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Maiara Taís Bazana

**NANOEMULSÕES DO EXTRATO DO CÁLICE DE *Physalis peruviana*:
DESENVOLVIMENTO, AVALIAÇÃO DA ESTABILIDADE E ESTUDOS
DE TOXICIDADE**

Santa Maria, RS
2019

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Tese apresentada ao Curso de Pós-Graduação em Ciência e Tecnologia dos Alimentos, da Universidade Federal de Santa Maria (UFSM, RS), como requisito parcial para obtenção do título de **Doutora em Ciência e Tecnologia dos Alimentos**.

Orientador: Prof. Dr. Cristiano Ragagnin de Menezes
Coorientadora: Prof^a. Dr^a. Cristiane de Bona da Silva

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E-mail: maiarabazana@gmail.com

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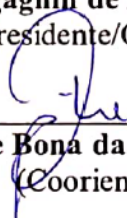
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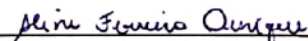
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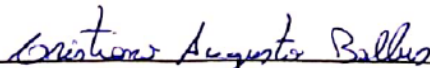
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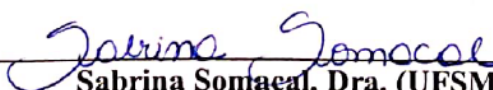
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Santa Maria, RS
2019

*Dedico aos meus
pais e ao Jeferson com
todo meu amor e carinho*

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*“A educação é a arma
mais poderosa
que você pode usar
para mudar o mundo.
Devemos
promover a coragem
onde há medo,
promover o acordo
onde existe conflito
e inspirar esperança
onde há desespero”*

Nelson Mandela

RESUMO

NANOEMULSÕES DO EXTRATO DO CÁLICE DE *Physalis peruviana*: DESENVOLVIMENTO, AVALIAÇÃO DA ESTABILIDADE E ESTUDOS DE TOXICIDADE

AUTORA: Maiara Taís Bazana
ORIENTADOR: Cristiano Ragagnin de Menezes
COORIENTADORA: Cristiane de Bona da Silva

A *Physalis peruviana* apresenta uma variedade de compostos biologicamente ativos principalmente em seu cálice, o qual é considerado um subproduto, com potencial para ser explorado e reutilizado. Entre as tecnologias mais promissoras na indústria de alimentos, destaca-se a nanotecnologia, com a capacidade de encapsular variados compostos que possuem limitações. No entanto, ainda requer maiores estudos quanto à nanotoxicologia das formulações e sua aplicação em resíduos gerados pelas indústrias. Nesse sentido, o objetivo deste trabalho consistiu em desenvolver nanoemulsões contendo o extrato do cálice de *P. peruviana* e avaliar a estabilidade do extrato e das nanoemulsões frente a diferentes condições de armazenamento, capacidade antioxidante, e toxicidade *in vitro* e *in vivo*, visando sua aplicação em alimentos. Os extratos foram preparados em etanol 60% e caracterizados quanto à capacidade antioxidante, capacidade redutora, composto majoritário, atividade antibacteriana e capacidade antibiofilme. As nanoemulsões foram preparadas pelo método de emulsificação espontânea e caracterizadas físico-quimicamente. Após, avaliou-se a estabilidade dos extratos e das nanoemulsões frente a diferentes condições de armazenamento, tais como temperaturas de 7 e 25°C com ausência ou incidência de luz por 120 dias. Também, verificou-se: a liberação do composto majoritário a partir das nanoemulsões e do extrato em meio gastrointestinal simulado; a toxicidade *in vitro* em linhagens de células tumorais (MCF-7 câncer de mama) e não tumorais (3T3 fibroblastos) e *in vivo* (utilizando o modelo *Caenorhabditis elegans*). O extrato do cálice de *P. peruviana* foi caracterizado quanto a sua capacidade redutora (610 mg Eq. de ácido gálico/100 g de cálice), capacidade antioxidante (138 µmol de Trolox/g de cálice) e teor de rutina (11,3 µg/mL). A concentração inibitória mínima variou de 3,15 a 30 mg/mL de extrato, demonstrando atividade bacteriostática frente a oito patógenos e atividade bactericida na concentração de 30 mg/mL frente a seis cepas. Ainda, o extrato inibiu a formação de biofilme produzido pela *Pseudomonas aeruginosa* PA01, porém não houve destruição. As nanoemulsões apresentaram tamanho de gotícula nanométrico (160-180 nm), índice de polidispersão abaixo de 0,15, potencial zeta (-8 a -11 mV), pH levemente ácido (5,4 a 6,3), distribuição do tamanho de gotículas (pico unimodal na faixa nanométrica), morfologia (formato esférico e superfície lisa), teor de rutina (11 µg/mL) e eficiência de encapsulamento de 85%. Entre as amostras testadas, a maior estabilidade foi observada para as nanoemulsões contendo o extrato do cálice de *P. peruviana* quando armazenadas à temperatura ambiente e com ausência de luz. O extrato e as nanoemulsões do extrato não mostraram toxicidade na linhagem celular não tumoral. No entanto, à medida que a concentração das nanoemulsões aumentou, elas demonstraram citotoxicidade contra a linhagem celular tumoral. No modelo *in vivo*, todas as amostras analisadas não causaram toxicidade para *C. elegans* no teste de sobrevivência. Em relação à resistência ao estresse oxidativo, as formulações exerceram seus efeitos antioxidantes em baixas concentrações *in vivo*. Além disso, a rutina presente no extrato nanoemulsionado apresentou maior estabilidade frente às condições gastrointestinais simuladas em relação a presente no extrato livre, controlando a liberação e aumentando os níveis de bioativo liberado no duodeno e íleo, onde ocorre a absorção. Portanto, este estudo contribui para que futuras aplicações na área de alimentos sejam estudadas e aprofundadas.

Palavras-chave: *Cape gooseberry*. Citotoxicidade. *C. elegans*. Nanotecnologia. Resíduo.

ABSTRACT

NANOEMULSIONS OF THE *Physalis peruviana* CALYX EXTRACT: DEVELOPMENT, STABILITY ASSESSMENT, AND TOXICITY STUDIES

AUTHOR: Maiara Taís Bazana
ADVISOR: Cristiano Ragagnin de Menezes
CO-ADVISOR: Cristiane de Bona da Silva

Physalis peruviana presents a variety of biologically active compounds, especially in its calyx, which is considered a byproduct with potential to be explored and reused. Among the most promising technologies in the food industry, nanotechnology stands out with its ability to encapsulate various compounds, which have limitations. However, this technology still requires further studies on the nanotoxicity of formulations and its use in waste generated by industries. Therefore, the objective of this study was to develop nanoemulsions containing the *P. peruviana* calyx extract and evaluate the stability of the extract and nanoemulsions against different storage conditions, antioxidant capacity, and *in vitro* and *in vivo* toxicity, aiming its application in food. The extracts were prepared in 60% ethanol and characterized by antioxidant capacity, reducing capacity, HPLC-quantified major compound, antibacterial activity, and antibiofilm capacity. Nanoemulsions were prepared by the spontaneous emulsification method and physically and chemically characterized. Afterwards, the stability of the extracts and nanoemulsions were evaluated under different storage conditions, such as temperatures of 7 and 25 °C with the absence or incidence of light for 120 days. Additionally, the release of the major compound from the nanoemulsions and extract in a simulated gastrointestinal environment, the *in vitro* toxicity in tumor cell lines (MCF-7 breast cancer) and non-tumor cell lines (3T3 fibroblasts), and *in vivo* (using the *Caenorhabditis elegans* model) were evaluated. The *P. peruviana* calyx extract was characterized by its reducing capacity (610 mg Eq. of gallic acid/100 g of calyx), antioxidant capacity (138 µmol Trolox/g of calyx), and rutin content (11.3 µg/mL). The minimum inhibitory concentration ranged from 3.15 to 30 mg/mL extract, showing bacteriostatic activity against eight pathogens and bactericidal activity at 30 mg/mL concentration against six strains. Nevertheless, the extract inhibited the formation of biofilm produced by *Pseudomonas aeruginosa* PA01, although there was no destruction. The nanoemulsions had nanometric droplet size (160-180 nm), polydispersity index below 0.15, zeta potential (-8 to -11 mV), slightly acidic pH (5.4 to 6.3), droplets (unimodal peak in the nanometer range), morphology (spherical shape and smooth surface), rutin content (11 µg/mL), and encapsulation efficiency of 85%. Among the tested samples, the highest stability was observed in nanoemulsions containing the *P. peruviana* calyx extract when stored at room temperature and in the absence of light. The extract and nanoemulsions of the extract showed no toxicity in the non-tumor cell line. However, as the concentration of nanoemulsions increased, they demonstrated cytotoxicity against the tumor cell line. In the *in vivo* model, all samples analyzed did not cause toxicity to *C. elegans* in the survival test. Regarding resistance to oxidative stress, the formulations exerted their antioxidant effects at low concentrations *in vivo*. In addition, the rutin present in the nanoemulsion extract showed greater stability against the degradation of simulated gastrointestinal conditions compared to the free extract, controlling the release and increasing the levels of bioactive compounds released in the duodenum and ileum, which is where absorption occurs. Therefore, this study contributes to future applications in the food area to be studied and deepened.

Keywords: Cape gooseberry. Cytotoxicity. *C. elegans*. Nanotechnology. Residue.

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LISTA DE ABREVIATURAS E SIGLAS

ABTS	2,2'-Azino-bis(3-ethylbenzothiazoline-6-sulfonic acid)
A/O	Água em óleo
<i>C. elegans</i>	<i>Caenorhabditis elegans</i>
CBM/MBC	Concentração bactericida mínima / “ <i>Minimum bactericidal concentration</i> ”
CIM/MIC	Concentração inibitória mínima / “ <i>Minimum inhibitory concentration</i> ”
CLAE/HPLC	Cromatografia líquida de alta eficiência / “ <i>High performance liquid chromatography</i> ”
DPPH	2,2-difenil-1-picril-hidrazil
MEV/SEM	Microscopia eletrônica de varredura / “ <i>Scanning Electron Microscopy</i> ”
mL	Mililitro
nm	Nanometro
O/A	Óleo em água
ORAC	Capacidade de desativação do radical peroxila / “ <i>Oxygen radical absorbance capacity</i> ”
<i>P. peruviana</i>	<i>Physalis peruviana</i>
PdI	Índice de polidispersão / “ <i>Polydispersity index</i> ”
pH	Potencial hidrogeniônico
TCM/MCT	Triglicerídeos de cadeia média / “ <i>Medium chain triglycerides</i> ”
µg	Micrograma
µm	Micrômetro

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APRESENTAÇÃO

A tese segue as normas estabelecidas na Estrutura e Apresentação do Manual de Dissertações e Teses da UFSM – MDT 2015. Os dados do estudo estão apresentados no formato de artigos científicos integrados, sendo composta a tese por dois artigos publicados (um de revisão bibliográfica publicado na Revista *Current Opinion in Food Science*, Qualis CAPES A1 e o outro com resultados da pesquisa publicado na Revista *Food Research International*, Qualis CAPES A1) e dois manuscritos com resultados da pesquisa, sendo que o manuscrito 1 foi submetido a Revista *Ciência Rural*, Qualis CAPES A4 e o manuscrito 2 será submetido a Revista *Journal of Functional Foods*, Qualis CAPES A1. Ao final, encontram-se os itens **DISCUSSÃO GERAL** e **CONCLUSÃO**, apresentando uma compilação de interpretações e comentários a respeito dos resultados demonstrados nos artigos e manuscritos. As **REFERÊNCIAS** referem-se às citações apontadas nos itens **INTRODUÇÃO**, **REFERENCIAL TEÓRICO** e **DISCUSSÃO** dessa tese.

1 INTRODUÇÃO

Os resíduos gerados no processamento de alimentos, geralmente são descartados ou empregados na alimentação animal, sendo, atualmente, o reaproveitamento um desafio tanto para as indústrias agroalimentares quanto para os pesquisadores. Por outro lado, na indústria alimentícia a busca por diferentes fontes de compostos bioativos que possam ter um efeito benéfico em nossa saúde, tem sido tendência emergente na ciência de alimentos. Esses subprodutos tem se mostrado como uma alternativa valiosa para extrair compostos naturais com bioatividade demonstrada (BALLESTEROS-VIVAS et al., 2019b).

Physalis peruviana L. também conhecida como *goldenberry* ou *cape gooseberry*, pertence à família Solanaceae, que possui cerca de 120 espécies, distribuídas em regiões tropicais e subtropicais da África, Ásia e América (KUSPKA; JÉLEN, 2016). O processamento industrial de *goldenberries* gera dois subprodutos, a baga (incluindo sementes e pele) e o cálice (RAMADAN, 2011). O cálice é um invólucro que sustenta os frutos comestíveis, protegendo-os naturalmente ao longo do seu desenvolvimento e amadurecimento, contra insetos, aves, doenças e situações climáticas adversas. Tal estrutura consiste em uma fonte essencial de carboidratos durante os primeiros 20 dias de crescimento do fruto (TAPIA; FRIES, 2007; PUENTE et al., 2011).

O cálice de *P. peruviana* é uma fonte potencial de compostos bioativos, tais como, fitoquímicos (terpenoides, fitoesteróis e derivados do fitol), flavonoides (destacando-se a rutina como um dos compostos majoritários), ácidos fenólicos e vitanolídeos (GIRONÉS-VILAPLANA et al., 2014; BALLESTEROS-VIVAS et al., 2019a; BALLESTEROS-VIVAS et al., 2019b; MEDINA et al., 2019). Além disso, possui potente atividade antioxidante (GIRONÉS-VILAPLANA et al., 2014; TORO et al., 2014) e anti-inflamatória comprovada (FRANCO et al., 2007; FRANCO et al., 2014; CASTRO; OCAMPO; FRANCO, 2015), dentre outros usos na medicina popular como anticancerígeno, antipirético, diurético, antimicrobiano e imunomodulador (FRANCO et al., 2007).

Essas substâncias com excelentes propriedades funcionais (compostos bioativos) possuem algumas limitações que dificultam sua aplicação em alimentos, incluindo baixa solubilidade em água, baixa absorção e biodisponibilidade (SHIN; KIM; PARK, 2015). Entre os compostos bioativos que apresentam essas desvantagens, está a rutina, um dos flavonoides presentes em maior quantidade no cálice de *P. peruviana* (GULLÓN et al., 2017). Além disso, esses compostos podem se degradar pelas condições de processamento e armazenamento de alimentos como aquecimento, acidificação, luz e oxigênio. Defeitos de

qualidade, como gosto adstringente também podem ser observados quando adicionados em produtos alimentares. Assim, alguns extratos não são utilizados devido a tais obstáculos (CODEVILLA et al., 2015).

Um dos setores que tem se destacado em pesquisas com alimentos e sua aplicação é a nanotecnologia (HAMAD et al., 2018), impulsionados pela necessidade de sistemas comestíveis capazes de encapsular, proteger e liberar compostos funcionais (ARANCIBIA et al., 2017). As nanoemulsões são utilizadas na indústria de alimentos na produção de edulcorantes, óleos aromatizados, molhos para saladas, bem como em outros alimentos processados (HAMAD et al., 2018). Assim, as nanoformulações são uma alternativa para impedir a degradação de compostos bioativos, melhorando sua estabilidade e biodisponibilidade (REZAEI; FATHI; JAFARI, 2019).

Por outro lado, os possíveis riscos para a saúde humana e ao meio ambiente das nanopartículas, como os impactos toxicológicos não acompanharam o rápido aumento nas pesquisas e utilização dos nanomateriais. Ainda são necessários estudos para ajudar a concluir os mecanismos de toxicidade das nanopartículas, bem como os mecanismos envolvidos na sua distribuição, acúmulo em células e tecidos e sua eliminação pelo organismo (MISSAOUI; ARNOLD; CUMMINGS, 2018).

Considerando o potencial do cálice de *P. peruviana* como uma fonte promissora de fitoquímicos bioativos, este trabalho visou obter um extrato desse derivado natural e caracterizá-lo. Assim como, desenvolver nanoemulsões contendo o extrato do cálice de *P. peruviana* e avaliar se, tanto as nanoemulsões como o extrato, são estáveis quanto às características físico-químicas, capacidade antioxidante e redutora, teor de rutina frente à temperatura ambiente (25°C) e refrigerada (7°C), submetidos à incidência ou ausência de luz. Além de, estudar a toxicidade *in vitro* e *in vivo* das formulações e verificar a liberação no trato gastrointestinal simulado.

2 OBJETIVOS

2.1 OBJETIVO GERAL

Desenvolver nanoemulsões contendo extrato do cálice de *Physalis peruviana* e avaliar sua estabilidade frente a diferentes condições de armazenamento e toxicidade *in vitro* e *in vivo*.

2.2 OBJETIVOS ESPECÍFICOS

- Preparar e caracterizar o extrato hidroalcoólico do cálice de *P. peruviana* frente à capacidade antioxidante e redutora, e o teor de rutina;
- Avaliar a atividade antibacteriana e antibiofilme do extrato;
- Desenvolver nanoemulsões contendo o extrato do cálice de *P. peruviana* e nanoemulsões brancas (sem o extrato associado às nanoestruturas);
- Avaliar as características físico-químicas das nanoemulsões quanto ao diâmetro médio de gotícula, índice de polidispersão, potencial zeta, distribuição do tamanho de gotícula, pH, teor de rutina, eficiência de encapsulamento e morfologia;
- Analisar a estabilidade dos extratos do cálice de *P. peruviana* e das nanoemulsões em diferentes condições de luminosidade e temperatura, por um período de 120 dias, através da capacidade redutora, capacidade antioxidante e teor de rutina;
- Avaliar a liberação *in vitro* da rutina a partir das nanoemulsões e do extrato livre em condições gastrointestinais simuladas;
- Verificar a citotoxicidade do extrato do cálice de *P. peruviana* associado ou não às nanoemulsões em linhagens de células tumorais (MCF-7 câncer de mama) e não tumorais (3T3 fibroblastos);
- Submeter às formulações e os extratos a testes de toxicidade e atividade antioxidante *in vivo*, utilizando o modelo de *Caenorhabditis elegans*.

3 REFERENCIAL TEÓRICO

3.1 *Physalis peruviana*

Produtores e comerciantes têm se interessado pelo cultivo de pequenas frutas no Brasil. Ainda são bastante incipientes e inovadoras, as atividades econômicas com a utilização de pequenas frutas. De modo geral, caracterizam-se, pelo baixo custo de implantação e de produção, acessível aos pequenos produtores, bom retorno econômico em curto prazo, boa adaptação às condições socioeconômicas e ao ambiente local, grande exigência de mão de obra, possibilidade de cultivo no sistema orgânico e demanda maior do que a oferta (MUNIZ et al., 2011).

A *P. peruviana* Linnaeus (Figura 1) é uma planta herbácea, semi-arbustiva, vertical e perene, podendo atingir 0,6 a 0,9 metros e em alguns casos pode alcançar até 1,8 m. As flores podem ser facilmente polinizadas por insetos, pelo vento e também por autopolinização. Os frutos possuem coloração laranja-amarelada e uma baga suculenta com formato ovoide. O diâmetro varia entre 1,25 e 2,50 cm e peso entre 4 e 10 g, contendo interiormente cerca de 100 a 200 pequenas sementes amareladas. Esse fruto é protegido pelo cálice que cobre completamente os frutos ao longo do seu desenvolvimento e amadurecimento (TAPIA; FRIES, 2007; PUENTE et al., 2011). O fruto de *P. peruviana* é considerado climatérico, ou seja, são frutos que podem amadurecer na planta ou fora dela se colhidos imaturos (RODRIGUES et al., 2012).

Figura 1 – *Physalis peruviana*



Fonte: <<https://www.sitiodamata.com.br/especies-de-plantas/plantas-ornamentais/physalis>> e <https://commons.wikimedia.org/wiki/File:Physalis_Nahaufnahme.JPG>

Vários são os nomes atribuídos ao fruto de *P. peruviana* ao redor do mundo, sendo conhecido como *uchuva* na Colômbia, *uvilla* no Equador, *cereza del Perú* ou *aguaymanto* no Peru, *topotopo* na Venezuela e *goldenberry* ou *cape gooseberry* em países de língua inglesa (PUENTE et al., 2011). No Brasil, é encontrado em algumas regiões como no Norte e Nordeste sendo comum nos quintais, nomeados como: camapu, camambu, camaru, joá-de-capote, joá-poca, balão-rajado, mulaca, saco-de-bode, bucho-de-rã e mata-fome (LICODIEDOFF, 2012). A planta *P. peruviana* é originária de países Andinos, sendo cultivada em países sul-americanos, especialmente Colômbia, Peru e Equador; e atualmente disponível em mercados internacionais (OLIVARES-TENORIO, 2016).

Na Colômbia é cultivada para exportação e o país lidera como o maior produtor seguido pela África do Sul (MAZORRA, 2006). A Colômbia produz 11.500 toneladas/ ano de frutos de *P. peruviana*, sendo que o excedente, não destinado à exportação, atingiu 50% do total da produção, pois esta fruta não é muito exportável devido o seu pequeno tamanho, por isso vem sendo usada para novos produtos desidratados (CASTRO; RODRIGUEZ; VARGAS, 2008). As pesquisas sobre esta planta, no Brasil, iniciaram em São Paulo, no ano de 1999, pela Estação Experimental Santa Luzia que foi a pioneira no cultivo. Para o comércio na capital paulista, a estação produz de 2 a 3 toneladas anuais (RUFATO et al., 2013). No município de Lages, SC, verificaram-se valores de produção de *P. peruviana* entre 2 a 8,67 ton.ha⁻¹ em áreas experimentais (BRIGHENTI et al., 2008).

A fruta de *P. peruviana* possui prestígio em alguns mercados internacionais. Os europeus, por exemplo, muitas vezes pagam preços “premium” para mergulhar as bagas no chocolate ou decorar bolos e tortas. Além de ser consumida *in natura*, a fruta pode ser apreciada de diferentes maneiras, sendo um ingrediente interessante em saladas, pratos cozidos, sobremesas e geleias (RAMADAN; MOERSEL, 2007).

A *P. peruviana* é uma boa fonte de provitamina A (vitamina A equivalente de atividade de retinol - RAE 163 µg/100g e vitamina A equivalente de retinol - RE 326 µg/100g), de minerais (cálcio, cobre, ferro, magnésio, manganês, zinco e em maior quantidade o potássio 382 mg/100g), vitamina C (17,72 mg/ 100g) e de vitaminas do complexo B (vitamina B6 0,321 mg/100g e riboflavina 0,023 mg/ 100g). A fruta contém 15% de sólidos solúveis (principalmente açúcares) e alta concentração de frutose. Assim como o nível de fósforo (33 mg/100g) também é elevado para uma fruta. Além disso, seu alto teor de fibra dietética é importantíssimo, pois a pectina das frutas atua como um regulador intestinal. Os valores de fibra alimentar total são 6,5 g/100g, fibra insolúvel 5,2 g/100g e fibra solúvel 0,9 g/100g (HASSANIEN, 2011; ROCKETT et al., 2018).

Outras partes da planta têm sido objeto de pesquisa, como a obtenção de 4 β -hidroxivitanolídeo, a partir do extrato das partes aéreas da planta (hastes e folhas). Esse composto é responsável por inibir o crescimento de células cancerígenas do pulmão, podendo ser um potencial quimioterápico (YEN et al., 2010). No cálice de *P. peruviana* foi, inicialmente, isolado um vitanolídeo (28-hidroxivitanolídeo E) (DINAN; SARKER; SIK, 1997). Ballesteros-Vivas e colaboradores (2019a) identificaram 56 fitoquímicos, incluindo principalmente compostos fenólicos, 17 tipos de vitanolídeos e ésteres de sacarose anti-inflamatórios, indicando a ampla variedade de compostos bioativos presentes nos cálices. De acordo com Severo et al. (2010) os teores de carotenoides totais e de fenólicos totais dos frutos foram avaliados e verificou-se que até o estágio de amadurecimento 4 (coloração amarelo-amarronzado) os teores foram crescentes. A atividade antioxidante se mostrou superior em frutos nos estágios iniciais de amadurecimento. O estudo verificou que os frutos de *P. peruviana* consistem em fontes relevantes de carotenoides e compostos fenólicos, contudo apresentam baixa atividade antioxidante, esta foi avaliada com o radical ABTS (SEVERO et al., 2010). Porém, Wu e colaboradores (2005) observaram que os extratos etanólicos de *P. peruviana* possuem boa atividade antioxidante, avaliada por três métodos diferentes: teste do ácido tiobarbitúrico, citocromo c e inibição da xantina oxidase, diferindo do estudo anterior, e além disso, um avaliou o fruto e o outro o extrato etanólico do fruto.

Portanto, *P. peruviana* tem um grande potencial para o processamento de novos alimentos funcionais, em decorrência de suas propriedades nutricionais, bem como os seus componentes biologicamente ativos como compostos fenólicos, vitanolídeos, carotenoides, compostos aromáticos e voláteis (HASSANIEN, 2011). O fruto apresenta uma abundância de constituintes, dentre os quais se destacam os flavonoides (ROP et al., 2012; SATHYADEVI; SUBRAMANIAN, 2015); vitamina E (RAMADAN; MORSEL, 2003), β -caroteno (RAMADAN; MORSEL, 2003; BRIONES-LABARCA et al., 2013; ETZBACH et al, 2018) e ácido ascórbico (ROP et al., 2012; BRIONES-LABARCA et al., 2013). De acordo com Teixeira e colaboradores (2016) o fruto apresenta compostos fenólicos, que são variáveis de acordo com o local de plantio, já que os frutos cultivados na região de Huánuco no Peru apresentaram $149,3 \pm 1,62$ mg/Eq de ácido gálico/100 g de fruto e nas demais regiões (Junín, Ancash e Cajamarca) se observou valores menores de compostos fenólicos.

Para a preservação das propriedades nutraceuticas é sugerida a aplicação de tecnologias inovadoras para a estabilização da polpa de *goldenberry*. Quanto às tecnologias, a aplicação de alta pressão hidrostática (APH) tem o potencial de produzir alimentos de alta qualidade que exibem características de produtos frescos, microbiologicamente seguros e têm

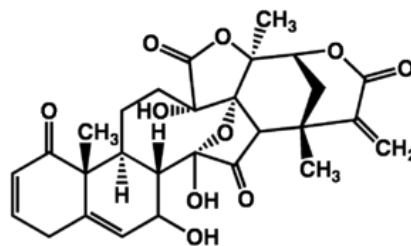
uma vida útil prolongada (FERRARI, MARESCA; CICCARONE, 2010; HUANG et al., 2014b). O emprego da encapsulação, também, é considerada uma alternativa muito promissora, como o estudo realizado em nosso grupo de pesquisa por Silva (2018) que desenvolveram nanoemulsões contendo o extrato da fruta, as quais apresentaram pH de 3,84, diâmetro médio de gotícula de 268 ± 7 nm, índice de polidispersão de 0,113 e potencial zeta negativo de 13,94 mV. A formulação se manteve estável pelo período de 60 dias sob refrigeração ao contrário do extrato livre que a partir do 30º dia perdeu 56% de compostos fenólicos. Além disso, as análises de toxicidade *in vitro* e *in vivo* demonstraram que tanto o extrato livre quanto associado às nanoemulsões não apresentaram toxicidade nas concentrações de 50, 200 e 400 µg/mL no teste de citotoxicidade frente a linhagem não tumoral (3T3 fibroblastos) e nas concentrações de 0,5; 1,0 e 2,0 mg/mL no teste de sobrevivência dos *Caenorhabditis elegans* (SILVA, 2018). Assim como, a microencapsulação do suco de *P. peruviana*, utilizando-se maltodextrina misturada com goma arábica, alginato e pectina para aumentar a estabilidade em fluídos de digestão simulados. As microcápsulas mostraram reter mais de 75% dos compostos fenólicos para todos os tipos de goma. *In vitro* foi possível verificar que a liberação de compostos fenólicos das microcápsulas foi maior no fluido intestinal simulado do que no meio gástrico (DAG; KILERCIOGLU; OZTOP, 2017).

3.2 CÁLICE DE *P. peruviana*

O nome *Physalis* provém do grego, em que “*Physa*” significa bolha ou bexiga, referindo-se ao cálice que sustenta seus frutos comestíveis (LICODIEDOFF, 2012). O cálice, formado por cinco sépalas, também conhecido como cesta de frutas ou lanterna chinesa é um involúcro que protege naturalmente o fruto ao longo do seu desenvolvimento e amadurecimento, protegendo-o contra pragas, doenças e clima. A estrutura, ainda, é considerada uma excelente fonte carboidratos durante os primeiros 20 dias de crescimento e desenvolvimento do fruto (TAPIA; FRIES, 2007; PUENTE et al., 2011). Outra característica que merece destaque relaciona-se à sua coloração que indica o ponto de colheita. Rodrigues e colaboradores (2012) enfatizam que o fruto de *P. peruviana* deve ser colhido quando o cálice apresentar coloração amarelo-esverdeado até amarelo-amarronzado. Nessas fases os frutos apresentam maiores massas, diâmetros e acúmulo de sólidos solúveis totais. Ademais, o cálice aumenta a conservação do fruto, pois a vida de prateleira aumenta, alcançando um mês; enquanto que sem o cálice é em torno de 4 a 5 dias (CEDEÑO; MONTENEGRO, 2004).

O cálice é importante, ainda, por ser o principal depósito das fisalinas que foram identificadas na espécie *P. alkekengi* (JI et al., 2012; XU et al., 2012; HUANG et al., 2014a), as quais são moléculas exclusivas do gênero *Physalis*. As fisalinas são moléculas de estruturas bastante complexas, apresentam lactona e γ lactona fundida ao anel D, sendo derivados esteroidais do tipo 13,14-seco-16,24 ciclo ergostano, carbonilados em C-15 (TOMASSINI et al., 2000). As concentrações desses compostos na planta são variáveis, sendo a fisalina A (Figura 2) a mais abundante e, dependendo das condições de conservação, mais estável em relação às fisalinas O e P (XU et al., 2012). Já foram descobertas várias fisalinas, sendo a mais recente a fisalina X (HUANG et al., 2014a) e também já foram identificadas as isofisalinas (JI et al., 2012). Além disso, estudos identificaram os vitanolídeos como moléculas da *Physalis*, extraídos por etanol, cuja atividade antitumoral, imunossupressora e hepatoprotetora vêm sendo estudada (GAUTAM; DWIVEDI; KUMAR, 2015). O local de colheita e a parte da planta de onde se obtém o extrato, consistem em outros fatores importantes, visto que já foi verificado que tanto o fruto como o cálice são os locais de maior acúmulo de compostos fenólicos, sendo mais proeminente no cálice, de acordo com o observado por Medina-Medrano e colaboradores (2015) que compararam as diferentes partes da planta (fruto, folha e cálice) de espécies de *Physalis* (*P. angulata*, *P. hederifolia* var. *hederifolia*, *P. solanacea*, *P. patula* e *P. subulata*).

Figura 2 – Estrutura química da fisalina A



Muitos utilizam o cálice em decoração (de doces finos ou outros produtos alimentícios), contudo na indústria de alimentos, muitas vezes, essa parte da planta é considerada subproduto (LICODIEDOFF, 2012). No entanto, alguns estudos relacionados ao cálice encontram-se descritos, destacando seus benefícios. Avaliou-se com o uso de um modelo de fagocitose *in vitro* (infecção por *Leishmania panamensis* em macrófagos murinos), o efeito anti-inflamatório dos extratos etanólicos ou etéreos dos cálices de *P. peruviana* no processo fagocítico. A atividade anti-inflamatória descrita neste modelo relaciona-se a um efeito imunomodulador exercido sobre macrófagos infectados, que "bloqueiam" direta ou

indiretamente sua capacidade de secretar mediadores pró-inflamatórios solúveis (MARTÍNEZ et al., 2010).

Castro e colaboradores (2015) avaliaram a atividade anti-inflamatória do extrato etéreo total de cálices de *P. peruviana* em protocolos preventivos e terapêuticos em um modelo de colite em ratos, induzido por ácido TNBS. O tratamento com o extrato produziu melhora significativa no tecido do cólon, tanto a nível macroscópico como histológico. Deste modo, o extrato apresentou atividade anti-inflamatória intestinal, demonstrando que o cálice desta espécie apresentou-se como uma fonte promissora de metabólitos que poderiam ser utilizados no tratamento da doença inflamatória intestinal.

Para Franco e colaboradores (2007), 38 frações secundárias foram logradas por cromatografia em coluna a partir da fração hidroalcoólica primária de extratos do cálice, das quais seis foram avaliadas no modelo de inflamação aguda em ratos. Identificou-se as principais frações responsáveis pela atividade anti-inflamatória, promissoras no desenvolvimento de fitoterápicos, porém, faz-se necessária a realização de estudos adicionais buscando evidenciar como isolar e identificar os compostos responsáveis pela atividade e para o mecanismo envolvido no efeito anti-inflamatório.

Os mesmos pesquisadores avaliaram a melhor fração obtida pela extração dos cálices de *P. peruviana*, a qual foi purificada por vários métodos cromatográficos para obter uma mistura inseparável de dois novos ésteres de sacarose denominados: peruviose A (1) e peruviose B (2). Elucidou-se as estruturas dos novos compostos utilizando métodos espectroscópicos e transformações químicas. A atividade anti-inflamatória da mistura de peruvioses foi avaliada pelo método de edema de pata em ratos. Como resultado, as peruvioses não produzem efeitos colaterais no fígado e rins. Além disso, atenuaram significativamente a inflamação induzida por λ -carragenina em uma dose-dependente. Este trabalho foi o primeiro a relatar a presença de ésteres de sacarose em cálices de *P. peruviana* demonstrando um potente efeito anti-inflamatório. Estes resultados sugerem o potencial dos ésteres de sacarose do gênero *Physalis* como uma alternativa natural para tratar doenças inflamatórias (FRANCO et al., 2014).

Conforme Toro e colaboradores (2014) a fração butanólica do cálice foi considerada promissora devido às suas atividades anti-inflamatórias e antioxidantes. Portanto, foi utilizada uma abordagem guiada por bioensaio para isolar e identificar a rutina e a nicotiflorina a partir dos dados de espectroscopia por ressonância magnética nuclear (RMN) e espectrometria de massas (MS). A identificação da rutina em cálices de *P. peruviana* apoia a possível utilização deste material de resíduos para preparações fitoterapêuticas, nutracêuticas e cosméticas.

Relevante, também, é o alto teor de compostos fenólicos e capacidade antioxidante no cálice. Estudo analisou separadamente os frutos e os cálices de *P. peruviana* obtidos da Colômbia. Verificou-se maior quantidade de compostos fenólicos e um teor muito maior no cálice ($195,44 \pm 3,88 \text{ mg} \cdot 100 \text{ g}^{-1}$ de peso seco) do que na fruta ($2,18 \pm 0,71 \text{ mg} \cdot 100 \text{ g}^{-1}$ de peso seco). A capacidade antioxidante foi avaliada por ORAC (*Oxygen Radical Absorbance Capacity*), sendo que o cálice de *P. peruviana* apresentou $24,29 \pm 3,11 \text{ mmol Trolox} \cdot 100 \text{ g}^{-1}$, enquanto o fruto resultou em uma menor capacidade antioxidante ($3,29 \pm 0,58 \text{ mmol Trolox} \cdot 100 \text{ g}^{-1}$). Ainda, foi possível identificar no cálice dois diferentes glicosídeos de quercetina (sendo a rutina o composto fenólico majoritário) e de kaempferol, mas apenas foram encontrados vestígios de rutina (quercetina 3-O-rutinosídeo) e kaempferol 3-O-rutinosídeo na fruta. Os ácidos 3-O-cafeoilquínico e 5-O-cafeoilquínico também foram identificados no cálice, contudo, na fruta não foram observados. Menciona-se, ainda, que o cálice apresentou as maiores concentrações de flavonóis e derivados de ácido hidroxicinâmico comparados aos frutos latino-americanos analisados, salientando a importância da utilização do cálice em novos produtos (GIRONÉS-VILAPLANA et al., 2014).

3.3 A IMPORTÂNCIA DOS BIOATIVOS EM RESÍDUOS DE FRUTAS E ANÁLISE DOS COMPOSTOS FENÓLICOS

No que tange à produção de frutas no mundo, o Brasil ocupa a terceira posição e com isso o processamento dos sucos de frutas nas indústrias resulta em um grande volume de resíduos, os quais podem ser explorados para a produção de substâncias altamente valorizadas (FORSTER-CARNEIRO et al., 2013). Um problema ocasionado por isso consiste no descarte desses resíduos, uma vez que é agravado por restrições ambientais atuais, tais como as que constam na Política Nacional de Resíduos Sólidos e na Resolução do CONAMA nº 313 de 29 de setembro de 2002, que dispõe sobre o Inventário Nacional de Resíduos Industriais. Assim, ressalta-se a importância da utilização destes resíduos, pois muitas vezes, representam uma importante fonte de açúcares, minerais, ácidos orgânicos, fibra dietética e compostos fenólicos. Além disso, esses compostos possuem várias ações benéficas para a saúde que incluem atividade antitumoral, antiviral, antibacteriana, cardioprotetora e antimutagênica (SONJA; JASNA; GORDANA, 2009). Nas frutas o processamento origina dois principais subprodutos, a casca e as sementes, sendo que o extrato desses resíduos contém uma quantidade considerável de compostos bioativos (GOOT et al., 2016). Para exemplificar, vários estudos relataram a exploração de resíduos de algumas frutas, como uma nova fonte de

ingredientes alimentares, tais como sementes de *Zizyphus spina-christi* (juzuba de espinhos de Cristo) (EMBABY; MOKHTAR, 2011), sementes de manga (TORRES-LEÓN et al., 2016), bagaço de maçã (SUDHA; BASKARAN; LEELAVATHI, 2007) e subproduto da polpa de *goldenberry* ou do processamento de suco (resíduo em pó) (MOKHTAR; SWAILAM; EMBABY, 2018). O processamento industrial de *P. peruviana* gera dois tipos de resíduos, a baga (incluindo sementes e pele) que representa 27 % do peso das frutas, e o cálice que representa 5% da fruta crua. Cerca de 33 toneladas de resíduos de cálice são gerados por hectare cultivado de *P. peruviana* (RAMADAN, 2011).

Entende-se como relevante o destaque à importância do conhecimento detalhado sobre esses resíduos e sua caracterização química, contribuindo para sua utilização como fonte de compostos bioativos (ABDENNACER et al., 2015). Esses, são encontrados em pequenas quantidades nos alimentos e são considerados como ingredientes não nutricionais, porém vitais para a manutenção da saúde humana (PATIL et al., 2009). Em meados de 1980, surgiu no Japão a terminologia de alimentos funcionais. O conceito se refere a alimentos usados como parte de uma dieta normal, fisiologicamente benéficos, reduzindo o risco de doenças crônicas, além de suas funções básicas nutricionais. Esses alimentos, designados para “uso específico de saúde” (FOSHU, do inglês *foods for specified health use*), apresentam um selo de aprovação do Ministério da Saúde e Bem-Estar japonês. Deste modo o conceito foi rapidamente adotado no mundo (COSTA; ROSA, 2016).

Entre os compostos bioativos, os fenólicos são compostos poli-hidroxilados, constituindo um dos grupos mais extensos de produtos químicos presentes no reino vegetal. Estes mostram diversidade estrutural variando de fenólicos simples a complexos, bem como compostos altamente polimerizados. Os compostos fenólicos de alto peso molecular com estrutura complexa são frequentemente referidos como polifenóis (BALASUNDRAM et al., 2006; SINGH et al., 2017). Além disso, eles são considerados os principais constituintes das frutas e dos seus resíduos (BATAGLION et al., 2015). Os polifenóis são compostos que possuem mais de 9000 substâncias identificadas e podem ser divididos em grupos de acordo com a sua estrutura química: flavonoides (isoflavonoides, antocianidinas, flavanois, flavonois, flavanonas e flavonas) e não flavonoides (ácidos hidroxicinâmicos e hidroxibenzoicos, estilbenos, lignoides e cumarinas) (TSAO, 2010). Em geral, a presença de compostos fenólicos em produtos alimentares é de extrema importância, pois eles possuem muitos benefícios para a saúde, podendo atuar como anticancerígenos, antitrombóticos, antiulcerativos, antiarteriogênicos, antialérgicos, anti-inflamatórios, antioxidantes,

imunomoduladores, antimicrobianos, agentes cardioprotetores, vasodilatadores e analgésicos (BALASUNDRAM et al., 2006; SINGH et al., 2017).

Para a extração de compostos fenólicos existem muitos métodos disponíveis, porém nenhum pode ser categorizado como um método padrão. Tal extração é principalmente afetada por vários parâmetros (como o tamanho de partícula das amostras, o tipo de solvente utilizado, a relação soluto/solvente, taxa de agitação, eficiência de transferência de massa e temperatura utilizada). A etapa importante é selecionar as condições ótimas necessárias para a extração. Além disso, o passo crucial na quantificação desses compostos é o método de preparação das amostras e a influência da precisão e da repetibilidade durante a análise (ZHAO et al., 2011). As amostras em pó fino melhoram consideravelmente a extração de compostos fenólicos devido ao aumento da área superficial e ruptura das paredes celulares presentes no material vegetal utilizado. Geralmente solventes orgânicos puros, tais como metanol, etanol, acetona e acetato de etila, bem como as suas combinações com água (dependendo da seletividade, preço e recuperação) são utilizados para extrair esses componentes (KHODDAMI; WILKES; ROBERTS, 2013; TALUKDAR, 2013).

As técnicas espectrofotométricas (como as que utilizam o reagente Folin-Ciocalteu) são muito empregadas para quantificação dos compostos fenólicos totais (KHODDAMI; WILKES; ROBERTS, 2013; TALUKDAR, 2013). Apesar de serem econômicas, simples e rápidas, não são seletivas e não quantificam compostos fenólicos individuais (KHODDAMI; WILKES; ROBERTS, 2013). Por outro lado, a cromatografia líquida de alta eficiência (CLAE) é comumente usada para elucidar os perfis de compostos fenólicos presentes em muitas espécies vegetais (KHODDAMI; WILKES; ROBERTS, 2013; TALUKDAR, 2013).

Portanto, o cálice de *P. peruviana* é um resíduo, que ainda pode ser mais explorado, o qual apresenta variados compostos bioativos, destacando-se o flavonoide (rutina), como um dos compostos majoritários, e pode ser analisado utilizando essas técnicas de análise (BALLESTEROS-VIVAS et al., 2019a; GIRONÉS-VILAPLANA et al., 2014; MEDINA et al., 2019).

3.4 CAPACIDADE ANTIOXIDANTE E RUTINA

As espécies reativas de oxigênio (EROs) são um subproduto do metabolismo celular normal em plantas. Porém, quando estas são submetidas a condições de estresse ocorrem alterações na produção de EROs afetando a homeostase. Estas espécies reativas formadas começam a inativar enzimas rapidamente, degradar macromoléculas importantes na

manutenção orgânica da planta, danificar organelas celulares vitais e desestabilizar as membranas celulares, principalmente, através da indução da degradação de pigmentos, proteínas, lipídios e ácidos nucleicos, acarretando o processo de morte celular (KARUPPANAPANDIAN et al., 2011).

Assim como, no corpo humano a molécula de oxigênio pode produzir uma reação gerando EROs desencadeados por alguns fatores exógenos e/ou processos metabólicos endógenos. As EROs incluem uma série de moléculas quimicamente reativas, como o peróxido de hidrogênio (H_2O_2), superóxido ($O_2^{\cdot-}$) e radical hidroxila ($\cdot OH$). Quando a geração de EROs é induzida por vários estímulos no organismo e excede a capacidade antioxidante do mesmo, resultará em uma variedade de processos fisiopatológicos (YANG; GUO; YUAN, 2008).

Nos últimos anos, inúmeros sequestradores naturais e sintéticos das EROs e antioxidantes vêm sendo pesquisados e desenvolvidos para proteger biomoléculas contra o ataque de EROs e/ou para suprimir o dano resultante (YANG; GUO; YUAN, 2008). Como por exemplo, extratos aquosos de 30 plantas foram investigados por suas propriedades antioxidantes usando o teste de capacidade de eliminação de radicais (DPPH e ABTS), teste da capacidade de desativação do radical peroxila (ORAC), ensaio de superóxido dismutase (SOD) e teste de potencial antioxidante redutor férrico (FRAP). O teor fenólico total também foi determinado pelo método de Folin-Ciocalteu. Sendo que, as propriedades antioxidantes e o conteúdo fenólico total diferiram significativamente entre as plantas selecionadas. Verificou-se que os extratos aquosos de carvalho (*Quercus robur*), pinho (*Pinus maritima*) e canela (*Cinnamomum zeylanicum*) possuíam as maiores capacidades antioxidantes na maioria dos métodos utilizados e, portanto, poderiam ser fontes ricas de antioxidantes naturais. Os extratos aquosos de erva mate (*Ilex paraguariensis*) e cravo (*Eugenia caryophyllus clovis*) também apresentaram fortes propriedades antioxidantes e um elevado teor fenólico. Foi possível observar uma relação significativa entre a capacidade antioxidante e o conteúdo fenólico total, indicando que os compostos fenólicos são os principais contribuintes para as propriedades antioxidantes dessas plantas (DUDONNÉ et al., 2009).

Flavonoides são conhecidos por suas significantes propriedades de eliminação de radicais de oxigênio *in vivo* e *in vitro*. Além desses efeitos importantes, eles possuem propriedades estabilizadoras da membrana e também afetam alguns processos de metabolismo intermediário (BOMBARDELLI; MORAZZONI, 1993). Além disso, muitos estudos demonstram a importância da ação contra radicais livres dos flavonoides. Dentre eles, extensas pesquisas com o objetivo de beneficiar seres humanos foram descritos, ressaltando

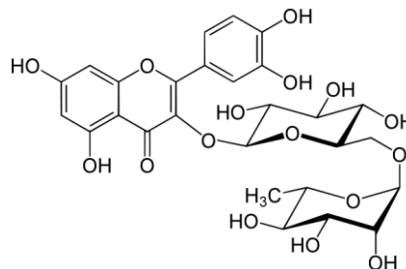
que os flavonoides possuem numerosas atividades biológicas, como antioxidantes, anti-inflamatórias, estrogênica, antitumoral, citotóxica, antiviral e outras (HARBORNE; WILLIAMS, 2000).

De acordo com Zhang e colaboradores (2010), antioxidantes sintéticos têm seu uso generalizado como aditivos alimentares em muitos países. No entanto, antioxidantes naturais são preferidos aos sintéticos, pois possíveis problemas de saúde podem estar relacionados aos sintéticos (HOU, 2003; PRIOR, 2004). Atualmente, compostos naturais com capacidade antioxidante, como tocoferóis e flavonoides vêm atraindo um interesse crescente (ERKAN; AYRANCI; AYRANCI, 2008).

A capacidade dos flavonoides de eliminar os radicais livres e quelar íons metálicos levaram a análise de suas atividades como antioxidantes na peroxidação lipídica. A atividade antioxidante está relacionada à sua estrutura química, coeficientes de partição e taxa de reação com os radicais de interesse (BOYLE et al., 2000).

Nesse sentido, a rutina (quercetina-3-rutinosídeo) (Figura 3) é um derivado natural da flavona que foi descoberto pela primeira vez no século XIX em trigo mourisco ou também chamado de trigo sarraceno. É um composto de baixo peso molecular amplamente distribuído em vegetais, frutas, bebidas derivadas de plantas, como o chá e vinho. Pouca ou nenhuma rutina dietética é absorvida intacta uma vez que a microflora intestinal metaboliza a rutina a uma variedade de compostos que podem ser absorvidos (YANG; GUO; YUAN, 2008). O banco de dados de suplementos alimentares lista mais de 860 produtos que contêm rutina atualmente comercializados nos EUA (GULLÓN et al., 2017).

Figura 3 – Estrutura química da rutina



Mais de 70 espécies de plantas contêm rutina. Trigo sarraceno (*Fagopyrum esculentum* Moench) da família Polygonaceae é uma importante fonte de rutina (KIM et al., 2005). O trigo sarraceno é de origem chinesa e tem um cultivo histórico devido ao seu interesse na medicina tradicional. Há relatos em vários livros agrícolas antigos e obras

medicinais (GULLÓN et al., 2017). No Japão, o trigo sarraceno é cultivado, não apenas para ser usado em alimentos tradicionais, mas também como ingrediente de alimentos saudáveis (SUZUKI et al., 2015). Outras fontes comerciais de rutina incluem *Ruta graveolens* L. (Rutaceae), *Sophora japonica* L. (Fabaceae) e *Eucalyptus* spp. (Myrtaceae) (CHUA, 2013).

A atividade antioxidante da rutina *in vitro* é comparável à do ácido ascórbico (YANG; GUO; YUAN, 2008) e superior à do alfa-tocoferol (FRANKEL; MEYER, 2000). A rutina tem sido amplamente utilizada no tratamento de doenças devido suas diversas atividades farmacológicas, incluindo atividade antialérgica (OLIVEIRA; FERNANDES; VIEIRA, 2006), anti-inflamatória e vasoativa, antitumoral, antibacteriana, propriedades antivirais e antiprotozoárias (CALABRÒ et al., 2005). Além disso, também foram relatadas atividades hipolipidêmica e citoprotetora (CASA et al., 2000), antiespasmódica e anticancerígena (WEBSTER; GAWDE; BHATTACHARYA, 1996) e prevenção da ulceração da mucosa gástrica em modelos animais (LIU et al., 2013). Outros estudos demonstraram um efeito dose-resposta da rutina na inibição da peroxidação das lipoproteínas de baixa densidade (LDL) (JIANG et al., 2007) e atividade antioxidante da rutina na reação de Fenton (CAILLET et al., 2007). Em um estudo *in vivo*, foi avaliada a inibição da peroxidação lipídica e a melhora de sua ação antioxidante no fígado, rim e cérebro de ratos diabéticos. Eles descobriram que a administração via oral de rutina (100 mg/kg) por um período de 45 dias teve um efeito antioxidante significativo na diabetes induzida experimentalmente por estreptozotocina (KAMALAKKANNAN; PRINCE, 2006).

Por outro lado, a principal desvantagem da rutina é a sua baixa biodisponibilidade, causada principalmente por sua baixa solubilidade em meio aquoso, baixa estabilidade e permeabilidade à membrana limitada (GULLÓN et al., 2017).

3.5 NANOTECNOLOGIA E A INDÚSTRIA DE ALIMENTOS

A demanda atual dos consumidores por produtos alimentícios saudáveis e mais naturais levou a um interesse crescente da indústria de alimentos na substituição de ingredientes sintéticos por outros mais naturais (WALKER, DECKER; MCCLEMENTS, 2015; MCCLEMENTS; BAI; CHUNG, 2017). Somado a isso, a indústria e os pesquisadores estão dando ênfase à nanotecnologia para abordar questões relevantes para a alimentação e nutrição, impulsionados pela necessidade de sistemas comestíveis capazes de encapsular, proteger e liberar compostos funcionais (SILVA, CERQUEIRA; VICENTE, 2012; ARANCIBIA et al., 2017).

Entre os conceitos de nanotecnologia existentes, destaca-se a ISO/TC 229, que a define como: 1) compreensão e controle da matéria e processos em nanoescala, em geral, mas não exclusivamente, abaixo de 100 nanômetros em uma ou mais dimensões, em que o aparecimento de fenômenos ligados ao tamanho geralmente possibilita novas aplicações; e 2) utilização de propriedades dos materiais em escala nanométrica que diferem das propriedades dos átomos, das moléculas e da matéria a granel, para o desenvolvimento de materiais, dispositivos e sistemas aprimorados que exploram essas novas propriedades. Ou seja, para um dispositivo ser considerado nanotecnológico, não basta ter dimensões nanométricas; ele deve apresentar propriedades diferenciadas associadas à nanoescala.

A nanotecnologia tem uma aplicação próspera em vários setores, e sua aplicação na ciência de alimentos é um evento recente. A segurança e a qualidade dos alimentos são uma grande preocupação, desta forma, os pesquisadores buscaram por várias tecnologias na tentativa de melhorar essas questões. Em especial, a aplicação da nanotecnologia na indústria alimentar levou à produção de alimentos com melhor estabilidade térmica, solubilidade e com maior biodisponibilidade oral. Além disso, a incorporação de elementos funcionais aos alimentos vem sendo pesquisada por um bom tempo, e a nanotecnologia preparou-se para isso, levando o desenvolvimento de nanoemulsões e nanocompósitos (HAMAD et al., 2018).

O aumento da vida de prateleira dos produtos alimentícios, o melhor rastreamento de contaminantes, o armazenamento aprimorado, a incorporação de suplementos saudáveis ou agentes antibacterianos nos alimentos, são alguns exemplos da grande contribuição da nanotecnologia na ciência dos alimentos (NEO et al., 2013). Independente do meio escolhido para a preservação é importante ressaltar que a qualidade e o sabor dos alimentos não devem ser interferidos e devem permanecer tão intactos quanto possível. A incorporação de nutracêuticos, a liberação de nutrientes, agentes de gelificação e viscosificantes, a fortificação com vitaminas e minerais, bem como a nanoencapsulação de sabor e aroma são algumas das formas de processamento de alimentos com nanomateriais (HUANG; YU; RU, 2010).

Vale salientar que o valor de mercado das embalagens de alimentos da indústria aumentou US\$ 2,5 bilhões em 2012 (SCOTT; CHEN, 2013). É pertinente notar que com esta tecnologia, o problema da escassez de alimentos em algumas partes do mundo pode ser resolvido com mais facilidade. Várias formas de nanossistemas, como nanopartículas sólidas, nanofibras, nanocápsulas e nanoemulsões são alguns dos nanomateriais que encontraram seu caminho nos setores de processamento de alimentos, embalagens e preservação (DUNCAN, 2011).

No desenvolvimento de embalagens, a aplicação da nanotecnologia pode fornecer diversas funções como: incorporação de nanopartículas com propriedades antimicrobianas; nanosensores capazes de detectar produtos químicos, agentes patogênicos e toxinas em alimentos; nanopartículas bioativas capazes de manter os compostos em condições ideais, até a sua migração para o produto alimentício; e nanocompósitos que melhoram as propriedades de flexibilidade, barreira a gases, umidade e quanto à absorção de irradiação UV dos materiais aos quais são incorporados, assim como a estabilidade frente à temperatura (ALMEIDA et al., 2015).

Entre os nanossistemas, neste trabalho foi dado ênfase às nanoemulsões, as quais são uma classe de emulsões com diâmetro de gotículas (O/A e A/O) na escala nanométrica. O tamanho de gotículas pode ser bastante variável, sendo considerados diâmetros de 1-100 nm (MASON et al., 2006), 20-200 nm (SOLANS et al., 2005), 20-300 nm (YILDIRIM, OZTOP; SOYER, 2017), 50-500 nm (CHAARI et al., 2018) e 50-1000 nm (EZHILARASI et al., 2013). Essas variações encontradas na literatura em relação ao diâmetro de gotícula se devem as diferenças na composição do sistema (tensoativos/ co-tensoativos, óleo e água) e no método de preparação utilizado (SALEM; EZZAT, 2018).

As nanoemulsões e microemulsões óleo-em-água são dispersões coloidais adequadas para encapsulação e entrega de componentes lipofílicos na indústria. É possível produzir tanto microemulsões como nanoemulsões, utilizando os mesmos componentes (óleo, água e tensoativo), porém usados em diferentes proporções, sendo que as nanoemulsões geralmente apresentam-se como um líquido opaco/ leitoso e as microemulsões possuem aspecto transparente/translúcido. A principal diferença entre essas duas formulações é que as nanoemulsões são termodinamicamente instáveis, enquanto as microemulsões são termodinamicamente estáveis. Contudo, deve salientar-se que as microemulsões são apenas termodinamicamente estáveis sob uma gama particular de composições e condições ambientais, e podem degradar se estas condições forem alteradas (por exemplo, devido à diluição, adição de ingredientes ou alterações de temperatura). Além disso, as microemulsões apresentam uma alta concentração de tensoativos, sendo uma característica negativa. (RAO; MCCLEMENTS, 2012).

As nanoemulsões, do ponto de vista físico-químico, são sistemas cineticamente estáveis. A alta estabilidade cinética pode ser obtida quando o método de preparação, a composição e os componentes do sistema são adequadamente selecionados (SOLANS; SOLÉ, 2012). A preparação pode ser realizada por uma variedade de métodos, classificados como métodos de alta energia e baixa energia. A abordagem de alta energia utiliza

dispositivos mecânicos capazes de gerar forças disruptivas intensas que conduzem à formação de gotículas de óleo, enquanto rompem-se as fases óleo e água (tais como, homogeneizadores de válvulas de alta pressão, microfluidizadores e métodos de sonicação). Já as abordagens de baixa energia dependem da formação espontânea de gotículas de óleo dentro de sistemas mistos de emulsionantes/óleo/água quando a solução ou as condições ambientais são alteradas, por exemplo, métodos de inversão de fase, deslocamento de solvente e emulsificação espontânea (MCCLEMENTS, 2012; SILVA; CERQUEIRA; VICENTE, 2012).

No presente trabalho, as nanoemulsões foram preparadas usando o mecanismo de emulsificação espontânea que ocorre quando uma fase orgânica e uma fase aquosa são misturadas, e após realiza-se a evaporação/eliminação do solvente orgânico. A fase orgânica é uma solução homogênea de óleo, tensoativo lipofílico e solvente miscível em água, a fase aquosa consiste em tensoativo hidrofílico e água. Destaca-se que a composição da fase orgânica inicial é de grande importância para o processo de emulsificação espontânea e, portanto, para as propriedades físico-químicas das emulsões obtidas (BOUCHEMAL et al., 2004).

A emulsificação espontânea pode ser utilizada para encapsular compostos bioativos na indústria de alimentos, bebidas e farmacêutica. Esse método possui várias vantagens em relação aos métodos de alta energia e outros de baixa energia, incluindo proteção de compostos sensíveis frente às condições severas dos métodos de alta energia, capacidade de minimizar a quantidade de tensoativo, remoção de cotensoativos e estabilidade térmica (MEHRNIA et al., 2016).

Na indústria de alimentos, as nanoemulsões são utilizadas na produção de edulcorantes, óleos aromatizados, molhos para saladas, bem como outros alimentos processados. A operação se dá a partir da liberação de diferentes aromas e sabores com vários estimulantes, como calor, pH, ondas ultrassônicas, dentre outros (HAMAD et al., 2018). Elas retêm as características de forma eficaz e impedem as reações enzimáticas, bem como as de oxidação. As nanoemulsões possuem várias vantagens em relação às emulsões convencionais, uma vez que são termicamente estáveis e de menor tamanho. Devido à sua grande área superficial, as nanoemulsões podem interagir com uma série de componentes biológicos, tais como as enzimas no trato gastrointestinal (TGI). Por exemplo, a lipase facilmente digere as pequenas gotículas de nanoemulsões no TGI (ZARIF, 2003). Além disso, as nanoemulsões sob a forma de carboidratos ou proteínas ajudam a melhorar a textura e levam à uniformidade do sorvete. Também, podem conter agentes antimicrobianos em sua composição, como os tensoativos não iônicos e alguns óleos essenciais, sendo utilizadas para evitar o crescimento

microbiano e conseqüentemente reduzir a deterioração dos alimentos, assim como, para fins de descontaminação de embalagens de alimentos (PRADHAN et al., 2015).

Até o momento, não foram encontradas pesquisas com nanoemulsões contendo extrato do cálice de *P. peruviana*, comprovando o ineditismo do estudo realizado. Somente, nanopartículas de prata (AgNPs) foram sintetizadas usando o cálice de *P. peruviana* (PATRA et al., 2018). Rashid e Sabir (2014) realizaram a biossíntese de nanopartículas de prata (AgNPs) utilizando o extrato aquoso das frutas comestíveis de *P. peruviana*. E em nosso grupo de pesquisa, Silva e colaboradores (2019) desenvolveram nanoemulsões contendo o extrato da fruta de *P. peruviana*. Alguns estudos de nanopartículas com outras espécies de *Physalis* e com outros focos de aplicação também foram encontrados. Nanopartículas de óxido de zinco (ZnO-NPs) foram sintetizadas usando *Physalis alkekengi* (QU et al., 2011). Assim como, nanopartículas de prata (AgNPs) foram sintetizadas com fotoindução verde utilizando o extrato aquoso das folhas de *Physalis angulata* (KUMAR et al., 2017).

3.6 NANOTOXICOLOGIA

Em relação a nanotecnologia, ainda há uma necessidade urgente de mais estudos farmacocinéticos e toxicológicos sobre os nanomateriais, especialmente antes de serem comercializados (MISSAOUI; ARNOLD; CUMMINGS, 2018). O perfil toxicológico dos nanomateriais engenheirados nos ecossistemas e na saúde humana deve ser cuidadosamente avaliado. A nanotoxicologia é uma área de pesquisa interdisciplinar dedicada à avaliação dos perigos associados à esses nanomateriais e está se expandindo rapidamente. Muitas técnicas físico-químicas e metodologias bioquímicas foram propostas e atualmente são usadas para caracterizar nanopartículas do ponto de vista toxicológico (BETTAZZI; PALCHETTI, 2019).

Testes preliminares são realizados *in vitro* como o de citotoxicidade (HILLEGASS et al., 2010). No entanto, há relatos que os ensaios *in vivo* seriam mais adequados para avaliar a biossegurança das nanopartículas. Porém, o custo mais alto e o tempo experimental mais longo dos ensaios *in vivo*, acabam limitando o uso da maioria das análises convencionais que utilizam mamíferos (WU et al., 2019).

O modelo animal que utiliza o nematódeo *Caenorhabditis elegans* (*C. elegans*) foi postulado em 1965 por Sydney Brenner, fornecendo aos pesquisadores a possibilidade de investigar os efeitos biológicos com custo mais baixo (BRENNER, 1974). Em termos de nanotoxicidade *in vivo*, a distribuição e liberação das nanopartículas, bem como seus impactos nos tecidos ou órgãos, são geralmente avaliados. *C. elegans* possui quatro sistemas orgânicos,

iguais aos dos vertebrados, incluindo o sistema neural, digestivo, imunológico e reprodutivo, os quais permitem que as descobertas com *C. elegans* sejam confiáveis e de grande valor. Além disso, esse modelo possui algumas vantagens, tais como, um ciclo de vida curto, um corpo minúsculo e transparente, fácil manuseio e informações genéticas bem caracterizadas. Foi o primeiro organismo multicelular a ter todo o seu genoma sequenciado e, a partir de 2012, foi o único organismo a ter o seu conectoma, isto é, diagrama de ligações neuronais concluído (WU et al., 2019).

Levando em consideração a temática abordada, neste estudo foi realizado o desenvolvimento de nanoemulsões contendo o extrato do cálice de *P. peruviana*. A estabilidade frente às diferentes condições de armazenamento foi analisada, quanto às características físico-químicas das formulações, bem como, a capacidade antioxidante e teor de rutina. Além disso, os aspectos de nanotoxicidade *in vitro* e *in vivo* foram avaliados e as formulações foram submetidas as condições gastrointestinais simuladas, a fim de avaliar as possíveis toxicidades e o perfil de liberação do composto bioativo majoritário nas diferentes etapas do trato.

**4 ARTIGO 1 – NANOENCAPSULATION OF BIOACTIVE COMPOUNDS:
CHALLENGES AND PERSPECTIVES**

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Nanoencapsulation of bioactive compounds: challenges and perspectives

Maiara Taís Bazana, Cristiane Franco Codevilla and Cristiano Ragagnin de Menezes

Nanotechnology is increasingly used in food science, and one of the lines of research is the nanoencapsulation of bioactive compounds. While these compounds promote improvements in human health, they are often improperly absorbed. Thus, nanostructured systems improve several characteristics, such as protection against degradation, solubility, stability, and bioavailability, among others. However, the development of nanostructures faces many challenges, from choosing the best method to obtain them to identifying the type of nanomaterial ideal for a bioactive compound of interest. In addition, the characterization of toxicological effects is sought by specific regulation for safety in human consumption and the environment, such as the use of green synthesis.

Address

Department of Food Science and Technology, Rural Science Center, Federal University of Santa Maria, Santa Maria, Brazil

Corresponding author:

de Menezes, Cristiano Ragagnin (cristiano.ufsm@gmail.com)

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Introduction

Nanotechnology is generally defined as the design, production, and application of structures, devices, and systems where the control of the size and shape of the material is on the scale of 10^{-9} m [1]. This technology has thriving applications in many sectors, and its rapid development has transformed many areas of food science, especially those involving processing, packaging, storage, transport, functionality, and other aspects of food safety [2]. In this sense, industry and academic researchers are emphasizing nanotechnology to address issues relevant to food and nutrition, driven by the need for edible systems capable of encapsulating, protecting, and releasing functional compounds [3,4*].

The current consumer demand for healthier and more natural food products has led to growing interest in the food industry in replacing synthetic with natural ingredients, such as bioactive compounds [5,6]. Thus, since many biologically active compounds are lipophilic and poorly soluble, nanoencapsulation facilitates the delivery of these poorly bioavailable compounds into functional food ingredients [7] by increasing their absorption into cellular structures through favorable particle properties of form, size, and surface. These properties allow nanoparticles to increase potential solubilization, alter absorption pathways by modifying the rate and site of release, influence gastrointestinal dispersion, and prevent premature metabolic degradation of a bioactive compound [8].

With gathering research in food nanotechnology, public concerns about the safety of these products for human consumption and use also arise [9]. Therefore, a comprehensive assessment of toxicity and potential health risks is essential before they are made commercially available. However, no universal guidelines have been specifically developed for the safety assessment of nanomaterials in food. In light of this, further studies and regulations on the impacts of these nanomaterials on human and environmental health must be conducted and established to ensure food safety [10*].

On the basis of what has been learned so far, this review will address the key benefits of, challenges for, regulatory aspects of, and future prospects for the nanoencapsulation of bioactive compounds, as well as examining different aspects of the toxicity and safety of nanocarriers.

Benefits of nanoencapsulation for food science and technology

Bioactive compounds, considered essential for human health, are mostly hydrophobic and poorly soluble. Examples include phenolic compounds, carotenoids, essential oils, essential fatty acids, and insoluble vitamins. The main challenges for the application of these compounds in the pharmaceutical and food industries include low bioavailability and stability [7].

The incorporation of functional elements in foods has long been researched. Nanotechnology is a useful tool for this purpose, with nanoemulsions and nanocomposites both being developed [11]. Nanoencapsulation is a favorable alternative to preserve and protect bioactive

compounds against inappropriate environmental circumstances and increase their bioavailability and stability, thus favoring their application in food and pharmaceutical products [7]. In addition, encapsulation decreases volatility and increases chemical and thermal stabilities; protects against oxygen, light, pH, moisture, and gastric digestion; masks unpleasant taste and aroma; promotes controlled release; improves the solubility of lipophilic compounds in aqueous media, and enables the prolonged absorption of nutrients; and may confer and facilitate the development of packaging materials with antimicrobial and antioxidant potential. All of these advantages are made possible by reducing the size of nanoparticles, thus increasing the area per unit volume [12] (Figure 1).

Increasing the shelf life of food products, better tracking contaminants, improving storage, and incorporating healthy supplements or antibacterial agents into food are just a few examples of nanotechnology's major contributions to food science [13]. Incorporation of nutraceuticals; release of nutrients, gelling agents, and viscosifiers; vitamin and mineral fortification; and flavor and aroma nanoencapsulation are some forms of food processing with nanomaterials [14].

Nanotechnological challenges

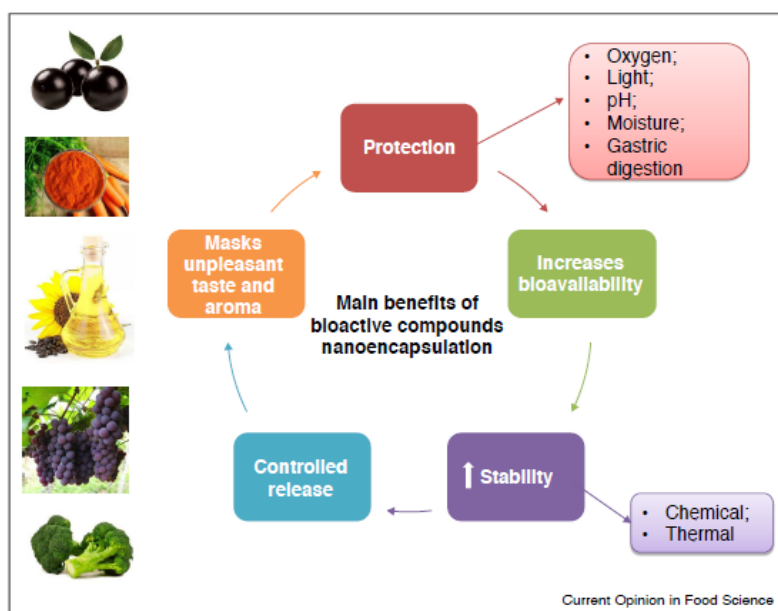
Methods of nanoencapsulation and green nanotechnology

Nanoencapsulation involves the encapsulation of substances in nanocarriers by absorption, incorporation,

chemical interaction, or dispersion [15,16]. Several nanoencapsulation techniques for bioactive compounds and nutraceuticals have been studied recently, including the production of different nanocarriers such as nanoemulsions; nanostructured lipid transporters; nanosuspensions; solid lipid nanoparticles (SLNs); liposomes and nanometer phytosomes; biopolymer SLNs; and micelles made of proteins, polysaccharides, or their complexes or conjugates [17]. In addition, the literature discusses other nanotransporters, such as inclusion complexes through cyclodextrins, amylose, and yeast cells, nanogels, nanofibers, nanosponges, or nanoparticles made with lipids and biopolymers [7] (Table 1). In light of this range of nanosystems, different technologies can clearly be used to nanoencapsulate food ingredients. The process requires some general equipment, including homogenizers, mills, and mixing devices, but implementing some techniques of nanoencapsulation requires specialized equipment, such as electrospunners, electrospray, nanospray dryers, and microfluidic devices [17].

Several natural and synthetic polymers are employed for nanoencapsulation. Natural encapsulating materials include polymers such as chitosan, alginates, cyclodextrins, and phospholipids, among others [30]. Synthetic polymers include biodegradable esters, such as poly(lactic-co-glycolic acid) (PLGA), polymers of poly-ε-caprolactone, and methacrylate [15]. Few studies have directly compared various methods or materials for a specific bioactive compound or ingredient, making it difficult

Figure 1



Benefits of bioactive compounds nanoencapsulation.

Table 1**Studies using different nanocarriers for the nanoencapsulation of bioactive compounds**

Type of nanocarrier	Method of preparation	Bioactive compound	Major findings	References
Nanocapsules	Spray drying and electrospraying	Folic acid	High encapsulation efficiency (83.9%). Improved folic acid stability under different storage conditions over 60 days.	[18]
Nanotubes	Nanotube synthesis	Caffeine	α -Lactalbumin nanotubes synthesized in the presence of Mn^{2+} were highly stable during freeze-drying. α -LA nanotubes were very efficient at caffeine encapsulation (around 100%). Low temperature and high pH prevented caffeine release from α -LA nanotubes.	[19]
Poly(ethylene oxide) (PEO) nanofibers	Electrospinning	Tea tree oil	The nanofibers were plasma-treated to modify their surface; after plasma treatment, the release efficiency of nanofibers and the antibacterial activity of tea tree oil improved. The nanofibers increased the shelf life of beef.	[20]
Zein nanofibers	Electrospinning	Fish oil and ferulic acid	Zein nanofibers exhibited nanoscale size of ~440 nm and showed high encapsulation efficiency (94%) and loading capacity of 20%. Loading ferulic acid into the nanofibers enhanced the oxidative stability of encapsulated fish oil. Fish oil loaded nanofibers showed an excellent release profile under gastrointestinal and enzymatic conditions.	[21]
Nanoemulsions	Low-energy emulsification method	Vitamin E	Revealed mean particle size around 86 nm; high encapsulation efficiency of near 100%; improved antioxidant and antimicrobial activity.	[22]
Cross-linked, pectin-coated solid lipid nano-particles (SLNs) and nanostructured lipid carriers (NLC)s	Solvent-diffusion and hot emulsification	Curcumin	SLNs and NLCs both exhibited high loading capacity. SLNs revealed excellent stability under storage conditions and in simulated gastrointestinal conditions.	[23]
Niosomes	Film-hydration method	α -Tocopherol	Niosomes were found on the nanometer scale (106.8–190 nm). The optimum condition for the encapsulation of α -tocopherol enhanced stability, encapsulation efficiency, and prolonged release.	[24]
Phytosome nanoparticles	Thin-film hydration method	Quercetin	The phytosome nanocapsules exhibited excellent encapsulation efficiency (98.4%). Phytosome nanocapsules enhanced the therapeutic benefits of quercetin in ovariectomized rats.	[25]
Nanogels	Self-assembly method	Rosemary essential oils	Particle size of less than 100 nm was achieved. The antibacterial activity of nanogel-encapsulated essential oils against <i>S. aureus</i> increased. Encapsulated essential oil in nanogels increased the transparency and tensile strength of starch-carboxy methyl cellulose films.	[26]
Biopolymer nanoparticles (nanospheres)	Thermal processing and electrostatic complexation of whey protein–pectin mixtures	Anthocyanins	Mean particle diameter was less than 200 nm. Exhibited high loading capacity before heating (55%). Enhanced thermal stability and lower antioxidant activity.	[27]
Nanoliposomes	Supercritical-assisted formation process	Olive pomace extract	The encapsulation efficiency of the supercritical-assisted liposome formation process was six times larger than the conventional thin-layer hydration method.	[28]
Cyclodextrin nanospheres (CDNS)	Solvent evaporation	Curcumin (CUR)	The photostability of CUR was enhanced 1.7 fold in pyromellitic dianhydride (PMDA)-crosslinked nanospheres. <i>In vitro</i> cytotoxicity study revealed increased toxicity of the drug nanosphere complex for MCF-7 cells at a lower concentration. PMDA-CDNS was found to be a potential nanocarrier for curcumin compared to di-phenyl carbonate (DPC)-CDNS.	[29]

to establish which method or material is superior for a specific bioactive compound [16]. Many encapsulation techniques and matrices can be applied to the same bioactive compound, which presents challenges for the development of nanoencapsulation. The choice of nano-carrier should be controlled by the nature of the bioactive compound, the presence of surface groups, temperature and pH during processing, release rate, and degree of cellular uptake, as well as economic considerations [12].

Nanoparticles can be synthesized using various methods, including (a) physical methods, such as mechanical grinding or milling; (b) chemical methods, such as coprecipitation or microemulsion; and (c) biological methods, using microorganisms, plants, or enzymes. Recently, various eco-friendly technologies to produce environmentally benign, non-toxic products are being developed using green nanotechnology and biotechnological tools. Nanoparticles synthesized using biological methods, which is also called green technology, have diverse natures, with greater stability and appropriate dimensions because they are synthesized through a one-step procedure [31].

The use of bacteria for the synthesis of nanomaterial has emerged as a novel approach for the synthesis of metal nanoparticles. Bacteria are known to produce inorganic materials either intracellularly or extracellularly. Microorganisms are considered as a potential biofactory for the synthesis of nanoparticles such as gold, silver, and cadmium sulphide [32]. Jayaseelan *et al.* [33] studied zinc oxide nanoparticles using bacteria *Aeromonas hydrophila* as eco-friendly reducing and capping agent. The synthesized ZnONPs had average particle size of around 57 nm, hydrophilic nature, highly stability and had significant antimicrobial activity.

The fungal mediated nanoparticle synthesis is a relatively recent research area. Fungi could be used as a source for the production of large amount of nanoparticles [32]. Chan and Don [34] selected and compared five species of fungi for their capability to synthesize Ag nanoparticles (AgNPs). Among the five fungi tested, *Pycnoporus sanguineus* and *Schizophyllum commune* were capable in synthesizing AgNPs with an average particle size of range from 52.8 to 103.3 nm, respectively.

The use of algae for the biosynthesis of nanoparticles is also possible. The ability of algae to accumulate metals and reduce metal ions makes them a good contender for the biosynthesis of nanoparticles [35]. Azizi *et al.* [36] synthesized zinc oxide nanoparticles using the brown marine macroalgae *Sargassum muticum* (*S. muticum*) aqueous extract. The ZnO NPs had hexagonal structures and the particle size ranges from 3 nm to 57 nm with a mean size of 42 nm.

The use of plant extract for synthesizing nanoparticles could be advantageous compared to other biological

methods, eliminating the elaborate process of maintaining cell cultures. While fungi and bacteria require a comparatively longer incubation time for the reduction of metal ions, water soluble phytochemicals do it in a much lesser time [32]. Extracts from plants may act as both reducing and stabilizing agents in the synthesis of nanoparticles (Table 2), perhaps due to the presence of phytochemicals. The main phytochemicals responsible have been identified as flavonoids, xanthophylls, carotenoids, anthocyanins, and phenolic acids [41].

Physical methods of synthesis frequently entail highly expensive instruments, high temperature and pressure, and high energy consumption. Chemical methods often require costly metal salts and toxic or hazardous solvents. Meanwhile, green methods synthesize nanoparticles through biological routes (microorganisms, plants, or viruses) or their byproducts (such as proteins and lipids) with the help of various biotechnological tools. Green nanotechnology has garnered wide interest due to its inherent features of rapidity, simplicity, eco-friendliness, and lower cost [31,46,47].

According to Srivastava and Bhargava [48], green nanotechnology has two key aspects. The first aspect involves nano-products that provide solutions for environmental challenges, and the second involves producing nanomaterials or products containing nanomaterials with an aim to minimize harm to human health or the environment. However, it is necessary to foster and promote the development of green nanotechnology for human and environmental sustainability. Its commercialization will require concerted effort from researchers, governments, and other stakeholders. The development of this environmentally friendly technology could greatly accelerate human welfare [32,48].

Toxicity of nanoparticles

Nanotechnology represents a very advanced field of research and application. Nanomaterials can be constituents in a wide range of manufactured commercial and domestic products [49–51]. Although beneficial, advances in nanotechnology are also associated with expectations of growing potential toxicity and ecotoxicity, largely due to nanoparticles' unusual properties (shape, small size, chemical composition, structure, and increased surface area) [7,51].

Considering their small size, nanoparticles can enter and translocate the circulatory and lymphatic systems, ultimately reaching bodily tissues and organs [52]. Some nanoparticles, depending on their composition and size, can irreversibly damage cells by oxidative stress or organelle injury. For example, small particle size and high specific surface area increases reactivity and facilitates their passage across biological barriers [51,52]. Understanding the impacts of nanomaterials for human health

Table 2

Nanoparticles synthesized using different plant extracts				
Reducing/stabilizing agent	Product	Average sizes	Comments	References
<i>Tecoma castanifolia</i> leaf	Zinc oxide nanoparticles	72.5 nm	Zinc oxide nanoparticles had antibacterial, antioxidant, and anticancer activity.	[37]
<i>Mangifera indica</i> leaves	Zinc oxide nanoparticles	47.7 nm	Zinc oxide nanoparticles exhibited cytotoxicity against lung cancer A549 cell lines and antioxidant activity.	[38]
Rosemary	Iron nanoparticles	50.5 nm	Iron nanoparticles showed potent cytotoxicity effects on 4T1 and C26 cancer cell lines.	[39]
<i>Pistacia Atlantica</i>	Gold nanoparticles	55.0 nm	Gold nanoparticles had great antibacterial and antioxidant effects.	[40]
<i>Alternanthera bettzickiana</i>	Gold nanoparticles	100.0 nm	Gold nanoparticles exhibited significant anti-microbial activity against <i>Bacillus subtilis</i> , <i>Staphylococcus aureus</i> , <i>Salmonella typhi</i> , <i>Pseudomonas aeruginosa</i> , <i>Micrococcus luteus</i> , and <i>Enterobacter aerogenes</i> .	[41]
Cinnamaldehyde	Nanoparticles	455.2 nm	Nanoparticles had high encapsulation efficiency and sustained release effect.	[42]
Turmeric	Silver nanoparticles	18.0 nm	Silver nanoparticles exhibited excellent antibacterial activity against foodborne pathogens (<i>Escherichia coli</i> O157:H7 and <i>Listeria monocytogenes</i>).	[43]
Olive leaf	Silver nanoparticles	30.0 nm	Silver nanoparticles significantly inhibited bacterial growth against multi-drug-resistant <i>Staphylococcus aureus</i> , <i>Pseudomonas aeruginosa</i> , and <i>Escherichia coli</i> .	[44]
<i>Viburnum opulus</i> fruit	Silver nanoparticles	25.0 nm	Silver nanoparticles presented potent anti-inflammatory activity.	[45]

and the environment is an important aspect of the development of nanotechnology [50].

According to Sia [49], nanomaterials or nanoparticles released into the atmosphere, soil, and surface water can persist in the environment for a long time or be absorbed by organisms, offering an eco-toxicological risk and biodegrading or bioaccumulating across the food chain. Nanomaterials consumed by humans can have a biological fate in the gastrointestinal tract. Nanomaterials should be completely digested and absorbed; if partially digested, encapsulated compounds may be incompletely released or it are resistant to digestion and either the compounds themselves or the nanomaterials may be thrown from the digestive system or cross to the epithelium, entering the blood stream and resulting in immunological reactions [7]. Nanomaterials can also be absorbed through inhalation or through the skin. They can cause oxidative and inflammatory stresses and cardiovascular or extrapulmonary complications, generating bioaccumulation and toxicity to human health [49,53].

Exact quantification of the release of nanoparticles into the environment and occupational exposure is quite a challenge. There is an urgent need for information to better understand nanoparticle–biological interactions and processes [53,54]. While a steady stream of *in vitro* studies indicates promising directions, *in vivo* studies of nanoparticles have not yet established a unified system (Table 3). *In vivo* evaluation, one step closer to clinical application, more directly reflects organism's adaptation and injury. Despite some large-scale progress in the field of nanotoxicology, there remains a void between

validation and evidence-based studies [53,54]. According to Naseer *et al.* [53], the current toxicological assays need to be revamped and new tools (such as proteomics, functional genomics, high-throughput screening, and metabolomics) should be increasingly incorporated into these studies. These advanced tools will minimize the number of false positives and accelerate and validate the evaluation of the toxicity of nanoparticles.

Nanomaterials and consumers

Currently, the use of nanomaterials is gaining increasing importance in industrial applications, including the food industry, as a way to provide new benefits to consumers and additional functionalities [63].

In 2011, according to the European Regulation Concerning the Provision of Consumer Food Information (N^o 1169/2011), the European Commission decided to label any ingredient containing nanomaterials as such, followed by the word 'nano' in parentheses. This regulation, which entered into force on 13 December 2014, seems to be difficult to apply in agreement with several consumer associations. Some French consumer associations, with the help of the French National Laboratory of Metrology and Testing (LNE), demonstrated the presence of nanomaterials in several products, with no labels on the list of ingredients [64*].

Nongovernmental organizations from different countries have widely reported on research studies showing some negative effects of nanomaterials. Two of the latter are the study by Bettini *et al.* [65] and Schoepf *et al.* [66], among others, shown in Table 4, forcing the Australian

Table 3

Selected examples of nanoparticles assessment by *in vitro* and *in vivo* studies

Type of studied nanoparticle	Type of approach employed	Main conclusion	References
Dihydromyricetin nanocapsules (NC-DMY)	<i>In vitro</i> toxicity, cell viability assay by MTT (peripheral blood mononuclear cells) for genotoxic effects, comet assay for interaction with DNA	NC-DMY did not demonstrate cytotoxicity and genotoxicity, presenting high interaction with DNA <i>in vitro</i> , suggesting DNA sequestration.	[55]
Pomegranate seed oil nanoemulsions (NE PSOB)	<i>In vitro</i> toxicity, cell viability assay by MTT (peripheral blood mononuclear cells) for genotoxic effects, comet assay	The results demonstrated no cytotoxicity or genotoxicity. NE PSOB increased cell viability by about 50%.	[56]
Kaempferol-loaded nanoemulsions (KPF-NE)	Isolated porcine nasal mucosa	KPF-NE caused no structural damage toward the nasal mucosa. The epithelial layer was intact, indicating that the formulation is biocompatible and safe for nasal administration.	[57]
Curcumin nanoemulsion (CNE) and curcumin mucoadhesive nanoemulsion (CMNE)	Human neuroblastoma cell line, SK-N-SH cells	The results suggest that CNE and CMNE did not inhibit growth and were non-toxic to the cells, suggesting they are safe for brain delivery.	[58]
Gold nanoparticles synthesized using leaf extract of <i>Alternanthera bettzickiana</i>	Zebrafish (<i>Danio rerio</i>) embryo model	Gold nanoparticles at 25 μm concentration caused low inhibition of hatching after exposing embryos. At 50 μm concentration, gold nanoparticles strongly inhibited hatching after treatment of 48 hour. The tail formation in treated zebra fish embryo was more affected than the control, and large amounts of dark material were found in the gut tract of treated embryos, but not in the control.	[41]
Chitosan-coated nanoliposome for encapsulating the flavonoid quercetin	<i>In vitro</i> toxicity in HepG2 cells, cell viability assay by MTT	At concentration of 10mg/mL, the encapsulated quercetin exhibited comparatively lower cell viability (40.92%) than native quercetin (46.67%).	[59]
Silver nanoparticle synthesized using black tea extract	Cytotoxicity in human ovarian (A2780) and colorectal (HCT116) carcinoma cell lines	The AgNPs show higher cytotoxic activity against A2780 when compared to HCT116. The observed IC50 in A2780 cells is 10 times lower than that for normal human primary fibroblasts.	[60]
Nanosilver produced by two macroalgae: <i>Sargassum boveanum</i> and <i>Ulva flexuosa</i> extracts	<i>Daphnia magna</i>	AgNPs from <i>U. flexuosa</i> was less toxic and more environmental friendly than AgNPs synthesized from <i>S. boveanum</i> . Both showed significant lower toxicity than chemical nanosilver.	[61]
Copper oxide nanoparticles synthesized using aqueous extract of <i>Pterospermum acerifolium</i> leaves	<i>Daphnia magna</i>	The results indicated that plant-synthesized CuO NPs are less toxic than engineered CuO NPs.	[62]

food safety agency and the French food safety agency to report on titanium dioxide and apatite (tricalcium phosphate), respectively. Unfortunately, non-governmental organizations did not report contradictory studies, leading to a climate of fear of nanotechnology by consumers and the implicit link between nanomaterial and toxicity, despite the message from the European Commission that explicitly stated that a nanomaterial is not intrinsically dangerous. Consumers are often unaware that food additives have been subjected to toxicological testing before food authorization and are regularly reassessed. As a consequence, several food industries have announced the ban on some food additives from their food products to maintain the confidence of their consumers [64*].

The reasons for this low market penetration are partly linked to the limited consumer acceptance and remaining

uncertainties regarding safety aspects and legislation [63]. To assist in these studies the projective methods are welcome and essential, these constitute a group of qualitative sensory techniques, which are increasingly used in research to investigate the perception of consumers. This group of techniques allow a condition of free interpretation and response to an ambiguous stimulus, seeking to understand the motivations, feelings, beliefs, attitudes and the deep reasons to select alternatives and answers within a specific topic [72].

Giles *et al.* [73] carried out a systematic review methodology to synthesize current knowledge regarding the acceptance or rejection of society regarding nanotechnology applied to agri-food production. They found that nanotechnology is more likely to be accepted by consumers when applied to the development of packaging

Table 4

***In vitro* and *in vivo* toxicity studies of nanoparticles**

Nanoparticle	Model	Effect/conclusion	References
Silver nanoparticles (AgNPs)	<i>Drosophila melanogaster</i>	Lethal doses of dietary AgNPs caused developmental delays and profound lethality in developing animals and young adults. Sublethal doses were not deadly to developing animals but shortened the adult lifespan and compromised their tolerance to oxidative stress. Dietary AgNPs caused a variety of ROS-mediated stress responses, including apoptosis, DNA damage, and autophagy.	[67]
Zinc oxide (ZnO) and titanium dioxide (TiO ₂) nanoparticles	Human lung epithelial cells (A549)	The results demonstrated that both materials generate reactive oxygen species (ROS), but nano-ZnO might also release zinc ions into the cell culture medium, potentially resulting in greater cell damage. Moreover, nano-ZnO has positive charge and can usually penetrate cells more readily and subsequently harm them than do negatively charged NPs, as nano-TiO ₂ .	[68]
Titanium dioxide nanoparticles (TiO ₂ NPs)	L929 fibroblasts	TiO ₂ NPs decreased cell viability and GSH and SOD. Increased ROS production.	[69]
Gold/citrate nanoparticles	Human dermal fibroblasts	Nanoparticles decreased cell proliferation rate, adhesion, and motility.	[70]
Zinc oxide nanoparticles	Female Wistar rats	A single installation of ZnONP induced alveolar and interstitial inflammatory cell infiltration, including eosinophilia, airway epithelial cell injury, regenerative goblet cell hyperplasia, bronchocentric pulmonary fibrosis, and atelectasis.	[71]

with distinct benefits and not when integrated directly into agri-food products. Trust and confidence in agri-food nanotechnology and its governance must be fostered through transparent regulation and development of socially beneficial impacts to increase consumer acceptance. Providing information to consumers about the benefits of nanotechnology, and securing public information could help reduce consumer concern and inspire purchases of food nanotechnology. However, surveys are needed to understand what consumers perceive to be beneficial as well as how they interpret risks. Adoption of theoretically sustained approaches to understanding consumer perceptions and attitudes will facilitate comparative analysis across different consumer groups, different food applications of nanotechnology, and enable the assessment of trends, consumer priorities and concerns over time [73].

Regulatory aspects and future perspectives

Safety and health concerns, as well as regulatory policies, should be considered during the manufacture, processing, and active packaging and consumption of nanoprocessed food products [2]. Regarding the regulatory aspects of nanostructures in the food and medical sectors, no specific legislation is applied globally. Most countries still have no specific regulations for the risk assessment of encapsulated nano-products. The lack of consistency in the exchange of information among countries is a reported risk to human health and the environment and may limit the marketing of new and beneficial products around the world [74]. Recently, European Union regulations have established that any food ingredient derived from

nanotechnology applications must undergo a safety assessment before receiving approval for its use [75]. Some guidelines have also been released by the U.S. Food and Drug Administration regarding the use of nanotechnology in food [10]. In an article published in the prestigious journal, *Nature Nanotechnology*, Associate Professor Steffen Foss Hansen [76] from DTU Environment called for the adoption of a new regulatory framework for nanomaterials. This framework is named REACT NOW, standing for Registration, Evaluation, Authorisation, Categorisation and Tools to Evaluate Nanomaterials – Opportunities and Weaknesses. All uses of nanomaterials should be evaluated according to the NanoRiskCat methodology [76].

Since there are no specified regulations for nanofoods, a database for those products currently released in the European market was established in 2012 [77]. To date, the database contains information about 3036 products, 128 of which concern nano-engineered food and beverages currently in the European market (www.nanodb.dk).

Further studies are needed to overcome the limitations of nanoencapsulation processes and to improve existing methods, formulations, and encapsulation systems, as well as to meet commercial demands and to explore the application of nano-products in food and gastrointestinal systems and their production at industrial scale. In addition, studies should emphasize the application of nanocarriers of bioactive compounds in food and biological systems, exploring their effects on cell viability and their adsorption, distribution, metabolism, and excretion

profiles in humans and other living systems [74*]. The prospects for the nanoencapsulation of bioactive compounds highlight the need for their worldwide regulation to facilitate their safe use and marketing. Various studies reporting the beneficial effects of nanostructured bioactive compounds support this trend and offer a promising future direction for research.

Conclusion

Nanotechnology applied to the encapsulation of bioactive compounds has several evident advantages from food processing to packing, such as improvements in stability and bioavailability. It also offers health benefits through the protection and controlled release of bioactive compounds. However, understanding the safety and toxicity of these nanomaterials requires more research, and global regulations must be drafted to promote the safe marketing of new nanotechnology products that are beneficial to health.

Conflict of interest statement

Nothing declared.

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5 ARTIGO 2 – DEVELOPMENT OF NANOEMULSIONS CONTAINING *Physalis peruviana* CALYX EXTRACT: A STUDY ON STABILITY AND ANTIOXIDANT CAPACITY

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Development of nanoemulsions containing *Physalis peruviana* calyx extract: A study on stability and antioxidant capacity



Maiara Taís Bazana^a, Suelen Santos da Silva^a, Cristiane Franco Codevilla^a, Cassandra de Deus^a, Bruna Nichelle Lucas^a, Gustavo Andrade Ugalde^b, Marcio Antonio Mazutti^b, Erico Marlon Moraes Flores^c, Juliano Smanioto Barin^a, Cristiane de Bona da Silva^d, Cristiano Ragagnin de Menezes^{a,*}

^a Department of Food Science and Technology, Rural Science Center, Federal University of Santa Maria, Santa Maria, Brazil

^b Department of Chemical Engineering, Technology Center, Federal University of Santa Maria, Santa Maria, Brazil

^c Department of Chemistry, Center for Natural and Exact Sciences, Federal University of Santa Maria, Santa Maria, Brazil

^d Department of Industrial Pharmacy, Health Sciences Center, Federal University of Santa Maria, Santa Maria, Brazil

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ABSTRACT

The aim of this study was to develop and evaluate the physicochemical and antioxidant stability of nanoemulsions containing a *Physalis peruviana* calyx extract (CPp-NE) and free extracts under different storage conditions (7 and 25 °C) and with absence or incidence of light for 120 days. The calyx extracts were prepared with ethanol 60% and characterized for later preparation of the nanoemulsions by spontaneous emulsification. The formulations presented nanometric sizes, low polydispersity index, negative zeta potential, acid pH, rutin content (11 µg mL⁻¹), and encapsulation efficiency of 85%. Regarding the stability, the droplet size and PDI of the CPp-NE stored at refrigeration temperature in the dark, room temperature in the dark, and refrigeration temperature with light incidence were stable for 120 days and with no visible changes in the formulations. The antioxidant capacity was related to the reducing capacity, and the best results were found for nanoemulsions stored at room temperature and in absence of light. In addition, CPp-NE presented higher antioxidant and reducing capacity in relation to the free extracts.

1. Introduction

Physalis peruviana Linnaeus is a highly functional fruit belonging to the *Solanaceae* family and genus *Physalis*. Its cultivation has expanded in tropical and subtropical countries, with Colombia being the largest producer in the world followed by South Africa, countries of which export fruit to other countries of America and most importantly Europe (Rufato, Rufato, Lima, & Muniz, 2013). *P. peruviana* is an herbaceous, semi-shrub, vertical, and perennial plant that adapts in subtropical areas. It produces orange-yellow fruit with a juicy berry and has a calyx, which is a casing that naturally protects the fruit during its development and maturity, protecting it against insects, birds, diseases, and adverse climatic conditions. The calyx represents an essential source of carbohydrates during the first 20 days of fruit growth and development (Puente, Pinto-Muñoz, Castro, & Cortés, 2011). In addition, it is a residue rich in bioactive compounds, such as flavonols and

hydroxycinnamic acid derivatives (Gironés-Vilaplana et al., 2014).

P. peruviana calyces are widely used in folk medicine due to their anticancer, antimicrobial, antipyretic, diuretic, immunomodulatory, and anti-inflammatory properties (Franco, Matiz, Calle, Pinzón, & Ospina, 2007), which are well established and reported in scientific studies using in vivo models (Castro, Ocampo, & Franco, 2015; Franco et al., 2007; Franco et al., 2014). Moreover, Toro, Aragón, Ospina, Ramos, and Castellanos (2014) verified that the crude ethanolic extract and some calyx fractions demonstrated promising anti-inflammatory and antioxidant activity.

According to Cömert and Gökmen (2018), food antioxidants have been extensively studied in food science because they provide health benefits in addition to the preservative effects on food, as there is growing interest among food scientists in developing in vitro and in vivo assays to improve knowledge on the physiological effects of antioxidants in the human body. The authors also reported that the

* Corresponding author at: Departamento de Tecnologia e Ciência dos Alimentos, Programa de Pós-Graduação em Ciência e Tecnologia dos Alimentos, Centro de Ciências Rurais, Universidade Federal de Santa Maria, Av. Roraima, 1000, CEP 97105-900 Santa Maria, RS, Brazil.

E-mail addresses: juliano@ufsm.br (J.S. Barin), csbona@smail.ufsm.br (C. de Bona da Silva), cristiano.ufsm@gmail.com (C.R. de Menezes).

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evolution of research on antioxidants by covering their occurrence, roles in various foods, and analytical methods for determining their antioxidant capacity and physiological effects.

The use of plant extracts or their derivatives is a growing trend for application in food and beverages in the food industry (Perumalla & Hettiarachchy, 2011; Valduga, Gonçalves, Magri, & Finzer, 2019). Oftentimes, these extracts are obtained by plant residues, which is a relevant factor for their use, as they are by-products and may present high nutritional values and economically viable reutilization. Therefore, the importance of these products is highlighted since they often represent a varied source of compounds with biological activity, therefore, being beneficial to health (Bilal & Iqbal, 2019).

However, these substances with excellent functional properties have several limiting elements that hinder their application in food for oral administration, including poor water solubility, low absorption, and bioavailability (Shin, Kim, & Park, 2015). Among these substances, rutin is one of the flavonoid compounds present in the highest quantity in the *P. peruviana* calyx in addition to having these disadvantages (Gullón, Lú-Chau, Moreira, Lema, & Eibes, 2017). Moreover, the inclusion of some bioactive compounds in food products may cause quality defects, such as astringent taste. Furthermore, they can act as substrates for darkening reactions, degradation, and/or deterioration by food processing and storage operations, such as heating, acidification, light, and oxygen (Codevilla, Bazana, Silva, Barin, & Ragagnin de Menezes, 2015). As a result, some extracts are not used due to such obstacles (Bonifácio et al., 2014).

To overcome the limitations of adding plant extracts or bioactive compounds to food products, innovative technologies, such as nanotechnology, become a promising alternative. In the food industry, the use of nanostructured systems can contribute to the processing, packaging, and preservation of food, in addition to providing longer shelf life to most products (Hamad, Han, Kim, & Rather, 2018). Nanoemulsions are one of the most interesting delivery systems in food industry. They are emulsions that have very small particle sizes, increased surface area, and less sensitivity to physical and chemical changes. These phenomena are thermodynamically unstable and can be kinetically stable over long time scales. Furthermore, nanoemulsion droplet sizes (O/W and W/O) are reported to be 20–500 nm, which is variable based on the composition of the system (surfactants/co-surfactants, oil, and water) and homogenization method used (Salem & Ezzat, 2018).

Nanoemulsions stand out among nanocarriers as a promising tool to develop delivery systems of lipophilic active compounds with the aim of facilitating their incorporation into food. Encapsulating compounds of natural origin in nanoemulsions may improve their stability and functional performance in either food matrices or gastrointestinal conditions (Acevedo-Fani, Soliva-Fortuny, & Martín-Belloso, 2017). Furthermore, nanoemulsions containing plant extracts and/or oils have shown advantages such as increased solubility, antioxidant efficiency, and slower phenolic compound release (Rezaei, Fathi, & Jafari, 2019). Other benefits also include increased stability and protection against degradation (Silva Junior et al., 2013), better antimicrobial activity (Paudel, Bhargava, & Kotturi, 2019), and improved stability through simulated gastrointestinal fluids (Li, Cai, Yang, & Sun, 2019).

In this context, the aim of this study was to develop nanoemulsions containing the *Physalis peruviana* calyx extract and evaluate extract and nanoemulsion stability under different storage conditions.

2. Material and methods

2.1. Material

The *P. peruviana* calyces were obtained from Italbraz (Vacaria, Brazil - 28:0:44 S and 50:56:02 W) in the 2016/2017 harvest. Sorbitan monooleate (Span 80[®]) was purchased from Fluka (São Paulo, Brazil), medium chain triglycerides (MCT) from Delaware (Porto Alegre,

Brazil), polysorbate 80 (Tween[®] 80) from Neon Comercial Ltda (São Paulo, Brazil), and Folin-Ciocalteu was purchased from Cromoline Química Fina Ltda (São Paulo, Brazil). Fluorescein sodium was purchased from Synth (São Paulo, Brazil), peroxy radical generator (AAPH: 2,2'-azobis-2-methyl-amidinopropane dihydrochloride) and standard antioxidant (Trolox: 6-hydroxy-2,5,7,8-tetramethylchromo-2-carboxylic acid 97%) from Sigma-Aldrich (USA). The other reagents were of PA grade for spectrophotometer analysis and HPLC grade for high performance liquid chromatography analyses.

2.2. Methods

2.2.1. Extraction

The calyx extracts were prepared following a modified version of the methodology proposed by Codevilla et al. (2018). The *P. peruviana* calyces (3 g) were separated from the fruit, submitted to knife milling (Mill MA 630/1-Marconi) (speed = 5 rpm, time = 10 s), added with 50 mL of 60% ethanol (v/v), and submitted to stirring in a homogenizer (Shaker model TE 421- Tecnal, Piracicaba, SP, Brazil) at a speed of 200 rpm for 2 h. Afterwards, the extracts were filtered in a qualitative filter (Fitec) for coarse separation of the calyx. Subsequently, the extract was subjected to filtration in a polypropylene syringe (0.22- μ m pore size) (Chromfilter) and kept in a refrigerated environment.

2.2.2. Characterization of the *P. peruviana* calyx extract

The extract was characterized in terms of phenolic compounds via quantification by high performance liquid chromatography (HPLC), where the standards of caffeic acid, chlorogenic acid, rutin, quercetin, gallic acid, catechin, coumarin, luteolin, and kaempferol were tested. In addition, the reduction capacity by Folin-Ciocalteu and antioxidant capacity by Oxygen Radical Absorbance Capacity (ORAC) were evaluated.

2.2.3. Nanoemulsion preparation

Nanoemulsions were prepared in triplicate by spontaneous emulsification based on the technique of Bouchemal, Briçon, Perrier, and Fessi (2004). The organic phase was composed of sorbitan monooleate (Span 80[®] - 0.1925 g), medium chain triglycerides (MCT - 0.400 g), *P. peruviana* calyx extract (2.083 mL), and acetone (65 mL) was stirred in a magnetic stirrer until complete solubilization of the components. The organic phase was injected, under stirring, into the aqueous phase consisting of polysorbate 80 (Tween 80[®] - 0.1925 g) and ultrapure water (135 mL). Magnetic stirring was maintained for 10 min. Then, the formulation was taken to the rotary evaporator at 40 °C to remove the organic solvent and adjusting of the extract concentration to 5 mg.mL⁻¹ and final volume to 25 mL. For comparison purposes, the formulations without the extract were also prepared and evaluated (blank nanoemulsions - NE).

2.2.4. Physicochemical characterization of nanoemulsions

2.2.4.1. PH determination. The pH determination was performed directly on the nanoemulsions using a potentiometer (MS Tecnopon) previously calibrated with buffer solutions pH (4.0 and 7.0). Droplet size distribution was evaluated by water dispersion of the samples until the obscuration index was around 13% using the refractive index of the MCT = 1.461 and density 0.98 g.cm⁻³ by laser diffraction (Mastersizer[®] 3000E, Malvern, UK). Determination of the mean droplet diameter and polydispersity index (PdI) was performed by Zetasizer[®] (Nano series, ZEN 3600, Malvern Instruments, UK) using photometric correlation spectroscopy. For this, the samples were diluted in ultrapure water in the proportion of 1:500 (v/v).

2.2.4.2. Determination of the zeta potential. the nanoemulsions were dispersed in a 10-mM NaCl solution (1:500, v/v) for further analysis of the zeta potential by electrophoretic mobility in the same equipment used to determine the mean diameter (Zetasizer[®]).

2.2.4.3. Morphological analysis. The morphology of the nanodroplets was evaluated by scanning electron microscopy (SEM) (Sigma 300 VP, Zeiss, Germany). For this, the sample was homogenized with 10% trehalose (cryoprotectant), frozen in a freezer at -18°C (Electrolux FE 26) for 24 h, and then lyophilized (LIOTOP L101, Brazil) for 24 h. The lyophilized nanoemulsion sample was stuck with a double-sided tape in aluminum stubs and covered by a thin layer of carbon. The shape and surface of the droplets were analyzed.

2.2.4.4. Extract content and encapsulation efficiency determination (EE). The content of the extract was determined by rutin analysis because, among the tested standards, this was the major compound according to high performance liquid chromatography (HPLC) following a modified version of the methodologies proposed by Gironés-Vilaplana et al. (2014) and Medina et al. (2019). For this, an analytical curve with the rutin standard of 0.1 to $5\ \mu\text{g}\cdot\text{mL}^{-1}$ was performed. Shimadzu Nexera XR chromatograph equipped with a SPD-M20A UV-DAD detector set at $370\ \text{nm}$, LC-20 CE pump module, and CTO-20A column oven (Shimadzu, Kyoto, JP) was used. As the stationary phase, the C18 Luna[®], $5\ \mu\text{m}$, $100\ \text{\AA}$ reverse phase column ($4.6\ \text{i.d.} \times 250\ \text{mm}$) (Phenomenex, Torrance, CA, USA) was used. The injection volume was $50\ \mu\text{L}$ and the elution gradient was composed of 0.45% formic acid in water (solvent A) and acetonitrile (solvent B) with the mobile phase at a constant flow rate ($0.6\ \text{mL}\cdot\text{min}^{-1}$). The linear gradient for solvent B was 0 min, 5%, 15 min, 100%, and 3 final min of column conditioning. For rutin extraction, 2-mL aliquots of nanoemulsions and 0.167-mL of extracts were diluted in 10 mL of methanol ($1\ \text{mg}\cdot\text{mL}^{-1}$ of calyx extract) and submitted to 20 min of ultrasound and 10 min of centrifugation ($48,000\ \text{g}$, 10°C). Afterwards, the solutions were filtered in a $0.22\text{-}\mu\text{m}$ membrane and injected into the chromatograph. Encapsulation efficiency was determined by a centrifugation-ultrafiltration technique (AMICON 10000 MW Ultra, Millipore, Bedford, USA). An aliquot of nanocapsules ($400\ \mu\text{L}$) was used and submitted to centrifugation-ultrafiltration at $2200\ \text{g}$. Rutin concentration in the ultrafiltrate was determined and encapsulation efficiency was calculated through the difference between total rutin concentration and ultrafiltrate concentration (free concentration).

2.2.5. Antioxidant capacity

2.2.5.1. Reducing capacity. The *P. peruviana* calyx extract and nanoemulsions were diluted in water to a concentration of $2\ \text{mg}\cdot\text{mL}^{-1}$. Afterwards, they were subjected to ultrasound for 20 min and centrifugation for 10 min at $48000\ \text{g}$. Subsequently, 1 mL of the sample was pipetted, followed by $0.250\ \text{mL}$ of the Folin-Ciocalteu reagent, followed by waiting for 5 min for the reaction. Next, 2 mL of 20% sodium carbonate was added and another 10 min was waited. The nanoemulsion samples were centrifuged for 10 min at $48000\ \text{g}$ in order to eliminate turbidity. Subsequently, spectrophotometer reading was performed at a wavelength of $730\ \text{nm}$. A standard curve prepared with gallic acid was used to quantify the reducing capacity (mechanism based on electron transfer) and the results are expressed in mg gallic acid Equivalent per 100 g of sample ($\text{mgGAEq}\cdot 100\ \text{g}^{-1}$) (Chandra & Mejia, 2004).

2.2.5.2. Oxygen radical absorbance capacity (ORAC). The antioxidant capacity was evaluated by the peroxy radical deactivation capacity method via the Oxygen Radical Absorbance Capacity (ORAC) method described by Ou, Hampsch-Woodill, and Prior (2001). The capturing capacity of the *P. peruviana* calyx (present in the extracts and nanoemulsions) was verified against the formation of a peroxy radical induced by 2,2'-azobis (2-amidinopropane) dihydrochloride (AAPH) at 37°C . The samples were diluted in phosphate buffer with pH 7.4 ($75\ \text{mmol}\cdot\text{L}^{-1}$) before being added to the microplate, subjected to ultrasound for 20 min, and centrifuged for 10 min at $48000\ \text{g}$. In a 96-well microplate, $25\ \mu\text{L}$ of diluted *P. peruviana* extract and nanoemulsion ($300\ \text{mg}\cdot\text{L}^{-1}$) in phosphate buffer pH 7.4 ($75\ \text{mmol}\cdot\text{L}^{-1}$) were added,

followed by incubation in a microplate, SpectraMax M5 (Molecular Devices, CA USA) for 10 min at 37°C with $150\ \mu\text{L}$ fluorescein working solution ($81\ \text{nmol}\cdot\text{L}^{-1}$). After incubation, $25\ \mu\text{L}$ of AAPH ($152\ \text{mmol}\cdot\text{L}^{-1}$) was added to form peroxy radicals. Fluorescence was monitored every minute (excitation and emission wavelengths of 485 and $528\ \text{nm}$, respectively) for 120 min at 37°C . Antioxidant capacity was determined using the area under the curve (AUC) for decay of sample fluorescence, standard, and blank versus time and the results were compared with a standard Trolox curve ($0\text{--}96\ \mu\text{mol}\cdot\text{L}^{-1}$) and expressed as $\mu\text{mol Trolox}\cdot\text{g}^{-1}$ calyx. The AUC was calculated as follows:

where

$$\text{AUC} = 1 + \frac{f^1}{f_0} + \frac{f^2}{f_0} + \frac{f^3}{f_0} + \dots + \frac{f^n}{f_0}$$

f^1 ... f^n : determined fluorescence every minute. f_0 : fluorescence at time zero.

2.2.6. Stability study

The stability study was carried out following a modified version of the methodologies proposed by Pedreño & Escibano, 2001, Xiu-li, Xue-li, Yuan-ping, & Qiang, 2015, Marques da Silva et al., 2018 and Nunes et al., 2018. The nanoemulsions and extracts from the *P. peruviana* calyx were packed in transparent flasks (exposed to light) and amber flasks wrapped in aluminum foil (absence of light). In addition, they were stored at room temperature ($25 \pm 0.5^{\circ}\text{C}$) or refrigerated ($7 \pm 0.5^{\circ}\text{C}$). Regarding the presence of light, a cold white lamp (luminous flux of $343\ \text{lm}$, power of $7\ \text{W}$, and light efficiency of $49\ \text{lm}\cdot\text{W}^{-1}$) was placed about $40\ \text{cm}$ above the samples. Stability evaluation was performed shortly after preparation (time zero) and over a period of 120 days (15, 30, 45, 60, 90, and 120 days). The parameters evaluated were: pH, mean droplet diameter, zeta potential, rutin content, reducing capacity, and antioxidant capacity.

2.2.7. Statistical analyses

The experiments were performed in triplicate. The results were evaluated by analysis of variance (ANOVA) and the means were compared by the Tukey test considering the level of significance of 5% ($p < 0.05$).

3. Results and discussion

3.1. Characterization of the *P. peruviana* calyx extract

The *P. peruviana* calyx extracts were characterized after preparation. Regarding the reducing capacity, the result was $610 \pm 20\ \text{mg gallic acid Eq}\cdot 100\ \text{g}^{-1}$ calyx. Regarding the antioxidant capacity by ORAC, they presented $137.93 \pm 2.10\ \mu\text{mol of Trolox}\cdot\text{g}^{-1}$ of calyx.

According to Ballesteros-Vivas, Alvarez-Rivera, Ibáñez, Parada-Alfonso, and Cifuentes (2019), the *P. peruviana* calyx extract was obtained under optimum conditions (at 125°C and 75% EtOH v/v) and exhibited moderate antioxidant activity (EC_{50} of $77.18\ \text{mg}\cdot\text{mL}^{-1}$ and $1.08\ \text{mM trolox}\ \text{g}^{-1}$). Analysis of LC-q-TOF-MS/MS from the extract allowed the quantification of 4-hydroxywithanolide E and withanolide E.

For analysis by HPLC, the major compound found was rutin with a content of $11.38 \pm 0.98\ \mu\text{g}\cdot\text{mL}^{-1}$ and rutin retention time of $9.45\ \text{min}$. The analytical method was linear in the concentration range of 0.1 to $5\ \mu\text{g}\cdot\text{mL}^{-1}$ ($y = 93,497x + 4979.8$; $r = 0.9978$), specific, accurate (DPR = 1.96%) and with a detection limit of $0.01\ \mu\text{g}\cdot\text{mL}^{-1}$ and $0.02\ \mu\text{g}$ rutin quantification mL^{-1} , according to the analytical conditions used.

According to Medina et al. (2019), ultra-high performance liquid chromatography triple quadrupole tandem mass spectrometry (UHPLC-QqQ-MS/MS) revealed novel active oxylipins denominated phytoprostanines in the *P. peruviana* calyces. In addition, the profile of phenolic

Table 1
Physicochemical characteristics of the nanoemulsions containing the *P. peruviana* calyx extract (CPp-NE- 5 mg.mL⁻¹ extract) and blank nanoemulsions (NE).

Formulation	Rutin concentration (µg.mL ⁻¹)	Droplet size (nm)	PdI ^a	Zeta potencial (mV)	pH	EE ^b
CPp-NE	10.70 ± 0.10	182 ± 1	0.14 ± 0.03	-10.7 ± 0.5	5.49 ± 0.08	84.13%
NE	-	165 ± 3	0.10 ± 0.02	-8.0 ± 1.2	6.33 ± 0.02	-

^a PdI: polydispersity index.

^b EE: Encapsulation efficiency.

compounds was characterized using LC-IT-DAD-MS/MS, where six phenolic derivatives were first described, of which the authors highlighted five cinnamoyl acid derivatives and one flavonoid (quercetin-3-O-glucoside). Among the compounds already identified, rutin was also the major phenolic compound.

It should be noted that there is a range of compounds in *P. peruviana* calyces, and Ballesteros-Vivas et al. (2019) reported a total of fifty-six phytochemicals, including major phenolic components, several withanolides (C28-isoprenoids) with a variety of biological activities, and a large family of anti-inflammatory sucrose esters tentatively identified, including rutin and quercetin as the major flavonoids found. These studies corroborate with the major compound found in our work because our focus was to find a marker for stability analyses.

3.2. Physicochemical characterization of nanoemulsions

The nanoemulsions presented homogeneous macroscopic appearance similar to a milky opalescent liquid with blueish reflex (Tyndall effect), which is typical of colloidal systems. The physicochemical characteristics of nanoemulsions are described in Table 1.

Nanoemulsions presented nanometric droplet sizes of 160–180 nm and polydispersity index below 0.15, which shows adequate homogeneity of the system. These values agree with other studies that characterize nanoemulsions because they have droplet sizes ranging between 20 and 300 nm (Komaiko & McClements, 2015; Yildirim, Oztop, & Soyer, 2017). In addition, the laser diffraction size results demonstrated nanometric distribution (D [4; 3]: 0.48 ± 0.01 and 0.44 ± 0.01 µm for CPp-NE and NE, respectively; and Span values of 0.62 ± 0.01 for both formulations, with values of Span ≤ 1 characterizing good formulation homogeneity), with a unimodal peak in this range and without the presence of micrometric drops. It is possible to observe that adding the extract to the formulations increased droplet sizes as demonstrated by Zorzi, Caregnato, Moreira, Teixeira, and Carvalho (2016) in the nanoemulsions containing the hydroalcoholic extract of macela (*Achyrocline satureioides*) in relation to the blank nanoemulsions. Furthermore, a more acidic pH was observed in the CPp-NE due to the extract acidity (CPp pH = 5.51 ± 0.01). In addition, among the phenolic compounds analyzed by HPLC, from the method used, the identified compound was rutin (concentration of 10.7 µg.mL⁻¹). According to the literature, the main disadvantage of the molecule is its poor solubility in the aqueous medium, which explains its low oral or topical bioavailability (Mauludin, Müller, & Keck, 2009) and being detrimental to its conversion into suitable dosage forms. The importance of nanoencapsulation in this case is emphasized, since it may make this compound have better apparent solubility in the aqueous medium, improving its bioaccessibility. The encapsulation efficiency obtained was around 85%, which is considered satisfactory, since other studies have reported lower efficiencies for bioactive compounds or plant extracts. For example, for exopolysaccharides extracted from brown algae (*Sargassum longifolium*) in nanoemulsions of orange oil, the EE found was 67% (Shofia, Jayakumar, Mukherjee, & Chandrasekaran, 2018). In another study, the authors obtained an EE of 68% for nanoemulsions containing phenolic constituents of the *Phyllanthus emblica* branch extract with 50% ethanol (Chaittitanan & Sripanidkulchai, 2014).

The nanoencapsulation of bioactive compounds has been a matter of

great interest due to its exclusive functionalities compared to other techniques, such as high encapsulation efficiency and loading capacity, enhanced bioavailability, improved stability, sustained release profile, and masking undesirable flavors (Shishir, Xie, Sun, Zheng, & Chen, 2018).

The zeta potential reflects the surface potential of the particles, which is influenced by changes in the interface with the environment-dispersing agent due to the dissociation of functional groups on the surface of the particle or the adsorption of ionic species present in the aqueous dispersion medium (Schaffazick, Guterres, Freitas, & Pohlmann, 2003). Regarding the zeta potential, the samples presented negative values. The presence of the calyx extract in the formulations increased the zeta potential in modulus, which may predict a greater physicochemical stability of the nanoemulsions, since large repulsive forces tend to avoid aggregation between droplets (Kovalchuk & Starov, 2012).

As for droplet morphology, the nanoemulsion micrographs can be observed at different magnifications in Fig. 1. The produced CPp-NE had a spherical shape and smooth surface. In addition, they are in agreement with the size data obtained by photon correlation spectroscopy (PCS), confirming the size in the nano range.

3.3. Stability study

In the stability study, the formulations were maintained under different conditions: Refrigeration temperature (7 °C) in the dark (CPp-NE_{RE}), room temperature (25 °C) in the dark (CPp-NE_{AE}), refrigeration temperature (7 °C) with light incidence (CPp-NE_{RL}), room temperature (25 °C) with light incidence (CPp-NE_{AL}), and analyzed for 120 days at pre-established time intervals. In relation to pH, all formulations presented acidic pH (5.49 ± 0.08) with variations occurring over time and the CPp-NE_{RL} sample being stable at the end of the study (5.39 ± 0.02). Meanwhile, the CPp-NE_{AE} and CPp-NE_{AL} formulations showed a significant reduction in pH (p < 0.05), being more pronounced in the last formulation that was exposed to light and at room temperature (supplementary material), which may be due to oxidations in the formulation components.

Droplet size and PdI of the CPp-NE_{RE}, CPp-NE_{AE}, and CPp-NE_{RL} nanoemulsions did not show significant variations (p > 0.05) in 120 days (Fig. 2), remaining stable for these parameters and without any visible changes in the formulations. In addition, the same nanoemulsions presented PdI values below 0.25 at all analysis times, indicating good system homogeneity. However, in the CPp-NE_{AL} sample, a significant increase in size was observed at 15 days. In addition, after 90 days of analysis, a destabilization of this formulation was verified occurring the separation of phases. This reflected in a significant increase in droplet size and PdI (p < 0.05) at 90 and 120 days, even reaching the micrometer scale (data not shown). From this, this sample was excluded from the other analyses. Among the mechanisms of destabilization, such as flocculation, coalescence, Ostwald ripening, and creaming, the Ostwald ripening is the dominant mechanism of destabilization for nanoemulsions, however, other phenomena may occur after, leading to the separation of phases. Several parameters can be related to these phenomena of instability, such as component structure and chemical properties, the concentration of emulsifiers and additives, and temperature, these of which cause changes in properties such as

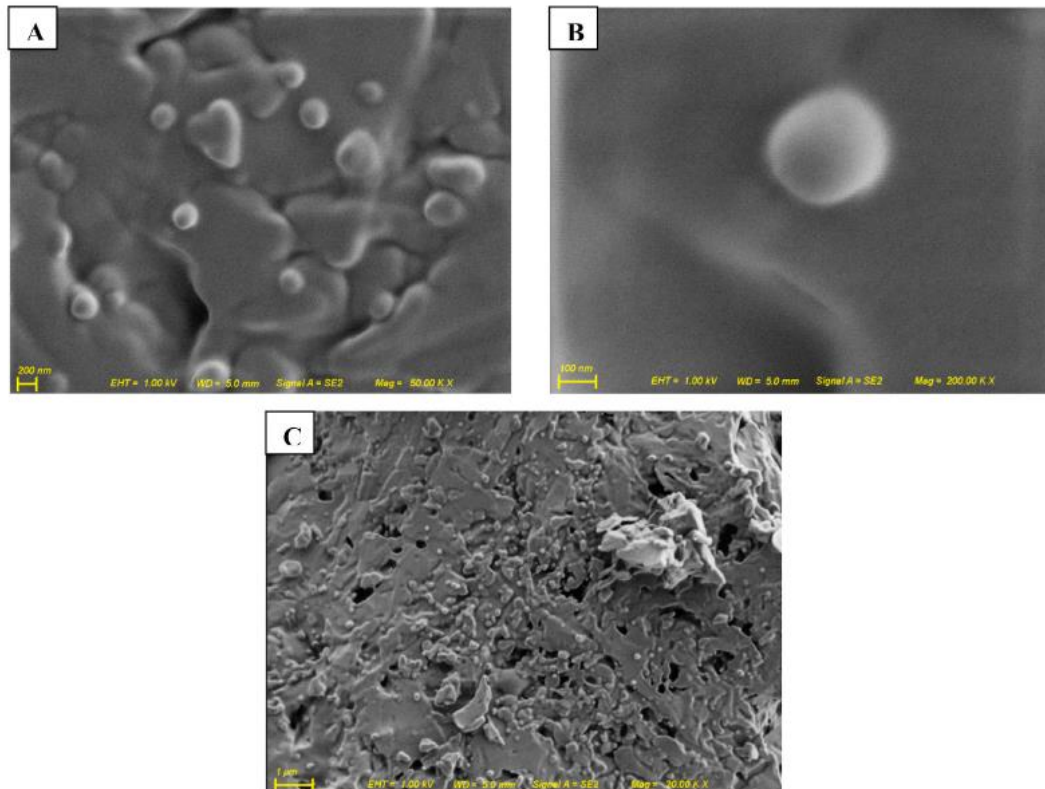


Fig. 1. Scanning electron microscopy of the nanoemulsions containing $5 \text{ mg}\cdot\text{mL}^{-1}$ of the *P. peruviana* calyx extract in different increases. (A) $50\times$, (B) $200\times$, and (C) $20\times$.

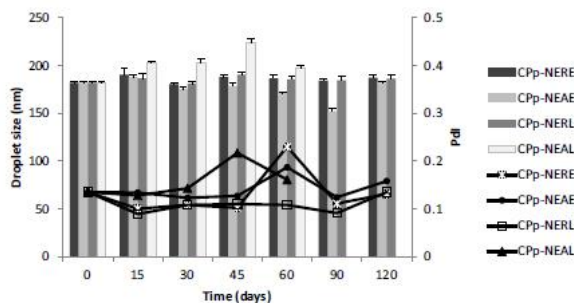


Fig. 2. Droplet size and polydispersity index (PDI) of the nanoemulsions containing the *P. peruviana* calyx extract ($\text{CPp-NE} - 5 \text{ mg}\cdot\text{mL}^{-1}$ extract) in different storage conditions: R: refrigeration (7°C); A: room (25°C); E: dark; L: presence of light.

interfacial tension, droplet elasticity, and potential droplet interaction (Gupta, Eral, Hatton, & Doyle, 2016). According to Godoi et al. (2017), the nanoemulsions containing *Eucalyptus globulus* oil and stored at room temperature (25°C) at 90 days showed increased sizes and reduced pH near 3, as verified in our study (sample $\text{CPp-NE}_{\text{AL}}$). Therefore, high temperatures or room temperatures and the incidence of light have been shown to be factors already reported to cause changes in nanoemulsion stability (Rebolledo et al., 2015).

Analyses of the zeta potential showed that the negative values remained stable during the study, regardless of the treatment. However, some modifications were observed over time. The samples $\text{CPp-NE}_{\text{RE}}$

and $\text{CPp-NE}_{\text{RL}}$ at the end of the experiment were stable with values of -10.46 ± 1.74 and -9.83 ± 1.45 , respectively, and similar to the initial ones (-11.23 ± 0.49) ($p > 0.05$). In the samples at 25°C with light incidence ($\text{CPp-NE}_{\text{AL}}$), a significant reduction in modulus was observed for -3.19 ± 0.29 ($p < 0.05$) at the zeta potential values at the end of the study, which is in agreement with the previous analyses that indicated instability in this formulation at 90 and 120 days. According to Giongo et al. (2016), the nanoemulsions containing the geranium oil presented phase separation at room temperature at 90 days, as observed in our study. In addition, both the nanoemulsions stored at $25 \pm 2^\circ\text{C}$ and at $45 \pm 2^\circ\text{C}$ showed increased droplet sizes and polydispersity index, which is expected since the temperature increases the kinetic energy of the system, increasing the possibility of occurring an instability phenomenon.

3.4. Antioxidant capacity

3.4.1. Reducing capacity

The values of the reducing capacity can be observed in Table 2. The $\text{CPp-NE}_{\text{RE}}$ formulation remained stable up to 90 days ($p > 0.05$). Afterwards, a reduction was observed and the experiment finished at around 85.22%. In contrast, the $\text{CPp-NE}_{\text{AE}}$ samples had a significant increase ($p < 0.05$) at 45 days of analysis, as well as the extract that was stored under the same conditions, which showed this increase after 90 days. Severo et al. (2010) already verified these changes in the reducing capacity in fruit stored at 23°C , which increased from 210 to $360 \text{ mg gallic acid Eq}\cdot 100 \text{ g}^{-1}$. On the other hand, at refrigeration temperature (4°C), the increase was lower, reaching an amount of $300 \text{ mg gallic acid Eq}\cdot 100 \text{ g}^{-1}$. Furthermore, in the same study, the

Table 2

Reducing capacity (mg gallic acid Eq. 100 g⁻¹ calyx) of the nanoemulsions containing the *P. peruviana* calyx extract (CPp-NE - 5 mg.mL⁻¹ extract) and CPp extracts under different storage conditions.

Samples	0 days	15 days	30 days	45 days	60 days	90 days	120 days
CPp-NE _{RE}	575 ± 28 ^{baB}	535 ± 9 ^{cdBC}	619 ± 2 ^{ba}	619 ± 12 ^{ba}	615 ± 32 ^{ba}	614 ± 56 ^{ca}	490 ± 26 ^{cc}
CPp-NE _{AE}	575 ± 28 ^{bdE}	584 ± 12 ^{ad}	500 ± 7 ^{de}	804 ± 24 ^{cc}	1696 ± 14 ^{ab}	1670 ± 19 ^{ab}	1814 ± 68 ^{ba}
CPp-NE _{RL}	575 ± 28 ^{ba}	534 ± 23 ^{cdAB}	523 ± 24 ^{cdBC}	520 ± 4 ^{dbc}	517 ± 3 ^{dbc}	514 ± 5 ^{dbc}	489 ± 15 ^{cc}
CPp-NE _{AL}	575 ± 28 ^{ba}	573 ± 9 ^{ba}	572 ± 1 ^{ba}	567 ± 4 ^{ca}	564 ± 17 ^{ca}	459 ± 11 ^{eb}	245 ± 29 ^{cc}
CPp- _{RE}	610 ± 20 ^{ba}	569 ± 6 ^{abcB}	549 ± 10 ^{bcc}	535 ± 4 ^{cd}	522 ± 1 ^{de}	515 ± 1 ^{de}	375 ± 9 ^{ef}
CPp- _{AE}	610 ± 20 ^{ab}	557 ± 6 ^{abcd}	552 ± 4 ^{bcc}	522 ± 10 ^{dd}	466 ± 11 ^{efe}	754 ± 8 ^{ba}	611 ± 13 ^{bb}
CPp- _{RL}	610 ± 20 ^{ba}	523 ± 11 ^{db}	524 ± 24 ^{cdB}	519 ± 8 ^{dbc}	485 ± 10 ^{decd}	479 ± 8 ^{ed}	379 ± 14 ^{de}
CPp- _{AL}	610 ± 20 ^{ba}	541 ± 19 ^{bcdB}	523 ± 18 ^{cdB}	481 ± 6 ^{cc}	441 ± 1 ^{fd}	386 ± 7 ^{ef}	382 ± 1 ^{de}

Data were expressed as mean ± standard deviation (n = 3). Means with the same letter (lowercase in the column and uppercase in the row) do not differ statistically from each other by the Tukey test at 5% significance.

CPp-NE_{RE}: nanoemulsions containing the *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) in the dark.

CPp-NE_{AE}: nanoemulsions containing the *P. peruviana* calyx extract maintained at room temperature (25 °C) in the dark.

CPp-NE_{RL}: nanoemulsions containing the *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) with light incidence.

CPp-NE_{AL}: nanoemulsions containing the *P. peruviana* calyx extract maintained at room temperature (25 °C) with light incidence.

CPp-_{RE}: *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) in the dark.

CPp-_{AE}: *P. peruviana* calyx extract maintained at room temperature (25 °C) in the dark.

CPp-_{RL}: *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) with light incidence.

CPp-_{AL}: *P. peruviana* calyx extract maintained at room temperature (25 °C) with light incidence.

Table 3

Antioxidant capacity by ORAC (μmol Trolox.g⁻¹ calyx) of nanoemulsions (NE) containing the *P. peruviana* calyx extract (CPp-NE - 5 mg.mL⁻¹ extract) and CPp extracts under different storage conditions.

Samples	0 days	15 days	30 days	45 days	60 days	90 days	120 days
CPp-NE _{RE}	112.93 ± 2.85 ^{ba}	109.67 ± 1.53 ^{cdB}	111.29 ± 1.65 ^{ba}	110.80 ± 2.43 ^{cc}	110.13 ± 3.27 ^{bd}	110.52 ± 3.78 ^{abd}	111.77 ± 3 ^{ba}
CPp-NE _{AE}	112.93 ± 2.85 ^{hb}	111.24 ± 1.05 ^{cdBC}	105.04 ± 3.00 ^{bcc}	104.34 ± 4.23 ^{abcc}	103.09 ± 4.79 ^{abcc}	102.26 ± 3.77 ^{bcc}	131.66 ± 4.14 ^{ba}
CPp-NE _{RL}	112.93 ± 2.85 ^{ba}	108.52 ± 5.99 ^{da}	105.51 ± 6.48 ^{bcA}	87.27 ± 2.41 ^{bb}	74.35 ± 6.46 ^{cbC}	67.49 ± 3.90 ^{dc}	65.6 ± 1.22 ^{cc}
CPp-NE _{AL}	112.93 ± 2.85 ^{ba}	107.13 ± 1.20 ^{daB}	94.65 ± 0.60 ^{bc}	91.03 ± 8.11 ^{cdC}	74.05 ± 8.55 ^{cd}	14.54 ± 0.53 ^{efE}	-
CPp- _{RE}	137.93 ± 2.10 ^{ba}	140.54 ± 0.41 ^{ba}	117.17 ± 2.84 ^{abB}	108.77 ± 3.42 ^{abcc}	98.28 ± 10.67 ^{abcc}	93.98 ± 8.49 ^{cc}	17.78 ± 0.49 ^{dd}
CPp- _{AE}	137.93 ± 2.10 ^{ba}	134.49 ± 11.83 ^{baB}	125.11 ± 4.38 ^{abcc}	110.36 ± 5.58 ^{abd}	103.53 ± 7.53 ^{abbd}	115.06 ± 0.96 ^{abcc}	67.05 ± 2.89 ^{ef}
CPp- _{RL}	137.93 ± 2.10 ^{ba}	118.63 ± 4.36 ^{bcB}	104.32 ± 8.57 ^{bcc}	96.99 ± 6.74 ^{bcdC}	95.26 ± 5.61 ^{abcc}	77.30 ± 0.64 ^{dd}	8.29 ± 0.59 ^{ef}
CPp- _{AL}	137.93 ± 2.10 ^{ba}	110.87 ± 4.88 ^{cdB}	107.93 ± 7.56 ^{bcb}	105.35 ± 1.87 ^{abB}	88.19 ± 4.24 ^{bcc}	67.39 ± 3.34 ^{dd}	2.95 ± 0.14 ^{ef}

Data were expressed as mean ± standard deviation (n = 3). Means with the same letter (lowercase in the column and uppercase in the row) do not differ statistically from each other by the Tukey test at 5% significance.

CPp-NE_{RE}: nanoemulsions containing the *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) in the dark.

CPp-NE_{AE}: nanoemulsions containing the *P. peruviana* calyx extract maintained at room temperature (25 °C) in the dark.

CPp-NE_{RL}: nanoemulsions containing the *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) with light incidence.

CPp-NE_{AL}: nanoemulsions containing the *P. peruviana* calyx extract maintained at room temperature (25 °C) with light incidence.

CPp-_{RE}: *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) in the dark.

CPp-_{AE}: *P. peruviana* calyx extract maintained at room temperature (25 °C) in the dark.

CPp-_{RL}: *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) with light incidence.

CPp-_{AL}: *P. peruviana* calyx extract maintained at room temperature (25 °C) with light incidence.

antioxidant capacity remained stable at room temperature and decreased at 4 °C from 1.55 to 1.3 mmol TE.g⁻¹, as assessed by the ABTS-2,2'-azino-bis (3-ethylbenzthiazoline-6-sulfonic acid). Based on this, it is emphasized that the storage temperature can influence reducing capacity values and consequently antioxidant capacity. The reducing capacity of the active substances present in the nanoemulsions with presence of light (CPp-NE_{RL}) remained for 45 days at 90%. Afterwards, it declined, finishing with 85.04% at 120 days. The CPp-NE_{AL} sample remained stable for 60 days (p > 0.05). Another relevant factor is the high phenolic compound content present in the calyx. Gironés-Vilaplana et al. (2014) analyzed separately *P. peruviana* calyces and fruit from Colombia and, by comparing them, found that more compounds and high levels of non-red polyphenols were observed in the calyces (195.44 ± 3.88 mg.100 g⁻¹ dry weight) than in the fruit (2.18 ± 0.71 mg.100 g⁻¹ dry weight). This stresses the important source of bioactive compounds present in the calyx, making it of great interest for new food products. In addition, the presence of phenolic compounds in food products is extremely important since they provide numerous health benefits and act as anticancer, antithrombotic, antiulcerative, antiarterogenic, anti-allergic, anti-

inflammatory, antioxidant, immunomodulatory, antimicrobial and cardioprotective agents, vasodilators, and analgesics (Singh, Singh, Kaur, & Singh, 2017).

Together, the nanoemulsions had greater reducing capacity than the free extracts. These data are in agreement with those observed in the antioxidant capacity (Table 3). It should be noted that the CPp-NE_{AE} samples had the antioxidant capacity potentiated and the CPp-NE_{RE} samples at 120 days maintained their antioxidant capacity (p > 0.05). These results demonstrate that nanostructured systems can protect the bioactive compounds present in the extract against degradation, which favors prolonged shelf-life of nanoencapsulated extracts. The flavonoid mechanism of antioxidant action is determined by its structure, in particular by hydroxyls that can donate electrons and support delocalization around the aromatic system (Dornas, Oliveira, Rodrigues-dos-Dores, Santos, & Nagem, 2007).

In the case of nanoemulsions that were subjected to light, CPp-NE_{RL} samples were stable for 30 days (p > 0.05) and the samples at room temperature (CPp-NE_{AL}) showed significant reduction (p < 0.05), however, they still had approximately 84% of the initial antioxidant capacity at 30 days.

Table 4

Rutin content ($\mu\text{g}\cdot\text{mL}^{-1}$) of nanoemulsions (NE) containing the *P. peruviana* calyx extract (CPP-NE - $5\text{ mg}\cdot\text{mL}^{-1}$ extract) and CPP extracts under different storage conditions.

Samples	0 days	15 days	30 days	45 days	60 days	90 days	120 days
CPP-NE _{RE}	10.7 ± 0.1 ^{aA}	10.10 ± 0.39 ^{dB}	4.58 ± 0.14 ^{cC}	0.64 ± 0.12 ^{cDE}	0.65 ± 0.01 ^{CD}	0.38 ± 0.10 ^{dDE}	0.10 ± 0.02 ^{eE}
CPP-NE _{AE}	10.7 ± 0.1 ^{aA}	11.15 ± 0.21 ^{abA}	10.80 ± 0.04 ^{abcA}	8.62 ± 0.13 ^{abB}	8.03 ± 0.21 ^{abC}	7.71 ± 1.14 ^{abC}	7.05 ± 0.85 ^{ac}
CPP-NE _{RL}	10.7 ± 0.1 ^{aA}	9.51 ± 0.17 ^{CA}	9.50 ± 0.08 ^{DA}	6.84 ± 0.12 ^{dB}	6.12 ± 0.78 ^{dB}	5.73 ± 0.13 ^{BB}	5.59 ± 0.95 ^{BB}
CPP-NE _{AL}	10.7 ± 0.1 ^{aA}	9.28 ± 0.18 ^{CB}	10.29 ± 0.08 ^{bcdA}	7.05 ± 0.04 ^{cdC}	6.08 ± 0.60 ^{bdD}	4.37 ± 0.15 ^{CE}	2.57 ± 0.09 ^{dF}
CPP- _{RE}	11.38 ± 0.98 ^{aA}	10.77 ± 0.16 ^{bCA}	11.74 ± 0.02 ^{aA}	9.11 ± 0.06 ^{abB}	8.95 ± 0.14 ^{abB}	6.22 ± 0.01 ^{BC}	5.42 ± 0.02 ^{BC}
CPP- _{AE}	11.38 ± 0.98 ^{aA}	10.44 ± 0.02 ^{cdAB}	11.27 ± 1.16 ^{abA}	9.84 ± 0.48 ^{aAB}	9.03 ± 0.74 ^{abB}	5.92 ± 0.34 ^{BC}	5.38 ± 0.16 ^{BC}
CPP- _{RL}	11.38 ± 0.98 ^{aA}	11.42 ± 0.01 ^{aA}	9.57 ± 0.03 ^{dB}	8.59 ± 0.39 ^{abBC}	8.3 ± 0.50 ^{ac}	7.63 ± 0.11 ^{ac}	6.08 ± 0.19 ^{bdD}
CPP- _{AL}	11.38 ± 0.98 ^{aA}	11.16 ± 0.10 ^{abA}	10.07 ± 0.01 ^{cdAB}	8.39 ± 1.24 ^{bcB}	6.55 ± 0.29 ^{bc}	5.39 ± 0.07 ^{bcCD}	4.01 ± 0.06 ^{CD}

Data were expressed as mean ± standard deviation (n = 3). Means with the same letter (lowercase in the column and uppercase in the row) do not differ statistically from each other by the Tukey test at 5% significance.

CPP-NE_{RE}: nanoemulsions containing the *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) in the dark.

CPP-NE_{AE}: nanoemulsions containing the *P. peruviana* calyx extract maintained at room temperature (25 °C) in the dark.

CPP-NE_{RL}: nanoemulsions containing the *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) with light incidence.

CPP-NE_{AL}: nanoemulsions containing the *P. peruviana* calyx extract maintained at room temperature (25 °C) with light incidence.

CPP-_{RE}: *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) in the dark.

CPP-_{AE}: *P. peruviana* calyx extract maintained at room temperature (25 °C) in the dark.

CPP-_{RL}: *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) with light incidence.

CPP-_{AL}: *P. peruviana* calyx extract maintained at room temperature (25 °C) with light incidence.

3.4.2. Oxygen Radical Absorbance Capacity (ORAC)

The antioxidant capacity of the *P. peruviana* calyx was also evaluated by ORAC according to Gironés-Vilaplana et al. (2014). The authors observed that the *P. peruviana* calyces presented higher antioxidant capacity than the fruit, which highlights the importance of working with this residue. In addition, the values found for the calyx extract samples analyzed in our study presented higher antioxidant capacity values than those observed in the previously mentioned study, which may be due to the samples being from another location, growing season, and different methodology to obtain the extract.

3.5. Rutin content

Regarding the rutin content found in the samples, the nanoemulsions maintained at room temperature in the dark remained stable ($p > 0.05$) for 30 days and at 90 days presented a loss of 28% of rutin (Table 4). According to Yuan, Gao, Zhao, and Mao (2008), β -carotene nanoemulsions exhibited good physical stability but were chemically degraded during storage and at the end of the four-week period (30 days), with about 14–25% of β -carotene being lost, but with higher degradation at 25 °C than at 4 °C. On the other hand, NE stored at room temperature in the dark (CPP-NE_{AE}) were more stable for rutin content than the refrigerated ones in our study. As for the free extracts, the treatments submitted to the dark were stable for 30 days, and those stored with light incidence were stable for 15 days. In this manner, it is possible to infer that the rutin extracted from the *P. peruviana* calyx extract did not degrade on the first days of analysis, as the free extracts were expected to be more unstable in adverse temperatures and light conditions. In this case, rutin was found to have a positive point of not instantly degrading.

Rutin was also found in fruit of the same species, varying according to maturation rates ($5.89\ \mu\text{g}\ \text{rutin}\cdot\text{g}^{-1}$ for beginning and $6.90\ \mu\text{g}\ \text{rutin}\cdot\text{g}^{-1}$ fruit at the end of maturation) (Licodiedoff, Koslowski, & Ribani, 2013). According to Gironés-Vilaplana et al. (2014), it was only possible to identify trace amounts of rutin (quercetin 3-O-rutinoside) and kaempferol 3-O-rutinoside in the fruit.

In the same study, two different quercetin glycosides were found in the *P. peruviana* calyx (among them rutin in greater quantity), two different kaempferol glycosides, 3-O-caffeoylquinic, and 5-O-caffeoylquinic acids. The calyx presented the highest concentrations of flavonols and derivatives of hydroxycinnamic acid among all the Latin American fruit analyzed (Gironés-Vilaplana et al., 2014). In addition, Toro et al. (2014) also found bioactive compounds, such as rutin and

nicotiflorin (kaempferol-3-O-rutinoside) in the *P. peruviana* chalice, corroborating with the previous studies and the major compound (rutin) found in the present study.

In the literature, other encapsulation techniques have already been performed with the *P. peruviana* plant, as the fruit juice was microencapsulated to increase its stability in simulated digestion fluids (Dag, Kilercioglu, & Oztop, 2017) and development and optimization of the microencapsulated butanolic fraction of the *P. peruviana* calyces to improve its hypoglycemic activity (Echeverry, Valderrama, Costa, Ospina-Giraldo, & Aragón, 2018). Future trends point towards evaluating extracts and nanoemulsions against the release of bioactive compounds in simulated gastrointestinal fluid and toxicity in vitro and in vivo models.

4. Conclusion

Different storage conditions, such as the incidence or absence of light and the temperature, influence stability, especially in relation to bioactive compound content, reducing capacity, and antioxidant capacity. In this study, the best storage conditions of the samples were at both temperatures in the dark, and the highest stability was observed for the nanoemulsions containing the *P. peruviana* calyx extract stored at room temperature and in the absence of light. These results demonstrate that developed nanoemulsions have potential use in preserving the antioxidant capacity of *P. peruviana* calyx extract, thus favoring its reuse, as they avoid waste and have bioactive compounds that are beneficial to health. Therefore, the use of nanotechnology together with plant residues is of great interest for the science of food and development of new food products.

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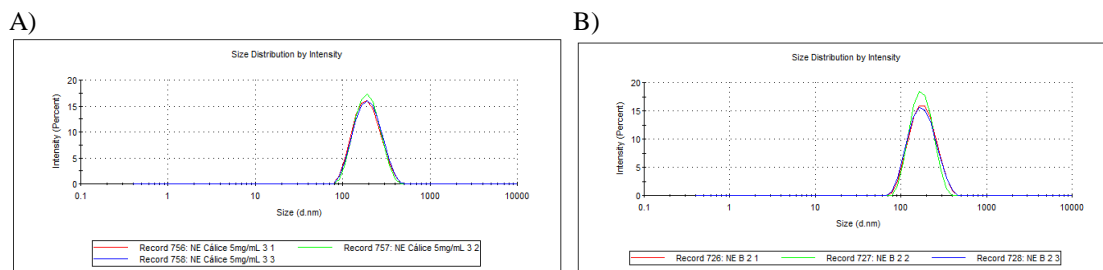
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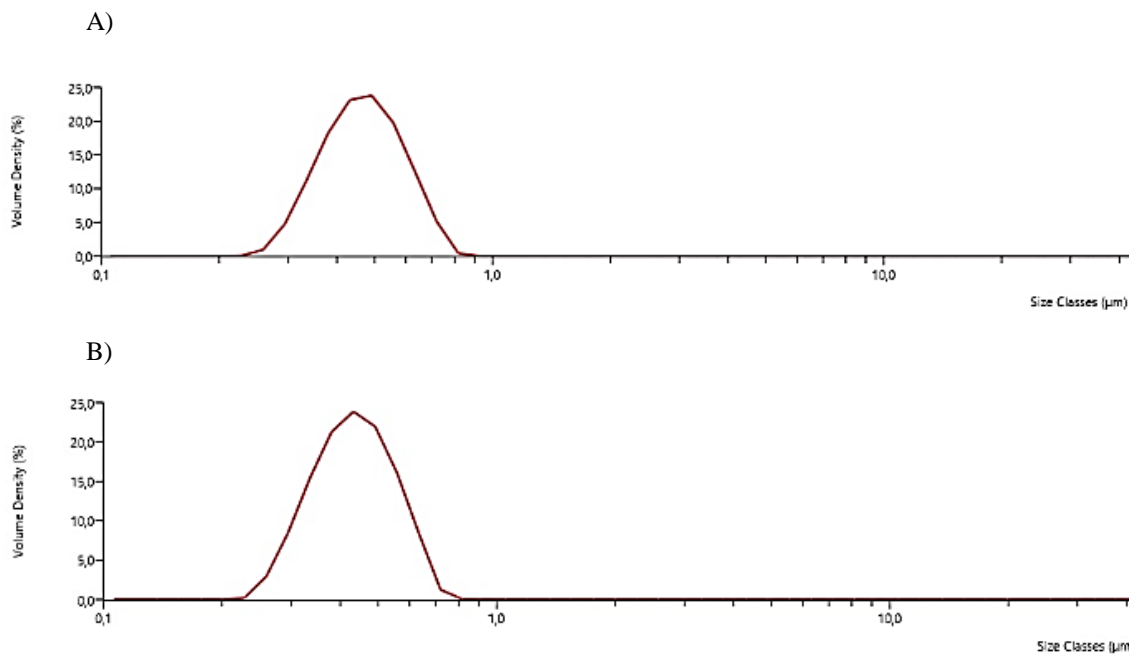
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MATERIAL SUPLEMENTAR DO ARTIGO 2

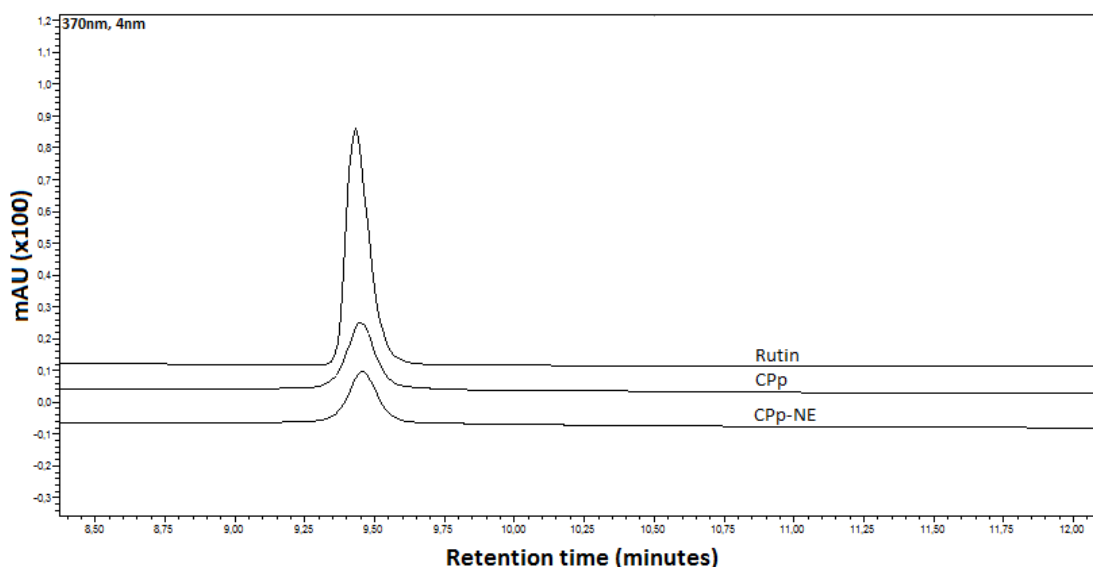
Supplementary data to this article can be found online at <https://doi.org/10.1016/j.foodres.2019.108645>.



Supplementary Fig. S1. Nanoemulsion droplet size distribution containing 5 mg mL^{-1} of the *P. peruviana* calyx extract (A) and blank nanoemulsion (B) analyzed by photon correlation spectroscopy.



Supplementary Fig. S2. Nanoemulsion droplet size distribution containing 5 mg mL^{-1} of the *P. peruviana* calyx extract (A) and blank nanoemulsion (B) analyzed by laser diffraction.



Supplementary Fig. S3. Overlapping chromatograms obtained by rutin standard HPLC from the *P. peruviana* calyx extract (CpP) and the nanoemulsions containing the *P. peruviana* calyx extract (CpP-NE).

Supplementary Table S1. pH measurements of the nanoemulsions containing the *P. peruviana* calyx extract (CpP-NE - 5 mg.mL⁻¹ extract) under different storage conditions.

Time (days)	CpP-NE _{RE}	CpP-NE _{AE}	CpP-NE _{RL}	CpP-NE _{AL}
0	5.49 ± 0.08 ^{cA}	5.49 ± 0.08 ^{aA}	5.49 ± 0.08 ^{cA}	5.49 ± 0.08 ^{aA}
15	5.26 ± 0.02 ^{dB}	5.34 ± 0.01 ^{bA}	5.33 ± 0.01 ^{dA}	3.16 ± 0.02 ^{cC}
30	5.23 ± 0.04 ^{dA}	4.95 ± 0.01 ^{cB}	5.35 ± 0.02 ^{dA}	3.32 ± 0.11 ^{bC}
45	5.46 ± 0.02 ^{cA}	4.65 ± 0.01 ^{dC}	5.37 ± 0.02 ^{dB}	3.24 ± 0.01 ^{bcD}
60	5.67 ± 0.01 ^{bA}	4.55 ± 0.01 ^{eC}	5.59 ± 0.02 ^{aB}	3.25 ± 0.03 ^{bcD}
90	5.89 ± 0.02 ^{aA}	4.38 ± 0.01 ^{fC}	5.53 ± 0.02 ^{abB}	3.25 ± 0.02 ^{bcD}
120	5.67 ± 0.01 ^{bA}	4.35 ± 0.02 ^{fC}	5.39 ± 0.02 ^{cdB}	3.28 ± 0.01 ^{bcD}

Data were expressed as mean ± standard deviation (n = 3). Means with the same letter (lowercase in the column and uppercase in the row) do not differ statistically from each other by the Tukey test at 5% significance.

CpP-NE_{RE}: nanoemulsions containing the *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) in the dark

CpP-NE_{AE}: nanoemulsions containing the *P. peruviana* calyx extract maintained at room temperature (25 °C) in the dark

CpP-NE_{RL}: nanoemulsions containing the *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) with light incidence

CpP-NE_{AL}: nanoemulsions containing the *P. peruviana* calyx extract maintained at room temperature (25 °C) with light incidence

CpP-_{RE}: *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) in the dark

CpP-_{AE}: *P. peruviana* calyx extract maintained at room temperature (25 °C) in the dark

CpP-_{RL}: *P. peruviana* calyx extract maintained at refrigeration temperature (7 °C) with light incidence

CpP-_{AL}: *P. peruviana* calyx extract maintained at room temperature (25 °C) with light incidence

**6 MANUSCRITO 1 – ANTIBACTERIAL AND ANTIBIOFILM ACTIVITY OF
Physalis peruviana CALYX EXTRACT**

Manuscrito submetido a Revista Ciência Rural.

1 **Antibacterial and antibiofilm activity of *Physalis peruviana* calyx extract**

2 **Atividade antibacteriana e antibiofilme do extrato do cálice de *Physalis peruviana***

3

4 **Maiara Taís Bazana^I Marcos Vinícius Missel^{II} Cassandra de Deus^I Vandrê Souza Pinto^I**

5 **Suelen Santos da Silva^I Cristiane Franco Codevilla^I Roberto Christ Vianna Santos^{II}**

6 **Cristiane de Bona da Silva^{III} Cristiano Ragagnin de Menezes^{I*}**

7

8 **ABSTRACT**

9 This study aimed to evaluate the antibacterial and antibiofilm activity of a *P. peruviana* calyx
10 extract. Goldenberry calyx extracts were prepared with 60% (v/v) ethanol. Minimum
11 inhibitory concentration (MIC) analyses were performed by the 96-well plate microdilution
12 method together with the minimum bactericidal concentration (MBC). Biofilm inhibition and
13 destruction was performed in microdilution plates. The *P. peruviana* calyx extract presented
14 antibacterial activity against the pathogens analysed (*Enterobacter aerogenes*, *Klebsiella*
15 *pneumoniae*, *Staphylococcus epidermidis*, *Streptococcus pneumoniae*, *Pseudomonas*
16 *aeruginosa* PA01, *Enterococcus faecalis*, *Escherichia coli*, *Shigella sonnei*, *Acinetobacter*
17 *baumannii*, *Streptococcus agalactiae*, *Acinetobacter baumannii*, *Salmonella sp.*, *Salmonella*
18 *enteridis* and *Staphylococcus aureus*). MIC varied from 3.15 to 30 mg/mL extract and
19 showed bacteriostatic activity against eight pathogens and bactericidal activity at 30 mg/mL
20 concentration against six strains. Regarding the biofilm tests, there was inhibition of biofilm
21 formation, but there was no destruction. According to these results, the potential antibacterial
22 activity of *P. peruviana* calyx extract was verified. This will enable further studies to be
23 carried out to contribute to its use in the food industry as a preservative of natural origin and
24 other clinical applications.

25 **Keywords:** biofilm, minimum inhibitory concentration, minimum bactericidal concentration.

26 ^IDepartamento de Tecnologia e Ciência dos Alimentos, Universidade Federal de Santa Maria (UFSM), Santa
27 Maria, RS, Brasil

28
29 ^{II}Departamento de Microbiologia e Parasitologia, Universidade Federal de Santa Maria (UFSM), Santa Maria,
30 RS, Brasil.

31
32 ^{III}Departamento de Farmácia Industrial, Universidade Federal de Santa Maria (UFSM), Santa Maria, RS, Brasil.

33
34 ^{I*} Departamento de Tecnologia e Ciência dos Alimentos, Universidade Federal de Santa Maria (UFSM), 97105-
35 900, Santa Maria, RS, Brasil. E-mail: cristiano.ufsm@gmail.com Autor para correspondência.

36

37

38 **RESUMO**

39 Este estudo teve como objetivos avaliar a atividade antibacteriana e antibiofilme do extrato do
40 cálice de *P. peruviana*. Os extratos do cálice de goldenberry foram preparados com etanol
41 60% (v/v). As análises de concentração inibitória mínima (CIM) foram realizadas pelo
42 método de microdiluição em placas de 96 poços, juntamente com a concentração bactericida
43 mínima (CBM). A inibição e destruição de biofilme foi realizada em placas de microdiluição.
44 O extrato do cálice de *P. peruviana* apresentou atividade antibacteriana frente aos patógenos
45 analisados (*Enterobacter aerogenes*, *Klebsiella pneumoniae*, *Staphylococcus epidermidis*,
46 *Streptococcus pneumoniae*, *Pseudomonas aeruginosa* PA01, *Enterococcus faecalis*,
47 *Escherichia coli*, *Shigella sonnei*, *Acinetobacter baumannii*, *Streptococcus agalactie*,
48 *Acinetobacter baumannii*, *Salmonella sp.*, *Salmonella enteridis* and *Staphylococcus aureus*).
49 A CIM variou de 3,15 a 30 mg/mL de extrato, demonstrando atividade bacteriostática frente a
50 oito patógenos e atividade bactericida na concentração de 30 mg/mL frente a seis cepas. Em
51 relação aos testes com biofilmes, houve inibição da formação de biofilme, porém não houve
52 destruição. De acordo com esses resultados, foi verificada a potencial atividade antibacteriana
53 do extrato do cálice de *P. peruviana*. Isto possibilitará que novos estudos sejam realizados
54 para contribuir na sua utilização na indústria de alimentos, como um conservante de origem
55 natural.

56 **Palavras-chave:** biofilme, concentração inibitória mínima, concentração bactericida mínima.

57

58 INTRODUCTION

59

60 Among the new natural sources, *Physalis peruviana* has been studied for its
61 antimicrobial potential. ERTÜRK et al. (2017) evaluated the antioxidant and antimicrobial
62 activity of ethanolic extracts from different parts of *P. peruviana* (fruit, seed, root, body and
63 leaf). Seeds and fruit were the most effective of the plant in terms of its antimicrobial activity.
64 In another study, the dimethyl sulfoxide extract of *P. peruviana* fruit inhibited the growth of
65 microorganisms used in the test (GÖZTOK; ZENGİN, 2013). All parts of the *P. peruviana*
66 plant may be potential sources of antimicrobial agents, the most potent being leaf extracts in
67 dichloromethane (KAMAU et al., 2017). In addition, the ripe berries and leaves of *P.*
68 *peruviana* were evaluated and the ethanolic extracts were more effective than the aqueous
69 ones (CUEVA et al., 2017).

70 *P. peruviana* fruit is usually marketed without its calyx; consequently, the waste
71 generated by its trade is substantial. Colombia is the world's leading producer of *P.*
72 *peruviana*, which represents the country's second largest exported product. However, waste
73 generated by the agro-industrial sector, particularly fruit and vegetable waste, is a major
74 economic and environmental issue for many companies, as it is up to them to assume the cost
75 of their management (MEDINA et al., 2019).

76 Contamination of foods is usually found when they are damaged or contaminated by
77 containing microorganisms such as bacteria, parasites or toxic substances that make them
78 unfit for consumption (HUSSAIN, 2016). *P. aeruginosa* PA01 is a Gram-negative rod,
79 considered a standard biofilm-producing strain and an important pathogen chosen for this
80 study. A biofilm is a complex of bacterial cells lined with a polysaccharide layer that acts as a
81 protective factor for bacteria against antimicrobial attack and the host immune system. The

82 formation of biofilms causes considerable problems in the medical and industrial area, as
83 these structures cause greater resistance to treatment with antibiotics and biocides. Resistance
84 to antibiotic therapy and the emergence of multiresistant strains is worrisome, and new
85 therapeutic approaches are required (LIMA et al., 2018).

86 Several chemical preservatives are used to control these foodborne pathogens (BAI et
87 al., 2016). However, due to the increased awareness of the side effects of these additives,
88 there is a growing consumer demand for natural products in place of chemical preservatives.
89 In recent years, research on natural antimicrobial compounds has increased at a rapid pace
90 (LEE et al., 2019).

91 There are a variety of natural preservatives from different sources and with different
92 preservation properties, such as plant extracts, essential oils, organic acids (acetic, ascorbic,
93 tartaric, malic, and citric), lactic acid bacteria and bacteriocins from microbiological sources,
94 and chitosan of animal origin (BAPTISTA; HORITA; ANA, 2019). Plants that possess these
95 properties have bioactive compounds that act to protect plants from microbiological attacks,
96 although they can also be exploited and employed by humans as sources for food and
97 medicine (ANTOLAK; KREGIEL, 2017).

98 Thus, *P. peruviana* calyces are an important byproduct that deserves further study so
99 that different sources of reuse can be explored and the database on their antimicrobial activity
100 expanded. Therefore, the aim of this study was to evaluate, for the first time, the antibacterial
101 activity against different microorganisms and antibiofilm activity against *Pseudomonas*
102 *aeruginosa* of *P. peruviana* calyx extract for therapeutic and food application.

103

104

105

106 MATERIAL AND METHODS

107

108 **Material:** *P. peruviana* calyces were obtained from Italbraz (Vacaria, Brazil - 28:0:44
109 S and 50:56:02 W) in the 2016/2017 harvest. Mueller Hinton broth (Himedia®) and Mueller-
110 Hinton agar (Sigma-Aldrich®) and crystal violet (Sigma Chemical®) were used.

111 **Microorganisms and inoculum:** American Type Culture Collection (ATCC)
112 bacterial strains and clinical isolates provided by the UFSM Department of Microbiology and
113 Parasitology were used [*Enterobacter aerogenes* ATCC 13048, *Klebsiella pneumonia* ATCC
114 1705, *Escherichia coli* (clinical isolate), *Staphylococcus epidermidis* (clinical isolate),
115 *Streptococcus pneumoniae* ATCC 99619, *Shigella sonnei* (clinical isolate), *Acinetobacter*
116 *baumannii* ATCC 19606, *Streptococcus agalactiae* (clinical isolate), *Acinetobacter baumannii*
117 (clinical isolate), *Pseudomonas aeruginosa* PA01, *Enterococcus faecalis* ATCC 29212,
118 *Salmonella sp.* (clinical isolate), *Salmonella enteridis* (clinical isolate), and *Staphylococcus*
119 *aureus* (clinical isolate)]. The bacterial inoculum sizes were standardized according to CLSI
120 guidelines (2017). Isolated colonies were grown for 18 and 24 h in Mueller Hinton Agar and
121 the suspension was prepared in saline solution (NaCl 0.85%) with density adjusted to 0.5 on
122 the McFarland scale (1.5×10^8 CFU/mL).

123 **Extraction:** The calyx extracts were prepared following Bazana et al. (2019). The *P.*
124 *peruviana* calyces (3 g) were separated from the fruit, submitted to knife milling (Mill MA
125 630/1-Marconi) (speed = 5 rpm, time = 10s), added with 50 mL of 60% ethanol (v/v), and
126 submitted to stirring in a homogenizer (Shaker model TE 421- Tecnal, Piracicaba, SP, Brazil)
127 at a speed of 200 rpm for 2 h. Afterwards, the extracts were filtered in a qualitative filter
128 (Fitec) for coarse separation of the calyx. Subsequently, the extract was subjected to filtration
129 in a polypropylene syringe (0.22- μ m pore size) (Chromfilter) and kept in a refrigerated
130 environment.

131 **Determination of the minimum inhibitory concentration (MIC) and minimum**
132 **bactericidal concentration (MBC):** The MICs were determined by the microdilution method
133 according to Clinical and Laboratory Standards Institute (CLSI, 2017). The assay was carried
134 out in 96-well-microplates using Mueller Hinton broth (MHB). Several concentrations of *P.*
135 *peruviana* calyx extract (60; 30; 15; 7.5; 3.75; and 1.87 mg/mL) was diluted in ethanol 60%
136 (v/v) and added in wells with MHB and the suspension with microorganism (0.5 MacFarland
137 scale). The positive control was considered the well with the bacterial suspension and MHB
138 while the negative control was MHB and extract. A control with ethanol 60% (v/v) was
139 performed to discard the diluent activity. The plates were incubated at 37°C and the minimum
140 inhibitory concentration (MIC) was recorded after 24 h of incubation. 2,3,5-
141 triphenyltetrazolium chloride was used as an indicator which develop a red colour in the
142 microbial grown. The assays were performed in triplicate. The MICs were defined as the
143 lowest concentration that inhibits visible bacterial growth. The determination of the MBC was
144 performed in conjunction with the MIC, using the method proposed by Courvalin (1985).
145 After the determination of MIC, wells containing visible or non-visible growth were
146 transferred to Petri dishes containing the solid Mueller-Hinton agar, then incubated at 37°C
147 for 18-24 hours. After this period, the number of colonies per plaque was determined and the
148 MBC was defined as the lowest concentration of the extract that presented 0.01% viable
149 bacteria.

150 **Biofilm inhibition:** For this analysis, 15 µL of inoculum were pipetted with 100 µL
151 MHB and 100 µL of extract, in MIC and subinhibitory concentrations into four wells of a
152 sterile flat-bottomed microtiter plate. After incubation for 24 h at 37°C, all of the planktonic
153 microorganisms were removed and washed with distilled water three times. The biofilm was
154 visualised by adding 200 µL of 0.1% crystal violet solution to each well. The microplates
155 were washed with distilled water and air-dried. To solubilize the biofilm, 200 µL of 95%

156 ethanol was added. The solution was transferred to a new microtiter plate and the biofilm
157 formation was revealed by measuring the absorbance at 570 nm in a microplate reader. For
158 the control culture broth was used for negative control and only *P. aeruginosa PA01*
159 inoculum for positive control (BONEZ et al., 2017).

160 **Biofilm destruction:** 15 µL of inoculum with 185 µL of MHB was pipetted into four
161 wells of a sterile flat-bottomed microtiter plate. After incubation for 24 h at 37°C, the
162 planktonic microorganisms were removed from the wells and the extracts were added in MIC
163 and subinhibitory concentrations. After incubation for 24 h at 37°C, all of the planktonic
164 microorganisms were removed and washed with distilled water three times. The biofilm was
165 visualised by adding 200 µL of 0.1% crystal violet solution to each well. The microplates
166 were washed with distilled water and air-dried. To solubilize the biofilm, 200 µL of 95%
167 ethanol was added. The solution was transferred to a new microtiter plate and the biofilm
168 formation was revealed by measuring the absorbance at 570 nm in a microplate reader. For
169 the controls, culture broth was used as the negative control and only *P. aeruginosa PA01*
170 inoculum was used as the positive control (BONEZ et al., 2017).

171 **Statistical analyses:** The experiments were performed in triplicate. The results were
172 evaluated by analysis of variance (ANOVA), and the means were compared by the Tukey
173 test, considering the level of significance at 5% ($p < 0.05$).

174

175 **RESULTS AND DISCUSSION**

176

177 **Minimal inhibitory concentration (MIC) and minimal bactericidal concentration (MBC)**

178 The antimicrobial activity of the ethanolic extracts of *P. peruviana* calyx is shown in
179 Table 1. The lowest minimum inhibitory concentration observed was against *Enterococcus*

180 *faecalis* bacteria, in which the extract concentration of 3.75 mg/mL was able to inhibit growth
181 of this bacterium. MBC = 30 mg/mL was found for *Enterobacter aerogenes*, *Klebsiella*
182 *pneumoniae*, *Staphylococcus epidermidis*, *Streptococcus pneumoniae*, *Pseudomonas*
183 *aeruginosa* PAO1e *Enterococcus faecalis*, indicating that for these bacteria the 30 mg/mL *P.*
184 *peruviana* calyx extract has a bactericidal action. For the other microorganisms, MBC \geq 30
185 mg/mL indicated that the extract has bacteriostatic action. These results suggest that the
186 bioactive compounds of calyx are responsible for antimicrobial action. According to
187 Ballesteros-Vivas et al. (2019), several compounds were identified in the *P. peruviana* calyx
188 as phytochemicals (terpenoids, phytosterols and phytol derivatives), flavonoids, phenolic
189 acids and withanolides. Phytosterols are reported as bioactive compounds of great interest due
190 to their antioxidant capacity and health impact. They are referred to as anti-inflammatory,
191 antitumor, antibacterial, antifungal and hypocholesterolemic compounds. Its presence in
192 significant levels in oil extracted from the skin and pulp of *P. peruviana* has already been
193 found (PUENTE et al., 2011). According to these values, the calyx extract can be considered
194 promising for use as a preservative of natural origin or to be part of a sanitizer for use in the
195 food industry.

197 Table 1. MIC and MBC values of *P. peruviana* calyx extract against bacterial strains
 198

Bacterial strains	* MIC (mg/mL)	** MBC (mg/mL)
<i>Enterobacter aerogenes</i> ATCC 13048	30	30
<i>Klebsiella pneumoniae</i> ATCC 1705	30	30
<i>Escherichia coli</i> (Clinical isolate)	7.5	≥ 30
<i>Staphylococcus epidermidis</i> (Clinical isolate)	15	30
<i>Streptococcus pneumoniae</i> ATCC 99619	30	30
<i>Shigella sonnei</i> (Clinical isolate)	15	≥ 30
<i>Acinetobacter baumannii</i> ATCC 19606	7.5	≥ 30
<i>Streptococcus agalactiae</i> (Clinical isolate)	7.5	≥ 30
<i>Acinetobacter baumannii</i> (Clinical isolate)	30	≥ 30
<i>Pseudomonas aeruginosa</i> PA01	7.5	30
<i>Enterococcus faecalis</i> ATCC 29212	3.75	30
<i>Salmonella sp.</i> (Clinical isolate)	15	≥ 30
<i>Salmonella enteridis</i> (Clinical isolate)	7.5	≥ 30
<i>Staphylococcus aureus</i> (Clinical isolate)	7.5	≥ 30

199 * Minimum Inhibitory Concentration

200 ** Minimum Bactericidal Concentration

201

202

203 **Biofilm inhibition and destruction**

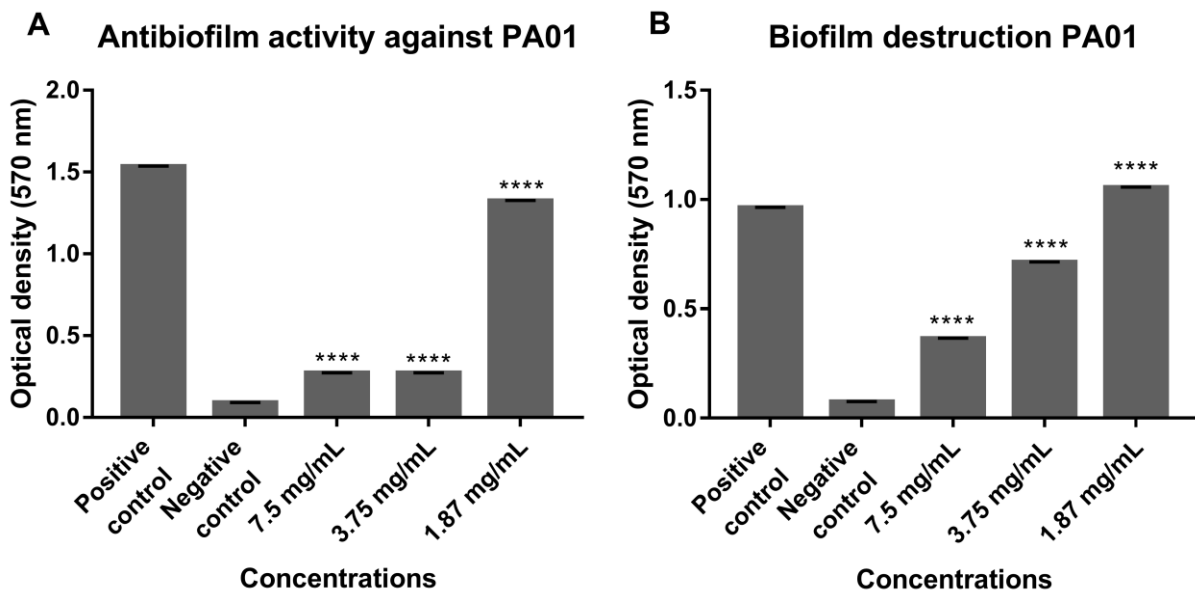
204

205 The results regarding biofilm are shown in Figure 1. The biofilm growth in our study
 206 was confirmed by the standard *P. aeruginosa* strain PA01 (control) with 100% growth. *P.*
 207 *aeruginosa* is known to be an opportunistic human pathogen capable of causing a wide range
 208 of acute and chronic infections that can be life-threatening, particularly in
 209 immunocompromised patients. It has been historically associated with lung infections in
 210 cystic fibrosis patients and is one of the major nosocomial pathogens affecting hospitalized
 211 patients, while being intrinsically resistant to a wide range of antibiotics (MORADALI;
 212 GHODS; REHM, 2017). In addition, the presence of *Pseudomonas* biofilms is also found in

213 food production and its facilities, causing food spoilage and generating various problems
 214 (MELIANI; BENSOLTANE, 2016).

215 *P. peruviana* calyx extract concentrations of 3.75 and 7.5 mg/mL inhibited biofilm
 216 formation in the early stages (Figure 1A). In Figure 1B, where the biofilm destruction was
 217 tested, it was observed that there was no condition that led to total destruction of the
 218 microorganism. The lowest extract concentration (1.87 mg/mL) did not have the ability to
 219 inhibit biofilm formation, i.e. the most effective concentrations to inhibit *P. aeruginosa* PA01
 220 biofilm formation were 3.75 and 7.5 mg/mL of calyx extract.

221



222

223

224 Figure 1. A: Inhibition of *P. aeruginosa* PA01 biofilm formation. B: Destruction of *P.*
 225 *aeruginosa* biofilm. Error bars are shown as one standard deviation and differences were
 226 considered statistically significant when $p < 0.0001$ between the positive control and the
 227 corresponding concentration. Each experiment was carried out three times, yielding similar
 228 results.

229

230

231 CONCLUSION

232

233 The *P. peruviana* calyx extract showed antibacterial activity against the pathogens
234 analysed. The minimum inhibitory concentration ranged from 3.15 to 30 mg/mL extract. The
235 highest activity demonstrated was bacteriostatic, but also had bactericidal activity at a
236 concentration of 30 mg/mL against *Enterobacter aerogenes*, *Klebsiella pneumoniae*,
237 *Staphylococcus epidermidis*, *Streptococcus pneumoniae*, *Pseudomonas aeruginosa* PA01 and
238 *Enterococcus faecalis*, indicating that, at this concentration, it acts as a bactericidal agent
239 against these pathogens. Regarding the biofilm tests, there was inhibition of biofilm formation
240 at the concentrations of 3.75 and 7.5 mg/mL of the calyx extract and a partial destruction at
241 the same concentrations. These results may be used in future research with an emphasis on the
242 use of *P. peruviana* calyx extract in therapeutic approaches and in the food industry as a
243 preservative of natural origin.

244

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246

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250

251 DECLARATION OF CONFLICTING INTERESTS

252

253 We have no conflicts of interest to declare.

254

255

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339

**7 MANUSCRITO 2 – *IN VITRO* AND *IN VIVO* TOXICITY ASSESSMENT AND
RELEASE IN SIMULATED GASTROINTESTINAL CONDITIONS OF *P. peruviana*
CALYX EXTRACT-LOADED NANOEMULSIONS**

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Functional Foods**

26 worms, and the formulations at low concentrations showed better results against resistance to
27 oxidative stress. In addition, release under simulated gastrointestinal conditions demonstrated
28 that nanoemulsions containing the extract protected rutin from degradation and showed
29 controlled release. Therefore, the nanoemulsions containing *P. peruviana* calyx extract
30 revealed slight to moderate *in vitro* cytotoxicity, which depends on the cell line, and no *in*
31 *vivo* toxicity under the evaluated conditions. Moreover, the proposed formulations protected
32 the bioactive compound against gastrointestinal conditions, allowing new studies to enable
33 applicability in the food industry.

34 Keywords: cytotoxicity, *C. elegans*, goldenberry, nanotoxicology, *in vitro* release

35

36

37 1. INTRODUCTION

38

39 *Physalis peruviana* L., which is also known as cape gooseberry, is an herbaceous plant
40 native to the Andes that belongs to the Solanaceae family (Álvarez-Flórez, López-
41 Cristoffanini, Jáuregui, Melgarejo, & López-Carbonell, 2017). The calyx grows into a
42 bladder-like organ during development of the fruit and surrounds it, providing protection
43 against insects, birds, diseases, and adverse weather conditions. In addition, this structure
44 represents an essential carbohydrate source during the first 20 days of fruit development
45 (Puente, Pinto-Muñoz, Castro, & Cortés, 2011).

46 The cape gooseberry fruit is usually marketed without the calyx, generating large
47 quantities of waste (Medina et al., 2019). However, this byproduct has attracted interest of
48 scientific studies and several bioactive compounds have already been identified as
49 phytochemicals (terpenoids, phytosterols, and phytol derivatives), flavonoids, phenolic acids,
50 and withanolides (Gironés-Vilaplana et al., 2014; Ballesteros-Vivas, Álvarez-Rivera, Del

51 Pilar Sánchez-Camargo, et al., 2019; Ballesteros-Vivas, Álvarez-Rivera, Ibáñez, Parada-
52 Alfonso, & Cifuentes, 2019; Medina et al., 2019). These compounds frequently have some
53 limitations, including low stability against light, heat, oxygen, low water solubility, low
54 absorption, and low bioavailability. Based on these factors, our research group developed
55 nanoemulsions containing *P. peruviana* calyx extract and evaluated the stability and
56 antioxidant capacity (Bazana et al., 2019). Despite advances in nanotechnology, the safety of
57 these systems has not yet been fully proven. As a result, the number of studies on nanoparticle
58 biosafety evaluation using different biological models has risen (Wu, Xu, Liang, & Tang,
59 2019).

60 Cytotoxicity assays are widely used in preliminary investigations of nanomaterial
61 toxicity (Hillegass et al., 2010). Established cell lines are useful alternative test systems for
62 these studies (Crespi, 1995). The degree of cell damage is measured by several assays that
63 evaluate aspects of cell viability, such as metabolic activity and plasma membrane integrity.
64 Among the most widely used tests is the MTT (3-[4,5-dimethylthiazol-2-yl]-2,5-
65 diphenyltetrazolium) reduction assay, which determines cellular metabolic activity. This
66 method measures the reduction of MTT salt to a colored insoluble formazan in active
67 mitochondria in viable cells and, in certain cases, outside the mitochondria. For reliable data
68 from cytotoxicity studies, comparative sensitivity analyses with more than one cell type are
69 strongly recommended (Nogueira, Mitjans, Infante, & Vinardell, 2011). Additionally, it is
70 recommended to consider the properties of the nanosystem and critical issues that may impact
71 its behavior *in vitro*, such as the concentration of components, population, and nanoparticle
72 structure/type (Mendes et al., 2015).

73 The *Caenorhabditis elegans* nematode (*C. elegans*) is a complete model and important
74 *in vivo* assay system for evaluating nanotoxicity (Ávila, Roncato, & Jacques, 2018; Velasques
75 et al., 2018; Silva et al., 2019). The worm has excellent features of a typical model system,

76 such as a short life cycle (2 to 3 weeks), a tiny and transparent body, easy cultivation, low
77 maintenance costs, and no bioethical concerns (Wu et al., 2019).

78 Finally, the gastrointestinal behavior of nanoemulsion-based delivery systems for
79 controlled or targeted release of bioactive compounds should also be considered when
80 designing nanoemulsions with oral administration purposes. Exposure to these different
81 regions may cause considerable changes in nanoemulsion properties or encapsulated bioactive
82 compounds (Zhang & McClements, 2018). In this context, this study aimed to evaluate
83 nanoemulsions containing *P. peruviana* calyx extract for cytotoxicity, toxicity in the *C.*
84 *elegans* model, and release in the simulated gastrointestinal tract.

85

86

87 **2. MATERIAL AND METHODS**

88

89 **2.1 Material**

90

91 *P. peruviana* calyces were obtained from Italbraz (Vacaria, Brazil - 28:0:44 S and
92 50:56:02 W). Ethanol and acetone were purchased from Dynamics (São Paulo, Brazil),
93 polysorbate 80 (Tween[®] 80) from Neon Comercial Ltda (São Paulo, Brazil), sorbitan
94 monooleate (Span[®] 80) from Fluka (São Paulo, Brazil), and medium chain triglycerides
95 (MCT) from Delaware (Porto Alegre, Brazil). The MTT (3-[4,5-dimethylthiazol-2-yl]-2,5-
96 diphenyltetrazolium), trypsin-EDTA solution, and DMSO were obtained from Sigma-Aldrich
97 (St. Louis, USA). Fetal bovine serum (FBS) and Dulbecco's Modified Eagle's Medium
98 (DMEM), which were supplemented with L-glutamine (584 mg/L) and antibiotic/antimicotic
99 (50 mg/mL gentamicin sulfate and 2 mg/L amphotericin B), were purchased from Vitrocell
100 (Campinas, SP, Brazil). The wild type strain of *C. elegans* N2 (var. Bristol) was supplied by

101 the Caenorhabditis Genetics Center (University of Minnesota, USA). The other reagents were
102 HPLC grade for high performance liquid chromatography analysis. Hydroalcoholic extracts of
103 the calyces and nanoemulsions with *P. peruviana* calyx extract (NE) and without (NB) were
104 performed according to Bazana et al. (2019).

105

106 2.2 *In vitro* toxicity

107

108 The *in vitro* cytotoxicity of *P. peruviana* calyx extract (CE), blank nanoemulsions
109 (NB), and nanoemulsions containing the extract (NE) was measured using the non-tumor cell
110 line 3T3 (murine Swiss albino fibroblasts) and tumor cell line MCF-7 (human breast cancer
111 cells). The 3T3 and MCF-7 cells were grown in DMEM medium supplemented with 10%
112 (v/v) FBS at 37 °C and 5% CO₂. After reaching approximately 80% confluence, the cells
113 were harvested using trypsin-EDTA and seeded into 96-well plates at a density of 1×10^5
114 cells/mL. After incubation for 24 h under 5% CO₂ at 37 °C, the spent medium was replaced
115 with 100 µL of fresh medium containing the extract, nanoemulsion containing extract or
116 blank nanoemulsion at the defined concentrations (10, 50, and 150 µg/mL) or with 100 µL
117 medium only for untreated control cells. After 24 h, the sample-containing medium was
118 removed and 100 µL of MTT diluted in medium without FBS (0.5 mg/mL) was added to the
119 cells. The plates were further incubated for 3 h, after which the medium was removed, and
120 100 µL of DMSO were then added to each well to dissolve the purple formazan product. After
121 10 min of shaking, the absorbance of the resulting solutions was measured at 550 nm using a
122 Multiskan FC (Thermo Fisher Scientific, Waltham, MA, EUA) microplate reader. The effect
123 of each treatment was calculated as a percentage of cell viability inhibition against the
124 untreated control cells (cells incubated with medium only).

125

126 2.3 *In vivo* toxicity

127

128 2.3.1 Worm maintenance

129 *C. elegans* strains N2 (wild-type) were manipulated and maintained at 20 °C in the
130 nematode growth medium (NGM) fed with *Escherichia coli* OP50 (Brenner, 1974). Worms in
131 the young adult stage used in all exposures were obtained by a synchronization process, which
132 consists of exposing the pregnant worms to a lysis solution (1% NaOCl, 0.25 M NaOH) to
133 separate the eggs from the worms. After 14 h, the eggs hatched and released the L1 larvae,
134 which were added to the surface of the NGM plates containing *E. coli* OP50, where they
135 remained at 20 °C until reaching young adult stage.

136

137 2.3.2 Treatment

138 *C. elegans* at young adult stage were exposed to 0.5, 1.0 or 2.0 mg/mL of CE, NE, NB
139 or vehicle for 2 h at 20 °C. The CE was diluted in ethanol (final concentration 1%) and the
140 NE and NB nanoemulsions were diluted in water. Treatments were performed with about
141 1000 worms per group in M9 buffer (42 mM Na₂HPO₄, 22 mM KH₂PO₄, 8.6 mM NaCl, and
142 1 mM MgSO₄). After 2 h, the worms were washed 3 times and transferred to the NGM plates
143 seeded with *E. coli* OP50. The evaluations were performed after 24 h at 20 °C.

144

145 2.3.3 Survival

146 Survival assay was performed following the protocol previously described, with some
147 modifications (Velasques et al., 2018). About 100 nematodes per group were evaluated for
148 viability under a Nikon E200 microscope (Tokyo, Japan). Animals that reacted to a
149 mechanical stimulus were classified as alive and non-responding animals were classified as

150 dead. Analyses were performed in three independent trials. Results were expressed as
151 percentage of survival.

152

153 2.3.4 Pharyngeal pumping rate

154 The number of pharyngeal pumping over a range of 10 s (Huang, Xiong, & Kornfeld,
155 2004) was measured with a Nikon E200 microscope. The analyses were performed in three
156 independent trials with ten worms per group. Results were expressed as pharyngeal
157 beats/minute.

158

159 2.3.5 Defecation cycle

160 The defecation frequencies were performed by observing the duration between bupic
161 steps (the contraction of the posterior body muscle) of three consecutive defecations
162 (Migliori, Simonetta, Romanowski, & Golombek, 2011) in a Nikon E200 microscope.
163 Analyses were performed in three independent trials with ten worms per group. Results were
164 expressed as defecation cycle/s.

165

166 2.3.6 Oxidative stress resistance assay

167 To perform the acute oxidative stress resistance assay, worms previously exposed for
168 2 h to the extracts, the nanoemulsions or vehicle were removed from their treatments and
169 immediately exposed for one hour to Juglone (5-hydroxy-1,4-naphthoquinone) at the final
170 concentration of 100 μ M (LC50) at 20 °C. Afterwards, the worms were washed from their
171 treatments and transferred to *E. coli* OP50 seeded NGM plates and, after 24 h, 100 nematodes
172 per group were evaluated for viability under a Nikon E200 microscope (Tokyo, Japan).
173 Animals that reacted to a mechanical stimulus were classified as alive and non-responding

174 animals were classified as dead. Analyses were performed in three independent assays.

175 Results were expressed as percentage of survivors.

176

177 2.4 Release under simulated gastrointestinal conditions

178

179 Evaluation of *in vitro* rutin release profile from calyx extracts (CE) and nanoemulsions
180 containing *P. peruviana* calyx extract (NE) were evaluated (n=3) by the dialysis bag method
181 and exposed to the simulated gastrointestinal conditions according to Madureira, Amorim,
182 Gomes, Pintado, & Malcata (2011), with modifications. The dialysis bag (MWCO = 12,000
183 Da, Sigma-Aldrich Corporation, MO, EUA) containing 1 mL of the samples (5 mg/mL of
184 calyx extract) was placed in an erlenmeyer containing 150 mL of dissolution medium. In the
185 esophagus–stomach step, 50 μ L of pepsin to 25 mg/mL (Sigma) was added. This solution was
186 prepared in 0.1 N HCl and added in equal aliquots at each step during the whole gastric phase
187 for a total of 90 min at 130 rpm agitation, with the pH being adjusted until 2, using 0.1 N HCl.
188 In the duodenum stage, 250 μ L of solution containing 2 g/L of pancreatin (Sigma) and 12 g/L
189 of bovine bile salts (Sigma) were diluted in 0.1 M NaHCO₃; this solution was added and the
190 pH adjusted up to 5.0 for 20 minutes at agitation of 45 rpm. Finally, the ileum stage was
191 performed for 90 minutes at 45 rpm agitation by raising the pH to 6.5. Analysis was
192 conducted in a refrigerated shaker incubator (TE-421, Tecnal, Brazil) maintained at 37 °C to
193 simulate human body temperature and mechanical agitation was used in parallel to simulate
194 intestinal peristaltic movements. At the end of each stage, aliquots were withdrawn and
195 filtered through a 0.22 μ m membrane for HPLC analyses. Rutin samples were evaluated by
196 HPLC according to the modified and validated method by Bazana et al. (2019).

197

198

199 2.5 Statistics

200

201 Statistical analyses were performed using *GraphPad Prism*® (Version 7.0, San Diego,
202 CA). All experiments were independently replicated and the results were plotted as the mean
203 \pm SEM of at least three individual experiments. Significance was determined using analysis of
204 variance (ANOVA) followed by Tukey post-hoc test. Values of $p < 0.05$ were considered
205 statistically significant.

206

207

208 **3. RESULTS AND DISCUSSION**

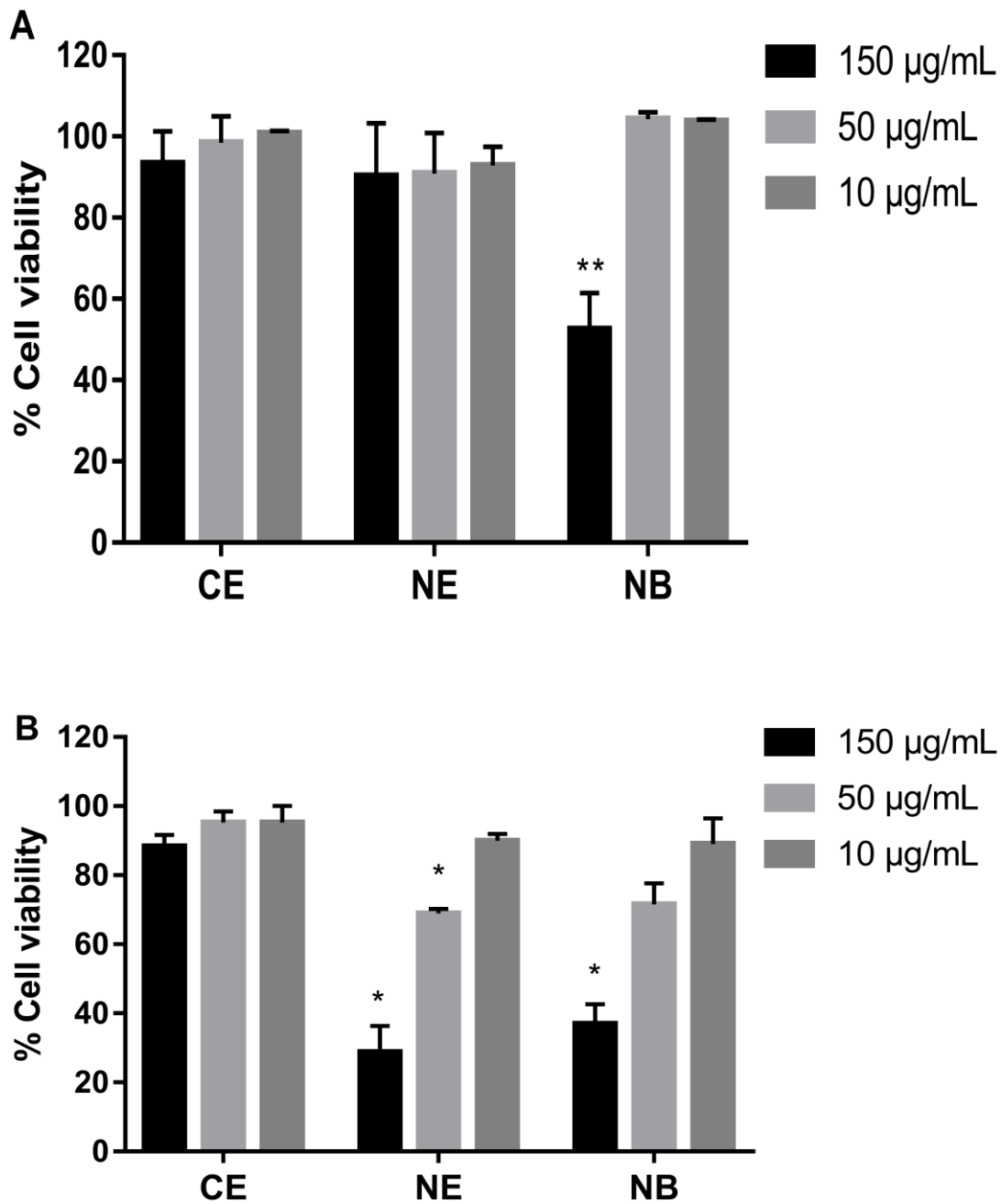
209

210 *3.1 In vitro* toxicity

211

212 The evaluation of *in vitro* cytotoxicity of a nanosystem is an important preliminary
213 test, even when the purpose of its formulation application in food products, thus verifying the
214 safety issues (Silva et al., 2018). In this study, two cell lines were chosen as model systems
215 because the use of cells from different species and with different embryogenic origins is an
216 important approach to understand the cell specific responses induced by nano-based materials
217 (Fröhlich, Meindl, Roblegg, Griesbacher, & Pieber, 2012).

218 The viability results obtained from 3T3 cells (Figure 1A) remained without significant
219 differences between the formulations tested, apart from NB at 150 $\mu\text{g}/\text{mL}$, which showed
220 significant reduction in cell viability.



221
 222 **Figure 1** Cell viability after exposure of the cells with *P. peruviana* calyx extracts (CE),
 223 nanoemulsions containing *P. peruviana* calyx extract (NE), and blank nanoemulsions (NB).
 224 (A) viability plot of 3T3 cells (non-tumor fibroblasts); (B) viability plot of MCF-7 cells
 225 (breast tumor). Each experiment was performed three times in triplicate. * means significant
 226 difference ($p < 0.05$) compared to the control after one-way ANOVA analysis followed by
 227 Tukey post-hoc test.

229 This cytotoxic response demonstrated that, at this concentration, NB was toxic to cells,
230 as viability was reduced to about 50%. In order to be considered cytotoxic, viability must be
231 below 75% (ISO 10993-5) (Wang et al., 2015). Previously, Codevilla et al. (2018) also
232 observed that the hydroalcoholic extract of goldenberry fruit was non-cytotoxic. Likewise,
233 studies conducted in our research group by Silva et al. (2019) reported that *P. peruviana* fruit
234 extract and the nanoemulsion containing the extract and blank nanoemulsion at concentrations
235 of 50, 200, and 400 µg/mL did not display cytotoxicity against 3T3 cells. Notably, the low
236 cytotoxic potential of nanoemulsions has already been reported when associated with essential
237 oils or plant extracts (Mendes et al., 2015).

238 Regarding MCF-7 (breast tumor) cells, free extract (CE) was unable to reduce the
239 viability of cancer cells (Figure 2B). Other studies also reported that the ethanolic extract of
240 goldenberry fruit was ineffective, i.e. Demir (2014) using 80% (v/v) ethanol to make the
241 extract, and Silva et al. (2019) at the concentrations of 50, 200, and 400 µg/mL extract made
242 with 60% (v/v) ethanol. In contrast, NE and NB showed higher inhibition rates in the growth
243 of these cells as the formulation concentrations increased (Figure 1B). According to Mendes
244 et al. (2015), the type and amount of surfactant in the formulation may influence their
245 cytotoxicity. Contri et al. (2016) reported decreased keratinocyte viability induced by blank
246 nanocapsules, which was similar to the one induced by polysorbate 80 solution, indicating
247 that nanocapsule toxicity is in fact linked to the nonionic surfactant used. In addition, the
248 presence of the nonionic surfactant (polysorbate 80) has also been indicated as a substance
249 that may cause toxicity in cell cultures, although it is better tolerated than ionic surfactants
250 (especially cationic surfactants) (Contri et al., 2016). Nevertheless, polysorbate 80 has already
251 been described as the least toxic surfactant for human fibroblasts compared to others, such as
252 those tested in this study, which can be classified in the following order of increasing
253 cytotoxicity: Tween 80 < Texapon N40 < Tween 60 < Texapon K1298 < Triton x100 <

254 benzethonium chloride (Arechabala, Coiffard, Rivalland, Coiffard, & De Roeck-Holtzhauer,
255 1999). Thus, cytotoxic responses have been observed even in blank nanoparticles during the
256 design of *in vitro* experiments using different cell lines (Mendes et al., 2014; Mendes et al.,
257 2015; Silva et al., 2019). Finally, it is worth mentioning that the different sensitivity of the
258 tested cell lines to the nanoemulsion treatments may be partly explained by healthy and tumor
259 cells having differences in their structures, functions, and more notably metabolic activity and
260 molecular composition (Frey et al., 2007). Therefore, the characteristics of each cell line may
261 lead to distinct mechanisms of defense and, consequently, different sensitivities to the toxic
262 effects of the applied treatment (Nogueira et al., 2011). This evidence highlights the relevance
263 of the combination of cell lines from different origins in order to obtain more information on
264 the potential toxic effects of novel nano-based systems.

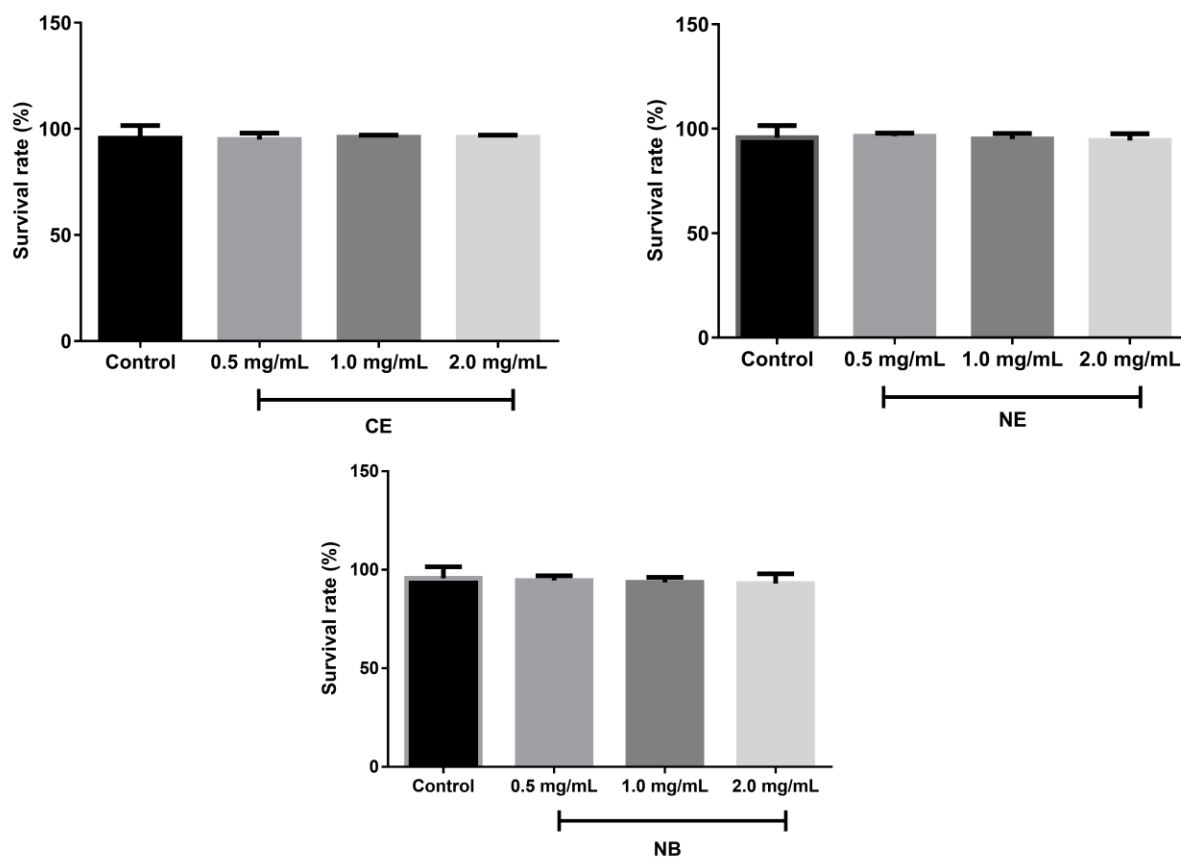
265

266 3.2 *In vivo* toxicity

267

268 The *in vivo* toxicity of the formulations was evaluated using the *C. elegans* model.
269 After the 2h-treatment, the survival rate of all animals remained around 100% (Figure 2),
270 regardless of the sample tested (CE, NE, and NB), indicating that exposure to the samples did
271 not cause lethality.

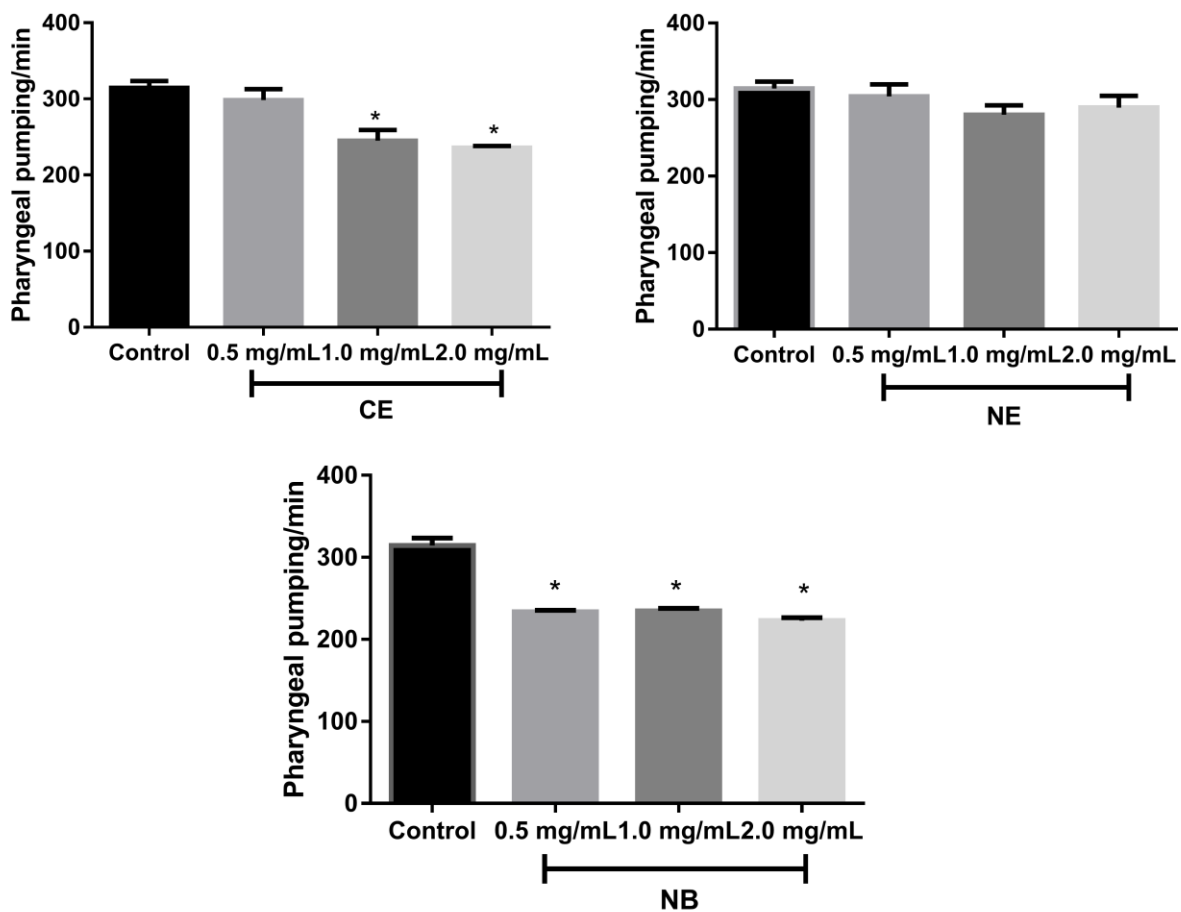
272 After acute treatment with the formulations, some *C. elegans* behaviors were analyzed
273 to evaluate the effects on toxicity. Animals exposed to the free extract (CE) at the
274 concentrations of 1 and 2 mg/mL had reduced pharyngeal pumping rates, whereas
275 nanoemulsions containing *P. peruviana* calyx extract (NE) did not cause any changes
276 (Figure 3).



277

278 **Figure 2** Survival rate of young adult worms of N2 strain exposed for 2h to different
279 concentrations of *P. peruviana* calyx extracts (CE), nanoemulsions containing *P. peruviana*
280 calyx extract (NE), and blank nanoemulsions (NB). The data are expressed as percent of
281 control; each experiment was performed three times in triplicate. (mean, SEM, n =
282 approximately 300 worms per group).

283



284

285 **Figure 3** Pharyngeal pumping/min of young adult worms of the N2 strain exposed for 2 h at
 286 different concentrations of *P. peruviana* calyx extracts (calyx extract), blank nanoelmsions
 287 (NB), and nanoelmsions containing *P. peruviana* calyx extract (NE) in liquid medium. Each
 288 experiment was performed three times in triplicate. (mean, SEM, n = approximately 15
 289 worms per group). * means significant difference ($p < 0.05$) compared to the control after
 290 one-way ANOVA analysis followed by Tukey post-hoc test.

291

292 However, NB showed a significant reduction in pharyngeal pumping rate, which has
 293 already been observed in the literature by other authors (Silva et al., 2019). Depending on the
 294 sample or concentration, there were changes in the feeding behavior of the worms in relation
 295 to the control group. According to You, Kim, Raizen, & Avery (2008), *C. elegans* varies its
 296 feeding rate by regulating the pumping rate, being a pharyngeal movement. Feeding rates are

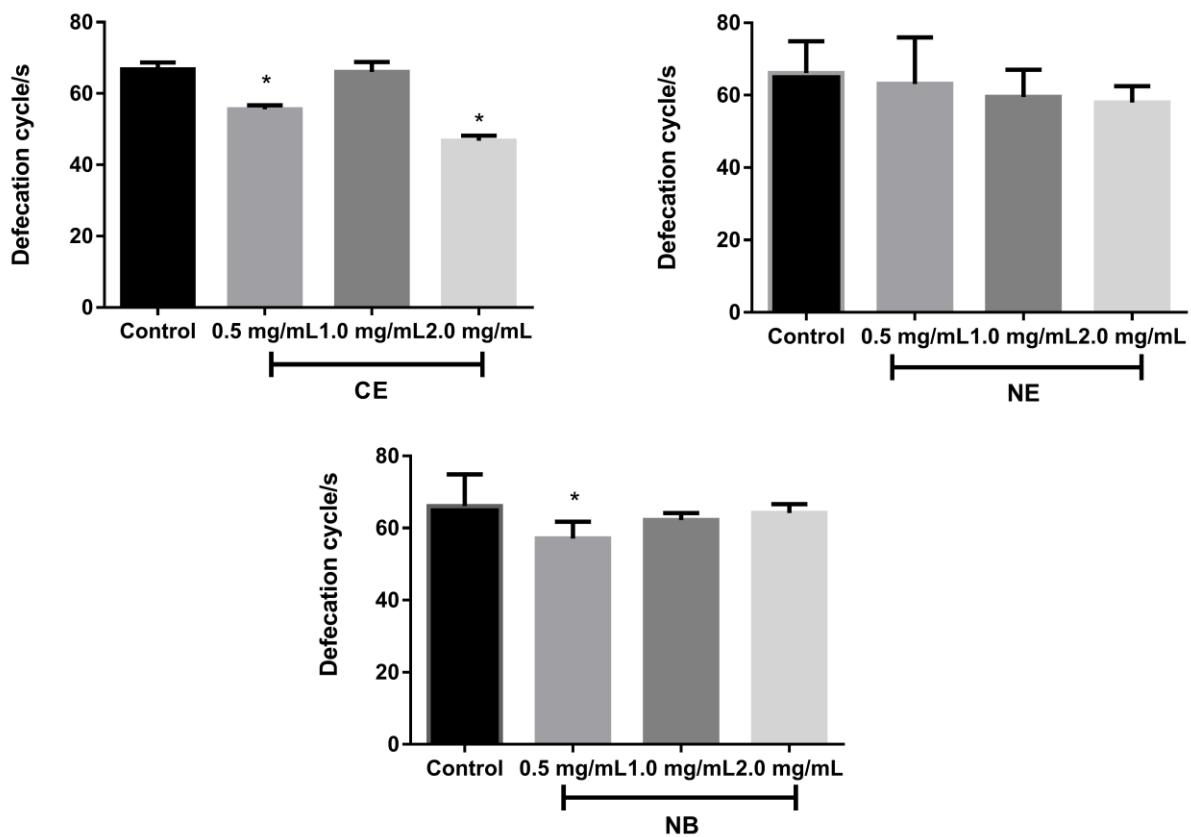
297 usually high in the presence of food and low in their absence. However, the nutritional status
298 of the animal, its feeding history, and food quality are influencing factors, just as the authors
299 suggest that quiescence is a satiety behavior of worms. Therefore, the reduced pharyngeal
300 pumping rates may have been influenced by these factors highlighted above and by the
301 diversity of the samples, since the control animals consumed only M9 buffer. Pharyngeal
302 pumping is an indicator of a healthy worm and is mainly controlled by cholinergic and
303 glutamatergic innervation and dopamine and serotonin (Raizen, Song, Trojanowski, & You,
304 2012). The nanoemulsion containing *P. peruviana* calyx extract (NE) did not cause changes in
305 the pharyngeal pumping of this organism. These findings are in agreement with the study by
306 Moraes et al. (2016), who reported that polymeric nanocapsules decrease the toxic effects of
307 clozapine in *C. elegans*, and with Pascoli et al. (2019), who demonstrated that zein
308 nanoparticles decreased the toxicity of neem oil in this organism.

309 Defecation cycle frequency decreased after free extract (CE) and blank nanoemulsion
310 (NB) treatments in some concentrations (Figure 4). This behavior is a stereotyped motor
311 program that occurs in the nematode approximately every 50 s (Thomas, 1990). Each cycle
312 begins with a posterior body contraction that directs the contents of the intestine to the
313 anterior part of the worm and is followed by a contraction of the nematode. The anterior
314 muscle pushes the bowel contents to the back of the animal, followed by a contraction of the
315 enteric muscle that expels the bowel contents (Riddle, Blumenthal, Meyer, & Priess, 1997).
316 The nanoemulsion containing *P. peruviana* calyx extract (NE) showed no change in the
317 frequency of defecation cycles in relation to the control (Figure 4B), which is a satisfactory
318 result for not presenting worm toxicity in this parameter.

319

320

321



322

323 **Figure 4** Defecation cycle / s of young adult worms of strain N2 exposed for 2h at different
 324 concentrations of *P. peruviana* calyx extracts (CE), nanoemulsions containing *P. peruviana*
 325 calyx extract (NE), and blank nanoemulsions (NB) in liquid medium. Each experiment was
 326 performed three times in triplicate. (mean, SEM, n = approximately 15 worms per group).

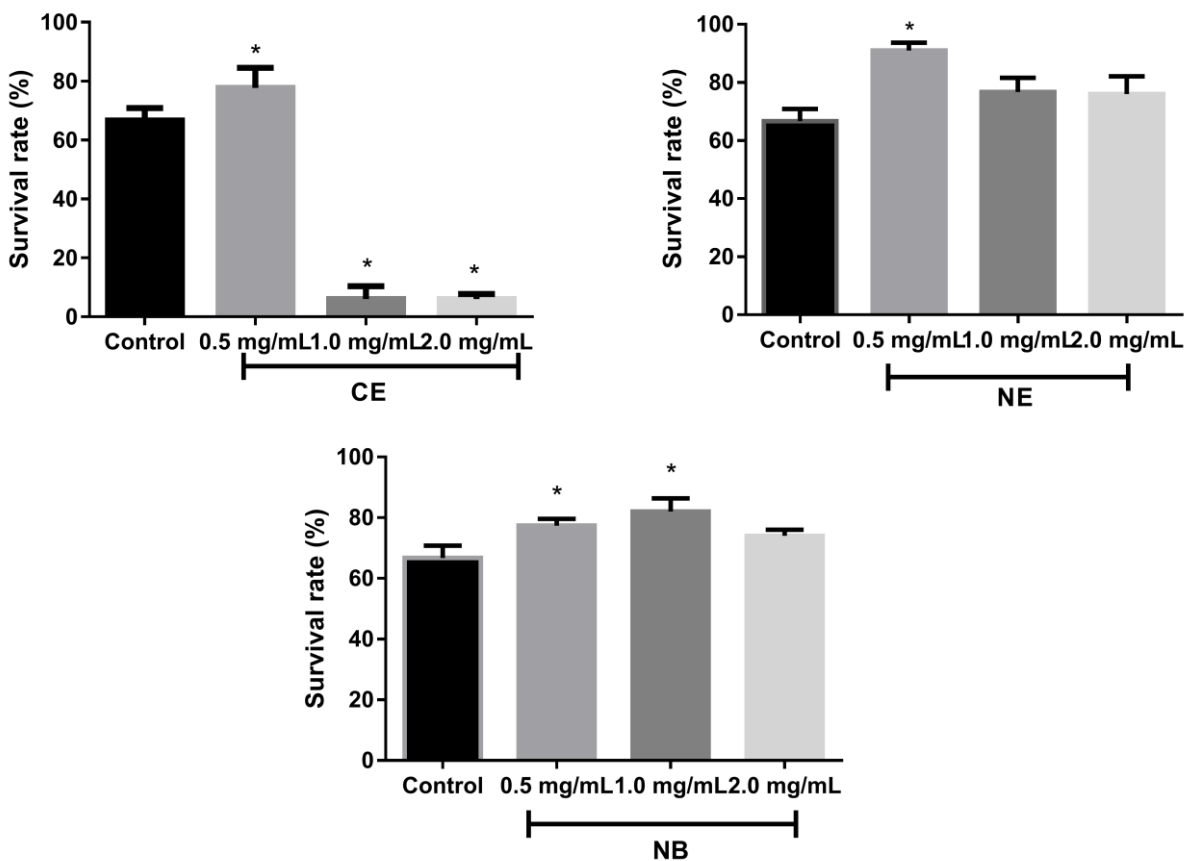
327 * means significant difference (p < 0.05) compared to the control after one-way ANOVA
 328 analysis followed by Tukey post-hoc test.

329

330 We therefore used a pro-oxidant juglone to evaluate the potential antioxidant effects of
 331 the samples in the nematode *C. elegans*. Juglone can induce oxidative stress through various
 332 mechanisms, such as superoxide (O_2^-), hydroxyl radical ($\bullet OH$), and hydrogen peroxide
 333 (H_2O_2) formation, triggering toxic effects by protein modifications (Inbaraj & Chignell, 2004;
 334 Stadtman, 2006), lipid oxidation (Fong, McCay, Poyer, Keele, & Misra, 1973), and nuclear
 335 DNA damage (Bjelland & Seeberg, 2003).

336 Both nanoemulsions reduced juglone toxicity, increasing the survival rate relative to
337 the control group or remaining similar to the juglone control group levels (Figure 5). The best
338 response to oxidative stress was found in the group treated with 0.5 mg/mL of nanoemulsion
339 containing *P. peruviana* calyx extract (NE), showing an increase in survival rate of 24.3%
340 compared to the juglone control group. For blank nanoemulsions (NB), concentrations of 0.5
341 and 1 mg/mL showed a significant increase in survival.

342 On the other hand, in the treatment with the calyx extract (CE), only 0.5 mg/mL
343 protected against juglone-induced oxidative stress, in addition to the concentrations of 1 and 2
344 mg/mL increasing juglone toxicity, which reduced the survival rate. This effect is likely due
345 to the vehicle used to obtain the extracts (ethanol) because a larger volume was used at these
346 concentrations. At higher concentrations, ethanol may have exacerbated the toxic effects of
347 juglone to *C. elegans*. These findings indicate that the formulations were ingested by the
348 worms, do not have toxic effects, and most likely exerted their pharmacological effects at low
349 concentrations *in vivo*.



350

351 **Figure 5** Survival of young adult worms of N2 strain exposed for 2h to different
 352 concentrations of *P. peruviana* calyx extracts (CE), nanoemulsions containing *P. peruviana*
 353 calyx extract (NE), blank nanoemulsions (NB), and then exposed for 1h to 100 μ M juglone.
 354 The data are expressed as percent of control; each experiment was performed three times in
 355 triplicate (mean, SEM, n = approximately 300 worms per group). * means significant
 356 difference ($p < 0.05$) compared to the control after one-way ANOVA analysis followed by
 357 Tukey post-hoc test.

358

359 3.3 Release under simulated gastrointestinal conditions

360

361 The results obtained are expressed in Figure 6. The free calyx extract (CE) released
 362 45% of rutin in the oesophagus/stomach stage and showed no statistical difference regarding
 363 the concentrations detected in the duodenum and ileum stages. Thus, these results indicate

364 that all rutin content present in the extracts was released in the gastric phase and that 55% of
365 the bioactive compound degraded due to acidic pH conditions.

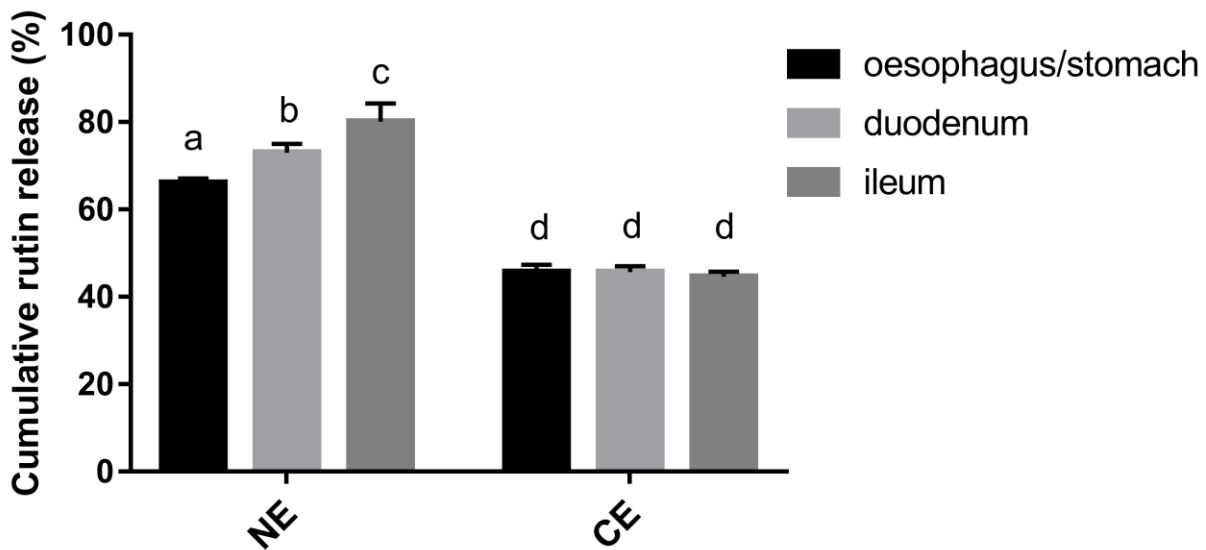
366 On the other hand, nanoemulsions (NE) were able to protect the *P. peruviana* calyx
367 extract, which can be observed by the % cumulative rutin release. Initially, around 65% of
368 rutin was released under gastric conditions. However, nanoemulsions continued to control the
369 release in the next steps and increased rutin contents was observed in the duodenum and
370 ileum, ending the test with 80% of rutin released and available for absorption in the ileum.
371 The remaining 20% may have been degraded during the process, as there was release in the
372 oesophagus/stomach stage. Furthermore, another hypothesis is that the bioactive compound
373 was not fully released by the end of the test and may slowly release the rutin from the NEs for
374 a longer time.

375 Wang, Xue, Hu, Zhou, & Luo (2017) evaluated the release rates of free and
376 encapsulated (solid lipid nanoparticles crosslinked with pectin - PSLN) curcumin in simulated
377 gastrointestinal conditions and found that curcumin release was better controlled by PSLN
378 than in the free form. About 20-40% of curcumin from nanoparticles was detected in release
379 medium after 2 h, and only about 10-20% more curcumin released during the next 4 h stage,
380 demonstrating that, in this study, not all content was released.

381 Note that the oral stage was not included, although there may be changes in the size
382 and interfacial characteristics of nanoemulsions, influencing their fate in the gastrointestinal
383 tract (McClements & Xiao, 2012) as the sample is liquid (and therefore chewing is not
384 relevant and residence time in the mouth is very short) and contains no starch (i.e. the primary
385 enzyme present in saliva, amylase would not act) (Silva et al., 2018).

386

387



388
 389 **Figure 6** *In vitro* rutin release profile of calyx extracts (CE) and nanoemulsions containing *P.*
 390 *peruviana* calyx extract (NE) using dialysis bag method exposed to the simulated
 391 gastrointestinal conditions throughout incubation time and pH values. Esophagus/stomach: 90
 392 min/pH=2.0; duodenum: 20 min/pH=5.0; ileum: 90 min/pH=6.5. Bars represent the mean \pm
 393 SD, n = 3. Two-way ANOVA - Tukey's multiple comparisons test ($p < 0.05$), same letters
 394 mean no statistically significant difference.

395

396

397 4. CONCLUSION

398

399 *In vitro* and *in vivo* tests were fundamental to evaluate the possible toxicity of
 400 nanoemulsions and *P. peruviana* calyx extract. The extracts and nanoemulsions containing the
 401 extract (NE) showed no toxicity in the non-tumor cell line. However, as the concentration of
 402 NE and blank nanoemulsions (NB) increased, they showed cytotoxicity against the tumor cell
 403 line. In the *in vivo* model, all samples caused no toxicity to *C. elegans* in the survival test. The
 404 pharyngeal pumping rate and worm defecation cycle in animals treated with NE were not
 405 altered, the other formulations caused a slight change in diet, which reduced some rates.

406 Regarding oxidative stress resistance, the formulations exerted their antioxidant effects at low
407 concentrations *in vivo*. In addition, the NE was able to protect the *P. peruviana* calyx extract
408 against degradation of gastrointestinal conditions, controlling rutin release and increasing the
409 levels of bioactive compound released in the duodenum and ileum, which is where absorption
410 occurs. Thus, future food applications of nanoemulsions containing calyx extract can be
411 further studied and explored.

412

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576

8 DISCUSSÃO GERAL

Os cálices são um importante subproduto do processamento do fruto de *goldenberry* e são uma promissora fonte de compostos bioativos, tais como o 4 β -hidroxivitanolídeo E e vitanolídeo E (BALLESTEROS-VIVAS et al., 2019b), cinquenta e seis fitoquímicos, incluindo os principais componentes fenólicos, vários vitanolídeos (C28-isoprenóides) com uma variedade de atividades biológicas e uma grande família de anti-inflamatórios (BALLESTEROS-VIVAS et al., 2019a). Novas oxilipinas ativas denominadas fitoprostanos, também, foram identificados. Além disso, seis derivados fenólicos foram descritos pela primeira vez. Entre os compostos já identificados, a rutina também foi o principal composto fenólico encontrado (MEDINA et al., 2019). Assim, baseado nesses estudos podemos observar a variedade de compostos já conhecidos no cálice e ressaltar a importância da escolha desse resíduo para nossa pesquisa, corroborando com os nossos achados em que o composto majoritário foi a rutina.

O extrato do cálice de *P. peruviana* foi caracterizado quanto sua capacidade redutora (610 ± 20 mg Eq. de ácido gálico/100 g de cálice), capacidade antioxidante por ORAC ($137,93 \pm 2,10$ μ mol de Trolox/g de cálice) e composto majoritário quantificado por CLAE (rutina: $11,38 \pm 0,98$ μ g/mL). Em relação à análise cromatográfica buscou-se por quantificar o composto predominante e usá-lo como marcador durante o estudo de estabilidade e liberação.

Estudos em relação ao potencial antimicrobiano de *P. peruviana* vêm sendo realizados. Extratos etanólicos de diferentes partes da planta (fruta, semente, raiz, corpo e folha) foram avaliados, sendo que semente e fruta foram as mais efetivas em termos de sua atividade antimicrobiana (ERTÜRK et al., 2017). Todas as partes da planta de *P. peruviana* podem ser fontes potenciais de agentes antimicrobianos, sendo o mais potente, os extratos das folhas em diclorometano (KAMAU et al., 2017). Além disso, os extratos etanólicos das bagas maduras e das folhas de *P. peruviana* foram mais efetivos do que os aquosos (CUEVA et al., 2017). Neste sentido, a atividade antibacteriana e antibiofilme do extrato do cálice de *P. peruviana*, também, foi avaliada.

De acordo com essas pesquisas até o momento nenhuma havia especificado a atividade antimicrobiana do extrato etanólico dos cálices. Assim, em nosso estudo verificamos que o extrato do cálice de *P. peruviana* apresentou atividade antibacteriana contra os patógenos analisados, possuindo atividade bactericida na concentração de 30 mg/mL frente a seis bactérias: *Enterobacter aerogenes*, *Klebsiella pneumoniae*, *Staphylococcus epidermidis*, *Streptococcus pneumoniae*, *Pseudomonas aeruginosa* PA01 e *Enterococcus*

faecalis. A CIM variou de 3,15 a 30 mg/mL de extrato e mostrou atividade bacteriostática frente a oito cepas: *Escherichia coli*, *Shigella sonnei*, *Acinetobacter baumannii* ATCC 19606, *Acinetobacter baumannii* (isolado clínico), *Streptococcus agalactiae*, *Salmonella sp.*, *Salmonella enteridis* and *Staphylococcus aureus*. Em relação à capacidade antibiofilme, houve inibição da formação de biofilme, mas não houve destruição.

De acordo com esses resultados, foi verificada a potencial atividade antibacteriana do extrato do cálice de *P. peruviana* e essa atividade não pode ser observada nas nanoemulsões, pois a maioria das atividades foram em concentrações de extrato acima de 7,5 mg/mL e a concentração de extrato usada nas nanoemulsões foi menor (5 mg/mL). Durante o desenvolvimento das NEs foram realizados testes com maior concentração de extrato, porém acima de 5 mg/mL havia a formação de gotículas no tamanho micrométrico e nosso objetivo era focar em uma formulação exclusivamente na escala nano. De qualquer forma, sabe-se que o extrato possui essa atividade e é de extrema importância, pois incentiva a realização de novos estudos para aplicabilidade de seu potencial antibacteriano na área de alimentos.

Em comparação com outras técnicas, a nanoencapsulação de compostos bioativos tem sido uma questão de grande interesse devido a suas funcionalidades exclusivas, tais como alta eficiência de encapsulamento, melhor biodisponibilidade, maior estabilidade, perfil de liberação sustentada e mascarar sabores indesejáveis (SHISHIR et al., 2018). Assim, com intuito de utilizar uma tecnologia inovadora, em nosso estudo aplicamos a nanotecnologia no extrato do cálice de *P. peruviana* desenvolvendo nanoemulsões.

As nanoemulsões obtidas apresentaram características físico-químicas satisfatórias com tamanho de gotículas nanométrico, baixo índice de polidispersão, potencial zeta negativo, pH ácido, rutina ($11,38 \pm 0,98 \mu\text{g/mL}$) como composto bioativo majoritário com eficiência de encapsulamento de 85% e morfologicamente apresentaram forma esférica e superfície lisa. Quanto ao estudo de estabilidade frente a diferentes condições de armazenamento (7 e 25°C) com ausência ou incidência de luz por 120 dias, observou-se que a formulação armazenada a temperatura ambiente e com incidência de luz (CPp-NE_{AL}) foi a que apresentou maiores alterações e instabilidades como a redução drástica do pH, aumento do tamanho de gotícula, aumento do índice de polidispersão e redução em módulo significativa do potencial zeta. Esses fenômenos de instabilidade podem estar relacionados a vários parâmetros como as propriedades químicas e a estrutura dos componentes, concentração de emulsificantes e aditivos, e a temperatura, que causam alterações nas propriedades, como tensão interfacial, elasticidade da gota e potencial interação da gota (GUPTA et al., 2016). Portanto, alta temperatura ou temperatura ambiente e a incidência de

luz são fatores já relatados como causadores de alterações na estabilidade de nanoemulsões (Rebolleda et al., 2015).

As melhores condições de armazenamento das amostras foram em ambas temperaturas no escuro, e a maior estabilidade foi observada para as nanoemulsões contendo extrato do cálice de *P. peruviana* armazenadas à temperatura ambiente e com ausência de luz. Além disso, a NE apresentou maior capacidade antioxidante e redutora em relação aos extratos livres. Esses resultados demonstram que as nanoemulsões desenvolvidas têm potencial para preservar a capacidade antioxidante do extrato do cálice de *P. peruviana*.

Entender a segurança e toxicidade dos nanomateriais é um grande desafio e requer muitas pesquisas, pois para promover a comercialização segura de novos produtos nanotecnológicos que sejam benéficos para a saúde, deveria existir um regulamento mundial (BAZANA; CODEVILLA; MENEZES, 2019). Com o intuito de estudar os potenciais de toxicidade das nanoemulsões desenvolvidas e do extrato, realizou-se testes *in vitro* e *in vivo*. Como investigação preliminar da toxicidade de um nanomaterial, os ensaios de citotoxicidade são amplamente utilizados (HILLEGASS et al., 2010).

No ensaio de citotoxicidade, quanto à viabilidade das células 3T3 (não tumorais) o extrato e as nanoemulsões contendo extrato não apresentaram toxicidade, somente a nanoemulsão branca na maior concentração (150 µg/mL) mostrou uma redução significativa na viabilidade celular. Em relação às células MCF-7 (tumorais), o extrato livre não reduziu a viabilidade das células cancerígenas. Já, as NEs e as NBs à medida que a concentração da formulação aumentou, demonstraram altos valores de inibição no crescimento dessas células (Figura 1, manuscrito 2).

Neste estudo, duas linhagens celulares foram escolhidas como sistemas modelo, porque o uso de células de diferentes espécies e com diferentes origens embriogênicas é uma abordagem importante para entender as respostas específicas das células induzidas por materiais nano-baseados (FRÖHLICH et al., 2012). A NB provavelmente exerceu esse efeito devido à presença do polissorbato 80, conforme já relatado por (CONTRI et al., 2016), pois o tipo e a quantidade de tensoativo na formulação podem influenciar sua citotoxicidade (MENDES et al., 2015). Assim, respostas citotóxicas foram observadas mesmo em nanopartículas brancas durante a realização de experimentos *in vitro* utilizando diferentes linhagens celulares (MENDES et al., 2014; MENDES et al., 2015; SILVA et al., 2019).

Portanto, vale ressaltar que a diferente sensibilidade das linhagens celulares testadas aos tratamentos de nanoemulsões pode ser parcialmente explicada pelo fato de que células

saudáveis e tumorais apresentam diferenças em sua estrutura e função, especialmente em sua atividade metabólica e composição molecular (FREY et al., 2007).

O modelo *C. elegans* foi utilizado para as análises de toxicidade realizadas *in vivo*. A taxa de sobrevivência de todos os vermes permaneceu em torno de 100% (Figura 2, manuscrito 2), independentemente da amostra testada (CE, NE e NB), indicando que a exposição às amostras não causou letalidade. A nanoemulsão contendo o extrato do cálice de *P. peruviana* (NE) não demonstrou alterações quanto aos demais parâmetros analisados (bombeamento faríngeo, ciclo defecação e resistência ao estresse oxidativo).

Além disso, a NE protegeu o extrato do cálice de *P. peruviana* frente à degradação das condições gastrointestinais simuladas, controlando a liberação de rutina e aumentando os níveis do composto bioativo liberado no duodeno e íleo, onde ocorre a absorção. Assim, futuras aplicações alimentares utilizando a nanotecnologia aliada aos resíduos da fruta de *P. peruviana* (cálice) podem ser mais e explorados pela ciência e tecnologia dos alimentos.

9 CONCLUSÃO

- As nanoemulsões contendo extrato do cálice de *P. peruviana* apresentaram características físico-químicas adequadas (tamanho nanométrico de gotículas, baixo índice de polidispersão, potencial zeta negativo, pH ácido, rutina como composto majoritário, eficiência de encapsulamento de 85% e morfologia esférica e lisa).
- Quanto ao estudo de estabilidade as melhores condições de armazenamento das amostras foram nas temperaturas de 7° C e 25 °C no escuro, sendo que a maior estabilidade foi observada para as nanoemulsões contendo extrato do cálice de *P. peruviana* armazenadas à temperatura ambiente e com ausência de luz.
- O extrato do cálice de *P. peruviana* apresentou atividade antibacteriana contra os patógenos analisados, demonstrando atividade bacteriostática frente a oito bactérias e atividade bactericida frente a seis cepas. Ainda, inibiu a formação de biofilme, porém não houve destruição do mesmo. Estes resultados possibilitam que novos estudos sejam realizados para contribuir na sua utilização na indústria de alimentos, como um conservante de origem natural e, também, na área clínica.
- As nanoemulsões contendo extrato do cálice de *P. peruviana* não apresentaram citotoxicidade *in vitro* frente a linhagem celular não tumoral (3T3 fibroblastos) e foram citotóxicas frente a linhagem celular tumoral (MCF-7 câncer de mama), conferindo resultados promissores. Também, não foi evidenciada toxicidade *in vivo* nas condições avaliadas. Além disso, a liberação sob condições gastrointestinais simuladas demonstrou que as nanoemulsões contendo o extrato apresentaram melhor controle da liberação em relação ao extrato livre, protegendo a rutina frente à degradação, propiciando que novos estudos viabilizem a aplicabilidade na área alimentos.
- As nanoemulsões desenvolvidas têm potencial para preservar a capacidade antioxidante do extrato do cálice de *P. peruviana*, favorecendo a reutilização desse subproduto. Portanto, o uso da nanotecnologia associado com resíduos de plantas é de grande interesse para a ciência de alimentos e o desenvolvimento de novos produtos alimentares.

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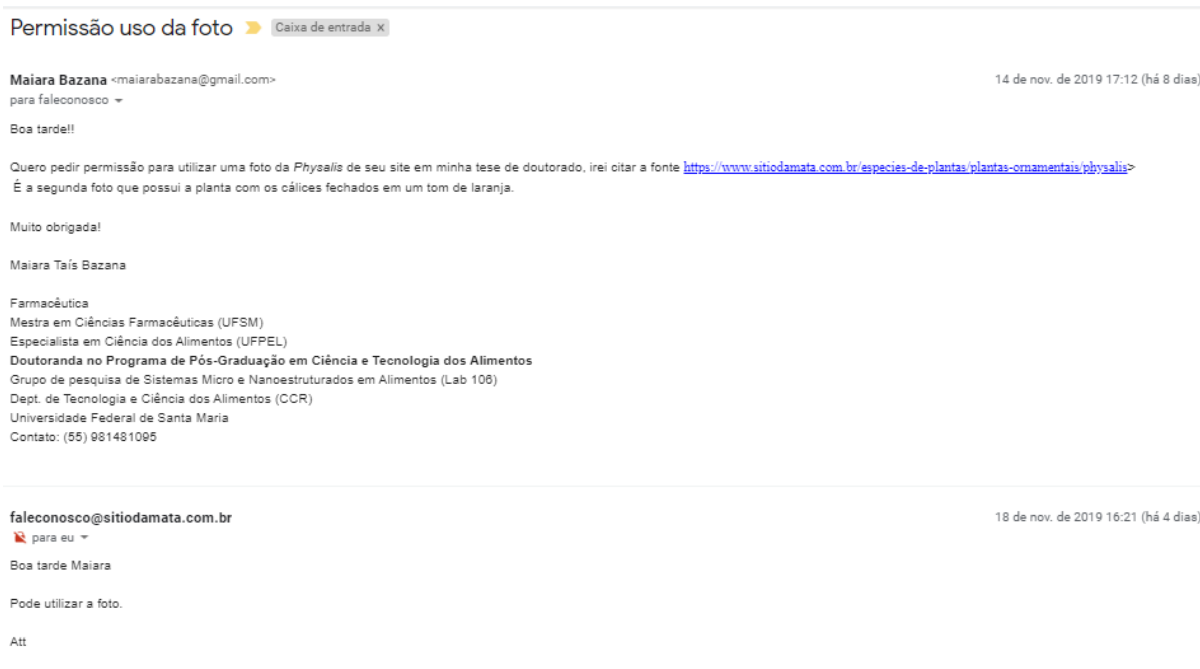
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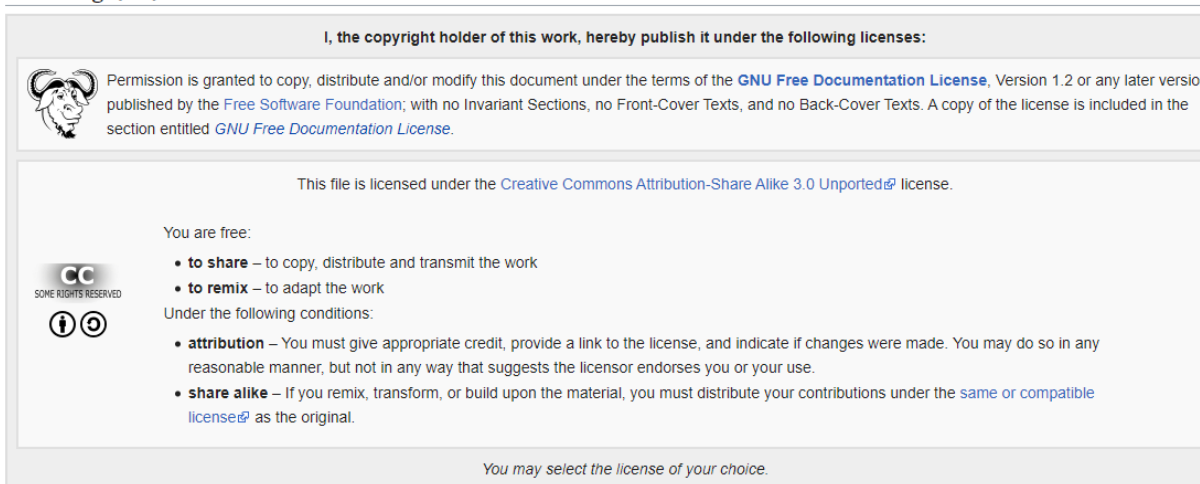
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Figura 2: https://es.wikipedia.org/wiki/Archivo:Physalin_AB.png

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ANEXO B – AUTORIZAÇÃO DO ARTIGO 1



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**Development of nanoemulsions containing *Physalis peruviana* calyx extract: A study on stability and antioxidant capacity****Author:**

Maiara Taís Bazana, Suelen Santos da Silva, Cristiane Franco Codevilla, Cassandra de Deus, Bruna Nichelle Lucas, Gustavo Andrade Ugalde, Marcio Antonio Mazutti, Erico Marlon Moraes Flores, Juliano Smanioto Barin, Cristiane de Bona da Silva et al.

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