43 ± 0.13	-
<i>HsBChE</i>	Vmax (Δ abs 415 nm/h/ mg protein)	156.0 ± 9.77	152.1 ± 20.34	-	124.3 ± 11.06	129.9 ± 5.25	138.6 ± 3.24	141.6 ± 2.27	139.7 ± 4.31	-	124.3 ± 10.73	129.9 ± 5.09	138.6 ± 3.15
	Km	0.27 ± 0.03	0.29 ± 0.06	-	0.17 ± 0.05	0.24 ± 0.03	0.25 ± 0.02	0.23 ± 0.01	0.25 ± 0.02	-	0.17 ± 0.05	0.24 ± 0.03	0.25 ± 0.02

Table S3. Validation of ChE structure models from protein homology modelling.

Enzyme	PROCHECK (most favorable regions)	Verify 3D	ProSa
<i>EeAChE</i>	87.1%	93.1%	-10.2
<i>EcBChE</i>	92.6%	90.1%	-10.8

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2.2 MANUSCRITO – Cytotoxic, antioxidative prospects and virtual screening of the new hybrid molecule (JM-20) on human blood cells

Cytotoxic, antioxidative prospects and virtual screening of the new hybrid molecule (JM-20) on human blood cells

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Abstract

Previous studies have shown that JM-20, a new 1,5-benzodiazepine fused to a dihydropyridine fraction, has different pharmacological properties of clinical interest. However, considering that studies on its possible toxic effects on blood cells have not yet been reported, the present study aimed to investigate for the first-time cytotoxicity through cell viability, and morphological changes on leukocytes. Additionally, we evaluated the production of reactive species and possible changes in the cell cycle (in human leukocytes), as well as its hemolytic effect in human erythrocytes. JM-20 antioxidant effects are still scarce in the literature; therefore, we verified the protective effect on the oxidative stress induced by Fe²⁺ in the leukocyte and the scavenger activity of free radicals. Besides, we predict ADMET parameters using *in silico* virtual screening tools. Human blood was obtained from healthy volunteers. Immediately after the blood collection the leukocytes or erythrocytes were isolated and treated with different concentrations of JM-20. The results showed no cytotoxic effect for red blood cells and no alterations in the cell cycle. A potent protective effect of JM-20 lipid peroxidation was also evidenced with an IC₅₀ value of around 1.0 μM. The *in silico* pharmacokinetic and toxicological properties of JM-20, derivatives, and nifedipine were studied. Our findings demonstrated that the JM-20 and its putative metabolites present similar characteristics to the nifedipine, as similar absorption by gastrointestinal tract (orally), distribution by permeation through BBB and in the CNS, metabolism, and toxicity. The *in vitro* and *in silico* data support the low toxicity of JM-20 in mammals.

key words: toxicity, oxidative stress, *in vitro*, *in silico*, swissADME, pkCSM

1. Introduction

The new molecule derived from a benzodiazepine and a pyridine fusion, JM-20 (3-ethoxycarbonyl-2-methyl-4-(2-nitrophenyl)-4,11-dihydro-1H-pyrido [2,3-b] benzodiazepine) (Figure 1) is arousing interest based on their pharmacological effects. For example, JM-20 has been demonstrated to reduce the aggressive behavior of socially isolated mice, increase the latency period for the beginning of the tonic crisis, and the lethality induced by pentylentetrazol (Nuñez-Figueroa et al., 2014b). Besides, the JM-20 treatment (2-10 mg/kg *p.o.*) reduces anxiolytic-like responses in mice (Nuñez-Figueroa et al., 2013).

JM-20 has also been described to exhibit *in vitro* and *in vivo* multi-target neuroprotective effects by reducing the neuronal excitotoxic injury and protecting the mitochondria from Ca²⁺-induced toxicity (Nuñez-Figueroa et al., 2014a; Ramírez-Sánchez et al., 2015; Wong-guerra et al., 2019). Recently, Silva et al (Silva et al., 2020) demonstrated that JM-20 is a potent *in vitro* acetylcholinesterase inhibitor; suggesting that this compound could be an option in the treatment of neurodegenerative pathologies.

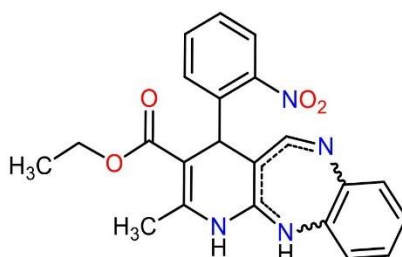


Figure 1. Structural formula of the compound 3-ethoxycarbonyl-2-methyl-4-(2-nitrophenyl)-4,11-dihydro-1H-pyrido [2,3-b] [1,5] benzodiazepine (JM-20).

Oxidative stress has been implicated in several pathological conditions, including cancer (Khansari et al., 2009), neurological disorders (Pohanka, 2014; Uttara et al., 2009), atherosclerosis (Li et al., 2014; Victor et al., 2009), hypertension (Briones and Touyz, 2010; Schulz et al., 2011), ischemia (Love, 1999; Misra et al., 2009), diabetes (Maritim et al., 2003), acute respiratory distress syndrome (Bartens et al., 1999), idiopathic pulmonary fibrosis (Kliment and Oury, 2010; Rahman et al., 1999), chronic obstructive pulmonary disease (Repine et al., 1997), and asthma (Nadeem et al., 2003).

In this context, reactive oxygen species (ROS) are physiologically produced by living organisms as a result of cellular metabolism, but at high concentrations, they cause damage to cell components, such as lipids, proteins, and DNA (Birben et al., 2012). Although aerobic organisms have natural antioxidant systems, which include enzymatic and nonenzymatic antioxidants that are, in normal conditions, effective in blocking the ROS harmful effects, the imbalance favoring oxidant species is termed, oxidative stress (Birben et al., 2012; Rahal et al., 2014).

In addition to the beneficial effects, synthetic compounds must be present minimal toxic effects for the cells and the organism in general. Thus, there is a need for reliable *in vitro* cytotoxicity tests to properly evaluated the toxicity of new compounds before *in vivo* pre-clinical or clinical studies (Hughes et al., 2011). Likewise, an accurate *in vitro* toxicity test can reduce the number of animal studies (Bhanushali et al., 2015). At the same time, *in vitro* cytotoxicity assessments can rapidly allow the screening of a large number of potential drug candidates (Hughes et al., 2011).

Another alternative developed to study de pharmacokinetics and toxicity of new compounds comprises computational approaches with the main function to guide compounds design and the modifications during the drug development process (Hu et al., 2020; Pires et al., 2018, 2015). The *in silico* programs used to discover drugs have a focus on the elucidation of the interaction between molecules and target proteins of interest (Daina et al., 2017; Hu et al., 2020; Pires et al., 2015). In addition, the *in silico* analysis of pharmacokinetics and toxicity can help in predicting the safety of new molecules (Daina et al., 2017; Hu et al., 2020; Pires et al., 2018, 2015).

Considering the pharmacological potential of JM-20 and the fact that no studies on its potential toxic effect on human blood cells have been reported so far, the present study was undertaken to investigate for the first time the cytotoxicity (in human leukocytes), as well as its potential hemolytic effect in human erythrocytes. Furthermore, we also investigated the antioxidant effect of the compound against Fe²⁺-induced lipid peroxidation in isolated leukocytes. Moreover, we also evaluated and predict a range of ADMET (absorption, distribution, metabolism, excretion, and toxicity) properties for the JM-20 and its derivative compounds.

2. Material and methods

2.1 Compound and reagents

JM-20 was synthesized, purified, and characterized as previously reported (Nuñez-figueroa et al., 2013). The compound was dissolved in dimethyl sulfoxide (DMSO). The chemicals reagents used in this work were purchased from Sigma-Aldrich (St. Louis, MO, USA).

2.2 Blood samples

Human blood was obtained from healthy male and female voluntary donors aged 20 to 60 years from the toxicological biochemistry laboratory at the Universidade Federal de Santa Maria (UFSM) for TBARS, cell viability, and hemolysis studies. The Ethical Committee of the UFSM approved the protocol (3.471.823) used in these studies. For the flow cytometry analysis (cell cycle arrest and reactive species production), human blood was obtained from healthy male and female voluntary donors aged 20 to 60 years from Instituto de Pesquisa Pelé Pequeno Príncipe (IPPP). The experiments carried out at IPPPP were approved by the Ethical Committee of the Faculdades Pequeno Príncipe (3.977.300). In all experiments, we used blood from different donors (n=3).

2.2.1 Isolated human leukocytes

Immediately after the blood collection, 2 mL of 5% dextran was added and the tubes with blood, and dextran were maintained in rest for 1h. After that, the supernatant was removed and centrifuged at 200 g for 10 min. The pellet was washed with 1 mL of lyse solution (150 mM NH₄Cl, 10 mM NaHCO₃, and 1mM EDTA), after this procedure the leukocytes were solubilized in Hank's buffer (5.4 mM KCl, 0.3 mM Na₂HPO₄, 0.4 mM KH₂PO₄, 4.2 mM NaHCO₃, 0.5 mM MgCl₂, 122.6 mM NaCl, 0.01 mM D-glucose, 10 mM Tris-HCl, and 1.3 mM CaCl₂, pH = 7.4). The samples were used for cell viability, TBARS, cell cycle arrest and ROS production.

2.3 Cytotoxicity assays

2.3.1 Hemolysis determination

Heparinized human blood was immediately centrifuged 200 g for 10 min and hemolysis evaluation was realized as reported elsewhere (Duarte et al., 2016; Menezes et al., 2014). The plasma was discarded, and the cell pellets were washed three times with phosphate-buffered saline (6.1 mM, 150 mM NaCl, pH 7.4). The hemolysis was estimated by measuring the resistance of red blood cells to the rupturing (lysis) and the release of their contents (cytoplasm) in a salt solution. Three hundred microliters of erythrocytes, 0.2 μ L of JM-20 solution (5–20 μ M), and 699.8 μ L of phosphate buffer saline (PBS) (6.1 mM, 150 mM NaCl, pH 7.4) were incubated for 3 h at 28 °C in a shaker. After incubation, samples were mixed and centrifuged at 2500 rpm for 10 min and the supernatant was transferred to a microplate and the erythrocytes lysis were measured using a spectrophotometer Spectra Max plate reader at 540 nm. The results were expressed as a percentage of the positive control (triton 0.05%). We discount the possible hemolysis caused by vehicle (DMSO) and used PBS like negative control.

2.3.2 Cell viability

The cell viability was determined using the Trypan blue exclusion method, which assumes that nonviable cells are stained in blue (Mishell et al., 1980). In summary, 1 μ L of JM-20 (10–50 μ M final concentration) was added to 249 μ L of leukocytes suspension (2000 cells/ μ L) and incubated at 37 °C for 3 h. Dimethyl sulfoxide (DMSO – 0.4%) was used as the control, tert-butyl hydroperoxide (1 mM) was used as the positive control, and water was used as blank. Thereafter, 50 μ L of this system were mixed with 50 μ L of 0.4% trypan blue and incubated for 5 min at room temperature. From the mixture, an aliquot of 10 μ L was placed in a Neubauer chamber and the viable cells were counted microscopically. The viability was expressed as viable cells in percent of the total cells.

2.3.3 Cell Cycle arrest

The DNA content of the cells was measured by flow cytometry using 7-amino actinomycin D (7-AAD) solution (BD via probe™). Leukocytes (2×10^5 cells/100 μ L) were incubated with JM-20 (10-50 μ M) for 3h at 37 °C. After the incubation, cells were carefully washed with PBS and fixed in cold ethanol (70%) for 30 minutes at -20 °C. Again, cells were washed with PBS and human albumin (2%) and incubated in 7-Aminoactinomycin D (7AA-D) (20 μ g/mL), RNase (350 μ g/mL), and Triton-X (0.1%) solution for 15 min under an ice bath. Cells were evaluated (10,000 events/sample in the granulocytes gate) in the PERCP channel ($E_{x_{max}}$ 482 nm/ $E_{m_{max}}$ 678 nm; blue laser) by FACSCanto II (BD Biosciences) cytometer and data were processed by Infinicyt Software V1.6.0.

2.3.4 RS formation

The RS generation was analyzed by flow cytometry using 2,7-dichlorodihydrofluorescein diacetate (DCFH-DA). Leukocytes (2×10^5 cells/100 μ L) were incubated with JM-20 (10-50 μ M) for 3h at 37 °C. After the incubation, cells were washed two times with PBS and incubated with 35 μ M DCFH-DA for 15 min at 37 °C in the dark. Cells were evaluated (10,000 events/sample in the granulocytes gate) in the FITC channel ($E_{x_{max}}$ 494 nm/ $E_{m_{max}}$ 520 nm; blue laser) by FACSCanto II (BD Biosciences) cytometer and data were processed by Infinicyt Software V1.6.0.

2.4 Antioxidant activity

2.4.1 Thiobarbituric Acid Reactive Substances (TBARS) Production

The TBARS production was performed as described by Ohkawa *et al.* (Ohkawa *et al.*, 1979). Isolated leukocytes were incubated with 100 μ M FeSO₄ in the presence or absence of JM-20 (0.5 – 5.0 μ M final concentration) at 37 °C for 1 h. Subsequently, 40 μ L of sodium dodecyl sulfate (8.1%), 100 μ L of acetic acid buffer (pH 3.4), and 100 μ L of 0.6% thiobarbituric acid (TBA) were added and incubated at 100 °C for 1 h. After cooling, the absorbance of the samples was measured at 532 nm using a

spectrophotometer Spectra Max plate reader. The quantification of protein in the isolated leukocyte samples was performed using a Nanodrop-1000 spectrophotometer. The average amount of protein was 4.82 mg / mL.

2.4.2 DPPH radical scavenging assay

The method proposed by Pereira et al (2014) (Pereira et al., 2014) was used for this test. For the time-dependent curve, the JM-20 compound was dissolved in DMSO and mixed with 0.3 mM DPPH in ethanol. To perform the concentration curve 0, 10, 20, 50, and 100 μ M of JM-20 were used; and mixed with 0.3 mM DPPH in ethanol. The reaction was divided into two parts. At the first moment, we monitored the absorbance in spectrophotometer SpectraMax at 518 nm for 30 minutes with an interval of 30 seconds, we called it a fast phase reaction. In the second moment, the slow phase reaction, the absorbance was recorded every 30 minutes for 180 minutes. α -tocopherol (100 μ M) was used as a positive control.

2.5 ADMET predictions

The platforms pkCSM and swissADME analyze pharmacokinetic and toxicological properties that rely on distance-based graph signatures in a free online system. This tool can be a powerful tracking and guidance mechanism showing the balance between power and security for researchers. The online platforms SwissADME (absorption, distribution, metabolism, and excretion) and pkCSM were used for ADMET (absorption, distribution, metabolism, excretion, and toxicity) profile determination of JM-20 and its putative metabolites. We compared the results of the two platforms with nifedipine (which is a therapeutically used antihypertensive drug and a Ca^{2+} channel blocker) and structurally like the JM-20 moiety. We also use the nifedipine metabolites described in the literature (Lee et al., 2014) to predict the possible metabolites of JM-20.

2.6 Statistical Analysis

Statistical analyses were performed using the GraphPad Prism 6 (version 6.01, GraphPad Software, Inc., USA). Results were expressed as the mean \pm standard error of the mean (SEM). Data were analyzed by one-way ANOVA followed by Newman Keuls post-hoc test. Results from TBARS were also calculated as the half-maximal inhibitory concentration (IC₅₀) according to nonlinear regression log (inhibitor) vs. response – variable slope in accordance with equation 1, using GraphPad, Prism 6.01.

$$\text{Equation 1: } Y = \text{Bottom} + (\text{Top} - \text{Bottom}) / (1 + 10^{-(\text{LogIC}_{50} - X) * \text{HillSlope}});$$

HillSlope describes the steepness of the family of curves (A HillSlope of -1.0 is standard).

Top and Bottom are plateaus in the units of the Y-axis.

Values of $p \leq 0.05$ were considered significantly different.

3. Results

Here we demonstrated the antioxidant activity of the JM-20 compound, as well as elucidating the low toxicity of this compound in human blood cells *in vitro*. Our results showed that JM-20 (even at high concentrations, i.e., 10-20 μM) did not cause hemolysis [F (3, 8) = 3.053; $p = 0.0919$] (Table 1). We consider positive control (triton 0.05%) to be 100% hemolysis.

Table 1. Determination of hemolysis after exposure to different concentration of JM-20 (1 – 20 μM).
Hemolysis (% of positive control)

PBS (negative control)	11.02 \pm 0.22
JM-20 (1 μM)	5.66 \pm 1.76
JM-20 (10 μM)	6.76 \pm 1.54
JM-20 (20 μM)	8.38 \pm 1.27

Data were expressed as mean \pm SEM (n = 3) and analyzed by One-way ANOVA followed by Newman Keuls post hoc test. No significant differences were observed ($p > 0.05$).

In the cell viability assay, we used leukocytes isolated from fresh whole blood from voluntary donors. The tert-butyl hydroperoxide (1 mM) was used as a positive control and damaged about 40% of the cells (i.e., 60% of leukocytes were viable) when compared to the vehicle (DMSO) group (Figure 2). At the lowest concentration tested (10 μ M), JM-20 did not change cell viability; but at the higher concentrations tested, 20 and 50 μ M, JM-20 caused a decrease of about 20% and 50% in leukocyte viability, respectively, when compared to the vehicle (DMSO) group (Figure 2). This result is also in accordance with the observed morphology of the leukocytes treated with JM-20 (Figure 3), in which the concentrations of 20 and 50 μ M seem to modify cell populations.

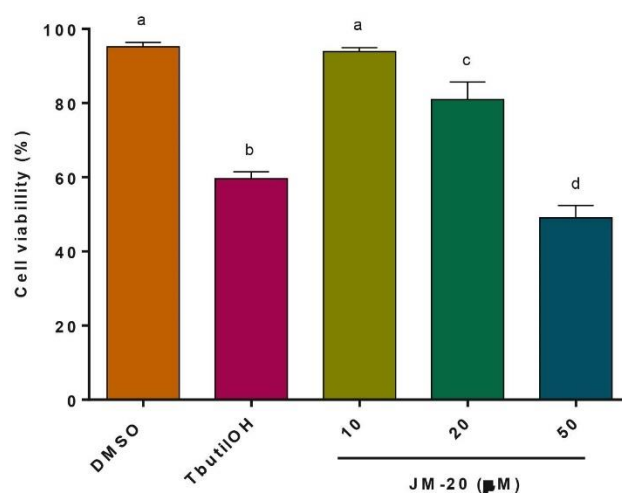


Figure 2. Cell viability after incubating different concentrations of JM-20 (10 – 50 μ M) with leukocytes. DMSO (0.4%) was used as the control, and tert-butyl hydroperoxide (TbutilOH-1mM) was used as the positive control. Data were expressed as mean \pm SEM (n = 3) and analyzed by One-way ANOVA followed by Newman Keuls post hoc test. Different letters indicated the differences between the compounds and/or JM-20 concentrations ($p \leq 0.05$).

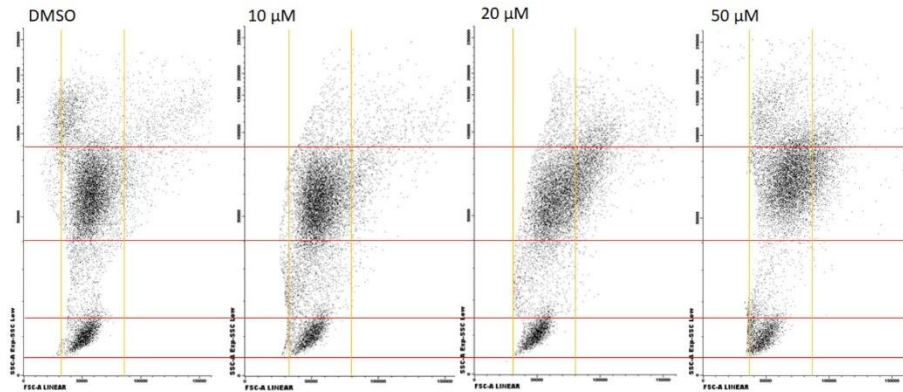


Figure 3. Morphological visualization of leukocytes exposed to JM-20 (10-50 μM). The leukocytes populations used to measure RS production via DCFH oxidation in this study.

The exposure to all JM-20 concentrations tested (10-50 μM) caused a significant increase in the intracellular reactive species (RS) levels (Figure 4). Although a decrease in the cell viability and an increase in the RS production was observed, the exposure to JM-20 (10-50 μM) did not alter the cell cycle (Figure 5).

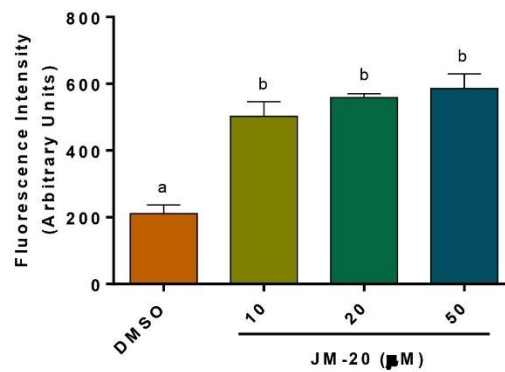


Figure 4: RS generation after incubating different concentrations of JM-20 (10 – 50 μM) with leukocytes. DMSO (0.4%) was used as the control. Data were expressed as mean \pm SEM (n = 4) and analyzed by One-way ANOVA followed by Newman Keuls post hoc test. Different letters indicated the differences between the DMSO and/or JM-20 concentrations ($p \leq 0.05$).

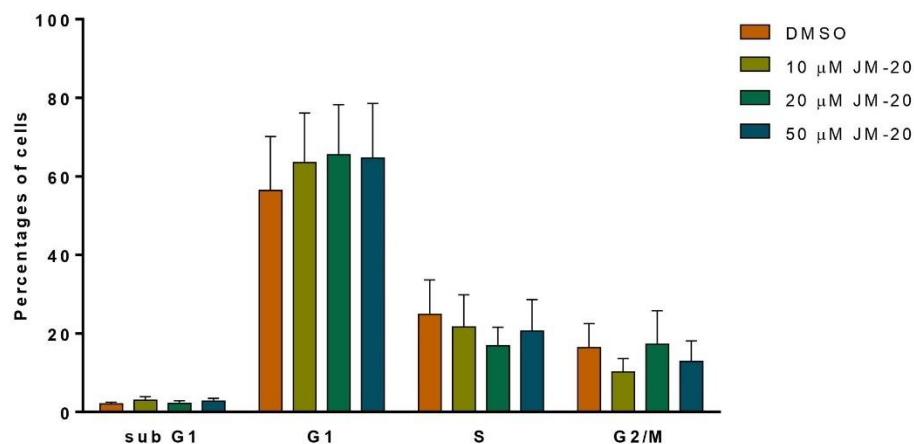


Figure 5: Cell cycle distribution after incubating different concentrations of JM-20 (10 – 50 μ M) with leukocytes. DMSO (0.4%) was used as the control. Data were expressed as mean \pm SEM (n = 6) and analyzed by One-way ANOVA followed by Newman Keuls post hoc test.

We also evaluate the inhibition of lipid peroxidation by JM-20. JM-20 inhibited the lipid peroxidation of isolated leukocytes induced by Fe^{2+} in almost all concentrations tested and was effective such as the positive control (α -tocopherol) (Figure 6). The calculated IC_{50} value for the inhibition of lipid peroxidation was $1.051 \pm 0.21 \mu\text{M}$ (Table 2). This result indicates that this compound is effective in preventing lipid peroxidation, even at low concentrations. Based on the TBARS results, we decided to test the compound for Fe-chelating activity, but it did not show significant effects (data not shown), suggesting another mechanism of action for the antioxidant effect observed in lipid peroxidation induced by Fe^{2+} probably related to the protection of this compound the cell membrane.

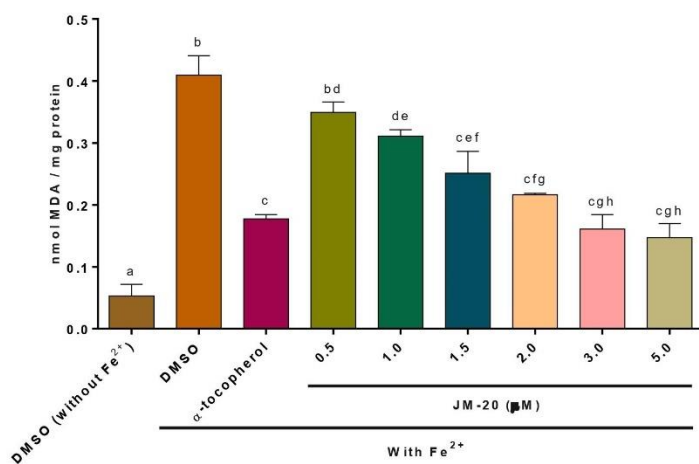


Figure 6. Effect of different concentrations of JM-20 (0.5 – 5 μM) on lipid peroxidation induced by iron. α -tocopherol (10 μM) was used as positive control. Data were expressed as mean \pm SEM ($n = 3$) and analyzed by One-way ANOVA followed by Newman Keuls post hoc test. Different letters indicated the differences between the compounds ($p \leq 0.05$).

Table 2. IC_{50} values for inhibition of lipid peroxidation induced by Fe^{2+} .

Compound	TBARS IC_{50} (μM)
α -tocopherol	1.065 ± 0.34
JM-20	1.051 ± 0.21

Regarding DPPH radical scavenging assay, JM-20 scavenged the DPPH radical at nearly all the tested concentrations (Figure 7). The calculated IC_{50} value for DPPH quenching color was $16.25 \pm 0.23 \mu\text{M}$ in the slow phase (12.600 seconds).

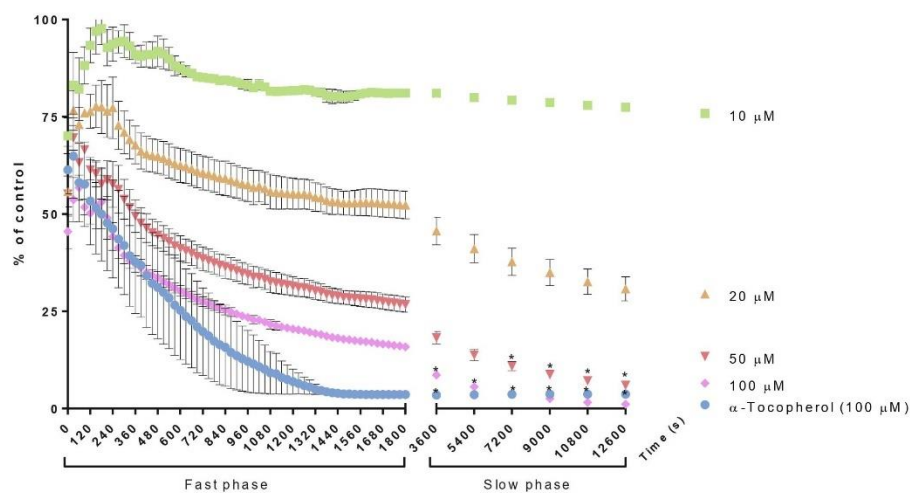


Figure 7. Time-dependent curve of JM-20 in the DPPH test. α -tocopherol was used as a positive control. Results were expressed as mean \pm S.E.M. ($n = 3$). Data were analyzed by repeated measurements Two-way ANOVA followed by Bonferroni post hoc test and differences are considered significant at $p < 0.05$. *Indicates equal values for α -tocopherol and JM-20 (50 and 100 μM) in the respective times ($p \leq 0.05$).

Here, for the first time, we demonstrate the *in silico* ADMET screening of JM-20 (Figure 8A) and some of its putative metabolites (Figure 9) using two different platforms (swissADME and pkCSM).

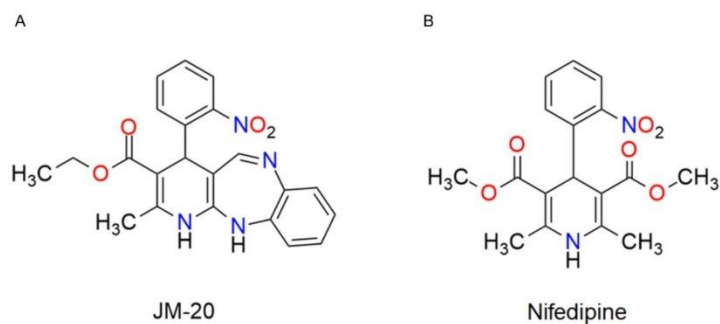


Figure 8. JM-20 and Nifedipine structure.

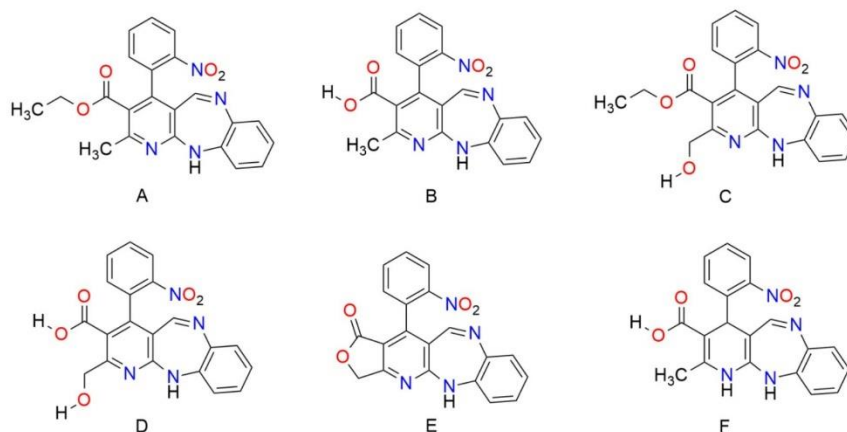


Figure 9: JM-20 putative metabolites. The metabolites were theoretically built considering the nifedipine.

We select the main parameters that represent the ADMET profile. On absorption, we consider water-solubility, Caco2 permeability, intestinal absorption (human), skin permeability, and p-glycoprotein substrate. Regarding distribution parameters, we observed two permeation processes: brain-blood barrier (BBB) and CNS. On metabolism, we highlight the effects on cytochrome P450 isoforms, which can be a substrate or responsible for the inhibition of isoforms. The excretion was predicted by total clearance and Renal OCT2 substrate. And finally, the toxicity, we regard the maximum tolerated dose (human) and total acute toxicity (LD50). The comparison between JM-20 and nifedipine in terms of ADMET is displayed in Table 3. We can also point out that JM-20 obtained values referring to good absorption by administration by the oral route, observed through CaCo-2 permeability and human intestinal absorption that refers to low absorption for drugs with a percentage below 30, here JM-20 and its metabolites showed intestinal absorption greater than 65%. The BBB permeability and in the CNS our compound also seems to be effective, $\log_{BB} > 0.3$ are capable of crossing the BBB, and $\log_{PS} > -2$ are penetrating substances in the CNS, the JM-20 had 0.23 \log_{BB} and -1.92 \log_{PS} . With regard to metabolism, exogenous substances are expected to have a certain degree of inhibition of enzymes responsible for the detoxification of the organism. Here we show that JM-20 was not an inhibitor of the CYP2D6 isoform, in the same way as nifedipine. We can also point out that our results are promising when compared to nifedipine. Although, we can see some

differences between swissADME and pkCSM we can observe that the JM-20 is very similar to Nifedipine.

Table 3. ADMET prediction of compound JM-20 and Nifedipine by SWISSADME and pkCSM.

	JM-20		Nifedipine	
	pkCSM	SwissADME	pkCSM	SwissADME
Water solubility	-4.23 log mol/L	3.26e-05 mol/L	-4.23 log mol/L	7.00e-04 mol/L
Caco2 permeability	1.24 log Papp in 10 ⁻⁶ cm/s	High	1.19 log Papp in 10 ⁻⁶ cm/s	High
Intestinal absorption (human)	93.53 % Absorbed	-	87.12 % Absorbed	-
Skin permeability	-2.74 log Kp	-6.32 cm/s	-2.82 log Kp	-6.85 cm/s
P-glycoprotein substrate	Yes	No	Yes	No
BBB permeability	0.23 log BB	No	-0.59 log BB	No
CNS permeability	-1.92 log PS	-	-2.53 log PS	-
CYP2D6 substrate	No	-	No	-
CYP3A4 substrate	Yes	-	Yes	-
CYP1A2 inhibitor	No	No	Yes	Yes
CYP2C19 inhibitor	Yes	No	No	Yes
CYP2C9 inhibitor	Yes	Yes	No	Yes
CYP2D6 inhibitor	No	No	No	No
CYP3A4 inhibitor	Yes	Yes	Yes	No
Total clearance	0.34 log mL/min/Kg	-	0.81 log mL/min/Kg	-
Renal OCT2 substrate	No	-	No	-
Max tolerated dose (human)	-0.08 log mg/Kg/day	-	-0.44 log mg/Kg/day	-
Oral acute toxicity (LD ₅₀)	2.53 mol/Kg	-	2.8 mol/Kg	-

4. Discussion

In vitro methodologies have advantages when compared to *in vivo* tests, for instance, reduction of the number of experimental variables, such as - weight, sex, and age of the experimental animals, as well as the stress of manipulation. Also, it is possible to obtain significant data in a short time (Dias et al., 2017; Emami, 2006; Hara et al., 1998). Here we used some techniques to evaluate the JM-20 in relation to antioxidant effects and toxicity. According to other studies that also focused on evaluating the toxicity of substances unknown, we analyzed hemolysis and cell viability in red blood cells and leukocytes, respectively (Bueno et al., 2018, 2013; Duarte et al., 2016; Prestes et al., 2019). The leukocytes isolated from fresh human blood as

previously described (Bueno et al., 2013; Duarte et al., 2016; Waczuk et al., 2015) were used to test the cytotoxicity and antioxidant effects of the JM-20 compound.

Toxic agents can trigger the hemolysis process, consequently causing hemoglobin release into the plasma (Anadón et al., 2014; Bowen et al., 2010; Çimen, 2008; Harvey, 1997). The hemolytic effect has been used as one of the screening methodologies for substances of clinical interest (Anadón et al., 2014; Bowen et al., 2010). Here, JM-20 exposure did not alter the hemoglobin homeostasis. The trypan blue exclusion methodology is widely used to assess cytotoxicity by measuring cell viability and cell membrane integrity (Adan et al., 2016; Weyermann et al., 2005). The test is based on the trypan blue absorption into the cytoplasm of damaged cells due to the plasma membrane loss of selectivity, while the living cells remain intact (Adan et al., 2016; Duarte et al., 2016; Menezes et al., 2014; Weyermann et al., 2005). Our results clearly showed that the JM-20 did not present toxic effects for erythrocytes and leukocytes except for the highest concentrations tested (20 and 50 μM). In addition, we evaluated the toxicity of the compound at considerably higher concentrations (micromolar) than the therapeutic concentrations, such as for the inhibition of acetylcholinesterase, which had an effect was on the range of nanomolar (Silva et al., 2020).

Moreover, other analyzes are important to determine the profile of a compound, such as the increases in the intracellular RS levels that are responsible for lipid, protein, and DNA damage leading to cellular death (Birben et al., 2012). Similarly, Fisher et al. 2004 (Fischer et al., 2004) demonstrated that an increase in the RS leukocyte generation occurred concomitantly with a decrease in cell viability. The main hypothesis is that the excessive production of RS causes the reduction of cell viability in leukocytes directly or indirectly (Bueno et al., 2018). Unexpectedly, in our study, the concentration of 10 μM JM-20 caused a significant increase in the RS species but did not alter the cell viability. These results call for attention and need to be explored.

Nevertheless, some synthetic antioxidant compounds showed a pro-oxidant effect at high concentrations (Nogueira and Rocha, 2010). In this context, we decided to test the antioxidant effect of the compound JM-20. Oxidative stress can lead to cell damage through lipid peroxidation of cell membranes (Kudryavtseva et al., 2015). The cell membrane consists mainly of phospholipids, which undergo peroxidation forms aldehydes, such as malondialdehyde (Kanno et al., 2009). The lipid peroxidation products react with thiobarbituric acid, forming a pink complex that can be quantified in

the TBARS assay (Ohkawa et al., 1979). In this case, JM-20 at concentrations of 3 and 5 μM was as effective as tocopherol (10 μM).

Furthermore, the study of ADMET of a given compound is important to understand or predict its potential toxicity and the future use of the clinical practice (Daina et al., 2017; Hu et al., 2020; Pires et al., 2018, 2015). Through the results of the analysis regarding absorption such as CaCo-2 permeability and intestinal absorption, it is possible to predict the absorption of a drug administered orally, for example (Pires et al., 2015). Besides, cytochrome P450 is the main detoxification enzyme present in the body and is found in the liver (Gregg, 2004; Shankar and Mehendale, 2014). Here we evaluated the inhibition for each isoform of this enzyme and point out that the CYP2D6 and CYP3A2 isoforms are the main ones involved in drug metabolism (Shankar and Mehendale, 2014). Moreover, the isoform CYP2D6 was not inhibited by our compound. In the analysis of toxicity *in silico*, we highlight the maximum tolerated dose, which predicts the limit of a toxic dose of chemical substances in humans, and the LD₅₀, which is a standard measure of acute toxicity. In this case, JM-20 was similar to nifedipine.

We also compared our results with nifedipine (Figure 8B) and its metabolites (Figure 10, Table S1, Table S2) (Lee et al., 2014). The choice of nifedipine to compare with JM-20 was based on the similarity between the nifedipine and nifedipine-like portion of the JM-20 structure. Likewise, we use published data from nifedipine and its known metabolites to propose the metabolites of our compound (Lee et al., 2014). As expected, the JM-20 showed an ADMET profile like nifedipine. Thus, emphasizing the potential therapeutic use of JM-20.

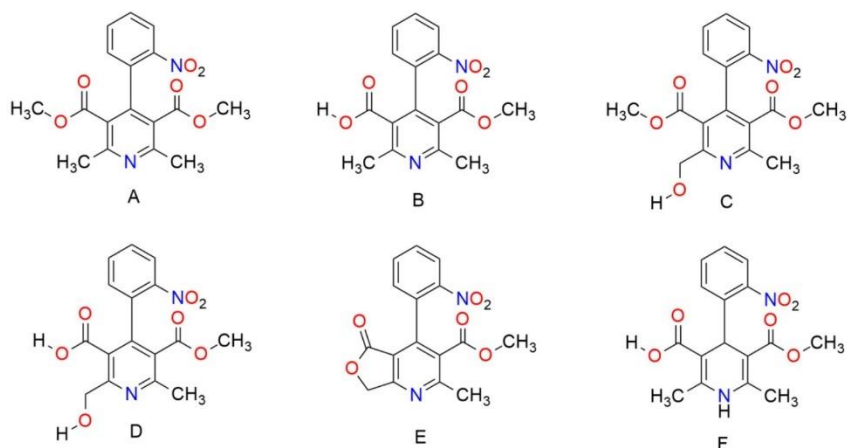


Figure 10: Nifedipine metabolites.

Conclusion

In conclusion, the results presented in this study demonstrated the *in vitro* antioxidant activity of JM-20 in human leukocytes, and the scavenger effect of free radicals. JM-20 was not cytotoxic to human erythrocytes at high concentrations nor altered the cell cycle in leukocytes. We also propose the metabolites of the compound based on previous studies in similar structures present in the literature and elucidate the pharmacokinetic and toxicological effects of the compound and its metabolites. Thus, our findings here with human cells *in vitro* and *in silico* give further support to new *in vitro* and *in vivo* studies from this using other types of cells and complex organisms such as rodent models.

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88882.182125/2018-01; 88882.182123/2018-01), and INCT-EN: for Cerebral Diseases, Excitotoxicity, and Neuroprotection.

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2.2.1 Material de apoio do Manuscrito - Cytotoxic, antioxidative prospects and virtual screening of the new hybrid molecule (JM-20) on human blood cells

Cytotoxic and antioxidative prospects of the new hybrid molecule (JM-20) on human blood cells

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The metabolites of JM-20 (Table S1) show high permeability for almost all, the exception to metabolite C and D, that platform swissADME consider low absorption. In addition, just metabolite D presents the value of intestinal absorption below 70% absorption. Table S2 shows the metabolites from nifedipine and like our compound, present high Caco2 permeability, except for metabolite D that present low absorption. The intestinal absorption we can see that the metabolite D e F presents a value below 70% absorption. The distribution profile observed through a blood-brain barrier membrane (BBB) permeability and CNS permeability was verified and the results showed that all structures were poor BBB permeability. The main enzymes responsible for the metabolism of drugs on the liver is the cytochrome P450 monooxygenase (CYP) enzymes superfamily. Here we evaluated different isoforms present in humans. According to the results for metabolites (Table S1, and Table S2), all molecules were seen to be negative to CYP2D6. However, our results show again disagreements between platforms pkCSM and swissADME for CYP1A2, CYP2C19, CYP2C9, and CYP3A4. In the same way, it is possible to observe that all molecules are substrates for CYP3A4, except metabolite B of JM-20. The total clearance determines the rates to achieve steady-state concentrations, which include hepatic clearance (metabolism in the liver and biliary clearance) and renal clearance (excretion via the kidneys). The values of JM-20 and your metabolites were lower than nifedipine and your metabolites (Table S1, and Table S2) for total clearance, but no molecule was substrate to renal OCT2. Toxicity has been a major concern to the safety of drug candidates and early identification of toxicity is of great relevance. Here we highlight that the values obtained for max tolerated dose (human) and oral acute toxicity (LD50) were similar for all molecules. In fact, it is possible to observe the similarity in the ADMET profile for the JM-20, Nifedipine, and its metabolites, although there are some disagreements between the platforms pkCSM and swissADME (Table S1, Table S2, and Table S3).

Table S1. ADMET prediction of the putative metabolites of JM-20 by SWISSADME and pkCSM.

	JM-20 metabolites											
	A		B		C		D		E		F	
	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME
Water solubility	-4.64 log mol/L	6.19e-06 mol/L	-2.99 log mol/L	1.72e-05 mol/L	-4.07 log mol/L	3.63e-05 mol/L	-3.30 log mol/L	1.84e-04 mol/L	-4.35 log mol/L	3.01e-05 mol/L	-3.87 log mol/L	9.44e-05 mol/L
Caco2 permeability	0.09 log Papp in 10 ⁵ cm/s	High	-0.29 log Papp in 10 ⁵ cm/s	High	0.09 log Papp in 10 ⁵ cm/s	Low	0.051 log Papp in 10 ⁵ cm/s	Low	0.16 log Papp in 10 ⁵ cm/s	High	-0.28 log Papp in 10 ⁵ cm/s	High
Intestinal absorption (human)	100 %	-	73.69 %	-	93.55 %	-	66.97 %	-	100 %	-	71.89 %	-
Skin permeability	Absorbed	-	Absorbed	-	Absorbed	-	Absorbed	-	Absorbed	-	Absorbed	-
P-glycoprotein substrate	-2.73 log Kp	-5.64 cm/s	-2.73 log Kp	-5.96 cm/s	-2.73 log Kp	-6.63 cm/s	-2.73 log Kp	-7.24 cm/s	-2.73 log Kp	-6.29 cm/s	-2.73 log Kp	-6.64 cm/s
BBB permeability	Yes	No	Yes	No	Yes	No	Yes	No	Yes	No	Yes	No
BBB permeability	-0.30 log BB	No	-0.08 log BB	No	-0.08 log BB	No	-1.38 log BB	No	-0.25 log BB	No	-0.71 log BB	No
CNS permeability	-1.94 log PS	-	-1.86 log PS	-	-2.30 log PS	-	-2.24 log PS	-	-1.92 log PS	-	-2.16 log PS	-
CYP2D6 substrate	No	-	No	-	No	-	No	-	No	-	No	-
CYP3A4 substrate	Yes	-	No	-	Yes	-	Yes	-	Yes	-	Yes	-
CYP1A2 inhibitor	Yes	No	Yes	No	Yes	No	No	No	Yes	Yes	No	No
CYP2C19 inhibitor	Yes	Yes	No	No	Yes	Yes	No	No	Yes	Yes	No	No
CYP2C9 inhibitor	Yes	Yes	Yes	Yes	Yes	Yes	No	Yes	Yes	Yes	No	Yes
CYP2D6 inhibitor	No	No	No	No	No	No	No	No	No	No	No	No
CYP3A4 inhibitor	No	Yes	No	No	No	Yes	No	No	Yes	Yes	No	No
Total clearance	0.17 log mL/min/Kg	-	0.07 log mL/min/Kg	-	0.112 log mL/min/Kg	-	0.067 log mL/min/Kg	-	0.031 log mL/min/Kg	-	0.27 log mL/min/Kg	-
Renal OCT2 substrate	No	-	No	-	No	-	No	-	No	-	No	-
Max tolerated dose (human)	0.28 log mg/Kg/day	-	0.65 log mg/Kg/day	-	0.44 log mg/Kg/day	-	0.76 log mg/Kg/day	-	0.11 log mg/Kg/day	-	-0.10 log mg/Kg/day	-
Oral acute toxicity (LD ₅₀)	2.85 mol/Kg	-	2.43 mol/Kg	-	2.68 mol/Kg	-	2.49 mol/Kg	-	2.79 mol/Kg	-	2.93 mol/Kg	-

Tables S2. ADMET prediction of metabolites of Nifedipine by SWISSADME and pkCSM.

	Nifedipine metabolites											
	A		B		C		D		E		F	
	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME
Water solubility	-4.78 log mol/L	1.98e-04 mol/L	-4.21 log mol/L	3.19e-04 mol/L	-4.32 log mol/L	1.59e-04 mol/L	-4.33 log mol/L	2.57e-04 mol/L	-4.12 log mol/L	5.57e-04 mol/L	-3.55 log mol/L	1.17e-03 mol/L
Caco2 permeability	1.19 log Papp in 10 ⁻⁶ cm/s	High	1.20 log Papp in 10 ⁻⁶ cm/s	High	1.21 log Papp in 10 ⁻⁶ cm/s	High	0.021 log Papp in 10 ⁻⁶ cm/s	Low	1.44 log Papp in 10 ⁻⁶ cm/s	High	0.99 log Papp in 10 ⁻⁶ cm/s	High
Intestinal absorption (human)	90.11 %	-	74.49 %	-	78.19 %	-	64.98 %	-	96.34 %	-	61.59 %	-
Skin permeability	Absorbed	-	Absorbed	-	Absorbed	-	Absorbed	-	Absorbed	-	Absorbed	-
P-glycoprotein substrate	-2.72 log Kp	-6.41 cm/s	-2.74 log Kp	-6.55 cm/s	-2.74 log Kp	-6.32 cm/s	-2.74 log Kp	-6.47 cm/s	-2.74 log Kp	-6.87 cm/s	-2.73 log Kp	-7.00 cm/s
BBB permeability	No	No	No	No	No	No	No	No	No	No	Yes	No
BBB permeability	-0.99 log BB	No	-1.03 log BB	No	-1.04 log BB	No	-0.87 log BB	No	-0.99 log BB	No	-0.53 log BB	No
CNS permeability	-2.53 log PS	-	-2.51 log PS	-	-3.45 log PS	-	-3.32 log PS	-	-2.56 log PS	-	-2.59 log PS	-
CYP2D6 substrate	No	-	No	-	No	-	No	-	No	-	No	-
CYP3A4 substrate	Yes	-	Yes	-	Yes	-	Yes	-	Yes	-	Yes	-
CYP1A2 inhibitor	No	Yes	No	No	No	Yes	No	No	No	Yes	No	No
CYP2C19 inhibitor	Yes	Yes	No	No	No	Yes	No	No	Yes	Yes	No	No
CYP2C9 inhibitor	Yes	Yes	No	Yes	No	Yes	No	No	No	Yes	No	Yes
CYP2D6 inhibitor	No	No	No	No	No	No	No	No	No	No	No	No
CYP3A4 inhibitor	No	No	No	No	No	No	No	No	No	No	No	No
Total clearance	0.76 log mL/min/Kg	-	0.74 log mL/min/Kg	-	0.68 log mL/min/Kg	-	0.73 log mL/min/Kg	-	0.63 log mL/min/Kg	-	0.81 log mL/min/Kg	-
Renal OCT2 substrate	No	-	No	-	No	-	No	-	No	-	No	-
Max tolerated dose (human)	0.23 log mg/Kg/day	-	0.06 log mg/Kg/day	-	0.53 log mg/Kg/day	-	0.06 log mg/Kg/day	-	0.08 log mg/Kg/day	-	-0.79 log mg/Kg/day	-
Oral acute toxicity (LD ₅₀)	2.84 mol/Kg	-	2.80 mol/Kg	-	2.06 mol/Kg	-	2.56 mol/Kg	-	2.84 mol/Kg	-	2.40 mol/Kg	-

Table S3. Physicochemical properties of JM-20, Nifedipine and respectively metabolites by pkCSM and swissADME.

JM-20 and metabolites														
	JM-20		A		B		C		D		E		F	
	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME
MW	404.43 g/mol	404.42 g/mol	402.41 g/mol	402.40 g/mol	374.36 g/mol	374.35 g/mol	418.41 g/mol	418.40 g/mol	390.35 g/mol	390.35 g/mol	372.34 g/mol	372.33 g/mol	376.37 g/mol	376.37 g/mol
LogP	4.16	2.56	4.95	2.80	4.47	1.77	4.13	2.45	3.65	2.03	4.13	2.09	3.68	1.59
Rotatable bonds	4	5	4	5	3	3	5	6	4	4	2	2	3	3
Acceptors	7	5	7	6	6	6	8	7	7	7	7	6	6	5
Donors	2	2	1	1	2	2	2	2	3	3	1	1	3	3
Surface area	172.67	108.54 Å ²	171.96	109.40 Å ²	158.91	120.40 Å ²	176.75	129.63 Å ²	163.70	140.63 Å ²	158.22	109.40 Å ²	159.62	119.54 Å ²
Nifedipine and metabolites														
	Nifedipine		A		B		C		D		E		F	
	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME	pkCSM	Swiss-ADME
MW	346.34 g/mol	346.33 g/mol	344.32 g/mol	344.32 g/mol	330.29 g/mol	330.29 g/mol	346.29 g/mol	346.29 g/mol	332.27 g/mol	332.26 g/mol	328.28 g/mol	238.28 g/mol	332.31 g/mol	332.31 g/mol
LogP	2.17	2.87	2.85	2.66	2.76	2.00	2.24	1.87	2.16	1.43	2.42	2.12	2.09	1.99
Rotatable bonds	4	6	4	6	4	5	4	6	4	5	3	4	4	5
Acceptors	7	6	7	7	6	7	8	8	7	8	7	7	6	6
Donors	1	1	0	0	1	1	1	1	2	2	0	0	2	2
Surface area	143.81	110.45 Å ²	143.10	111.31 Å ²	136.42	122.31 Å ²	141.53	131.54 Å ²	134.85	142.54 Å ²	135.73	111.31 Å ²	137.13	121.45 Å ²

4 DISCUSSÃO

Esta tese buscou descrever o potencial multialvo de uma nova molécula denominada JM-20 como inibidora de enzimas colinesterases e antioxidante, determinar sua toxicidade em células sanguíneas humanas, assim como averiguar *in silico* utilizando ferramentas de *docking molecular* e de *screening virtual* a conformação ideal do ligante para interação com a AChE e prever efeitos ADMET, respectivamente. Usando enzimas colinesterases de diferentes fontes, foi definido o mecanismo de ação do JM-20 nestas enzimas, a especificidade da molécula e o tipo de inibição enzimática que induz. Na busca de efeitos adicionais do JM-20, descrevemos o potencial antioxidante desta molécula como trapeadora de radicais livres e inibidora da peroxidação de fosfolípidios. A toxicidade do JM-20 foi avaliada através de abordagens *in vitro*, utilizando células sanguíneas saudáveis do organismo, e *in silico* (pkCSM e swissADME) predizendo além da toxicidade, a absorção, distribuição, metabolismo e excreção.

As substâncias capazes de inibir colinesterases são denominadas anticolinesterásicas, e são terapêuticamente administrados, por exemplo, para o tratamento dos sintomas de doenças neurodegenerativas como na doença de Alzheimer (DA) (H. FERREIRA-VIEIRA et al., 2016; POHANKA, 2012). Os medicamentos anticolinesterásicos incluem a tacrina, donepezil, galantamina e rivastigmina (YIANNPOULOU; PAPAGEORGIOU, 2013). De acordo com a hipótese colinérgica da DA, a degeneração de neurônios colinérgicos localizados, principalmente, no córtex cerebral ocasionam um prejuízo na neurotransmissão colinérgica. Este prejuízo contribui significativamente para o déficit da função cognitiva observada em pacientes com DA (FRANCIS et al., 1999; ITO et al., 2010; WARREN; FLETCHER; GOLDEN, 2012; YAN et al., 2008). Portanto, um inibidor de colinesterase retardaria a degradação da ACh promovendo a permanência deste neurotransmissor na fenda sináptica, estimulando uma transmissão colinérgica mais intensificada (H. FERREIRA-VIEIRA et al., 2016; POHANKA, 2012). O fármaco de referência para inibição da colinesterase, a tacrina, apresenta efeitos adversos variados principalmente relacionados a hepatotoxicidade, inviabilizando seu uso na prática clínica atual (BAZSON et al., 1995; LANCTÔT; RAJARAM; HERRMANN, 2009; OSSENI et al., 1999). Neste sentido, essas novas substâncias devem ser eficazes e ao mesmo tempo apresentar de efeitos adversos mínimos.

Adicionalmente, a ACh também pode ser encontrada fora do sistema nervoso, fazendo parte do sistema colinérgico não-neuronal (WESSLER et al., 2003; WESSLER;

KIRKPATRICK, 2016, 2008; WESSLER; KIRKPATRICK; RACK, 1998). Seus efeitos podem ser mediados através de receptores nicotínicos e muscarínicos extracelulares, ligados à membrana celular da célula onde foi sintetizada ou de células vizinhas, ou ainda interagir diretamente com proteínas de sinalização intracelular (WESSLER; KIRKPATRICK; RACK, 1998). A enzima que inativa a ACh, a AChE, é abundantemente expressa nas células de mamíferos, isto é, células neuronais e não neuronais (WESSLER; KIRKPATRICK; RACK, 1998). Inicialmente, a AChE presente nos eritrócitos foi descrita como um biomarcador da integridade da membrana dessas células (SALDANHA, 2017). Entretanto, foram observadas alterações na atividade dessa enzima quando não estavam relacionadas com a integridade da membrana dos eritrócitos, por exemplo, a atividade da AChE aumentada em diferentes doenças e baixa atividade da AChE eritrocitária em pacientes expostos a pesticidas (SALDANHA, 2017).

Os primeiros estudos que tinham como objetivo avaliar a atividade da AChE utilizavam tecidos animais que continham a enzima e, somente a partir da década de 30, que a enzima isolada passou a ser utilizada por pesquisadores (NACHMANSOHN; ROTHENBERG, 1945). A técnica mais utilizada para determinação da atividade da ChE foi descrita por Ellman e colaboradores em 1961 (ELLMAN et al., 1961). O método proposto por Ellman e colaboradores baseia-se na taxa de hidrólise da acetiltiocolina (ACh modificada) pelas ChE, formando acetato e tiocolina. A tiocolina reage com o ânion carboxilato do DTNB produzindo o 2-nitrobenzoato 5-mercaptopiocolina e o 5-tio-2-nitrobenzoato (ânion de coloração amarela) que é quantificado por espectrofotômetro no comprimento de onda de 412 nm (ELLMAN et al., 1961).

Nesta tese, ampliamos o conhecimento sobre uma nova molécula, o JM-20, na inibição específica da enzima acetilcolinesterase. Estudos prévios demonstraram valores de IC_{50} para inibição da AChE iguais a 193.2 ± 10.4 nmol/L e 214.8 ± 3.1 nmol/L para *HsAChE* e *EeAChE*, respectivamente (WONG-GUERRA et al., 2019). Além do mais, pela primeira vez foi demonstrada uma análise cinética desse efeito em diferentes fontes da enzima (artigo 1). A cinética enzimática ou o estudo das velocidades das reações enzimáticas compreende informações indiretas sobre os mecanismos de ação catalítica, especificidade das enzimas, fatores que podem afetar a velocidade da reação e a determinação quantitativa de seus efeitos (MOTTA, 2003).

Nós utilizamos a acetilcolinesterase isolada e purificada do órgão elétrico do peixe *Electrophorus electricus*, isolada das membranas de eritrócitos humanos e dos eritrócitos

totais. Independentemente da fonte da enzima o composto foi capaz de inibir a atividade da AChE em concentrações nanomolares. Para avaliação cinética, utilizamos diferentes concentrações de substrato da reação (acetiltiocolina) e três concentrações do composto previamente selecionadas. Os resultados demonstraram que o composto foi capaz de inibir a atividade da AChE concentração-dependente. Além disso, o JM-20 aumentou os valores de K_m em função de sua concentração enquanto a V_{max} diminuiu, sugerindo um tipo de inibição mista. Os inibidores de tipo misto fazem parte dos inibidores reversíveis, os quais ligam-se a um sítio distinto do sítio ativo (local onde o substrato se liga) (NELSON; COX, 2014). O inibidor pode ligar-se tanto a enzima quanto ao complexo enzima-substrato. Neste tipo de inibição observamos a variação de K_m e V_{max} (NELSON; COX, 2014).

Para avaliar se a inibição do JM-20 era especificamente para a AChE ou se poderia inibir também a BChE, foi utilizada a enzima isolada e purificada de *Equus caballus* e de plasma humano (artigo 1). O JM-20 foi capaz de inibir a *EcBChE* somente em concentrações micromolares (artigo 1). Por isso, consideramos que a inibição do JM-20 sob enzimas colinesterases é específica para a AChE devido ao fato da ausência de inibição para a BChE na faixa nanomolar (artigo 1).

A fim de entendermos melhor a interação do composto com a AChE, realizamos análises *in silico* de simulação molecular (artigo 1). Essa tecnologia computacional é uma potente ferramenta capaz de prever a orientação de uma ligante quando este interage com um receptor ou uma enzima (CHEN, 2015). Surpreendentemente, a análise estrutural do JM-20 indicou a presença de um carbono quiral na quarta posição, o qual pode apresentar ressonância com o átomo de hidrogênio do grupo amina do anel diazepínico variando entre as posições 6 e 11, caracterizando um tautômero. Adicionalmente, a presença de um átomo de nitrogênio com hibridização sp^3 (posição 6 ou 11) no anel diazepínico revela uma geometria não planar, possibilitando duas conformações adicionais diferentes para a molécula. No total, existe a possibilidade de o JM-20 apresentar pelo menos oito isômeros que não haviam sido estudados até o momento (artigo 1).

Sabe-se que podem ocorrer variações entre os isômeros de acordo com a especificidade e a atividade de ligação em relação as enzimas, assim, a determinação do isômero com maior afinidade pela enzima é crucial para elucidar o mecanismo de ação e as estratégias farmacológicas potencias do composto (AARTHY et al., 2017; ARKIN; WELLS; FRANCISCO, 2004; HOPKINS, 2008; JANG et al., 2018; MAKOLA et al., 2016; SINKO et al., 2011). Por este motivo, realizamos a simulação molecular com cada um dos oito isômeros

do JM-20 interagindo com a AChE de *Homo sapiens* (HsAChE) e de *Electrophorus electricus* (EeAChE) e com a BChE de *Homo sapiens* (HsBChE) e de *Equus caballus* (EcBChE) (artigo 1) com a finalidade de determinar a postura de ligação de cada isômero e a probabilidade do isômero mais ativo.

Para as análises aplicamos dois protocolos distintos de *docking molecular*. O primeiro protocolo consistiu na ausência moléculas de água, o qual demonstrou que todos os isômeros do JM-20 podem acessar o sitio ativo das enzimas, com um padrão específico de interações. Na ausência de água o JM-20 se liga ao fundo do *gorge* próximo a tríade catalítica das enzimas HsAChE e HsBChE. Por outro lado, o segundo protocolo que englobou simulações de docking molecular na presença de moléculas de água (utilizando o HsAChE e HsBChE), também demonstraram que o JM-20 interage no sítio ativo de ambas as colinesterases, mas com modo de ligação diferente quando comparado com o docking sem moléculas de água. ROSENBERRY et al., 2017 descrevem que a ausência de água em simulações moleculares com colinesterases resulta em posições de ligação não confiáveis e, além disso, as moléculas de água tem importância na catálise destas enzimas (DVIR et al., 2002; KOELLNER et al., 2000; NICOLET et al., 2003; ROSENBERRY et al., 2017). Desta forma, aqui destacamos os resultados da docagem na presença de água. As análises das interações e da energia de ligação mostraram que os isômeros a, b, e e f apresentam uma conformação de ligação muito semelhante, e energia de ligação mais favorável quando comparados com os isômeros c, d, g e h, indicando que os isômeros 4-R interagem melhor com o HsAChE (material de apoio do artigo 1).

Além de inibidor de AChE, o JM-20 também possui efeito antioxidante através da inibição da peroxidação lipídica e trapeador de radicais livres, que foram avaliadas utilizando as técnicas colorimétricas de produção de substâncias reativas ao ácido tiobarbitúrico (TBARS) e de trapeamento do radical 2,2-difenilpicrilhidrazil (DPPH) (manuscrito 1). No teste de TBARS, foram utilizados leucócitos isolados do sangue periférico e foi induzida a peroxidação com Fe^{2+} . O JM-20 foi capaz de proteger os leucócitos da peroxidação lipídica. Para entender o mecanismo de ação do JM-20 sob a inibição da peroxidação lipídica, foi avaliada a capacidade deste composto em quelar ferro, entretanto, o JM-20 não apresentou efeitos significativos (dados não mostrados). Estes resultados sugerem outro mecanismo de ação do JM-20 como inibidor da peroxidação lipídica induzida pelo Fe^{2+} , provavelmente relacionado à proteção da membrana celular por este composto. A avaliação da capacidade de trapeamento do JM-20 foi verificada em duas partes. Na primeira parte, monitoramos a absorvância em espectrofotômetro SpectraMax a 518 nm durante 30 minutos com intervalo de

30 segundos, denominamos reação de fase rápida. No segundo momento, a reação de fase lenta, a absorvência foi registrada a cada 30 minutos durante 3 horas. O JM-20 teve efeito somente na fase lenta, nas concentrações de 50 e 100 μM .

Para torna-se um medicamento de uso clínico, uma substância precisa possuir efeito farmacológico e também apresentar baixa toxicidade ao organismo humano (PIRES; KAMINSKAS; ASCHER, 2018). É importante ressaltar que o JM-20 não induziu mortalidade em camundongos fêmeas tratados com o composto (2000 mg/Kg) por via intragástrica (gavagem) (NUÑEZ-FIGUEREDO et al., 2013). Desse modo, elucidamos a ausência de toxicidade do JM-20 *in vitro* em altas concentrações nos eritrócitos (manuscrito 1). Substâncias ou compostos desconhecidos podem desencadear o processo de ruptura de glóbulos vermelhos ou eritrócitos e, conseqüentemente, a liberação de seu conteúdo (hemoglobina) no plasma (AMIN; DANNENFELSER, 2006). Assim, o teste que visa avaliar o efeito hemolítico tem sido comumente utilizado como uma das metodologias de triagem para muitas substâncias de interesse clínico (AMIN; DANNENFELSER, 2006). Adicionalmente, o ensaio de viabilidade celular usando azul de tripan é um método amplamente descrito na literatura e usado para avaliar a citotoxicidade através da integridade da membrana (ADAN; KIRAZ; BARAN, 2016; ALTMAN; RANDERS; RAO, 1999; BHATIA; YETTER, 2008). O teste é baseado na absorção do azul de tripan no citoplasma das células mortas devido à perda de seletividade da membrana da mesma, enquanto as células vivas permanecem intactas (ALTMAN; RANDERS; RAO, 1999). Nossos resultados mostraram que o JM-20 não causou hemólise, entretanto, as concentrações 20 e 50 μM do composto foram responsáveis por uma diminuição de cerca de 20% e 50% na viabilidade dos leucócitos, respectivamente, quando comparado ao grupo controle (DMSO) (manuscrito 1).

Neste estudo também exploramos a existência de efeitos citotóxicos do JM-20 em leucócitos através da produção de espécies reativas (ER) e ausência de alterações no ciclo celular (manuscrito 1). A citometria de fluxo pode ser usada para a detecção de ER através de sondas fluorescentes (ERUSLANOV; KUSMARTSEV, 2010). A oxidação do diacetato de 2'-7'-diclorodihidrofluoresceína (DCFH-DA) é uma das técnicas amplamente utilizadas para detectar a produção de ER. Essa técnica apresenta algumas vantagens como reprodutibilidade, sensibilidade e preço acessível (ERUSLANOV; KUSMARTSEV, 2010). A exposição a todas as concentrações de JM-20 (10-50 μM) causou um aumento significativo nos níveis de espécies reativas intracelulares (ER). Embora tenha sido observada diminuição da viabilidade celular e aumento da produção de ER, a exposição ao JM-20 (10-50 μM) não alterou o ciclo

celular. Por outro lado, o JM-20 apresentou IC_{50} similar ao α -tocoferol na inibição da peroxidação lipídica, sendo $1,051 \pm 0,21$ e $1,065 \pm 0,34$ μM , respectivamente. Logo, apresentando uma faixa de segurança ampla entre possíveis concentrações tóxicas e terapêuticas. Além do mais, como inibidor da AChE o JM-20 teve seu efeito em concentrações nanomolares.

Ferramentas computacionais também podem colaborar na triagem dos feitos de uma substância nova (HU et al., 2020; PIRES; KAMINSKAS; ASCHER, 2018). Neste contexto, nós avaliamos o perfil ADMET (absorção, distribuição, metabolismo, excreção e toxicidade) do JM-20 utilizando plataformas gratuitas online como o swissADME e o pkCSM. Por exemplo, parâmetros como permeabilidade CaCo-2 e absorção intestinal podem colaborar para elucidação sobre a absorção de drogas por via oral (DAINA; MICHIELIN; ZOETE, 2017; PIRES; BLUNDELL; ASCHER, 2015). Além disso, com relação ao metabolismo, é possível verificar efeitos de inibição em enzimas responsáveis pelo processo de desintoxicação do organismo, encontradas principalmente no fígado, como as isoformas do citocromo P450 (DAINA; MICHIELIN; ZOETE, 2017; PIRES; BLUNDELL; ASCHER, 2015). Adicionalmente comparamos o JM-20 com a nifedipina, uma molécula estruturalmente similar ao nosso composto e atualmente utilizada na prática clínica para tratar angina e hipertensão, compondo a lista de medicamentos essenciais da organização mundial da saúde (LEE et al., 2014; OMS, 2019). Conforme esperado, o JM-20 apresentou efeitos farmacocinéticos e toxicológicos similares a nifedipina de acordo com as análises computacionais, embora tenha havido divergências entre as plataformas utilizadas. Assim, também salientamos a importância da utilização de mais de um método computacional de avaliação.

4 CONCLUSÃO

Assim sendo, nosso trabalho demonstrou que o composto JM-20 foi eficaz na inibição da AChE, pois apresentou efeito em concentrações nanomolares. Enquanto que na BChE não foi constatada inibição nas mesmas concentrações, logo, o JM-20 apresenta seletividade para a AChE. Os ensaios cinéticos mostraram que, provavelmente, o tipo de inibição enzimática é misto. O JM-20 também teve um potente efeito antioxidante como trapeador de radicais livres e inibidor da peroxidação lipídica induzida pelo Fe^{2+} nos leucócitos, entretanto esse efeito não ocorre pela atividade quelante de Fe^{2+} . Os ensaios de citotoxicidade utilizando células sanguíneas humanas indicaram que o JM-20 não causa hemólise, nem altera o ciclo celular,

porém, parece ser pró-oxidante em altas concentrações. Através de testes *in silico* comparamos a absorção, distribuição, metabolismo, excreção e toxicidade (ADMET) do JM-20, da nifedipina e de seus respectivos metabólitos e, conforme esperado, ambos os compostos apresentaram perfil ADMET semelhante. Por fim, nossos resultados reforçam a utilização do JM-20 como um composto multialvo na terapia, de distúrbios do SNC, principalmente.

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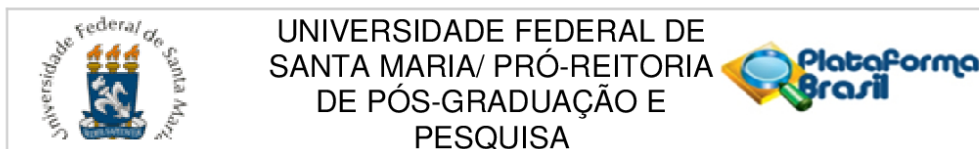
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ANEXO A - PARECER DO COMITÊ DE ÉTICA EM PESQUISA COM SERES HUMANOS (CEP)



PARECER CONSUBSTANCIADO DO CEP

DADOS DO PROJETO DE PESQUISA

Título da Pesquisa: Avaliação da toxicidade do JM-20 e seus derivados na viabilidade de leucócitos humanos.

Pesquisador: João Batista Teixeira da Rocha

Área Temática:

Versão: 2

CAAE: 14656919.4.0000.5346

Instituição Proponente: Universidade Federal de Santa Maria/ Pró-Reitoria de Pós-Graduação e

Patrocinador Principal: MINISTERIO DA CIENCIA, TECNOLOGIA E INOVACAO

DADOS DO PARECER

Número do Parecer: 3.471.823

Apresentação do Projeto:

Projeto vinculado ao PPG de Bioquímica Toxicológica do Centro de Ciências Naturais e Exatas da UFSM.

As amostras serão obtidas de 20 doadores voluntários saudáveis mediante consentimento. Será coletado 10 mL de cada participante em tubo contendo heparina. O tubo com sangue venoso será centrifugado e o plasma obtido será utilizado para o ensaio de viabilidade celular. O plasma obtido será purificado através de sucessivas lavagens com solução de lise de hemácias. Após, os leucócitos serão mantidos em tampão Hank's para manter a viabilidade dos mesmos. Os leucócitos isolados (2×10^6 /mL-1) serão incubados durante três horas com DMSO (0,5%), 1 mM de t-butil hidroperóxido (controle positivo) ou o composto indicado em uma solução de tampão de Hank's. Após o termino do período de incubação será adicionado o azul de Tripan e será realizada uma contagem com auxílio do microscópio ótico dos leucócitos viáveis e não viáveis. Contém critérios de inclusão e exclusão dos sujeitos de pesquisa.

Os dados obtidos serão expressos como média \pm erro padrão. A análise estatística dos dados será realizada através de análise de variância (ANOVA) de duas vias, seguida pelo teste de Newman Keuls quando for necessário. Todas as análises serão realizadas utilizando o programa Prism®. Os

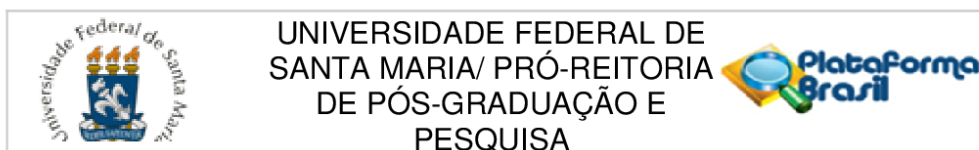
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PESQUISA

Continuação do Parecer: 3.471.823

valores de $p < 0.05$ serão considerados significativos.

Apresenta cronograma de execução.

Objetivo da Pesquisa:

Avaliar o efeito do composto JM-20 e seus derivados na viabilidade celular de leucócitos humanos.

Avaliação dos Riscos e Benefícios:

Considerando-se as características do projeto, a descrição de riscos e benefícios apresentada pode ser considerada suficiente.

Comentários e Considerações sobre a Pesquisa:

.

Considerações sobre os Termos de apresentação obrigatória:

Os termos de apresentação obrigatória podem ser considerados suficientes.

Recomendações:

Veja no site do CEP - <http://w3.ufsm.br/nucleodecomites/index.php/cep> - na aba "orientações gerais", modelos e orientações para apresentação dos documentos. ACOMPANHE AS ORIENTAÇÕES DISPONÍVEIS, EVITE PENDÊNCIAS E AGILIZE A TRAMITAÇÃO DO SEU PROJETO.

Conclusões ou Pendências e Lista de Inadequações:

.

Considerações Finais a critério do CEP:

Este parecer foi elaborado baseado nos documentos abaixo relacionados:

Tipo Documento	Arquivo	Postagem	Autor	Situação
Informações Básicas do Projeto	PB_INFORMAÇÕES_BÁSICAS_DO_PROJETO_1362786.pdf	01/07/2019 14:47:17		Aceito

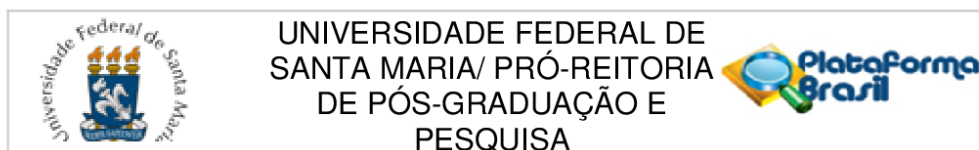
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Continuação do Parecer: 3.471.823

Outros	INSTRUMENTOSDECOLETADEADODOS.pdf	01/07/2019 14:46:27	João Batista Teixeira da Rocha	Aceito
Projeto Detalhado / Brochura Investigador	Projetocepjm20.pdf	01/07/2019 14:46:08	João Batista Teixeira da Rocha	Aceito
TCLE / Termos de Assentimento / Justificativa de Ausência	TERMODECONSENTIMENTOLIVREESCLARECIDOcor.pdf	01/07/2019 14:45:46	João Batista Teixeira da Rocha	Aceito
Outros	Pendencias.pdf	28/06/2019 10:19:16	João Batista Teixeira da Rocha	Aceito
Outros	TERMODECONFIDENCIALIDADEcorreto.pdf	28/05/2019 08:14:45	João Batista Teixeira da Rocha	Aceito
Folha de Rosto	JM20folharosto3.pdf	27/05/2019 13:46:51	João Batista Teixeira da Rocha	Aceito
Outros	projeto63983gap.pdf	24/05/2019 15:24:31	João Batista Teixeira da Rocha	Aceito
TCLE / Termos de Assentimento / Justificativa de Ausência	TERMODECONSENTIMENTOLIVREESCLARECIDO.pdf	24/05/2019 15:23:15	João Batista Teixeira da Rocha	Aceito
Declaração de Instituição e Infraestrutura	JM20autorizacao.pdf	22/05/2019 17:29:37	João Batista Teixeira da Rocha	Aceito
Projeto Detalhado / Brochura Investigador	Projetojm20.pdf	22/05/2019 17:29:17	João Batista Teixeira da Rocha	Aceito

Situação do Parecer:

Aprovado

Necessita Apreciação da CONEP:

Não

SANTA MARIA, 26 de Julho de 2019

Assinado por:
CLAUDEMIR DE QUADROS
(Coordenador(a))

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