

UNIVERSIDADE FEDERAL DE SANTA MARIA
CENTRO DE CIÊNCIAS NATURAIS E EXATAS
PROGRAMA DE PÓS-GRADUAÇÃO EM CIÊNCIAS BIOLÓGICAS:
BIOQUÍMICA TOXICOLÓGICA

**EFEITOS DO EXTRATO AQUOSO DE FOLHAS DE
Syzygium cumini SOBRE A OXIDAÇÃO E GLICAÇÃO
DE LIPOPROTEÍNAS DE BAIXA DENSIDADE**

DISSERTAÇÃO DE MESTRADO

Matheus Mülling dos Santos

Santa Maria, RS, Brasil
2014

**EFEITOS DO EXTRATO AQUOSO DE FOLHAS DE
Syzygium cumini SOBRE A OXIDAÇÃO E
GLICAÇÃO DE LIPOPROTEÍNAS DE BAIXA
DENSIDADE**

Matheus Mülling dos Santos

Dissertação apresentada ao Curso de Mestrado do Programa de Pós-Graduação em Ciências Biológicas: Bioquímica Toxicológica da Universidade Federal de Santa Maria (UFSM, RS), como requisito parcial para obtenção do grau de
Mestre em Ciências Biológicas: Bioquímica Toxicológica.

Orientadora: Prof^a. Dr^a. Nilda Berenice de Vargas Barbosa

Santa Maria, RS, Brasil
2014

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Mülling dos Santos, Matheus

Efeitos do extrato aquoso de folhas de *Syzygium cumini* sobre a oxidação e glicação de lipoproteínas de baixa densidade. / Matheus Mülling dos Santos.-2014.
66 p.; 30cm

Orientadora: Nilda Berenice de Vargas Barbosa
Dissertação (mestrado) - Universidade Federal de Santa Maria, Centro de Ciências Naturais e Exatas, Programa de Pós-Graduação em Ciências Biológicas: Bioquímica Toxicológica, RS, 2014

1. Lipoproteínas de baixa densidade 2. Aterosclerose
3. Diabetes 4. *Syzygium cumini* 5. Estresse oxidativo I.
de Vargas Barbosa, Nilda Berenice II. Título.

Universidade Federal de Santa Maria
Centro de Ciências Naturais e Exatas
Programa de Pós Graduação em Ciências Biológicas:
Bioquímica Toxicológica

A comissão examinadora, abaixo assinada, aprova a Dissertação de Mestrado

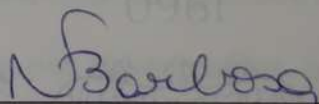
EFEITOS DO EXTRATO AQUOSO DE FOLHAS DE *Syzygium cumini* SOBRE
A OXIDAÇÃO E GLICAÇÃO DE LIPOPROTEÍNAS DE BAIXA DENSIDADE

Elaborada por

Matheus Mülling dos Santos

Como requisito básico para a obtenção do grau de
Mestre em Ciências Biológicas: Bioquímica Toxicológica

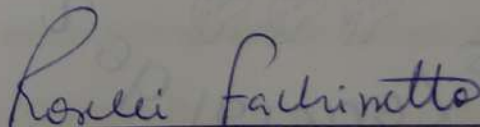
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Profª Drª Roselei Fachinnetto (UFSM)

Santa Maria, 28 de Agosto de 2014.

AGRADECIMENTOS

Agradeço primeiramente a Deus pelo dom da vida e pela companhia constante fazendo com que eu tenha a certeza de nunca estar sozinho.

Agradeço imensamente à minha mãe Leni e a meu pai João por todo amor, carinho e apoio que sempre me deram e que foi fundamental para eu chegar até aqui.

Quero agradecer de forma especial à minha orientadora Prof^a Nilda por todos os ensinamentos, pela compreensão, disponibilidade, confiança e apoio.

Também gostaria de agradecer ao Prof. João Batista pela oportunidade de ingressar no meio científico e com isso descobrir que é o que quero pra minha vida.

Muito obrigado a todos os colegas de laboratório, inclusive os que não estão mais conosco, por todos os momentos vividos. Alegro-me muito em fazer parte desse grupo de pesquisa e em tê-los como colegas.

Um agradecimento especial aos grandes amigos Alessandro e Claudinha pela parceria que vai muito além dos experimentos. Muito obrigado pela certeza de poder contar sempre com vocês.

Muito obrigado aos colegas dos laboratórios dos Prof. João, Félix, Gilson e das Prof^a Cristina e Rose pela parceria em muitos momentos.

Agradeço à CAPES pela bolsa de estudos durante todo o curso.

Por fim, agradeço à Universidade Federal de Santa Maria e ao Programa de Pós Graduação em Ciências Biológicas: Bioquímica Toxicológica pela possibilidade de realização deste curso.

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LISTA DE ABREVIATURAS

AGEs: Produtos finais de glicação avançada

apoB-100: Apolipoproteína B-100

AVC: Acidente vascular cerebral

DC: Dieno conjugado

DCV: Doenças carvasculares

glyLDL: LDL glicada

LDL: Lipoproteína de baixa densidade

MCP-1: Proteína quimiosstática de monócitos-1

M-CSF: Fator estimulador de colônia de macrófagos

MDA: Malondialdeído

MG: Metilglioxal

MG-H1: hydroimidazolone N₅-(5-hydro-5-methyl -4-imidazolon-2-yl)-ornithine

mm-LDL: LDL minimamente modificada

nLDL: LDL nativa

OMS: Organização Mundial de Saúde

oxLDL: LDL oxidada

PUFAs: Ácidos graxos poliinsaturados

RESUMO

Dissertação de Mestrado
Programa de Pós-Graduação em Ciências Biológicas: Bioquímica Toxicológica
Universidade Federal de Santa Maria, RS, Brasil

Efeitos do extrato aquoso de folhas de *Syzygium cumini* sobre a oxidação e glicação de lipoproteínas de baixa densidade

AUTOR: Matheus Mülling dos Santos
ORIENTADORA: Nilda Berenice de Vargas Barbosa
LOCAL E DATA DA DEFESA: Santa Maria, 28 de agosto de 2013

Modificações estruturais nas lipoproteínas de baixa densidade (LDL) são consideradas uma das principais causas para o desenvolvimento da aterosclerose. Dentre essas modificações, destacam-se as causadas por oxidação e/ou glicação. A glicação é favorecida em indivíduos diabéticos devido à hiperglicemia e torna o Diabetes mellitus um dos principais fatores de risco para a aterosclerose. Tendo em vista as relações entre a hiperglicemia e a aterosclerose, bem como o papel do estresse oxidativo em ambas as condições, este trabalho teve como objetivo avaliar o efeito *in vitro* do extrato aquoso de folhas de *Syzygium cumini* (S.cExt), uma planta utilizada popularmente no tratamento do diabetes, sobre parâmetros de oxidação e glicação em LDL isoladas de plasma humano. Os resultados obtidos demonstraram que o S.cExt exibiu um potente efeito antioxidante sobre a peroxidação lipídica induzida por CuSO₄ em LDL, soro e plasma humanos avaliada através da formação de dienos conjugados e da produção de espécies reativas ao ácido tiobarbitúrico (TBARS). Além da ação antioxidante exibida contra a peroxidação lipídica, a análise da perda da fluorescência do triptofano indicou que o S.cExt é também capaz de reduzir os danos oxidativos causados pelo CuSO₄ à porção protéica da LDL. Diferentemente dos parâmetros de oxidação, o S.cExt não exibiu potencial antiglicante sobre a glicação de LDL induzida pelo metilglioxal (MG), que foi avaliada através da mobilidade eletroforética na eletroforese em gel de agarose e através da análise fluorimétrica da formação de produtos finais de glicação avançada (AGEs). De forma geral, os dados obtidos mostram que o S.cExt pode ser considerado um agente antiaterogênico promissor devido à sua atividade antioxidante sobre a oxidação de LDL *in vitro*.

Palavras-Chave: LDL, aterosclerose, oxidação, glicação, estresse oxidativo, *Syzygium cumini*.

ABSTRACT

Dissertation of Master's Degree
Graduate Course in Toxicological Biochemistry
Federal University of Santa Maria, RS, Brazil

Effect of *Syzygium cumini* aqueous-leaf extract on oxidation and glycation of low density lipoproteins

AUTHOR: Matheus Mülling dos Santos
ADVISOR: Nilda Berenice de Vargas Barbosa
PLACE AND DATE OF THE DEFENSE: Santa Maria, 28th august, 2014.

Structural modifications of low density lipoproteins (LDL) are considered one of the main causes to atherosclerosis development. Among these modifications, highlights the modifications caused by oxidation and/or glycation, the latter being favored in diabetic subjects due to the hyperglycemia status and consequent increased availability of glucose, making diabetes mellitus one of the risk factors of atherosclerosis. Considering the relationships established between atherosclerosis and hyperglycemia, as well as the role of oxidative stress in both conditions, this work aimed to evaluate the effect *in vitro* of the *Syzygium cumini* aqueous-leaf extract (S.cExt), a plant popularly used to treat diabetes, on parameters of oxidation and glycation in low density lipoproteins isolated from human plasma. The results obtained demonstrated that S.cExt exhibited a potent antioxidant effect on CuSO₄-induced lipid peroxidation in human LDL, serum and plasma assayed by conjugated dienes formation and thiobarbituric acid reactive substances (TBARS) production. Besides the antioxidant activity exhibited against lipid peroxidation, the assay of loss of tryptophan fluorescence showed that S.cExt is also able to diminish the oxidative damage caused by CuSO₄ in the protein moiety of LDL. In contrast to oxidation parameters, S.cExt did not modify the LDL glycation induced by methylglyoxal (MG), which was evaluated by eletrophoretic mobility in agarose gel eletroforesis and by the fluorimetric analysis of the formation of advanced glycation end products (AGEs). In general, the data obtained here suggest that S.cExt can be considered a promising antiatherogenic agent due its antioxidant activity on LDL oxidation *in vitro*.

Key words: LDL, atherosclerosis, oxidation, glycation, oxidative stress, *Syzygium cumini*.

APRESENTAÇÃO

No item **INTRODUÇÃO**, está descrito uma sucinta revisão bibliográfica sobre os temas abordados nessa dissertação.

O **DESENVOLVIMENTO** está apresentado sob a forma de um manuscrito, o qual se encontra alocado no item **MANUSCRITO**.

As seções Materiais e Métodos, Resultados e Discussão encontram-se no próprio manuscrito e representam a íntegra deste estudo.

Os itens **CONCLUSÕES** e **PERSPECTIVAS** são encontrados no final desta dissertação e apresentam interpretações e comentários gerais sobre a investigação desenvolvida.

As **REFERÊNCIAS BIBLIOGRÁFICAS** referem-se somente às citações que aparecem no item **INTRODUÇÃO**, uma vez que o manuscrito contém as suas próprias referências.

1. INTRODUÇÃO

Segundo dados da Organização Mundial de Saúde (OMS), as doenças cardiovasculares (DCV) são atualmente a principal causa de morte em todo o mundo. Em 2008, estimou-se que cerca de 17,3 milhões de pessoas tenham morrido em decorrência de DCV, o que representa 30% do total de mortes no mundo (WHO, 2011a). Dentre essas, estima-se que em torno de 7,3 milhões ocorreram devido à doença cardíaca coronária e 6,2 milhões devido à ocorrência de acidente vascular cerebral (AVC) (WHO, 2011b). Um terço dos óbitos decorrentes de DCV ocorre de maneira precoce, em indivíduos na faixa etária de 35 a 64 anos (Ishitani et. al, 2006).

O infarto agudo do miocárdio bem como o AVC são considerados as principais manifestações das doenças cardiovasculares. Ambas as manifestações costumam ocorrer em decorrência de um processo vascular denominado aterosclerose. A aterosclerose pode ser definida como uma doença inflamatória crônica caracterizada pela formação de ateromas em artérias de grande e médio calibre (Libby et al., 2002). Esses ateromas são placas contendo basicamente lipídeos e tecido fibroso que se acumulam no endotélio arterial causando uma diminuição gradativa do lúmen do vaso sanguíneo, a qual pode culminar com a obstrução total do mesmo e com consequente prejuízo no fornecimento de sangue tecidual (Montecucco et al., 2009) (Figura 1). Além disso, o trombo formado pode se desprender e se alojar em artérias de menor calibre, ocluindo-as e causando um infarto. A trombose oriunda de uma placa aterosclerótica é considerada uma das principais causas dos eventos cardiovasculares que ocorrem de forma súbita (Libby, 2002).

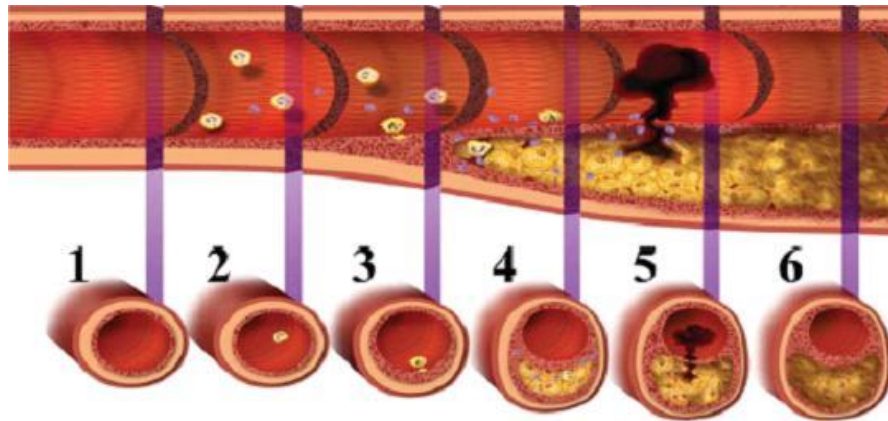


Figura 1: Progressão da lesão aterosclerótica. 1- Artéria em seu estado normal. 2- Acúmulo inicial de lipídeos na camada mais interna da artéria. 3- Estágios iniciais da placa. 4- Formação de uma placa complexa. 5- Ruptura da placa aterosclerótica com formação de trombo. 6- Reabsorção do trombo com conseqüente aumento no tamanho da placa aterosclerótica. (<http://radlivres2010-1.blogspot.com.br/2010/07/aterosclerose-e-sua-relacao-com.html> acessado em 18/08/2014)

Dados epidemiológicos vêm demonstrando que a aterosclerose é uma doença diretamente ligada ao envelhecimento, cujo aparecimento prematuro pode estar relacionado a diferentes condições clínicas como: excesso de lipoproteínas de baixa densidade (LDL) (Figura 2), diabetes, hipertensão, tabagismo, entre outras (Smith et al., 2004). Uma das principais dificuldades para o diagnóstico da aterosclerose se deve ao fato da mesma ser uma doença silenciosa. Várias evidências sugerem que o processo de formação da placa aterosclerótica inicia na infância, porém, a doença progride de maneira assintomática até a idade adulta (Françoso e Coates, 2002).

Durante as últimas décadas, inúmeros estudos vêm tentando elucidar os mecanismos celulares e moleculares envolvidos na aterogênese, a ponto de serem criadas diferentes hipóteses para explicar tal evento. Uma das hipóteses que tenta explicar como ocorre o processo aterogênico é a chamada hipótese da modificação oxidativa (Steinberg et al., 1989).

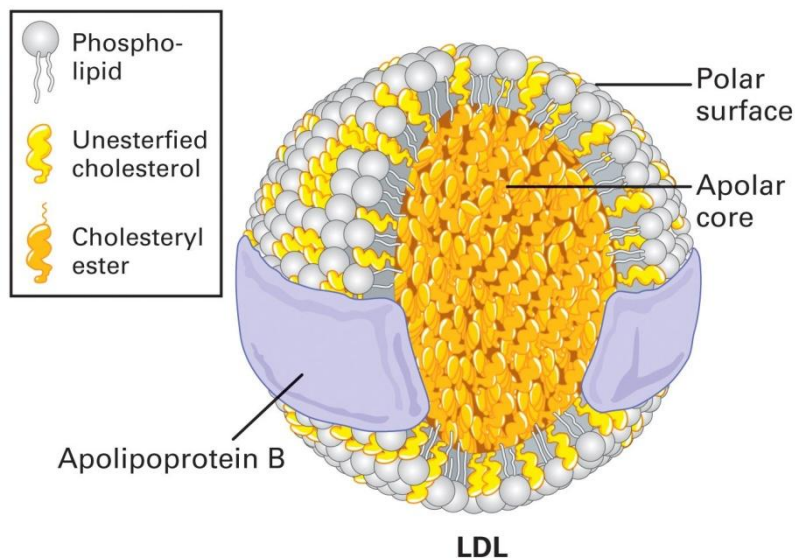


Figura 2: Lipoproteína de baixa densidade (LDL).
http://healthyprotocols.com/2_ldl.htm acessado em 15/08/2014)

De acordo com a hipótese da modificação oxidativa, as moléculas de LDL oxidada se tornam aterogênicas através (I) do recrutamento de monócitos circulantes para a túnica íntima da artéria, (II) da inibição da capacidade dos macrófagos residentes abandonarem este espaço, (III) da sua internalização pelos receptores scavengers dos macrófagos com consequente formação de células espumosas e (IV) da citotoxicidade e indução de perda da integridade endotelial (Steinberg et al., 1989).

Os mecanismos envolvidos na modificação da LDL ainda não estão completamente elucidados; no entanto, um dos eventos iniciais é a peroxidação lipídica de folfolídeos e ésteres de colesterol, os quais possuem ácidos graxos poliinsaturados (PUFAs). Inicialmente ocorre a remoção de um hidrogênio seguida pelo rearranjo das duplas ligações formando um dieno conjugado (DC). Este por sua vez reage com o oxigênio e desencadeia uma sequência de reações culminando com a formação de hidroperóxidos lipídicos. Pode então haver a clivagem das

macrófagos (Brown et al., 1983). Quando ligadas a esses receptores, as moléculas de LDL são internalizadas nos macrófagos formando as chamadas células espumosas que são as células predominantes na lesão aterosclerótica inicial conhecida como estria gordurosa (Osterub et al., 2003). Diferentemente dos receptores de LDL que são inibidos por retroalimentação, os receptores *scavenger* dos macrófagos não são regulados negativamente com o aumento no conteúdo de colesterol nos macrófagos. Dessa forma, os macrófagos continuam capturando moléculas de LDL oxidada (oxLDL), com conseqüente acúmulo de uma quantidade significativa de lipídeos na parede arterial (Osterub et al., 2003).

Diversos trabalhos têm sugerido que o aumento de oxLDL é um fator pró-aterogênico por desencadear um processo inflamatório (Witztum, 1993). Quando oxidada, a LDL pode induzir a ativação de células endoteliais e musculares-lisas; a secreção de mediadores inflamatórios e a expressão de moléculas de adesão (Witztum e Steinberg, 1991).

Antes de ocorrer a oxidação da apoB-100, quando apenas a porção lipídica é afetada e a apoB-100 ainda se encontra íntegra, a LDL é conhecida como “minimamente modificada” (mm-LDL). Nesse estágio, por não haver modificação na estrutura da apoB-100, as moléculas de LDL ainda são reconhecidas pelos receptores de LDL e não pelos receptores *scavenger* (Mashima e cols., 2001). As mm-LDL podem induzir a expressão da proteína quimiostática de monócitos-1 (MCP-1) e também do fator estimulador de colônia de macrófagos (M-CSF) pelas células endoteliais que então iniciam o recrutamento de monócitos para a parede arterial e promovem a diferenciação de monócitos em macrófagos (Cushing et al., 1990; Villa-Colinayo et al., 2000). A figura 3 ilustra o processo de oxidação da LDL e formação das células espumosas (foam cells).

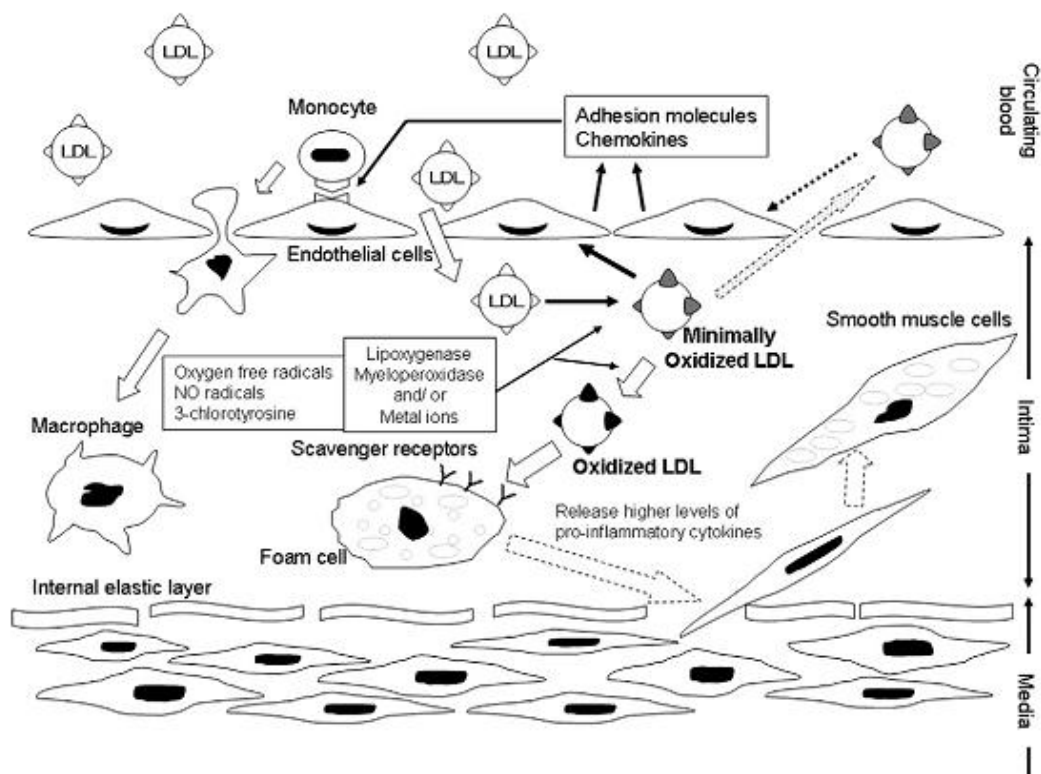


Figura 3: Oxidação da LDL e formação de células espumosas (foam cells) (Hioshida e Hisugi, 2010).

A importância da oxidação da LDL no desenvolvimento da aterosclerose torna a mesma uma das inúmeras doenças relacionadas ao estresse oxidativo (Dhalla et al., 2000). Nesse contexto, muitas pesquisas demonstram o papel benéfico do consumo de antioxidantes no combate à aterogênese, principalmente antioxidantes oriundos de produtos naturais como frutas e vegetais (Cherubini et al., 2005; Kardinaal et al., 1993; Rimm et al., 1993; Smith et al., 1989).

Dentre os fatores de risco para o desenvolvimento da aterosclerose, destaca-se o Diabetes mellitus, uma doença que acomete um número grande de pessoas no mundo e tem como evento chave um quadro de hiperglicemia crônica (D'Agostino et al., 2004). Segundo Lyons et al. (1986) o risco para o desenvolvimento de aterosclerose é aumentado aproximadamente três vezes em pacientes diabéticos. Além disso, diversos trabalhos vêm demonstrando o papel da hiperglicemia como

precursora de eventos oxidativos em diferentes tecidos, dentre eles a peroxidação lipídica, a qual contribui para a formação da ox-LDL (Jain et al., 1989).

Além da oxidação da LDL, outro tipo de reação extremamente importante no que diz respeito ao processo aterogênico é a glicação não-enzimática da lipoproteína, favorecida em condições hiperglicêmicas pela alta disponibilidade de glicose. Nas reações de glicação não-enzimática, os grupos carbonila de carboidratos reagem com os grupamentos amino de proteínas formando inicialmente bases de *Schiff*. Estas, por sua vez, sofrem rearranjos espontâneos formando os chamados produtos de Amadori, sendo esses denominados os produtos iniciais da glicação (Yeboah et al., 1999). Posteriormente, os produtos de Amadori podem ser degradados e formar compostos α -dicarbonílicos tais como o glicoxal, o metilglicoxal e o 3-deoxiglicosona, cujos grupamentos carbonila são ainda mais reativos frente aos grupos amino de proteínas e como resultado de tal interação, ocorre a formação dos produtos finais de glicação avançada (AGEs) (Onorato et al., 1998) (Figura 4).

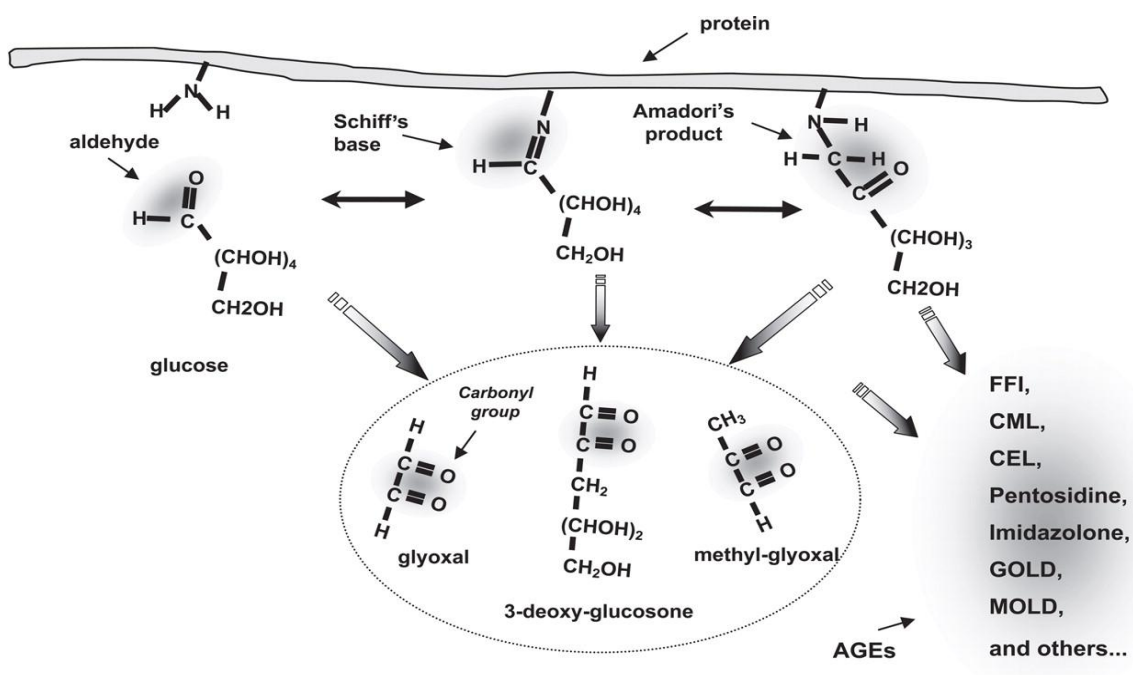


Figura 4: Possíveis vias para formação de AGEs (Basta et al., 2004).

Dentre os compostos α -dicarbonílicos precursores da formação de AGEs, destaca-se o metilglioxal (MG) que pode ainda ser formado em condições hiperglicêmicas através da degradação de trioses fosfato durante a glicólise (O'Brien et al., 2010). A concentração plasmática de MG é aumentada de duas a cinco vezes em pacientes diabéticos (McLellan et al., 1994). A reação de glicação entre os resíduos de arginina da LDL com MG leva à formação do AGE hydroimidazolone N₅-(5-hydro-5-methyl -4-imidazolon-2-yl)-ornithine (MG-H1), o qual é considerado o AGE presente em maior quantidade nos sistemas fisiológicos (Thornalley, 2005) (Figura 5)

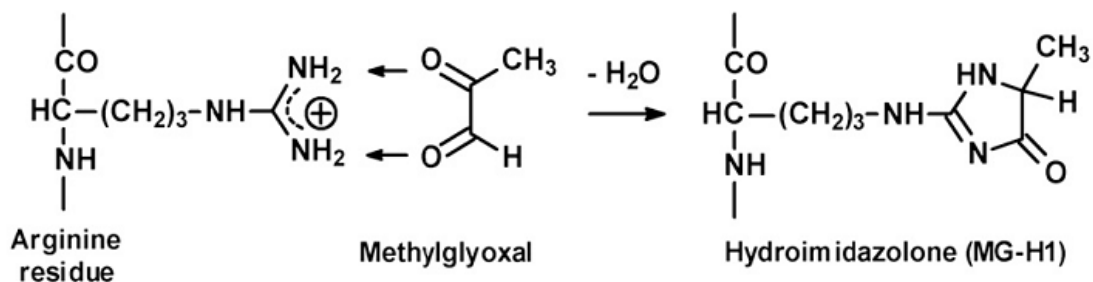


Figura 5: Reação entre resíduos de lisina e metilglioxal formando o AGE MG-H1 (Rabbani, N. et al., 2011).

Pesquisas recentes vêm demonstrando que muitas reações de glicação necessitam de condições pró-oxidantes para ocorrer, sendo por isso, muitas vezes utilizado o termo “glicooxidação” para designar tal processo (Vlassopoulos et al., 2013). Além disso, a LDL glicada (glyLDL) parece ser mais susceptível à oxidação que a a LDL nativa (nLDL) (Hunt et al., 1990; Kawamura et al., 1994). Esses fatos fazem com que muitas vezes seja difícil distinguir potenciais agentes antiglicantes de antioxidantes. Porém, modelos de glicação com compostos carbonílicos como o

metilglicoxal são modelos de glicação não-oxidativa (Baynes, 1991; Cervantes-Laurean et al., 1996). Desta forma, o modelo de glicação induzida pelo MG oferece uma vantagem em relação a modelos onde a glicação é induzida pela glicose, por permitir a verificar a ação de inibidores das reações de glicação que atuam independentemente de sua ação antioxidante.

Considerando o conhecimento obtido sobre o papel oxidante/glicante do processo aterogênico, a busca por substâncias com potencial antioxidante e antiglicante vêm crescendo muito nas últimas décadas. Neste contexto, destacam-se os produtos naturais, por serem normalmente de fácil acesso à população e amplamente usados na medicina tradicional.

A *Syzygium cumini*, conhecida popularmente como “jambolão”, é uma planta pertencente à família Asteraceae que é utilizada na medicina popular brasileira no tratamento da hiperglicemia (Oliveira et al., 2005). Na literatura, diversos trabalhos têm demonstrado a eficácia de diferentes partes e preparações da planta em modelos experimentais relacionados com condições diabetogênicas. Existem evidências que o tratamento com extrato aquoso de sementes e/ou etanólico de folhas da planta reduz os níveis de glicose sanguínea e a hipercolesterolemia em ratos diabéticos (Mahapatra et al., 1985; Vikrant et al., 2001). Além destas, outras propriedades descritas na literatura incluem: a ação antioxidante (Banerjee et al., 2005), antiinflamatória (Chaudhuri et al., 1990), antifúngica (Braga et al., 2007), anti-HIV (Kusumoto et al., 1995), antiulcerogênica (Ramirez et al., 2003), entre outras. No entanto, não há estudos acerca do efeito de nenhuma parte da planta frente à oxidação e/ou glicação de lipoproteínas de baixa densidade e conseqüentemente sobre o seu potencial como modulador do processo aterogênico.

2. OBJETIVOS

2.1 *Objetivo Geral*

O objetivo geral deste trabalho é avaliar o efeito do extrato aquoso de folhas da planta *Syzygium cumini* (S.cExt) sobre parâmetros de oxidação induzida por CuSO₄ na porção lipídica e protéica de LDL, bem como sobre parâmetros de glicação induzida por metilglioxal nas mesmas.

2.2 *Objetivos Específicos*

- Avaliar o efeito do S.cExt sobre a peroxidação lipídica induzida por CuSO₄ em LDL, soro e plasma humanos através da formação de dienos conjugados e produção de TBARS.
- Analisar o efeito do S.cExt sobre o dano oxidativo induzido por CuSO₄ na porção protéica da LDL utilizando a fluorescência do triptofano.
- Verificar a susceptibilidade da LDL isolada de plasma pré-incubado com o S.cExt à oxidação lipídica e protéica.
- Avaliar o efeito do S.cExt sobre a glicação da LDL induzida pelo metilglioxal através da eletroforese em gel de agarose e formação de AGEs fluorescentes.

3. DESENVOLVIMENTO

O desenvolvimento referente a essa dissertação está apresentado sob a forma de um manuscrito. Os itens Materiais e Métodos, Resultados, Discussão e Referências encontram-se no próprio manuscrito.

MANUSCRITO: Efeitos do extrato aquoso de folhas de *Syzygium cumini* sobre a oxidação e glicação de lipoproteínas de baixa densidade *in vitro*

Effects of aqueous-leaf extract of *Syzygium cumini* on oxidation and glycation of low density lipoproteins *in vitro*

Matheus M. dos Santos¹, Alessandro S. Prestes¹, Rômulo P. Barcelos¹, Aline A. Boligon², João B.T. da Rocha¹, Nilda B.V. Barbosa^{1*}

¹ Departamento de Bioquímica e Biologia Molecular, Centro de Ciências Naturais e Exatas, Universidade Federal de Santa Maria, Santa Maria, RS, Brazil.

² Departamento de Farmácia Industrial, Universidade Federal de Santa Maria, Santa Maria, RS, Brazil

*Corresponding author: Dr^a Nilda B. de Vargas Barbosa, Centro de Ciências Naturais e Exatas, Programa de Pós-Graduação em Bioquímica Toxicológica, 97105-900 Santa Maria - RS – Brazil. Tel.: 55-55-3220-8140 Fax: 55-55-3220-8978, e-mail: nvbarbosa@yahoo.com.br

Abstract

Structural modifications elicited by oxidation and glycation of low density lipoproteins (LDL) are considered crucial in the atherosclerosis development. Glycated LDL (glyLDL) levels are elevated in diabetic patients and it is known that diabetes mellitus is a risk factor to atherosclerosis. In view of the importance of LDL glycation and oxidation in the pathogenesis of atherosclerosis, the antioxidant and anti-glycant effect of many synthetic and natural products have been studied. In this work we evaluated the effects of *Syzygium cumini* aqueous-leaf extract (S.cExt) on oxidation and glycation of human LDL *in vitro*. Our results show that the S.cExt was effective in blunting LDL, plasma and serum oxidation evaluated by different parameters, namely: conjugated dienes formation, thiobarbituric acid reactive substances (TBARS) levels and tryptophan (Trp) fluorescence. On the other hand, the data obtained in the assays evaluating glycation reactions suggest that the S.cExt does not exhibit anti-glycant effect. These findings indicate that S.cExt has potential as antiatherogenic agent.

Key words: LDL, atherosclerosis, oxidation, glycation, oxidative stress, *Syzygium cumini*

1. Introduction

Cardiovascular diseases (CVD) are considered the major cause of death in the world (WHO, 2014). The atherosclerosis, a chronic inflammatory process characterized by formation of atherosclerotic plaque in large and medium size arteries, has been pointed as the underlying cause of coronary heart disease and most strokes (Steinberg and Witztum, 2002). Several studies have suggested that modifications in the structure of low density lipoprotein (LDL) are crucial in the atherosclerosis development (Stocker and Keane, 2004; Nishi et al., 2002; Matsuura et al., 2006), especially the oxidative modifications of LDL to form oxidized LDL (oxLDL) (Chisolm and Steinberg, 2000). OxLDL acts as an atherogenic particle by inducing endothelial cell activation, secretion of inflammatory mediators and expression of adhesion molecules; events that promote the recruitment of monocytes and their differentiation into macrophages in the intima of the arteries (Witztum and Steinberg, 1991). It is well known that macrophages are able to cause oxidation of LDL via oxidant enzyme systems, such as myeloperoxidase, NADPH oxidase and nitric oxide synthase (NOS) (Parhami et al., 1993). The total oxidation of LDL culminates with an increase in the apoB-100 negative charge, which promotes the recognition of oxLDL by scavenger receptors on macrophages and its internalization to form the foam cells, an important hallmark of the atherosclerotic lesion (Enriksen et al., 1981).

A growing body of evidence from experimental and clinical studies shows a close association between atherosclerosis and Diabetes Mellitus (DM). In fact, there is evidence that the risk for atherosclerosis manifestations is increased by approximately three-fold in diabetic patients (Lyons et al., 1986). DM is a metabolic disorder characterized by chronic hyperglycemia, resulting from defects in insulin

secretion, insulin action or both (WHO, 1999). Another important type of LDL modification, which is favored under hyperglycemic condition, is the LDL glycation with consequent formation of glycated LDL (glyLDL) (Veiraiah, 2005). Protein glycation is a complex series of Mailard reactions where glucose or others reactive carbonyl compounds such glyoxal (GO) and methylglyoxal (MG) react with protein amino groups to form initially reversible Schiff bases (Thornalley et al., 2001). Over time, the Schiff bases are able to form irreversible Amadori products and in the last stage can be formed the advanced glycation end products (AGEs), which trigger pathological events as oxidative stress and pro-inflammatory responses (Lee et al., 2009).

MG can be formed via oxidation of reducing sugars under hyperglycemic conditions and through triose phosphate intermediates in glycolysis (O'Brien et al., 2010). The plasma concentration of MG is increased two- to five fold in diabetic patients (McLellan et al., 1994). Glycation of LDL by MG is directed to the arginine residues present in apoB-100 and culminates with the formation of the hydroimidazolone N^ε-(5-hydro-5-methyl -4-imidazolone-2-yl)-ornithine residues (MG-H1) which are considered the major AGE in physiological systems (Thornalley, 2005).

LDL glycation and oxidation are closely linked and glycated LDL has been shown to be more sensitive to oxidation than native LDL (nLDL) (Hunt et al., 1990; Kawamura et al., 1994). However, MG-mediated protein glycation is considered a nonoxidative model (Pashikanti et al., 2010).

Considering the important role played by oxidative stress in the atherosclerosis pathogenesis, numerous *in vitro* and *in vivo* studies have been demonstrated the efficacy of antioxidants therapies (Bocan et al., 1992; Fruebis et al., 1994; Shaish et al., 1995;

Portella et al., 2008; de Bem et al., 2009; Barcelos et al, 2011). This scenario highlights the use of medicinal plants, which are promising sources of potent antioxidants and widely used around the world.

Syzygium cumini, popularly known in Brazil as “jambolão” belongs to the family Asteraceae and the hot infusion of its leaves is popularly used for Brazilian people in the treatment of hyperglycemia (Oliveira et al., 2005). Literature findings show that different parts and preparations of *S. cumini* exhibit antihyperglycemic (Ayyanar et al., 2013), antioxidant (Banerjee et al., 2005), anti-inflammatory (Chaudhuri, 1990), antifungal (Braga et al., 2007), anti-HIV (Kusumoto et al., 1995) and antiulcer properties (Ramirez et al., 2003). However, there is no data about the effects of *S. cumini* on the glycation and/or oxidation LDL. Taken the close relationship established among LDL, atherosclerosis and hyperglycemia, this study was undertaken to evaluate the possible antiatherogenic potential of *S. cumini* aqueous-leaf extract on oxidation and glycation parameters in human LDL.

2. Materials and Methods

2.1 Plant material

S. cumini leaves were collected from botanical garden of Universidade Federal de Santa Maria (UFSM). The leaves were then used to prepare the hot infusion: 30 g of leaves in 100 ml of distilled water at 100°C. Afterwards, the hot infusion was submitted to lyophilization process. The product was stored at –20°C and resuspended in distilled for the use in the different assays.

2.2 HPLC of *S.cumini* Extrac (*S.cExt*)

High performance liquid chromatography (HPLC-DAD) was performed with the HPLC system (Shimadzu, Kyoto, Japan), Prominence Auto Sampler (SIL-20A), equipped with Shimadzu LC-20AT reciprocating pumps connected to the degasser DGU 20A5 with integrator CBM 20A, UV-VIS detector DAD (diode) SPD-M20A and Software LC solution 1.22 SP1. Reverse phase chromatographic analyses were carried out under gradient conditions using C₁₈ column (4.6 mm x 250 mm) packed with 5 µm diameter particles; the mobile phase was water containing 2% acetic acid (A) and methanol (B), and the composition gradient was: 5% of B until 2 min and changed to obtain 25%, 40%, 50%, 60%, 70% and 100% B at 10, 20, 30, 40, 50 and 80 min, respectively, following the method described by [Laghari et al. \(2011\)](#) with slight modifications. The lyophilized aqueous extract of leaves of *Syzygium cumini* was analyzed, dissolved in water at a concentration of 5 mg/mL. The presence of six phenolics compounds was investigated, namely: gallic, chlorogenic and caffeic acids and the flavonoids quercetin, rutin and kaempferol. The identification of these compounds was performed by comparing their retention time and UV absorption spectrum with those of the commercial standards. The flow rate was 0.5 ml/min, injection volume 40 µl and the wavelength were 254 nm for gallic acid, 325 nm for

caffeic and chlorogenic acids, and 365 nm for quercetin, rutin and kaempferol. All the samples and mobile phase were filtered through 0.45 µm membrane filter (Millipore) and then degassed by ultrasonic bath prior to use. Stock solutions of standards references were prepared in the HPLC mobile phase at a concentration range of 0.020 – 0.200 mg/ml for kaempferol, quercetin and rutin; and 0.025 – 0.250 mg/ml for gallic, caffeic and chlorogenic acids. The chromatography peaks were confirmed by comparing its retention time with those of reference standards and by DAD spectra (200 to 400 nm). Calibration curve for gallic acid: $Y = 10237x + 1454.8$ ($r = 0.9998$); caffeic acid: $Y = 11372x + 1366.0$ ($r = 0.9996$); rutin: $Y = 11473x - 1375.7$ ($r = 0.9999$); quercetin: $Y = 12135x - 1392.6$ ($r = 0.9997$) and kaempferol: $Y = 129623x - 1343.9$ ($r = 0.9991$). All chromatography operations were carried out at ambient temperature and in triplicate.

2.3 LDL isolation

LDL isolation was performed by discontinuous density-gradient ultracentrifugation using fresh human plasma as described by [Silva et al. \(1998\)](#), with minor modifications. Plasma of non-fasted healthy normolipidemic voluntary donors was collected and pooled with EDTA (1 mg/ml). To prevent LDL aggregation was added sucrose at final concentration of 0.5 %. Solid KBr (0.326 g/ml) was used to adjust the EDTA-plasma to a density of 1.22 g/ml. Five milliliters of this mixture (EDTA-plasma-KBr) was placed in a centrifuge tube and 5 ml of a sodium chloride solution (density 1.006 g/mL) was overlaid on the top of the mixture. Ultracentrifugation was run at 65,000 rpm for 2 h at 4°C using a Himac CP80MX ultracentrifuge. LDL particles were collected aspirating the yellow/orange band at the center of saline layer. Isolated LDL was submitted to exhaustive dialysis with 10 mM phosphate buffer (pH 7.4) during 24 h at 4°C. After dialysis, isolated LDL was stored at -20°C for less than 2 weeks.

Protein quantification of the isolated LDL was performed by the method described by [Lowry et. al. \(1951\)](#).

2.4 LDL oxidation

2.4.1 Conjugated dienes formation

Isolated LDL (50 µg protein/ml) was incubated at 37°C with different concentrations (0.25-1.5 µg/ml) of S.cExt in a medium containing phosphate buffer (10 mM, pH 7.4). After 5 minutes, the oxidation was initiated by the addition of CuSO₄ (10 µM). In another set of experiments, we verified the effect of S.cExt (1.0 µg/ml) after the initiation of the conjugated dienes formation. For this, S.cExt (1.0 µg/ml) was added at 60 and 70 minutes after CuSO₄ (10 µM) addition. We also evaluated the conjugated dienes formation in LDL isolated from human plasma incubated with S.cExt. Human plasma was incubated without or with 200, 400 or 600 µg/ml of S.cExt during 5 h at 37 °C. Afterwards, LDL was isolated as described in the section 2.2. Conjugated dienes formation was evaluated spectrophotometrically by changes in the absorbance at 234 nm as described previously ([Esterbauer et. al., 1989](#)). Absorbance was measured every 20 minutes.

2.4.2 Lag phase determination and oxidation rate

Several parameters can be obtained from the conjugated dienes formation vs. time profile. One of them is the lag phase which is commonly determined graphically by the intercept of the tangents to the slow and fast increase of the dienes absorption. Another parameter is the maximum oxidation rate, given by the peak of the first derivative, i.e., change of absorbance at 234 nm as a function of time ([Giese and Esterbauer, 1994](#)).

2.4.3 LDL-tryptophan fluorescence

A single band centered at approximately 332 nm is observed in the fluorescence spectra of native LDL. This is attributed to the tryptophan (Trp) residues in ApoB-100 (Giessauf et al., 1995). So, loss of Trp fluorescence is considered a marker of oxidative damage in the protein moiety of LDL. In this study, LDL-Trp fluorescence measurement was performed incubating LDL (50 µg protein/ml) at 37°C in the presence or absence of CuSO₄ (10 µM) and different concentrations of S.cExt (0.25-8.0 µg/ml) in a medium containing phosphate buffer (10 mM, pH 7.4). We also evaluated the loss of Trp fluorescence in LDL isolated from human plasma incubated with S.cExt. Human plasma was incubated without or with 200, 400 or 600 µg/ml of S.cExt during 5 h at 37 °C. After, LDL was isolated as described in the section 2.2. The fluorescence (excitation at 282 nm and emission at 331 nm) was measured every 20 minutes in a Shimadzu spectrofluorometer as previously described (Giessauf et al., 1995; Reyftmann et al., 1990). The results were expressed as the percentage of the decrease in Trp fluorescence. The time required for reaching half Trp fluorescence (t_{1/2}) was also calculated.

2.4.4 TBARS production

Lipid peroxidation levels of LDL were evaluated by thiobarbituric acid reactive substances (TBARS), following the method described by Okhawa et. al (1979) with minor modifications. LDL (50 µg protein/ml) was incubated for 2 h at 37°C in the presence or absence of CuSO₄ (10 µM) and different concentrations of S.cExt (0.25-2.0 µg/ml) in a medium containing phosphate buffer (10 mM, pH 7.4). Afterwards, 300 µl of SDS 8.1 %, 500 µl of phosphoric acid and 500 µl of TBA 0.8 % were added. TBARS formation was also evaluated in the LDL samples isolated from human plasma incubated without or with 200, 400 or 600 µg/ml of S.cExt for 5 h at 37 °C.

The color reaction was performed by the incubation for 1 h at 95°C. TBARS production was measured spectrophotometrically at 532 nm and the results were expressed as percentage of control.

1.5 Serum oxidation

Conjugated dienes formation was used to analyze the serum oxidation. To evaluate the serum oxidation levels, venous blood was drawn from nonfasted healthy normolipidemic voluntary donors into tubes containing no anticoagulant and centrifuged for 10 minutes at 1000 x g. The serum (diluted 100-fold) was incubated in the presence or absence of different concentrations of *S.cExt* (0.25-1.0 µg/ml) and CuSO_4 (30 µM) in a medium containing phosphate buffer (10 mM, pH 7.4). In another set of experiments, we verified the effect of *S.cExt* (1.0 µg/ml) after the initiation of the conjugated dienes formation. For this, *S.cExt* (1.0 µg/ml) was added at 50 and 60 minutes after CuSO_4 (30 µM) addition. Conjugated dienes formation was monitored by changes in the absorbance at 245 nm as previously described ([Schnitzer et. al., 1998](#)).

2.6 Plasma oxidation

TBARS levels were used to evaluate the plasma oxidation. Venous blood was drawn from nonfasted healthy normolipidemic voluntary donors into tubes containing anticoagulant (heparin) and centrifuged for 10 minutes at 1000 x g. Fresh plasma (diluted 1:4) was incubated for 3 h at 37°C in the presence or absence of CuSO_4 (100 µM) and different concentrations of *S.cExt* (4.0-24.0 µg/ml) in a medium containing phosphate buffer (10 mM, pH 7.4). Then, 300 µl of SDS 8.1 %, 500 µl of phosphoric acid and 500 µl of TBA 0.8 % were added. The color reaction was performed by the

incubation for 1 h at 95°C. TBARS production was measured spectrophotometrically at 532 nm and the results were expressed as percentage of control.

2.7 LDL glycation

2.7.1 AGEs formation by LDL glycation

Fluorescent AGEs formation was performed incubating the isolated LDL (1.4 mg protein/ml) and methylglyoxal (10 mM) for 72 h at 37°C in the presence or absence of a medium containing phosphate buffer (10 mM, pH 7.4), 10 mM of aminoguanidine (positive control) and different concentrations of S.cExt (0.1-1.0 mg/ml). Afterwards, the samples were dialyzed for 24 h at 4°C against 2 l of 10 mM phosphate buffer (pH 7.4). The samples were diluted (1:10) in phosphate buffer and the fluorescence ($\lambda_{EX} = 370$ nm, $\lambda_{EM} = 440$ nm) was measured using a Shimadzu spectrofluorometer as previously described ([Pashikanti et al, 2010](#)). The results were expressed as fluorescence units (F.U).

2.7.2 Electrophoresis

Electrophoresis was performed with the same samples used for AGEs measurement (see 2.7.1). The electrophoresis was conducted with agarose gel (0.7 %) for 60 minutes at 100 V and 400 mA ([Younis et. al., 2009](#)).

2.8 Statistical analyzes

Data are expressed as means \pm SD. Statistical analysis was performed using a one-way analysis of variance (ANOVA), followed by Tukey's multiple comparison test when appropriate. Differences were considered significant when $p < 0.05$.

3. Results

3.1 Effects of *S.cExt* on LDL oxidation

3.1.1 Conjugated dienes formation

S.cExt was able to delay the conjugated dienes formation CuSO_4 -induced, causing a marked increase in the lag phase in a concentration dependent manner ($p < 0.05$, Fig. 1A and 1B). Indeed, the addition of *S.cExt*, 60 and 70 min, after the start of conjugated dienes formation blocked the progression of the oxidation elicited by CuSO_4 (Fig. 1C). *S.cExt* also reduced the maximum oxidation rate, at all tested concentrations, when compared to CuSO_4 ($p < 0.05$, Table 1). In the experiments performed with LDL isolated from plasma incubated with *S.cExt*, the lag phase was significantly increased by *S.cExt* (400 and 600 $\mu\text{g/ml}$) when compared to LDL exposed to CuSO_4 alone ($p < 0.05$, Fig. 4A and 4B).

3.1.2 TBARS production

The levels of lipid peroxidation in isolated LDL, evaluated by TBARS production, were significantly increased by CuSO_4 . *S.cExt*, at concentrations of 0.5 to 2.0 $\mu\text{g/ml}$, was able to blunt this effect induced by CuSO_4 ($p < 0.05$, Fig. 2). Similarly, TBARS content was decreased in the LDL isolated from plasma incubated with 400 and 600 $\mu\text{g/ml}$ of *S.cExt* when compared to the values found in LDL exposed to CuSO_4 alone ($p < 0.05$, Fig. 4C).

3.1.3 LDL-Trp Fluorescence

The data of Fig. 3 show that *S.cExt* was effective in reducing the loss of LDL-Trp fluorescence CuSO_4 -induced. *S.cExt* increased significantly the required time for reaching half Trp fluorescence ($t_{1/2}$) in concentrations up to 0.75 $\mu\text{g/ml}$ and for the

concentrations 2 to 8 $\mu\text{g/ml}$ the Trp $t_{1/2}$ obtained was higher than the time period of the experiment (180 min) ($p < 0.05$). Similar to previous results obtained with conjugated dienes formation and TBARS production, the incubation of human plasma with 400 and 600 $\mu\text{g/ml}$ of S.cExt prevented partially the loss of LDL-Trp fluorescence and the Trp $t_{1/2}$ was significantly higher than LDL incubated with CuSO_4 alone ($p < 0.05$, Fig. 5A and 5B).

3.2 Serum Oxidation

In this set of experiment, the S.cExt was also able to delay the conjugated dienes formation induced by CuSO_4 , causing a significant increase in the lag phase in a concentration dependent manner ($p < 0.05$, Fig. 6A and 6B). However, when S.cExt was added after the start of conjugated dienes formation, it was not effective in blocking the progression of the oxidation (Fig. 6C). Here, S.cExt also caused a significant reduction in the maximum oxidation rate when compared to CuSO_4 ($p < 0.05$, Table 2). This effect was verified in all concentrations tested.

3.3 Plasma Oxidation

The levels of human plasma oxidation CuSO_4 -induced were evaluated by TBARS production measurement. In this parameter was possible observe that all tested concentrations of S.cExt were able to blunt the plasma oxidation induced by CuSO_4 ($p < 0.05$, Fig. 7).

3.4 Effect of *S.cExt* on LDL glycation

3.4.1 AGEs formation

As expected, the incubation of LDL with MG (10 mM) for 72 h caused an increase in fluorescent AGEs formation, which was completely blocked by AG, used as an anti-glycant control. On the other hand, the *S.cExt* was not able to reduce the AGEs levels induced by LDL glycation ($p < 0.05$, Fig. 8)

3.4.2 Electrophoresis

Incubation of LDL with MG (10 mM) for 72 h culminated with an increase in LDL electroforetic mobility, which was represented by higher distance traveled into agarose gel when compared to the control. This increased dislocation of LDL was prevented by AG treatment. Different, no effect of *S.cExt* (0.1-1.0 mg/ml) was observed in this parameter (Fig. 9).

3.5 *S.cExt* composition by HPLC

HPLC fingerprinting of *Syzygium cumini* lyophilized aqueous extract revealed the presence of the gallic acid ($t_R = 14.72$ min; 3.46%; peak 1), chlorogenic acid ($t_R = 25.81$ min; 2.09%; peak 2), caffeic acid ($t_R = 34.27$ min; 1.57%; peak 3), rutin ($t_R = 39.64$ min; 4.95%; peak 4), quercetin ($t_R = 54.43$ min; 3.37%; peak 5) and kaempferol ($t_R = 47.28$ min; 0.62%; peak 6) (Fig. 10).

4. Discussion

Several lines of evidence suggest that modifications in the structure of low density lipoproteins implicate in an increase risk to development of atherosclerosis (Witztum, 1994). LDL oxidation which culminates in the formation of oxLDL is a key factor in the atherosclerosis development principally via stimulation of macrophage foam cell formation (Ross, 1999). Elevated levels of copper in plasma have been associated with existing atherosclerosis and it represents a risk factor to the development of disease (Ferns et al., 1997). Thus, the use of copper ions to induce LDL peroxidation *in vitro* is considered a valid method to study the events that occur in atherosclerotic artery wall (Smith et al., 1992). There are several assays to evaluate the extension of LDL peroxidation (Esterbauer et. al., 1992). However, due to its convenience and objectivity, the formation of conjugated dienes (CD) monitored by the changes in absorbance at 234 nm is the most used method to monitor LDL oxidation *in vitro*. After the initiation of lipid peroxidation, there is a rearrangement in the double bonds of the lipids which leads to the formation of CD. In view of this, CD formation is considered an early event in the chain reaction (Kappus, 1985). In this study, we demonstrated that the S.cExt delayed the LDL conjugated dienes in a concentration dependent manner (Fig. 1A). This effect is clearly observed analyzing the increase in the lag phase (Fig 1B). The lag phase determination represents the resistance of LDL to oxidation expressed by the time of the consumption of the antioxidants. It is determined graphically by the intercept of the tangents to the slow and fast increase of the diene absorption. This parameter is widely used to test antiathrogenic compounds (Witztum, 1994; Giesege and Esterbauer, 1994). We also investigated the effect of S.cExt after the start of CD formation. The results expressed in Fig 1C demonstrate the efficacy of S.cExt, at 1 µg/ml, in blocking the

progression of CD formation in LDL at two different times (50 and 60 minutes). Another parameter that can be obtained from dienes absorbance vs. time profile is the oxidation rate which is given by the peak of first derivative, i.e., change of A_{234} as a function of time (Giese and Sterbauer, 1994). In this parameter, our results showed that S.cExt, at all tested concentrations, was able to prevent the reduction of oxidation rate caused by CuSO_4 (Table 1). To better mimic the physiological conditions and to verify a possible direct interaction between S.cExt and LDL, we incubated human plasma with different concentrations of S.cExt during 5 h at 37 °C and after LDL was isolated and submitted to oxidation. Herein, we showed that the exposure of plasma to S.cExt (400 and 600 $\mu\text{g/ml}$) caused a significant increase in lag phase of oxidized LDL (Fig. 4A and 4B).

Since CD formation is an early event in lipid oxidation reactions, TBARS production can be used as a tool to verify the levels of malondialdehyde (MDA), an end product of the chain reaction (Okhawa et al., 1979). In this study, S.cExt exhibited a potent effect in inhibiting MDA formation in LDL. Similarly to the effect on CD formation, LDL isolated from plasma incubated with 400 and 600 $\mu\text{g/ml}$ of S.cExt was more resistant to MDA formation CuSO_4 -induced.

Beside lipid oxidation in isolated LDL, we also evaluated the effect of S.cExt on the oxidation of LDL protein moiety by the loss of Trp fluorescence. It has been reported that the fluorescence spectrum of native LDL displays a single band centered at approximately 332 nm which is assigned to the Trp residues present in apoB-100. So, the loss of Trp fluorescence is considered a marker of oxidative damage in the protein core of LDL (Reyftmann et al., 1990). Here, S.cExt exhibited a potent effect in preventing the loss of Trp fluorescence causing a considerable increase in the required time for reaching half Trp fluorescence ($t_{1/2}$). In the end of the

experimental period (180 minutes), S.cExt at 8 µg/ml maintained about 95% of Trp fluorescence (Fig. 3). Moreover, LDL isolated from plasma incubated 5 h with S.cExt (400 and 600 µg/ml) was more resistant to the loss of Trp fluorescence than native LDL (Fig. 5). These observations demonstrate that S.cExt can protect both lipid and protein core of LDL from CuSO₄-induced oxidation.

In the literature, there are controversial studies about the use of antioxidants as an alternative therapy for atherogenesis (Mashima et al., 2001). This conflict may contribute for our limited understanding about LDL oxidation in atherogenic process and about how antioxidants could act in preventing CVD. However, several findings demonstrate that the use of natural antioxidants and herbal medicine can inhibit the formation of oxLDL and consequently decrease the development and/or progression of atherosclerosis, attributing the effects to its phytochemical composition (Ahmadvand et al., 2012; Jeong et al., 2008; Lee et al., 2006). Among the most studied phytochemical constituents, polyphenolic compounds have demonstrated to possess various pharmacological properties, including antioxidant, anti-inflammatory, anticarcinogenic and antiatherogenic activities (Graziano et al., 1992; Surh, 2003). Indeed, *in vitro* studies have demonstrated that polyphenolic compounds from different plants can inhibit LDL oxidation (Teissedre et al., 1996; Waterhouse et al., 1996; Yoshida et al., 1999). Here, HPLC analyses showed the presence of six polyphenolic compounds in S.cExt, namely: Gallic acid, chlorogenic acid, caffeic acid, rutin, quercetin and kaempferol (Fig. 10). It is plausible assume that the potent effect of S.cExt in inhibiting LDL oxidation could be attributed, at last in part, to its polyphenolic composition. However, more studies are needed to verify if the effect is offered by a specific compound or by a synergic effect among them. In this way, Wu et al. (2010) demonstrated that the polyphenol epigallocatechin gallate (EGCG) binds

to LDL and protect it from oxidation and glycation in samples isolated from plasma incubated during 3 h with EGCG. Here we showed that LDL isolated from plasma incubated during 5 h with *S.cExt* (400 and 600 µg/ml) was more resistant to CuSO₄-induced lipid and protein oxidation than native LDL (Fig. 4 and 5). We suggest that probably one or more polyphenolic compound of *S.cExt* can bind to LDL, similarly to EGCG, and confer the protection against the oxidation.

LDL *in vivo* is present in the interstitial space of the artery wall, a place where there is various antioxidants from plasma/serum (Schwarz et al., 1991; Witztum and Steinberg, 1991). In view of this, Spranger et al. (1998) suggested that LDL oxidation *in vivo* can be more adequately tested *in vitro* using whole plasma/serum. In this work, we evaluated the effect of *S.cExt* on conjugated dienes formation in human serum and TBARS production in human plasma. Similarly to isolated LDL, *S.cExt* caused a significant delay in CD CuSO₄-induced increasing the lag phase in a concentration dependent manner (Fig. 6A and 6B). On the other hand, no effect was observed when *S.cExt* was added after the initiation of CD formation (Fig. 6C). It is plausible assume that the endogenous antioxidants have been responsible for the reduction in CD content. In human plasma, similarly to isolated LDL, *S.cExt* caused a significant protection against TBARS production. The higher concentration of *S.cExt* was able to maintain lipid peroxidation at basal levels (Fig. 7).

Previous studies have been reported that hyperglycemia can accelerate LDL oxidation and glycation with consequent AGEs formation (Wu et al., 2009). Similarly to oxLDL, the surface charge of glyLDL is also more negative and apoB-100 conformation is altered, impairing its recognition by the classical LDL receptor (Jenkins et al., 2004). The negative residual charges are responsible to increase the LDL eletrophoretic mobility. In accordance, the Fig. 8 clearly shows the increase in

electrophoretic mobility of LDL MG-glycated in comparison with native LDL. In contrast to oxidative parameters, S.cExt did not alter the glycation of LDL induced by MG. LDL samples incubated with AG, a classical inhibitor of glycation process, were maintained in a position similar to native LDL in agarose gel. Additionally, we evaluated the AGEs formation based in the fact that the fluorescence at 370 nm of excitation and 440 nm of emission is a general measure to detect the presence of AGEs (Pashikanti, S. et al., 2010). Corroborating with the results obtained in agarose gel electrophoresis, S.cExt was not effective in blunting MG-induced AGEs formation (Fig. 8).

Protein glycation mediated by glucose can occur in both oxidative and nonoxidative conditions (Fu et al., 1994). Because of this, many potential antiglycating candidates are difficult to distinguish from general antioxidants. In contrast, reactive carbonyl compounds like MG mediate nonoxidative glycation reactions (Baynes, 1991; Cervantes-Laurean et al., 1996). Thus, this model offers an important advantage on glucose-induced glycation because it allows us to identify pure glycation inhibitors, which can act independent of its antioxidant potential. This fact may explain the reason by which S.cExt did not show any effect on MG-induced LDL glycation.

In conclusion, the findings obtained in this work showed that S.cExt possesses a strong antioxidant capacity in inhibiting human LDL, serum and plasma oxidation induced by CuSO₄. S.cExt was also able to maintain the integrity of LDL apoB-100. LDL isolated from plasma incubated with S.cExt was more resistant to lipid and protein oxidation suggesting a possible binding of one or more S.cExt constituents with LDL. On the other hand, S.cExt was not effective in protecting LDL from glycation induced by MG. Although this result indicates that S.cExt did not exhibit anti-glycating

effect, further studies are needed to verify if the antioxidant capacity of S.cExt could inhibit LDL glycation in an oxidative model. Taken together, our data suggest that S.cExt may be considered a promising antiatherogenic agent.

5. References

- Ahmadvand, H. et al. Effects of olive leaves extract on LDL oxidation induced-CuSO₄ in vitro. **Pak J Pharm Sci** 25:571-575, 2012.
- Ayyanar, M. et al. Syzygium cumini (L.) Skeels., a novel therapeutic agent for diabetes: Folk medicinal and pharmacological evidences. **Comp Ther M** 21:232-243, 2013.
- Banerjee, A. et al. In vitro study of antioxidant activity of Syzygium cumini fruit. **Food Chem** 90:727-733, 2005.
- Barcelos, R.P. et al. Thiosemicarbazone derivatives protects from AAPH and Cu²⁺ - induced LDL oxidation. **Life Sci** 89:20-28, 2011.
- Baynes, J.W. Role of oxidative stress in development of complications in diabetes. **Diabetes** 40: 405-412, 1991.
- Bocan, T.M. et al. Antiatherosclerotic effects of antioxidants are lesion-specific when evaluated in hypercholesterolemic New Zealand white rabbits. *Exp Mol Pathol* 57:70-83, 1992.
- Braga, F.G. et al. Antileishmanial and antifungal activity of plants used in traditional medicine in Brazil. **J Ethnopharmacol** 111:396-402, 2007.
- Cervantes-Laurean, D. et al. Glycation and glycooxidation of histones by ADP-ribose. **J Biol Chem** 271:10461-10469; 1996.
- Chaudhuri, A.K.N. Antiinflammatory and related actions of Syzygium cumini seed extract. **Phytother R** 4:5-10, 1990.
- Chisolm, G.M and Steinberg, D. The oxidative modification hypothesis of atherogenesis: an overview. *Free Radical Bio Med* 28:1815-1826, 2000.

de Bem, A.F. et al. Diphenyl diselenide, a simple glutathione peroxidase mimetic, inhibits human LDL oxidation *in vitro*. *Atherosclerosis* 201:92-100, 2008.

Esterbauer, H. et al. The role of lipid peroxidation and antioxidants in oxidative modifications of LDL. **Free Radic Biol Med** 13:241-290, 1992.

Esterbauer, H. et. al. Continuous monitoring of in vitro oxidation of human low density lipoprotein. **Free Radical Res Com** 6:67-75, 1989.

Ferns, G.A.A. et al. The possible role of copper ions in atherogenesis: the Blue Janus. **Atherosclerosis** 133:139-52, 1997.

Fruebis, J. et al. A comparison of antiatherogenic effects of probucol and of a structural analogue of probucol in low density lipoprotein receptor-deficient rabbits. **J Clin Invest** 94:392-398, 1994.

Fu, M.X. Et al. Glycation, glycooxidation, and cross-linking of collagen by glucose: kinetics, mechanisms, and inhibition of late stages of the Maillard reaction. **Diabetes** 43:676-683, 1994.

Gaziano, J. M. et al. Dietary antioxidants and cardiovascular disease. **Ann N Y Acad Sci** 669:249-259, 1992.

Giese, S.P and Esterbauer, H. Low density lipoprotein is saturable by pro-oxidant copper. **FEBS Lett** 343:188-194, 1994.

Giessauf, A. et al. Early destruction of tryptophan residues of apolipoprotein B is a vitamin E- independent process during copper-mediated oxidation of LDL. **Biochim Biophys Acta** 1256:221-232, 1995.

Hunt, J.V. et al. Auto-oxidative glycosylation and possible involvement of peroxides and free radicals in LDL modification by glucose. **Diabetes** 39:1420-1424, 1990.

Jenkins, A.J. et al. Lipoproteins, glycooxidation and diabetic angiopathy. **Diabetes-Metab Res** 20:349-368, 2004.

- Jeong, T.S. et al. Low density lipoprotein (LDL)-antioxidant flavonoids from roots of *Sophora flavescens*. **Biol Pharm Bull** 31:2097-2102, 2008.
- Kappus, H. Lipid peroxidation: mechanisms, analysis, enzymology and biological relevance. In: Sies, H., ed. Oxidative stress. **London: Academic Press** pp. 273-310, 1985.
- Kawamura, M. et al. Pathophysiological concentrations of glucose promote oxidative modification of low density lipoprotein by a superoxide-dependent pathway. **J Clin Invest** 94:771-778, 1994.
- Kusumoto, I.T. et al. Screening of various plant extracts used in ayurvedic medicine for inhibitory effects on human immunodeficiency virus type I (HIV-I) protease. **Phytother R** 12:488-493, 1995.
- Laghari, A.H. et al. Determination of free phenolic acids and antioxidant activity of methanolic extracts obtained from fruits and leaves of *Chenopodium album*. **Food Chem** 126:1850-1855, 2011.
- Lee, J. H. et al. LDL-antioxidant pterocarpanes from roots of *Glycine max* (L.) Merr. **J Agric Food Chem** 54:2057-2063, 2006.
- Lee, O. et al. Fructose and carbonyl metabolites as endogenous toxins. **Chem Biol Interact** 178:332-339, 2009.
- Lowry, O.H. et al. Protein measurement with the Folin phenol reagent. **J Biol Chem** 193:265-275, 1951.
- Lyons, T.J. et al. Glycosylation of low-density lipoprotein in patients with type I (insulin dependent) diabetes: correlations with other parameters of glycemic control. **Diabetologia** 29:685-691, 1986.
- Mashima, R. et al. Oxidants and antioxidants in atherosclerosis. **Curr Opin Lipidol** 12:411-418, 2001.

Matsuura, E. et al. Oxidative modification of low-density lipoprotein and immune regulation of atherosclerosis. **Prog Lipid Res** 45:466-486, 2006.

McLellan, A.C. et al. Glyoxalase system in clinical diabetes mellitus and correlation with diabetic complications. **Clin Sci** 87:21-29, 1994.

Nishi, K. et al. Oxidized LDL in carotid plaques and plasma associates with plaque instability. **Atheroscler Thromb Vasc Biol** 22:1649-1654, 2002.

O'Brien, P.J. et al. Fructose-derived endogenous toxins, in: O'Brien, P.J. and Bruce, W.R. (Eds.). *Endogenous Toxins: Targets for Disease Treatment and Prevention*. **Wiley-VCH Verlag GmbH & Co. KGaA**, Weinheim, pp. 173-212, 2010.

Okhawa, H. et al. Assay for lipid peroxides in animal tissue by thiobarbituric acid reaction. **Anal Biochem** 95:351-358, 1979.

Oliveira, A.C.P. et al. Effect of the extracts and fractions of *Baccharis trimera* and *Syzygium cumini* on glycaemia of diabetic and non-diabetic mice. **J Ethnopharmacol** 102:465-469, 2005.

Parhami, F. et al. Minimally modified low density lipoprotein-induced inflammatory responses in endothelial cells are mediated by cyclic adenosine monophosphate. **J Clin Invest** 92:471-478, 1993.

Pashikanti, S. et al. Rutin metabolites: Novel inhibitors of nonoxidative advanced glycation end products. **Free Radic Biol Med** 48:656-663, 2010.

Portella, R.L. et al. Oximes as inhibitors of low density lipoprotein oxidation. **Life Sci** 83:878-885, 2008.

Ramirez, R.O. et al. The gastroprotective effect of tannins extracted from duhat (*Syzygium cumini* Skeels) bark on HCl/ethanol induced gastric mucosal injury in Sprague-Dawley rats. **CI Hemorrh M** 29:253-261, 2003.

Reyftmann, J.P. et al. Sensitivity of tryptophan and related compounds to oxidation induced by lipid autooxidation. Application to human serum low- and - high density lipoproteins. **Biochim Biophys Acta** 1042:159-167, 1990.

Ross, R. Atherosclerosis – An inflammatory disease. **N Engl J Med** 340:115-126, 1999.

Schnitzer, E. et al. Lipid oxidation in unfractionated serum and plasma. **Chem Phys Lipids** 92:151-170, 1998.

Schwartz, C. et al. The pathogenesis of atherosclerosis: an overview. **Clin Cardiol** 14:11-16, 1991.

Shaish, A. et al. Beta-carotene inhibits atherosclerosis in hypercholesterolemic rabbits. **J Clin Invest** 96:2075-2082, 1995.

Silva, L.E. et al. Inhibition of mammalian 15-lipoxygenase-dependent lipid peroxidation in low-density lipoprotein by quercetin and quercetin monoglucosides. **Arch Biochem Biophys** 349:313-320, 1998.

Smith, C. et al. Stimulation of lipid per-oxidation and hydroxyl-radical generation by the contents of human atherosclerotic lesions. **Biochem J** 286:901-905, 1992.

Spranger, T. et al. How different constituents of human plasma and low density lipoprotein determine plasma oxidizability by copper. **Chem Phys Lipids** 91:39-52, 1998.

Steinberg, D. and Witztum, J.L. Is the oxidative modification hypothesis relevant to human atherosclerosis? Do the antioxidant trials conducted to date refute the hypothesis? **Circulation** 105:2107-2111, 2002

Stocker, R. and Keane, J.F. Role of oxidative modifications in atherosclerosis. **Physiol Rev** 84:1381-1478, 2004.

Surh, Y. J. Cancer chemoprevention with dietary phytochemicals. **Nat Rev Cancer** 3:768-780, 2003.

Teissedre, P.I. et al. Inhibition of in vitro human LDL oxidation by phenolic antioxidants from grapes and wines. **J Sci Food Agric** 70:55-61, 1996.

Thornalley, P.J. Dicarbonyl intermediates in the Maillard reaction. **Ann N Y Acad Sci** 1043:111-117, 2005.

Thornalley, P.J. et al. Suppression of the accumulation of triosephosphates and increased formation of methylglyoxal in human red blood cells during hyperglycaemia by thiamine in vitro. **Biochem J** 129:543-549, 2001.

Veiraiah, A. Hyperglycemia, lipoprotein glycation, and vascular disease. **Angiology** 56:431-438, 2005.

Waterhouse, A.L. et al. Antioxidants in chocolate. **Lancet** 348:834, 1996.

WHO (World Health Organization). Definition, diagnosis and classification of Diabetes Mellitus and its complications. WHO/NCD/NCS, 1999.

WHO (World Health Organization). The top 10 causes of death. Fact sheet 310, may 2014. <http://www.who.int/mediacentre/factsheets/fs310/en/>

Witztum, J.L. and Steinberg, D. Role of oxidized low density lipoprotein in atherogenesis. **J Clin Invest** 88:1785-1792, 1991.

Witztum, J.L. The oxidation hypothesis of atherosclerosis. **Lancet** 344:793-795, 1994.

Wu, C-H. et al. Epigallocatechin gallate (EGCG) binds to low-density lipoproteins (LDL) and protects them from oxidation and glycation under high-glucose conditions mimicking diabetes. **Food Chem** 121:639-644, 2010.

Wu, C-H. et al. Low-density-lipoprotein (LDL)-bound flavonoids increase the resistance of LDL to oxidation and glycation under pathophysiological concentrations of glucose in vitro. **J Agric Food Chem** 57:5058-5064, 2009.

Yoshida, H., et al. Inhibitory effect of tea flavonoids on the ability of cells to oxidize low density lipoprotein. **Biochem Pharmacol** 58:1695-1703, 1999.

Younis, N. et al. Glycation of LDL in non-diabetic people: Small dense LDL is preferentially glycated both in vivo and in vitro. **Atherosclerosis** 202:162-168, 2009.

Figure 1

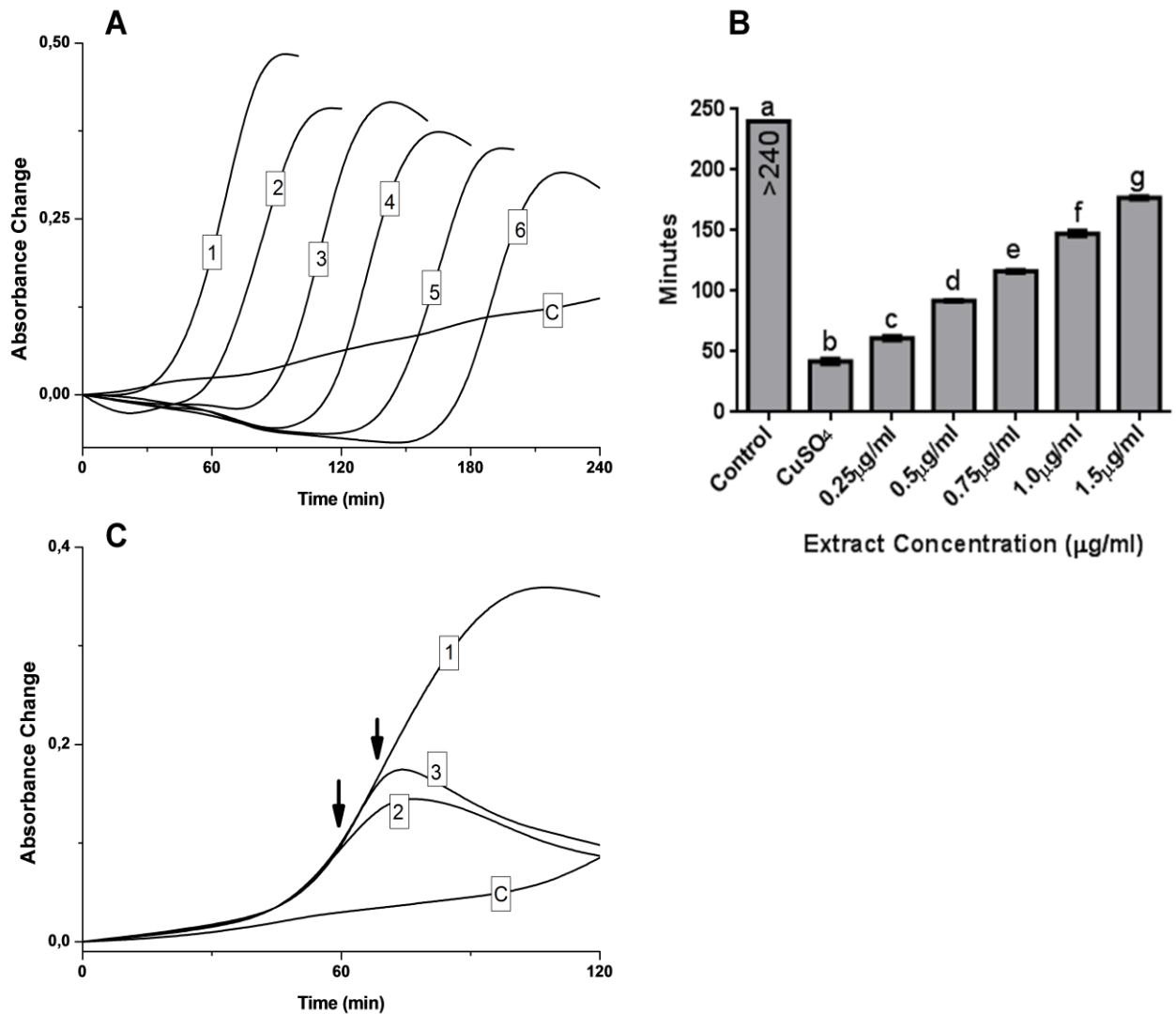


Fig. 1. Effect of S.cExt on conjugated dienes formation in human LDL. **(A)** Representative figure of the dienes formation. The incubation medium without S.cExt (1) or with 0.25 µg/ml (2), 0.50 µg/ml (3), 0.75 µg/ml (4), 1.0 µg/ml (5) or 1.5 µg/ml (6) of S.cExt and CuSO₄. Letter C represents the control without S.cExt and CuSO₄. **(B)** Effect of S.cExt on lag phase values for conjugated dienes formation in isolated human LDL. **(C)** S.cExt added after the initiation of conjugated dienes formation. The incubation medium without S.cExt (1) or with S.cExt 1.0 µg/ml added at 60 minutes (2) and S.cExt 1.0 µg/ml added at 70 minutes (3). Letter C represents the control without S.cExt and CuSO₄. The values are expressed as mean ± SD of four independent experiments in duplicate. Different letters indicate statistically significant differences (p<0.05).

Figure 2

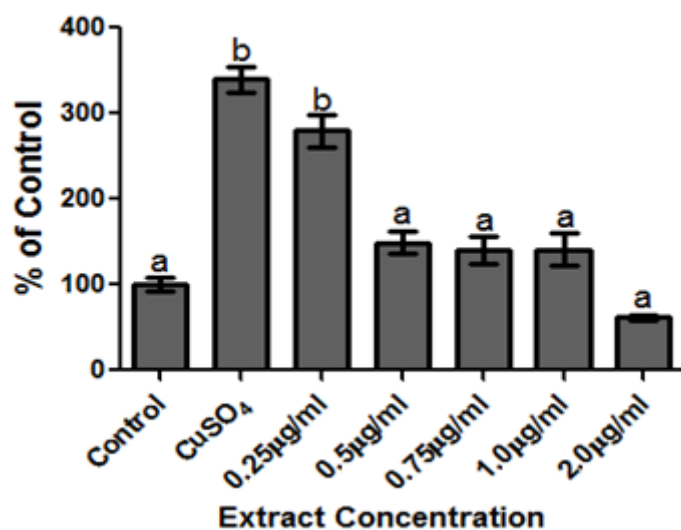


Fig. 2. . Effect of S.cExt concentrations on time-dependent changes in TBARS production during LDL oxidation induced by CuSO₄. LDL (50 µg protein/ml) was incubated for 2 h at 37 °C in the presence or absence of 10 µM CuSO₄ and S.cExt. The values are expressed as mean ± SD of four independent experiments in duplicate. Different letters indicate statistically significant differences (p<0.05).

Figure 3

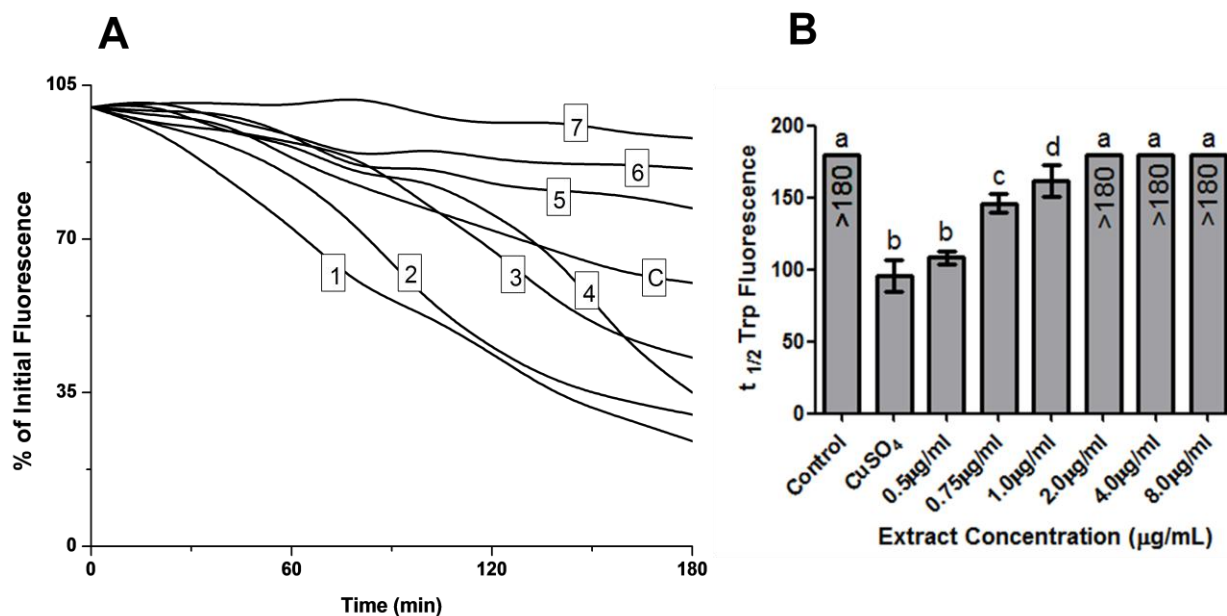


Fig. 3. Effect of S.cExt on Trp fluorescence in human LDL. **(A)** Representative figure of loss in Trp fluorescence. The incubation medium without S.cExt (1) or with 0.50 µg/ml (2), 0.75 µg/ml (3), 1.0 µg/ml (4), 2.0 µg/ml (5), 4.0 µg/ml (6) or 8.0 µg/ml (7) of S.cExt and CuSO₄. Letter C represents the control without S.cExt and CuSO₄. **(B)** Oxidation t_{1/2} values for Trp fluorescence expressed as the time (minutes) required to cause 50% of loss of the LDL tryptophan. The values are expressed as mean ± SD of four independent experiments in duplicate. Different letters indicate statistically significant differences (p<0.05).

Figure 4

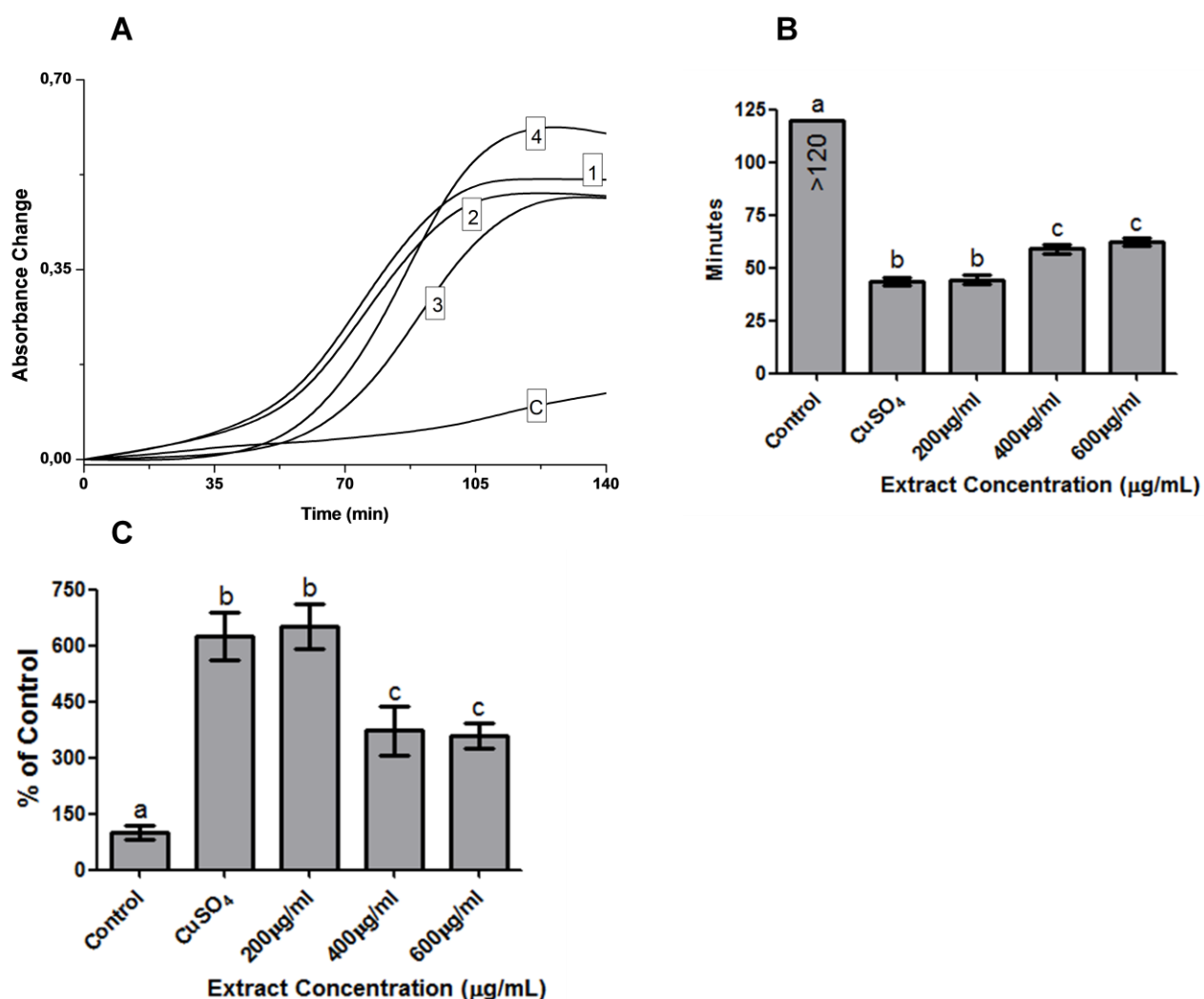


Fig. 4. Oxidative parameters of human LDL isolated from plasma incubated for 5 h with S.cExt . **(A)** Representative figure of the dienes formation. The incubation medium without S.cExt (1) or with 200 µg/ml (2), 400 µg/ml (3), or 600 µg/ml (4) of S.cExt and CuSO₄. Letter C represents the control without S.cExt and CuSO₄. **(B)** Effect of incubation of human plasma with S.cExt on lag phase values for conjugated dienes formation in isolated human LDL. **(C)** TBARS levels of human LDL isolated from plasma incubated with S.cExt. The values are expressed as mean ± SD of four independent experiments in duplicate. Different letters indicate statistically significant differences (p<0.05).

Figure 5

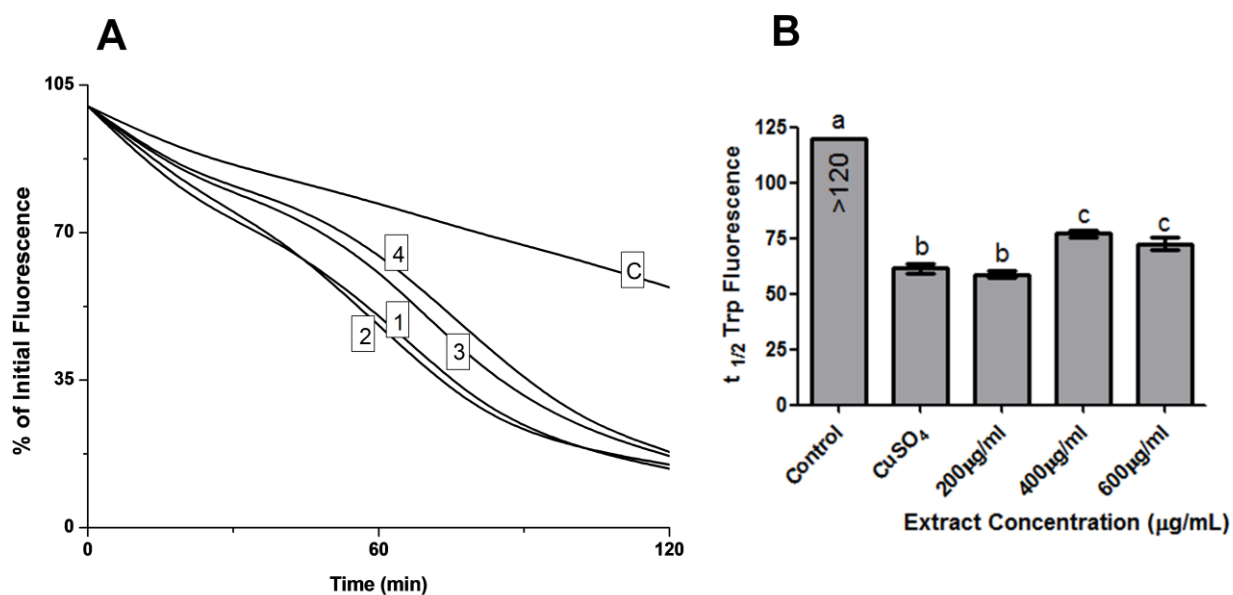


Fig. 5. Loss of Trp fluorescence of human LDL isolated from plasma incubated for 5 h with *S.cExt*. **(A)** Representative figure of Trp fluorescence. The incubation medium without *S.cExt* (1) or with 200 µg/ml (2), 400 µg/ml (3), or 600 µg/ml (4) of *S.cExt* and CuSO₄. Letter C represents the control without *S.cExt* and CuSO₄. **(B)** Oxidation $t_{1/2}$ values for Trp fluorescence expressed as the time (minutes) required to cause 50% of loss of the LDL tryptophan. The values are expressed as mean \pm SD of four independent experiments in duplicate. Different letters indicate statistically significant differences ($p < 0.05$).

Figure 6

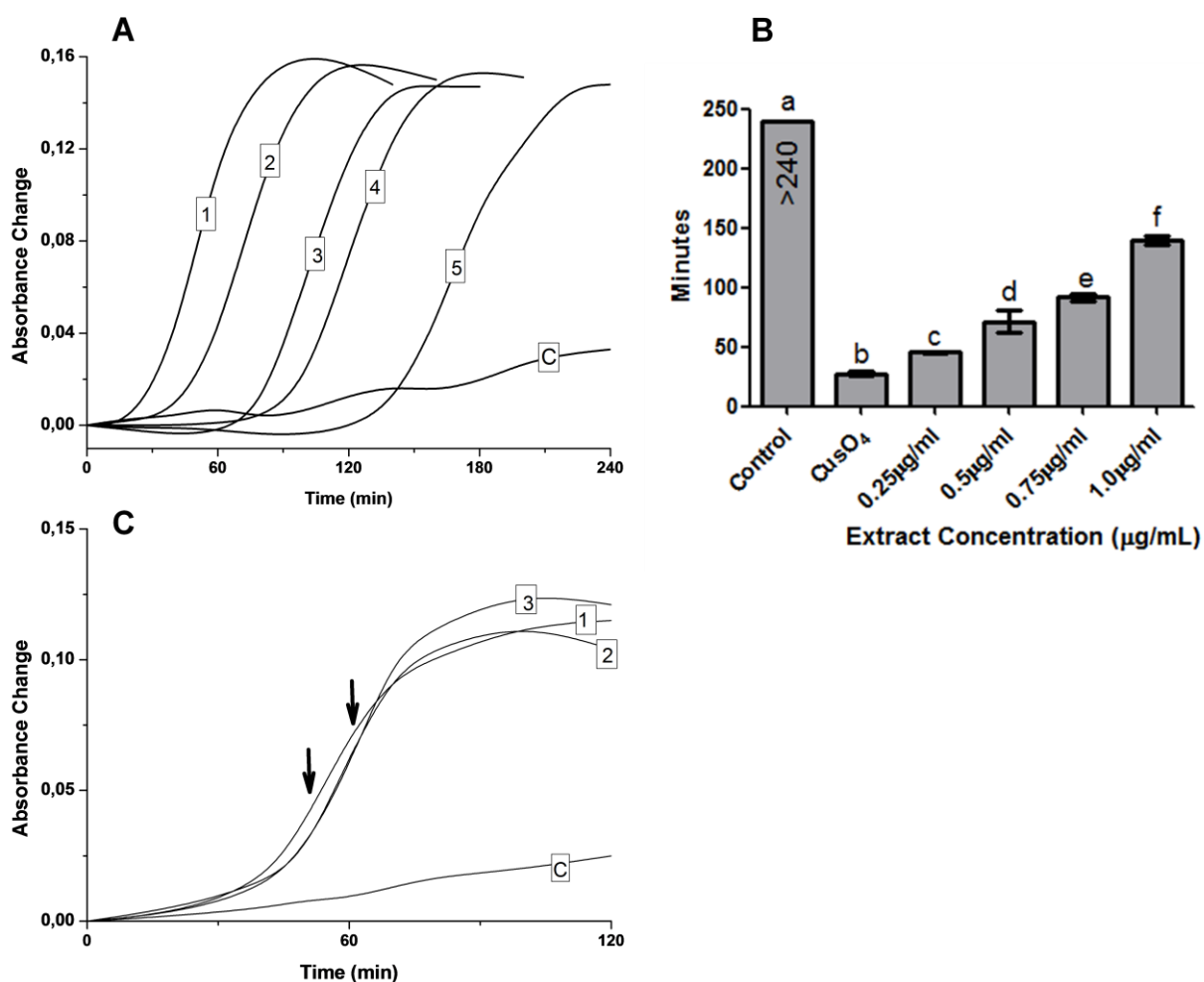


Fig. 6. Effect of S.cExt on conjugated dienes formation in human serum. **(A)** Representative figure of the dienes formation. The incubation medium without S.cExt (1) or with 0.25 µg/ml (2), 0.50 µg/ml (3), 0.75 µg/ml (4) or 1.0 µg/ml (5) of S.cExt and CuSO₄. Letter C represents the control without S.cExt and CuSO₄. **(B)** Effect of S.cExt on lag phase values for conjugated dienes formation in human serum. **(C)** S.cExt added after the initiation of conjugated dienes formation. The incubation medium without S.cExt (1) or with S.cExt 1.0 µg/ml added at 50 minutes (2) and S.cExt 1.0 µg/ml added at 60 minutes (3). Letter C represents the control without S.cExt and CuSO₄. The values are expressed as mean ± SD of four independent experiments in duplicate. Different letters indicate statistically significant differences (p<0.05).

Figure 7

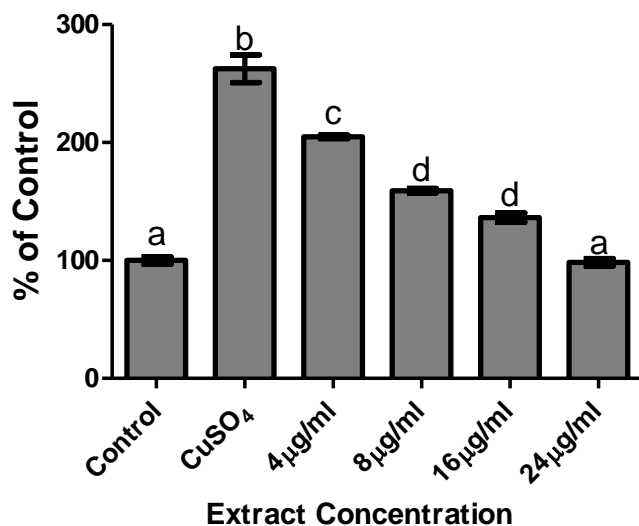


Fig. 7. Effect of *S.cExt* concentrations on time-dependent changes in TBARS production during plasma oxidation induced by CuSO₄. Plasma (diluted 1:4) was incubated for 3 h at 37 °C in the presence or absence of 10 µM CuSO₄ and *S.cExt*. The values are expressed as mean ± SD of four independent experiments in duplicate. Different letters indicate statistically significant differences ($p < 0.05$).

Figure 8

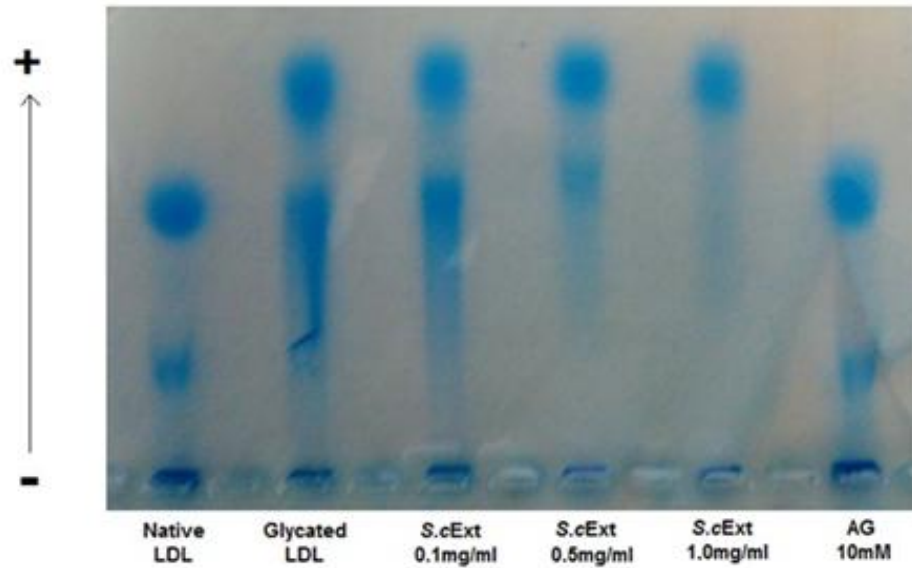


Fig. 8. Effect of S.cExt concentrations on LDL glycation evaluated by agarose gel electrophoresis. LDL (50 μ g protein/ml) was incubated for 48 h at 37 $^{\circ}$ C in the presence or absence of 10 mM of MG and S.cExt or 10 mM aminoguanidine (positive control).

Figure 9

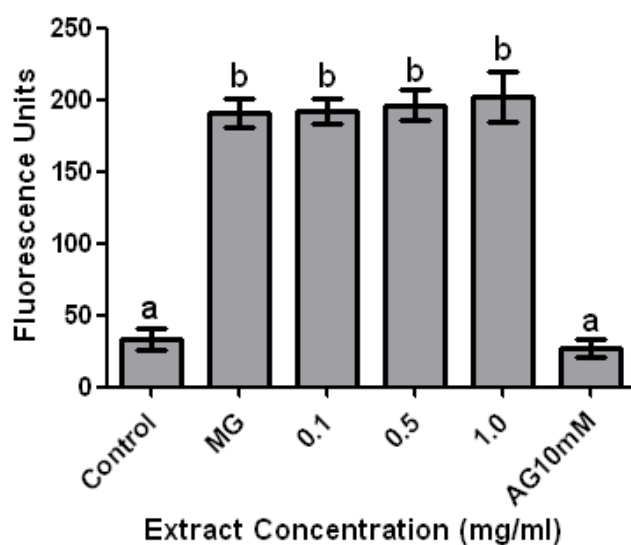


Fig. 9. Effect of S.cExt concentrations on fluorescent AGEs formation. LDL (50 μ g protein/ml) was incubated for 48 h at 37 $^{\circ}$ C in the presence or absence of 10 mM of MG and S.cExt or 10 mM aminoguanidine (positive control). The values are expressed as mean \pm SD of four independent experiments in duplicate. Different letters indicate statistically significant differences ($p < 0.05$)

Figure 10

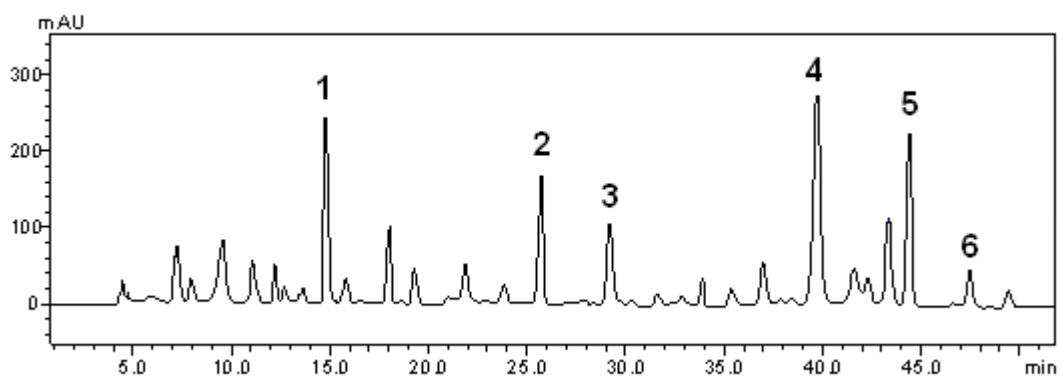


Fig. 10. Representative high performance liquid chromatography profile of *Syzygium cumini*, detection UV was at 325nm. Gallic acid (peak 1), chlorogenic acid (peak 2), caffeic acid (peak 3), rutin (peak 4), quercetin (peak 5) and kaempferol (peak 6). Chromatographic conditions are described in the Methods section.

Table 1. Effect of *S.cExt* on the oxidation rate for conjugated dienes formation in human LDL.

	Oxidation Rate (Δabs/20min)
Control	ND
CuSO₄	0.313 \pm 0.006 a
0.25 μg/ml	0.220 \pm 0.001 b
0.50 μg/ml	0.297 \pm 0.006 c
0.75 μg/ml	0.287 \pm 0.006 d
1.00 μg/ml	0.240 \pm 0.010 e
1.50 μg/ml	0.257 \pm 0.006 f

LDL was oxidized in the presence of CuSO₄ 10 μ M with or without different concentrations of *S.cExt*. The values are expressed as mean \pm S.D. of three independent experiments in duplicate. Different letters indicate statistically significant differences ($p < 0.05$). ND: not determined.

Table 2. Effect of *S.cExt* on the oxidation rate for conjugated dienes formation in human serum.

	Oxidation Rate (Δabs/20min)
Control	ND
CuSO₄	0.079 \pm 0.013 a
0.25 μg/ml	0.066 \pm 0.007 b
0.50 μg/ml	0.055 \pm 0.001 c
0.75 μg/ml	0.054 \pm 0.002 d
1.00 μg/ml	0.050 \pm 0.005 e

LDL was oxidized in the presence of CuSO₄ 30 μ M with or without different concentrations of *S.cExt*. The values are expressed as mean \pm S.D. of three independent experiments in duplicate. Different letters indicate statistically significant differences ($p < 0.05$). ND: not determined.

4. CONCLUSÕES

De acordo com os resultados obtidos neste trabalho é possível inferir que o *S.cExt*:

- Reduz a formação de dienos conjugados (DC) e aumenta o período de fase lag de maneira concentração-dependente na oxidação de LDL e soro humanos induzida por CuSO_4 .
- É capaz de bloquear a propagação da formação de DC na LDL.
- Diminui a formação de TBARS induzida por CuSO_4 em LDL e plasma humanos.
- Previne danos oxidativos à porção protéica da LDL.
- Quando incubado com o plasma, torna a LDL menos susceptível à oxidação lipídica e protéica.
- Não possui efeito sobre a glicação da LDL induzida por metilglicoxal.

Este conjunto de conclusões mostram a eficácia do *S.cExt* nos modelos experimentais avaliados e apontam o *S.cExt* como um agente antiaterogênico promissor. No entanto, mais estudos com o intuito de elucidar os mecanismos envolvidos na proteção contra a oxidação da LDL bem como os efeitos do *S.cExt* em outros modelos de glicação são necessários.

5. PERSPECTIVAS

Tendo em vista os resultados obtidos nesse trabalho, temos como perspectivas

- Avaliar os efeitos do *S.cExt* sobre a formação de células espumosas em cultura de macrófagos.
- Verificar se o *S.cExt* inibe a oxidação da LDL induzida por outros agentes pró-oxidantes.
- Investigar o efeito do *S.cExt* sobre enzimas e marcadores inflamatórios *in vivo*
- Avaliar os efeitos do *S.cExt* em outros modelos de glicação.

6. REFERÊNCIAS BIBLIOGRÁFICAS

- Banerjee, A. et al. In vitro study of antioxidant activity of *Syzygium cumini* fruit. **Food Chem** 90:727-733, 2005.
- Basta, G. et al. Advanced glycation end products and vascular inflammation: implications for accelerated atherosclerosis in diabetes. **Cardiovasc Res** 63:582-592, 2004.
- Baynes, J.W. Role of oxidative stress in development of complications in diabetes. **Diabetes** 40: 405-412, 1991.
- Braga, F.G. et al. Antileishmanial and antifungal activity of plants used in traditional medicine in Brazil. **J Ethnopharmacol** 111:396-402, 2007.
- Brown, M.S. et al. Lipoprotein metabolism in the macrophage: implication for cholesterol deposition in atherosclerosis. **Ann Rev Biochem** 52:223-261, 1983.
- Cervantes-Laurean, D. et al. Glycation and glycooxidation of histones by ADP-ribose. **J Biol Chem** 271:10461-10469; 1996.
- Chaudhuri, A.K.N. Antiinflammatory and related actions of *Syzygium cumini* seed extract. **Phytother R** 4:5-10, 1990.
- Cherubini, A. et al. Role of antioxidants in atherosclerosis: epidemiological and clinical update. **Curr Pharm Des** 11:2017-2032, 2005.
- Cushing, S.D et al. Minimally modified low density lipoprotein induces monocyte chemotact protein 1 in human endothelial cells and smooth muscle cells. **Proc Natl Acad Sci USA** 87:5134-5138, 1990.
- D'agostino, R.B.Jr. et al. Cardiovascular disease risk factors predict the development of type 2 diabetes: the insulin resistance atherosclerosis study. **Diabetes Care** 27:2234-2240, 2004.

Dhalla, N. et al. Role of oxidative stress in cardiovascular diseases. **J Hypertens** 18:655-673, 2000.

Françoso, L. A.; Coates, V. Anatomicopathological evidence of the beginning of atherosclerosis in infancy and adolescence. **Arq Bras Cardiol** 78:131-142, 2002.

Hiohida, H & Kisugi, R. Mechanisms of LDL oxidation. **Clin Chim Acta** 411:1875-1882, 2010.

Hunt, J.V. et al. Auto-oxidative glycosylation and possible involvement of peroxides and free radicals in LDL modification by glucose. **Diabetes** 39:1420-1424, 1990.

Ishitani, L.H. et al. Desigualdade social e mortalidade precoce por doenças cardiovasculares no Brasil. **Revista Saúde Pública** 40:684-691, 2006.

Jain, S.K et al. Erythrocyte membrane lipid peroxidation and glycosylated hemoglobin in diabetes. **Diabetes** 38:1539-1543, 1989.

Kardinaal, A. F. et al. Antioxidants in adipose tissue and risk of myocardial infarction: the EURAMIC Study. **Lancet** 342:1379-1384, 1993.

Kawamura, M. et al. Pathophysiological concentrations of glucose promote oxidative modification of low density lipoprotein by a superoxide-dependent pathway. **J Clin Invest** 94:771-778, 1994.

Kusumoto, I.T. et al. Screening of various plant extracts used in ayurvedic medicine for inhibitory effects on human immunodeficiency virus type I (HIV-I) protease. **Phytother R** 12:488-493, 1995.

Libby, P. Inflammation in atherosclerosis. **Nature** 420:868-874, 2002.

Lyons, T.J. et al. Glycosylation of low-density lipoprotein in patients with type I (insulin dependent) diabetes: correlations with other parameters of glycemic control. **Diabetologia** 29:685-691, 1986.

Mahapatra P.K. et al. Preliminary studies on glyceemic effect of *Syzygium cumini* seeds. **IRCS Med Sci-Biochem** 13:129-33, 1985.

Mashima, R. et al. Oxidants and antioxidants in atherosclerosis. **Curr Opin Lipidol** 12: 411-418, 2001.

McLellan, A.C. et al. Glyoxalase system in clinical diabetes mellitus and correlation with diabetic complications. **Clin Sci** 87:21-29, 1994.

Montecucco, F. et al. Atherosclerosis is an inflammatory disease. **Semin Immunopathol** 31:1-3, 2009.

O'Brien, P.J. et al. Fructose-derived endogenous toxins, in: O'Brien, P.J. and Bruce, W.R. (Eds.). *Endogenous Toxins: Targets for Disease Treatment and Prevention*. **Wiley-VCH Verlag GmbH & Co. KGaA**, Weinheim, pp. 173-212, 2010.

Onorato, J.M. et al. Immunohistochemical and ELISA Assays for Biomarkers of Oxidative Stress in Aging and Disease. **Ann NY Acad Sci** 854:277-290, 1998.

Osterub, B. et al. Role of monocytes in atherogenesis. **Physiol Rev** 83:1069-1112, 2003.

Rabbani, N. et al. Glycation of LDL by methylglyoxal increases arterial atherogenicity. **Diabetes** 60:1973-1980, 2011.

Ramirez, R.O. et al. The gastroprotective effect of tannins extracted from duhat (*Syzygium cumini* Skeels) bark on HCl/ethanol induced gastric mucosal injury in Sprague-Dawley rats. **CI Hemorrh M** 29:253-261, 2003.

Rimm, E. B. et al. Vitamin E consumption and the risk of coronary heart disease in men. **N Engl J Med** 328:1450-1456, 1993.

Smith, S. C. et al. Atherosclerotic Vascular Disease Conference: Writing Group II: risk factors. **Circulation** 109:2613-2616, 2004.

Smith, W. C. et al. Concomitants of excess coronary deaths--major risk factor and lifestyle findings from 10,359 men and women in the Scottish Heart Health Study. **Scott Med J** 34:550-555, 1989.

Steinberg, D. et al. Beyond cholesterol. Modifications of low-density lipoprotein that increase its atherogenicity. **N Engl J Med** 320:915-924, 1989.

Steinbrecher, U. et al. Decrease in reactive amino groups during oxidation or endothelial cellmodification of LDL. Correlation with changes in receptor-mediated catabolism. **Arteriosclerosis** 7:135-143, 1987.

Thornalley, P.J. et al. Suppression of the accumulation of triosephosphates and increased formation of methylglyoxal in human red blood cells during hyperglycaemia by thiamine in vitro. **Biochem J** 129:543-549, 2001.

Vikrant, V. et al. Treatment with extracts of *Momordica charantia* and *Eugenia jambolana* prevents hyperglycemia and hyperinsulinemia in fructose fed rats. **J Ethnopharmacol** 76:139-143, 2001.

Villa-Colinayo, V. et al. Genetics of atherosclerosis: the search for genes acting at the level of the vessel wall. **Curr Atheroscler Rep** 2:380-389, 2000.

Vlassopoulos, A. et al. Role of oxidative stress in physiological albumin glycation: A neglected interaction. **Free Radical Biol Med** 60:318-324, 2013.

Witztum, J.L. and Steinberg, D. Role of oxidized low density lipoprotein in atherogenesis. **J Clin Invest** 88:1785-1792, 1991.

Witztum, J.L. The oxidation hypothesis of atherosclerosis. **Lancet** 344:793-795, 1994.

World Health Organization. Global atlas on cardiovascular disease prevention and control. (http://whqlibdoc.who.int/publications/2011/9789241564373_eng.pdf - acessado em 19/08/2014). 2011 b.

World Health Organization. Global status report on noncommunicable diseases 2010.

(http://www.who.int/nmh/publications/ncd_report_full_en.pdf - acessado em 19/09/2014), 2011 a.

Yamaguchi, Y. et al. Assay methods of modified lipoproteins in plasma. **J Chromatogr B Analyt Technol Biomed Life Sci** 781: 313-330, 2002.

Yeboah, F.K. et al. Reactivities of D-glucose and D-fructose during Glycation of Bovine Serum Albumin. **J Agric Food Chem** 47:3164-3172, 1999.